Objectives

- Study transport of CBL DMS through clouds
- Study the distribution of SO₂ in and around clouds
- Estimate loss of SO₂ in clouds using DMS and other conservative tracers
- Study CCN formation around clouds

Atmospheric Pressure Ionization Mass Spectrometry (APIMS)

- ions are formed by ion molecule reactions initiated by electrons emitted by nickel-63
- DMS is determined using proton transfer ionization to form DMSH⁺
- SO₂ is determined as SO₅⁻ formed by reactions initiated with ozone, carbon dioxide, and oxygen
- ion molecule reactions are fast: leads to high rate determinations with ambient species and internal standard alternately measured (25 samples per second for each ion)
- calibration is achieved with isotopically labeled internal standards: deuterated DMS and ³⁴SO₂

Advantages of the internal isotopically labeled internal standard

- calibration for every sample
- determination of the ambient analyte without knowing the instrument absolute sensitivity
- can determine the lower limit of detection for every sample
- internal standard acts as a carrier for low concentrations of the analyte

Combined effect: can determine DMS and SO₂ both in and out of cloud with high precision and accuracy at high rates









