

# VOC observations during TORERO and comparison with CAM-chem results

24-Jun-14 TORERO Data Workshop

**TORERO TOGA – CAM-chem science team:**

Eric Apel, Becky Hornbrook, Alan Hills (NCAR/ACD)

Jean- Francois Lamarque, Doug Kinnison, Simone Tilmes

# The Trace Organic Gas Analyzer (TOGA)

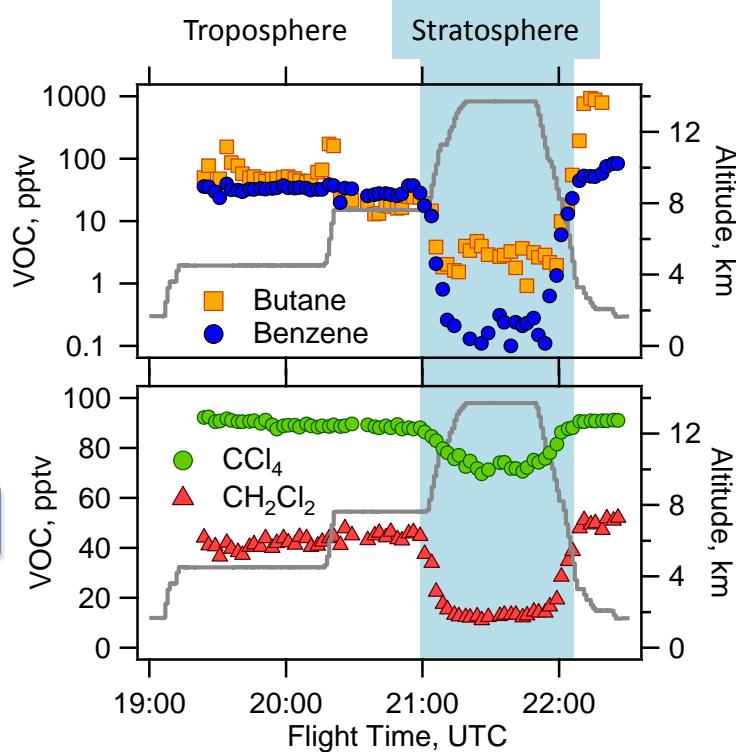
Eric Apel (PI), Alan Hills, Rebecca Hornbrook (ACD/NESL/NCAR)  
Dan Riemer (Co-PI; University of Miami)

- VOCs needed to understand chemistry leading to trop O<sub>3</sub> and aerosols. Halogenated species can impact both trop and lower strat
- Designed specifically for the G-V
- Maiden research voyage – **TORERO** 2 min time res
- Designed to have very low **LOD ppt to sub – pptv** detection limits, over 45 VOC measured simultaneously



Sample data:  
DC3 Test Flight 2

Pollutants



Greenhouse gases

Installed on the G-V

# TOGA compounds

<b>Hydrocarbons</b>	Propane 1-Butene <i>i</i> -Butene Butane <i>i</i> -Butane Benzene Toluene Ethyl Benzene <i>t</i> -2-Butene <i>c</i> -2-Butene Pentane 1,3-Butadiene <b>Limonene</b>	Isoprene <i>t</i> -2-Pentene <i>c</i> -2-Pentene <i>i</i> -Pentane <i>o</i> -Xylene <i>m/p</i> -Xylene 1,3,5-Trimethylbenzene 1,2,4-Trimethylbenzene <b><math>\alpha</math>-Pinene</b> <b><math>\beta</math>-Pinene</b> <b>Camphene</b> <b>Myrcene</b>	
<b>Oxygenates</b>	Acetaldehyde Propanal Butanal Pentanal <b>Methacrolein</b> <b>Methyl Vinyl Ketone</b> Methyl Butenol	Methanol Ethanol Acetone Butanone 2-Pantanone 3-Pantanone Methyl t-Butyl Ether	
<b>Halocarbons</b>	Chloroform ( $\text{CHCl}_3$ ) Methylene chloride ( $\text{CH}_2\text{Cl}_2$ ) Methyl chloride ( $\text{CH}_3\text{Cl}$ ) Methyl bromide ( $\text{CH}_3\text{Br}$ ) Tetrachloroethane ( $\text{CH}_2\text{Cl}_4$ ) Tetrachloroethylene ( $\text{C}_2\text{Cl}_4$ ) Bromoform	Tetrachloromethane ( $\text{CCl}_4$ ) CFC-113 HCFC-141b HCFC-134a 1,2-Dichloroethane ( $\text{C}_2\text{H}_4\text{Cl}_2$ ) Methyl Iodide ( $\text{CH}_3\text{I}$ ) iodoform	dibromomethane diodomethane bromochloromethane bromoiodomethane chloroiodomethane
<b>Nitrogen and sulfur compounds</b>	Acetonitrile Dimethyl Sulfide (DMS)	DMSO?	

## Preliminary Lucy Carpenter/Stephen Andrews analysis vs. various standards

Compound	vs. NCAR Lab	vs. NOAA Air spike	vs. NOAA gravimetric	vs. NCAR in-flight cals
$\text{CH}_3\text{I}$	3.50	2.75		3.71
$\text{CH}_2\text{Br}_2$	2.25	2.04	2.39	2.13
$\text{CHBr}_3$		5.14	5.89	4.60
$\text{CH}_2\text{I}_2$	0.83	1.12	0.50	0.72
$\text{CH}_2\text{BrCl}$	7.43	5.87		6.71
$\text{CHBr}_2\text{Cl}$		3.53		1.64
$\text{CH}_2\text{IBr}$	1.79	1.49	2.03	1.72

NOAA Gases

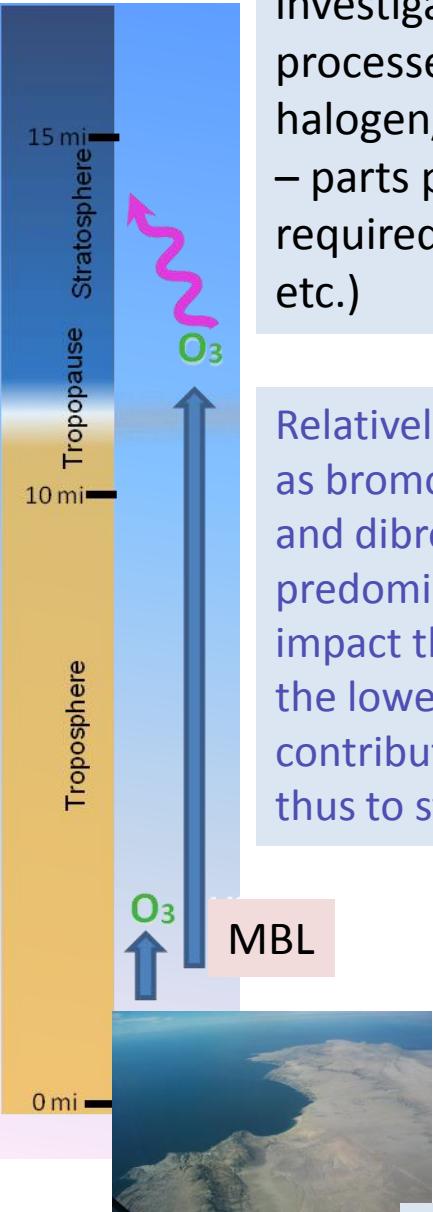


NCAR Gases



Canister Gases



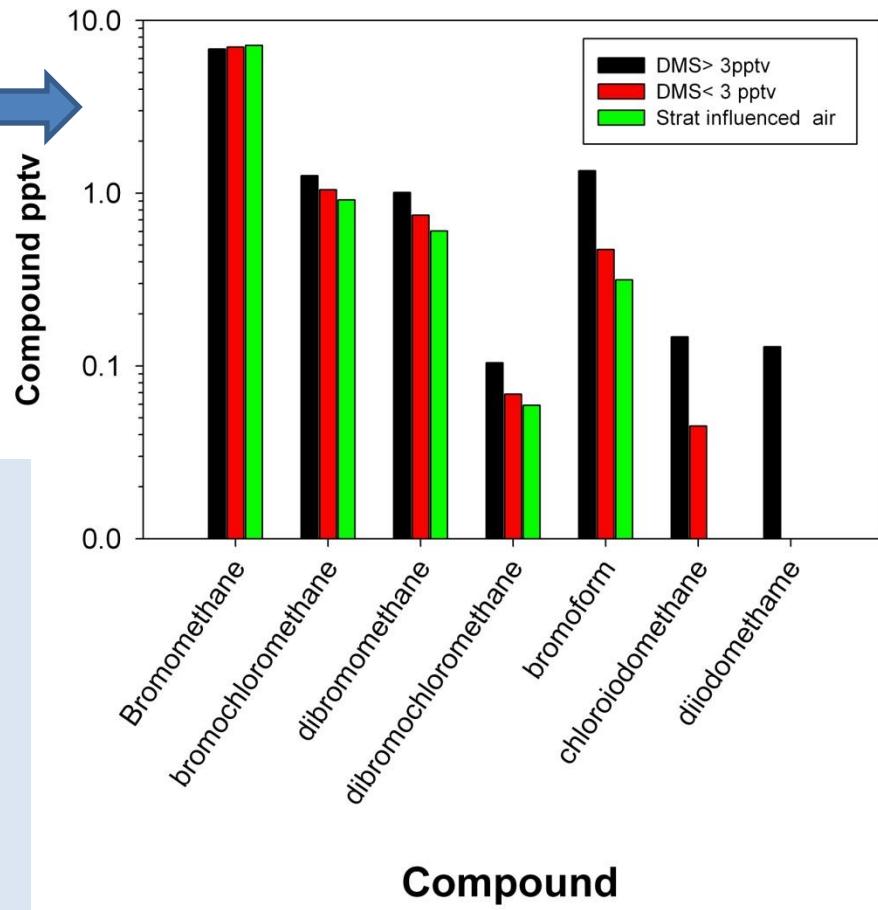


Ultra-high sensitivity needed to investigate some chemical processes such as the inorganic halogen/organo-halogen species – parts per quadrillion sensitivity required (see Carpenter, Atlas, etc.)

Relatively stable organic halogens such as bromomethane, bromoform ( $CHBr_3$ ) and dibromomethane ( $CH_2Br_2$ ), emitted predominantly from the oceans, can impact the MBL and be transported to the lower stratosphere and make a contribution to total bromine levels and thus to stratospheric ozone depletion.

Build on previous studies

## TORERO TOGA Organohalogen Measurements



**VSL – defined as less than 6 mo.**

Bromomethane  $\approx$  1 year

Bromoform  $\approx$  1 month

Dibromomethane  $\approx$  4 months

Chloroiodomethane  $\approx$  2 hours  
(LOD TOGA = 0.03 pptv)

Diiodomethane  $\approx$  5 minutes (LOD  
TOGA 0.03 pptv)

# Recent updates to modeling VSL halogen chemistry in CAM-Chem

D. Kinnison & J.-F. Lamarque (NCAR, USA)

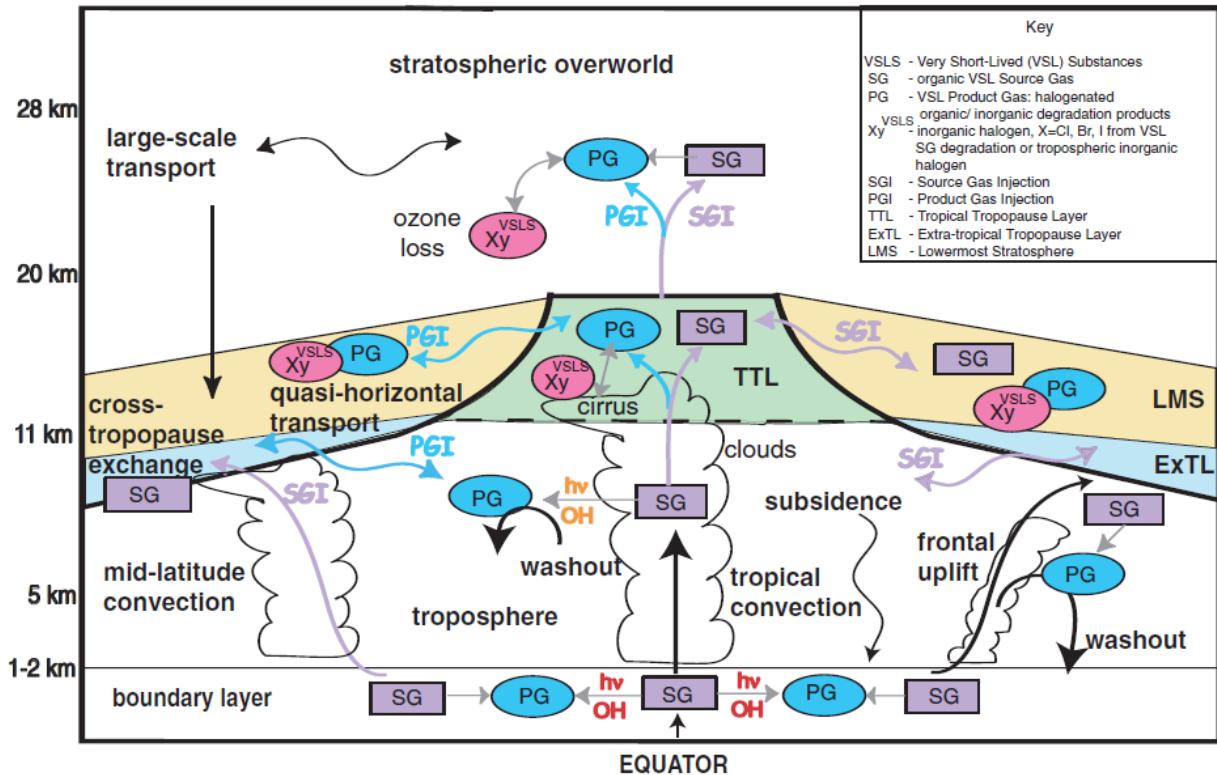
A. Saiz-Lopez (CSIC, Spain)

R. Fernandez (UNC, Argentina)

R. Salawitch (U. Maryland, USA)

# Motivation

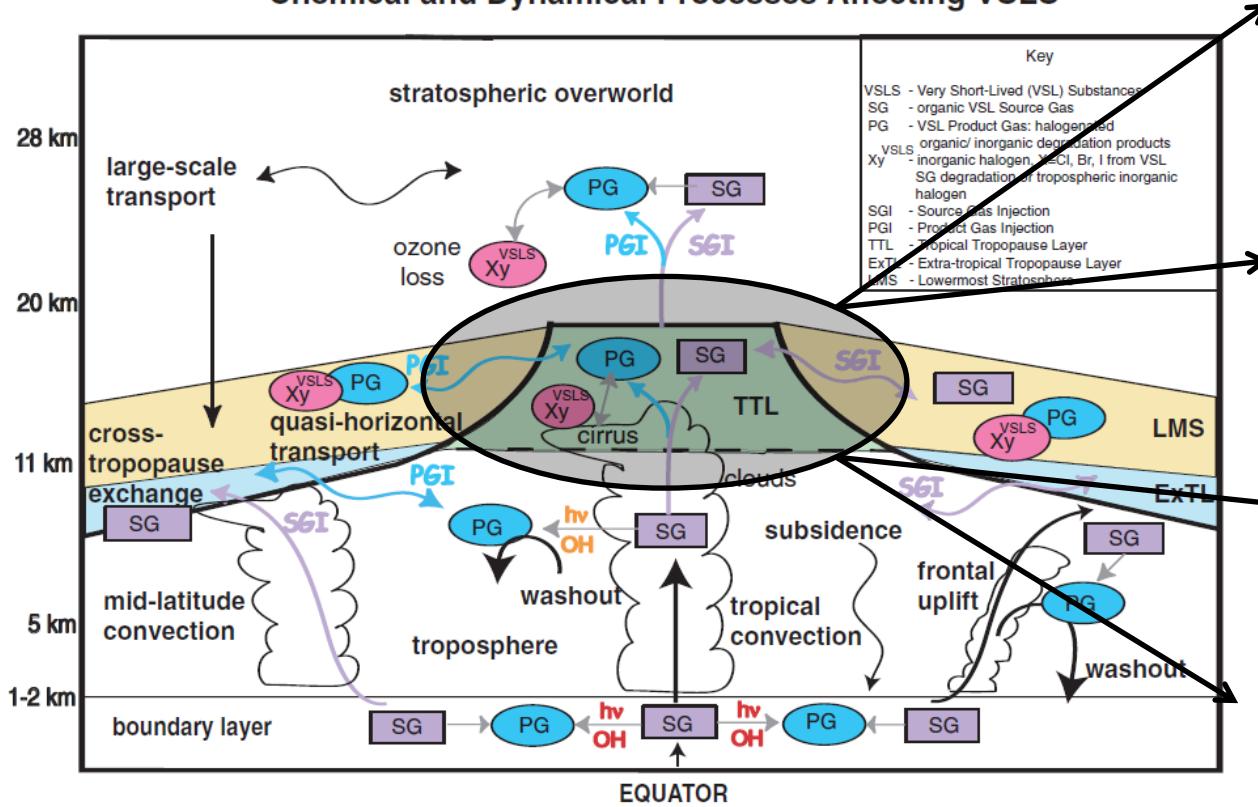
## Chemical and Dynamical Processes Affecting VSLs



$SG_{VSL}$  bromocarbons are converted to  $PG_{VSL} = Br_y$  in the TTL before/while being injected to the stratosphere

# Physical-chemistry processes

Chemical and Dynamical Processes Affecting VSLs



Heterogeneous recycling  
Over ice particles

Washout efficiencies  
and  
photochemical lifetime

Strength of convection

Entrainment from the MBL

$\text{SG}_{\text{VSL}}$  bromocarbons are converted to  $\text{PG}_{\text{VSL}} = \text{Br}_y$  in the TTL before/while being injected to the stratosphere

# On-going work

- Updated (to the CONTRAST model setup and version) simulations for TORERO and CONTRAST at horizontal resolution of  $\approx 1^\circ$
- Planning (end of summer) high-resolution ( $\approx 1/4^\circ$ ) for both periods

# Extensive recent developments in the representation of VSL halogen chemistry and emissions

- Ordonez, C., J.-F. Lamarque, S. Tilmes, D. Kinnison, E. Atlas, D. Blake, G. Sousa Santos, G. Brasseur and A. Saiz-Lopez, *Atmos. Chem. Phys.*, 12, 1423–1447, 2012.
- Saiz-Lopez, A., J.-F. Lamarque, D. Kinnison, S. Tilmes, C. Ordonez, J. J. Orlando, A. J. Conley, J. M. C. Plane, A. Mahajan, G. Sousa Santos, E. Atlas, D. R. Blake, S. P. Sander, S. M. Schauffler, A. M. Thompson and , *Atmos. Chem. Phys.*, 12, 3939–3949., 2012.
- Fernandez, R. P., R. Salawitch, D. E. Kinnison, J.-F. Lamarque and A. Saiz-Lopez, Bromine partitioning in the tropical tropopause layer: implications for stratospheric injection. Submitted to *Atmos. Chem. Phys. Disc.*, 2014.

# Organohalogens and DMS

Dibromomethane –  $\text{CH}_2\text{Br}_2$

Bromoform –  $\text{CHBr}_3$

Others...

methyl iodide  $\text{CH}_3\text{I}$

chloroiodomethane –  $\text{CH}_2\text{ICl}$

dibromochloromethane –  $\text{CHBr}_2\text{Cl}$

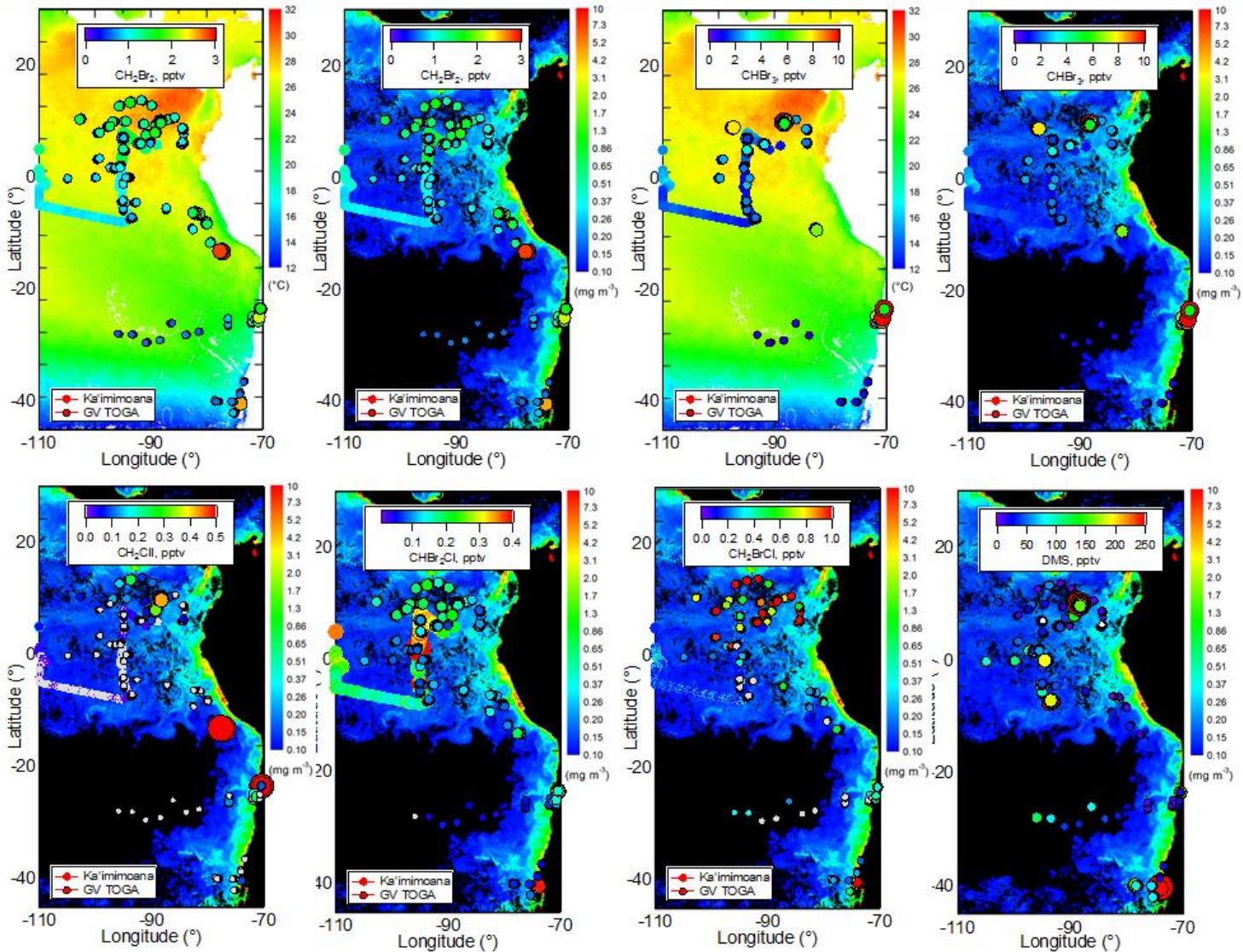
From 2012

Fall AGU poster:

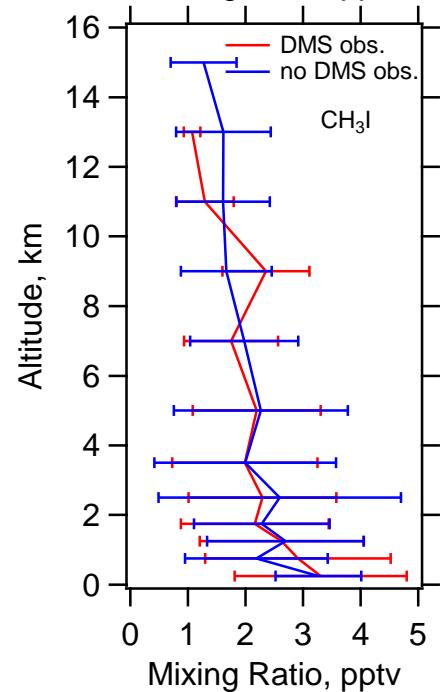
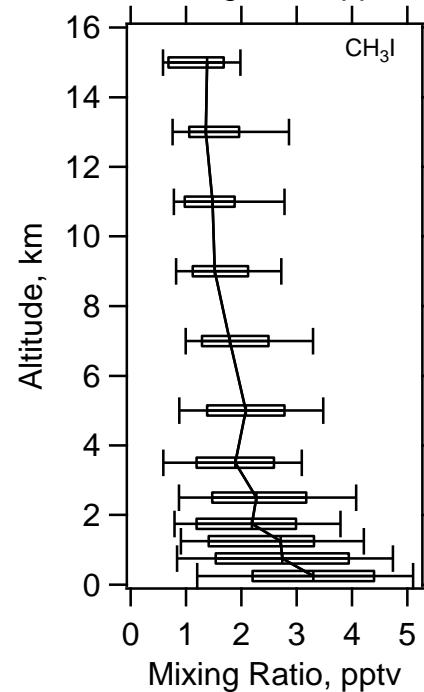
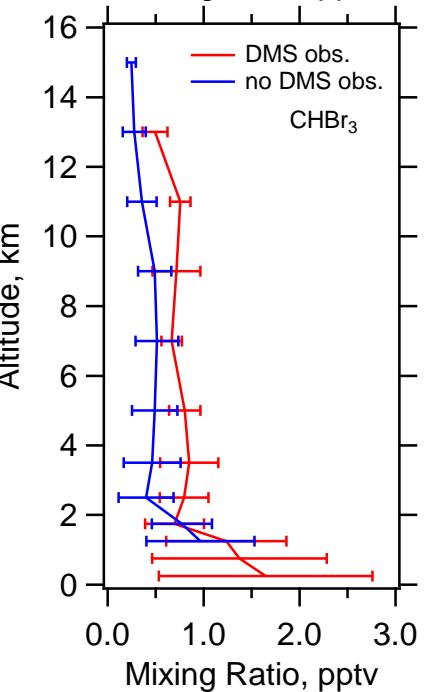
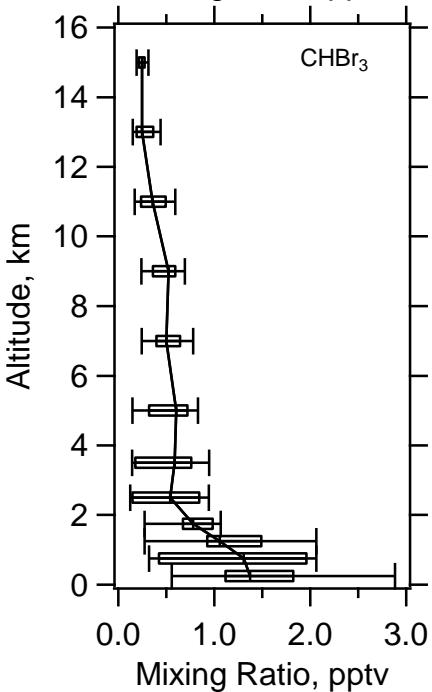
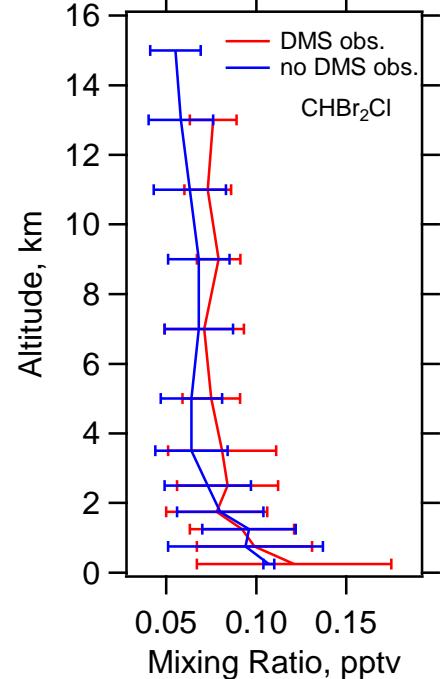
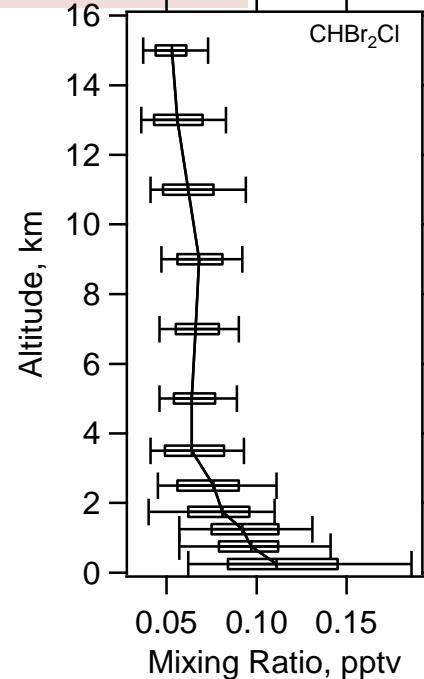
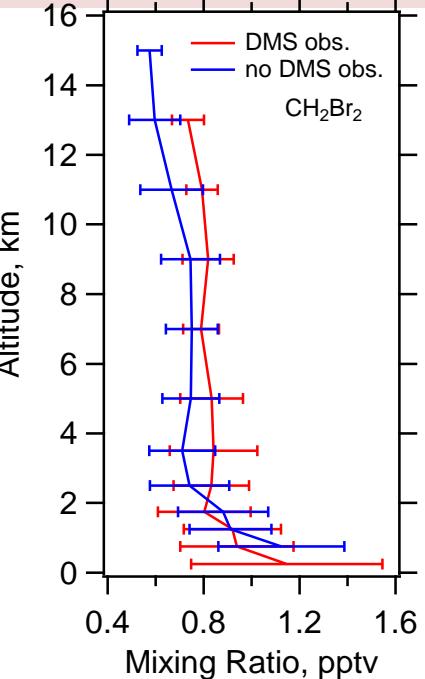
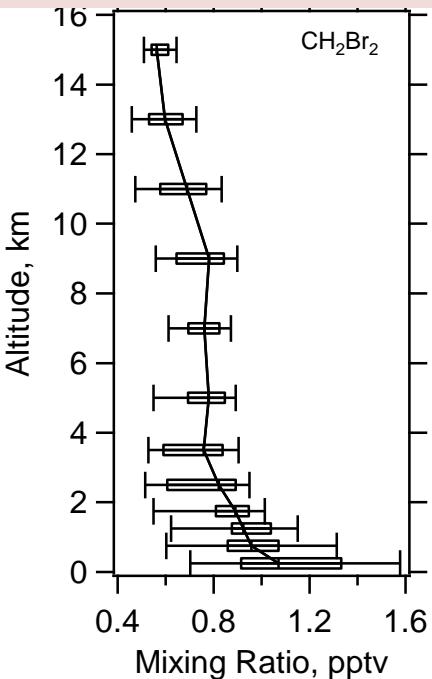
# Marine Boundary Layer Organohalogens

**Figure 3.** TOGA data from the MBL (< 500 m) and U. of York data on surface maps of MODIS sea surface temperatures (SST) and Chlorophyll-a (Chl-a).

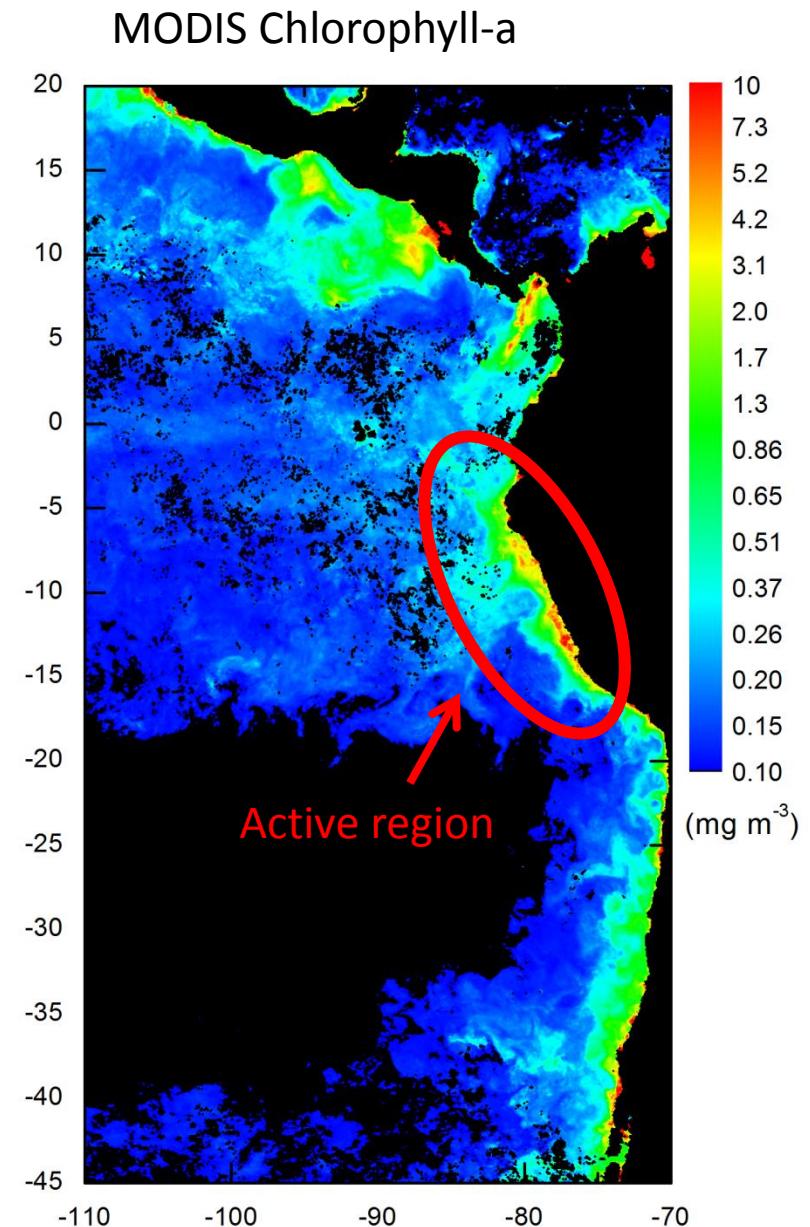
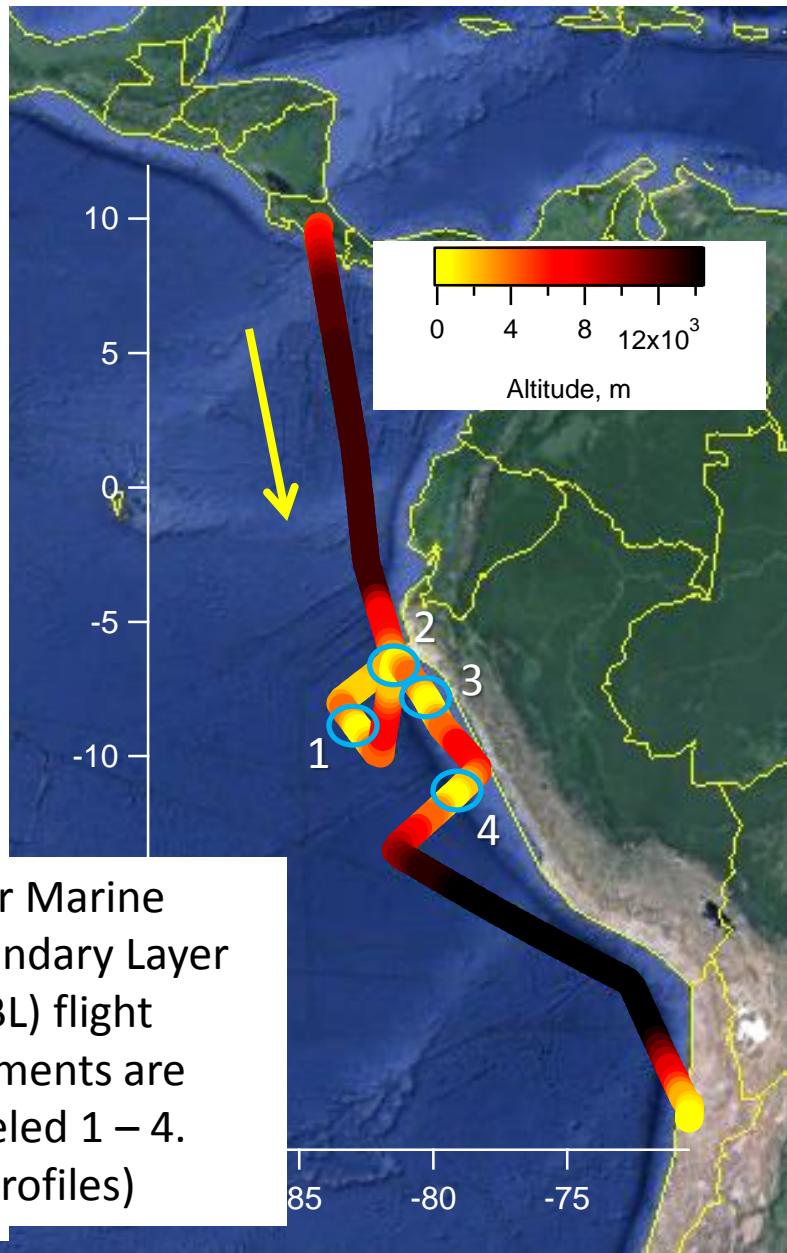
Emissions of Organohalogens and DMS are Similar, and yet not fully dependent on sea surface T (SST), or Chlorophyll-a, but there are similarities.



# Impact of convection on brominated VSLs and methyl iodide

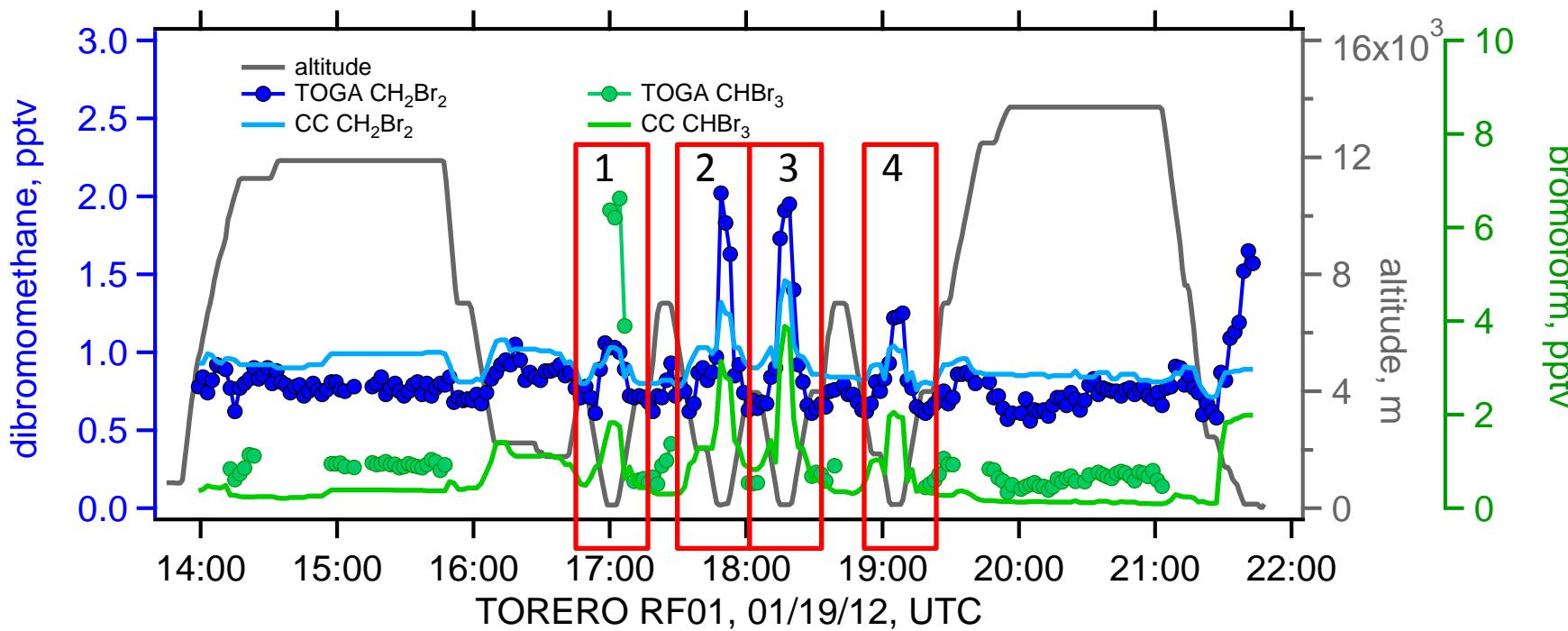


# RF01 – coastal emissions, continental outflow



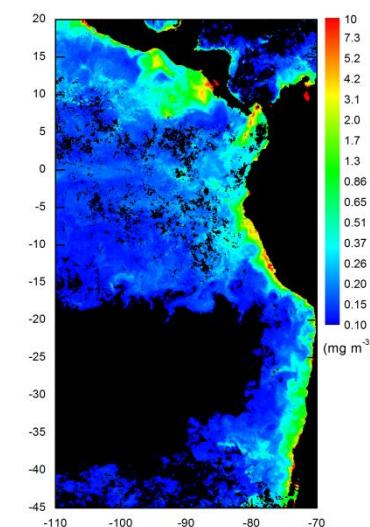
TORERO RF01.

Transit from Costa Rica to Chile, with profiling off the western coast of Peru.

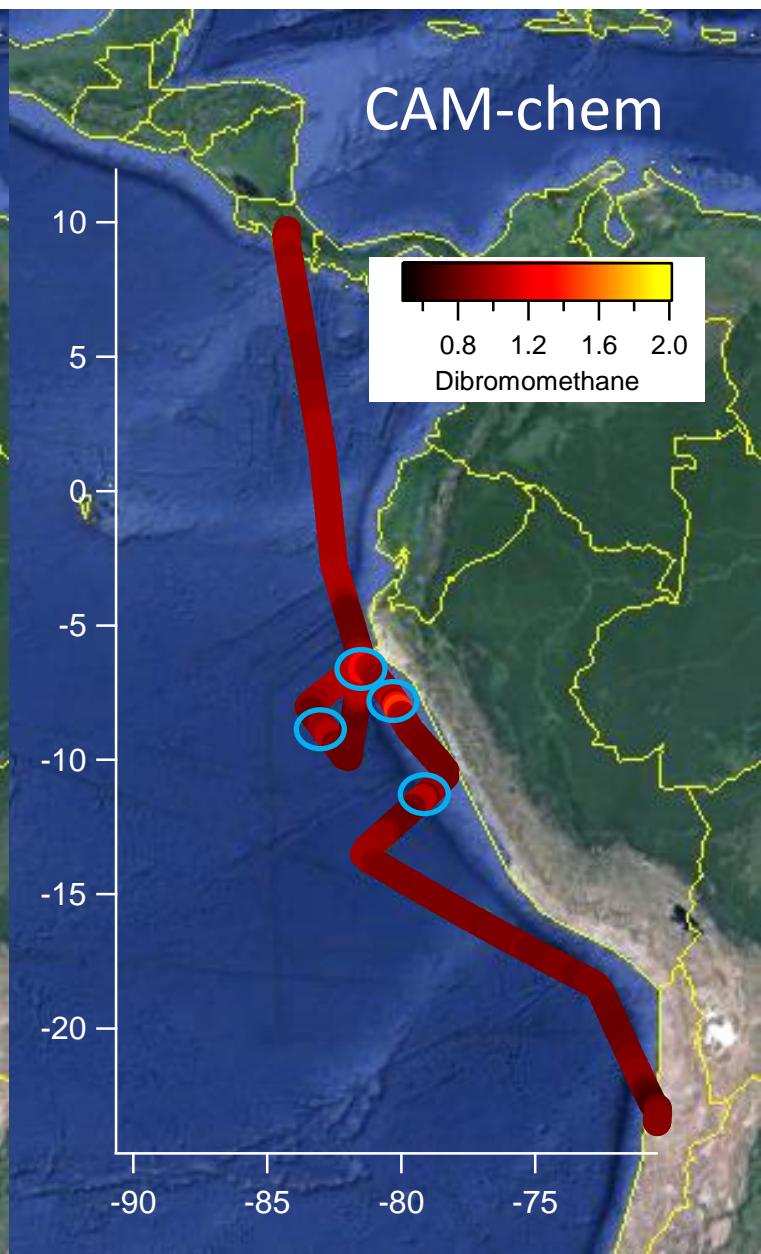
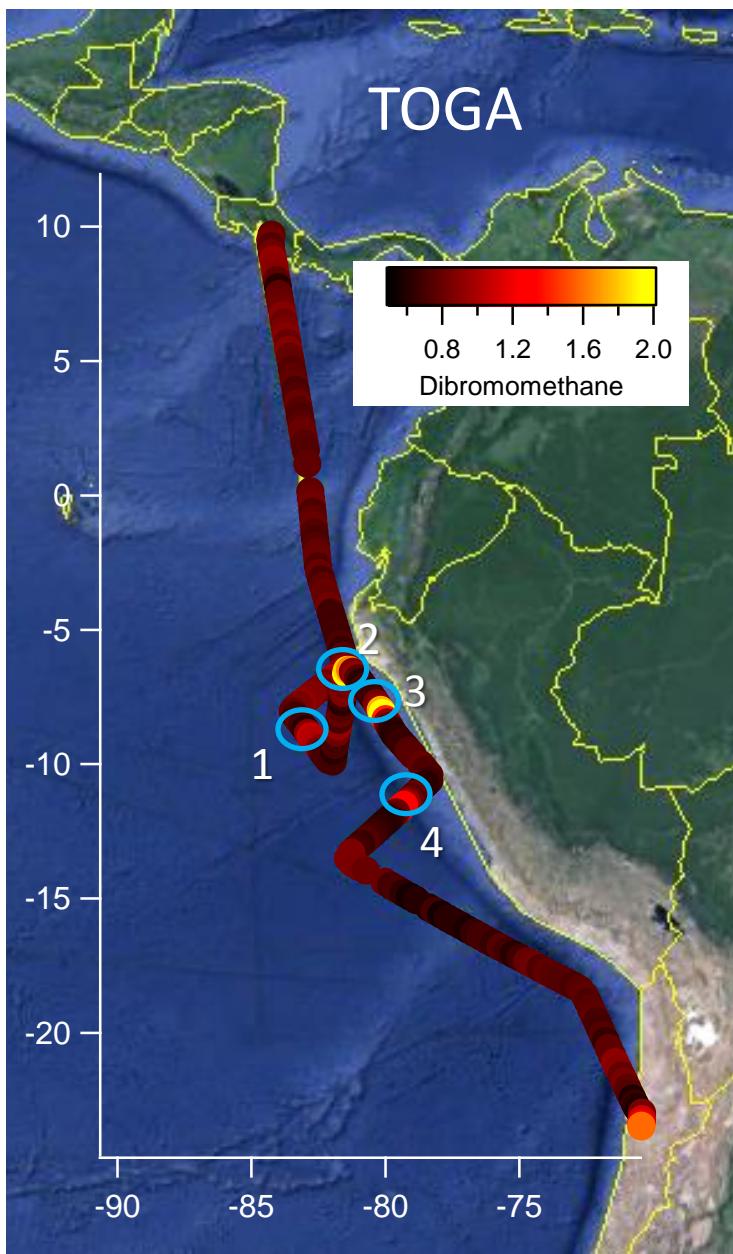


Four MBL flight segments are labeled 1 – 4.

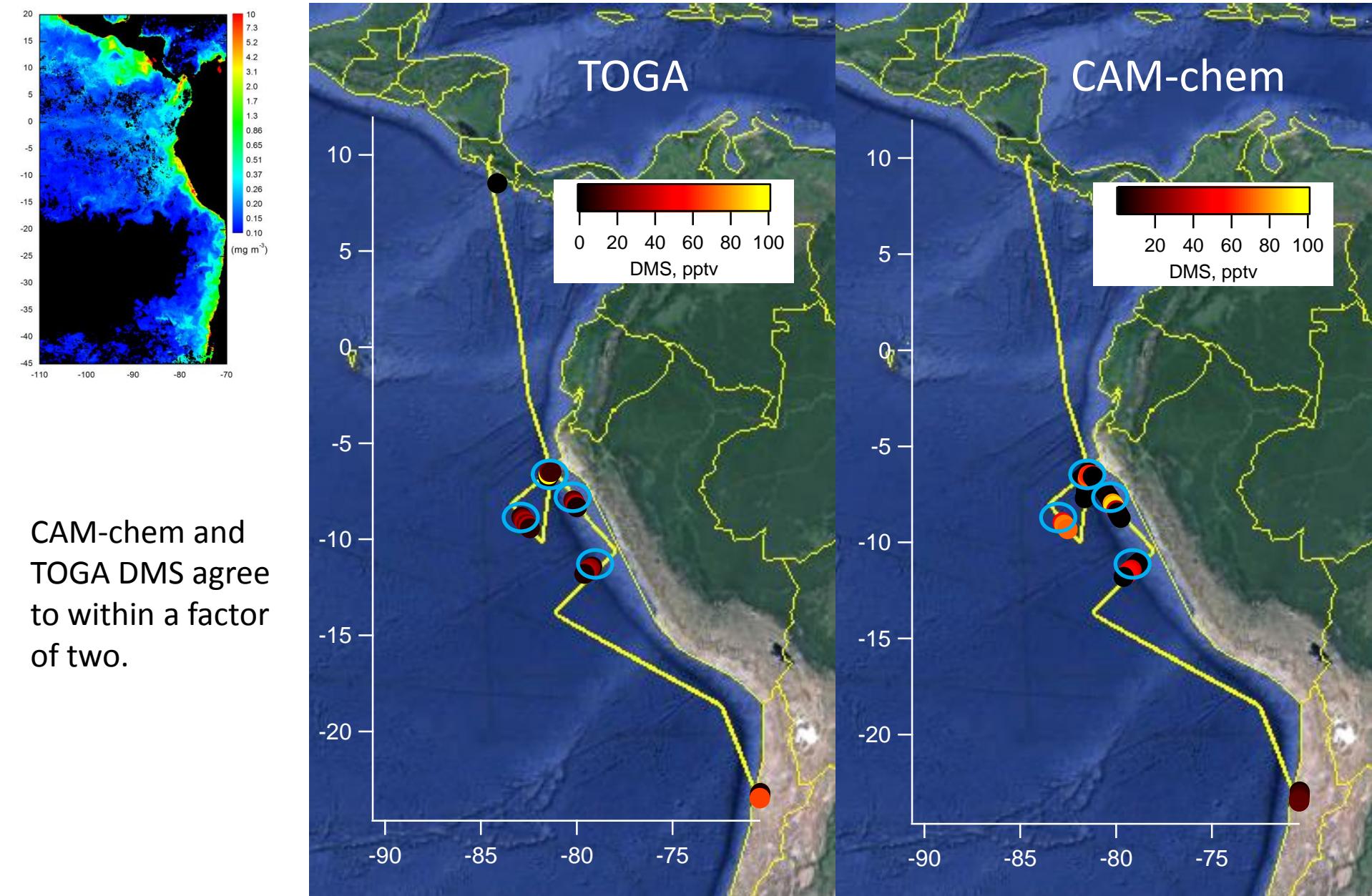
# From RF01



CAM-chem and  
TOGA CH<sub>2</sub>Br<sub>2</sub>  
agree to within a  
factor of two.

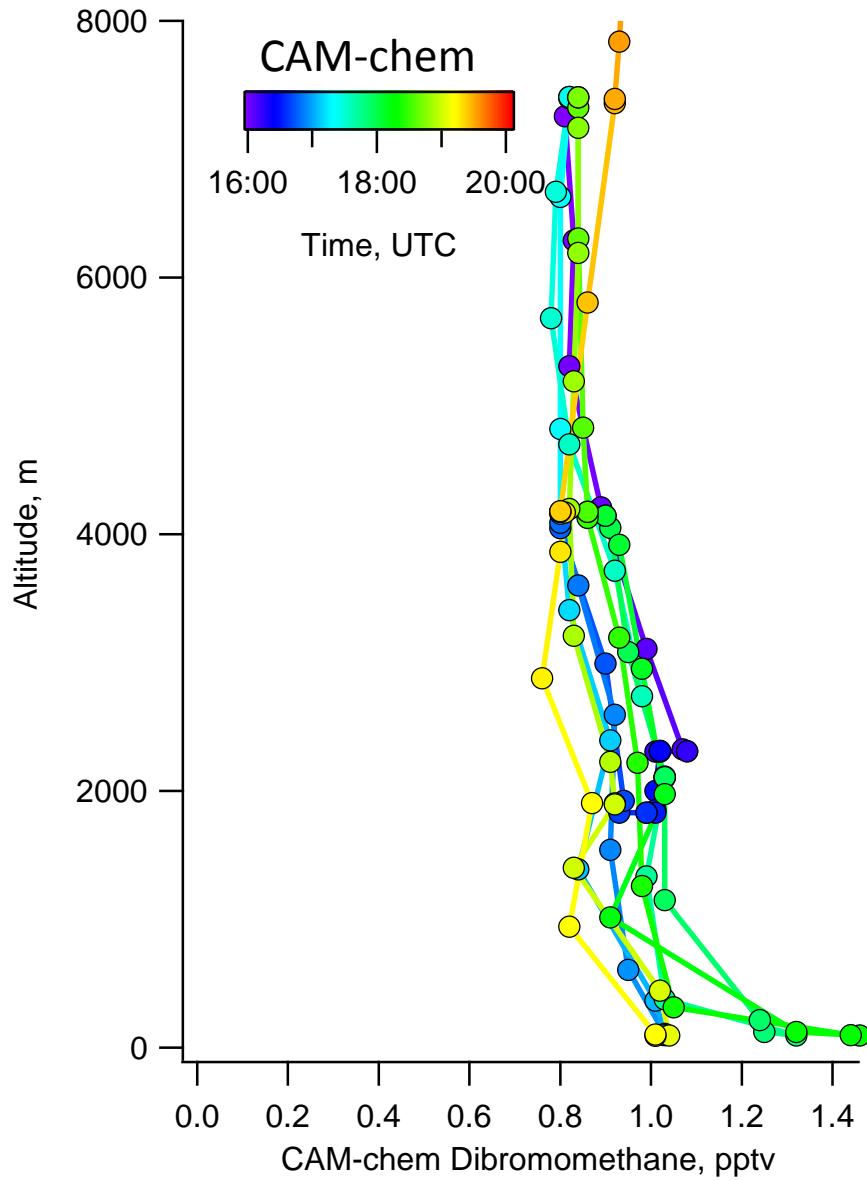
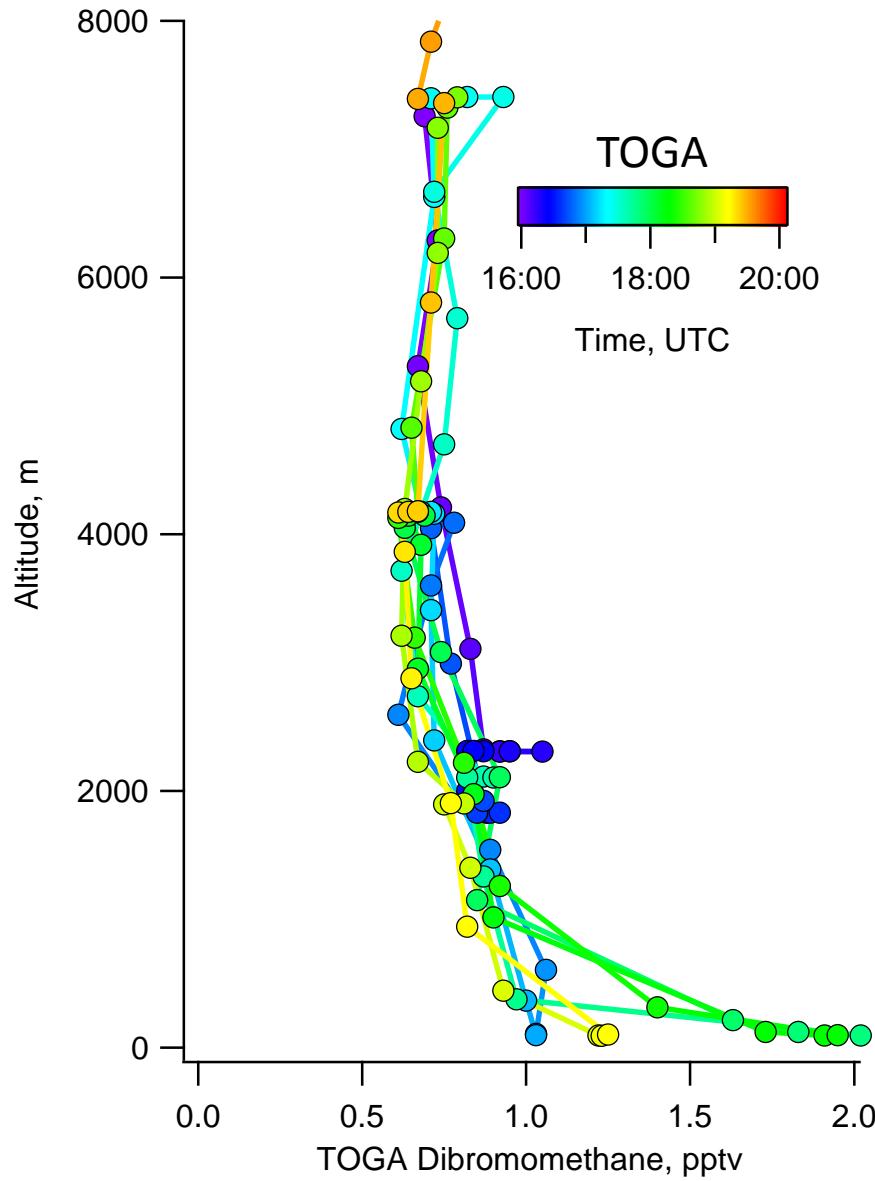


# From RF01



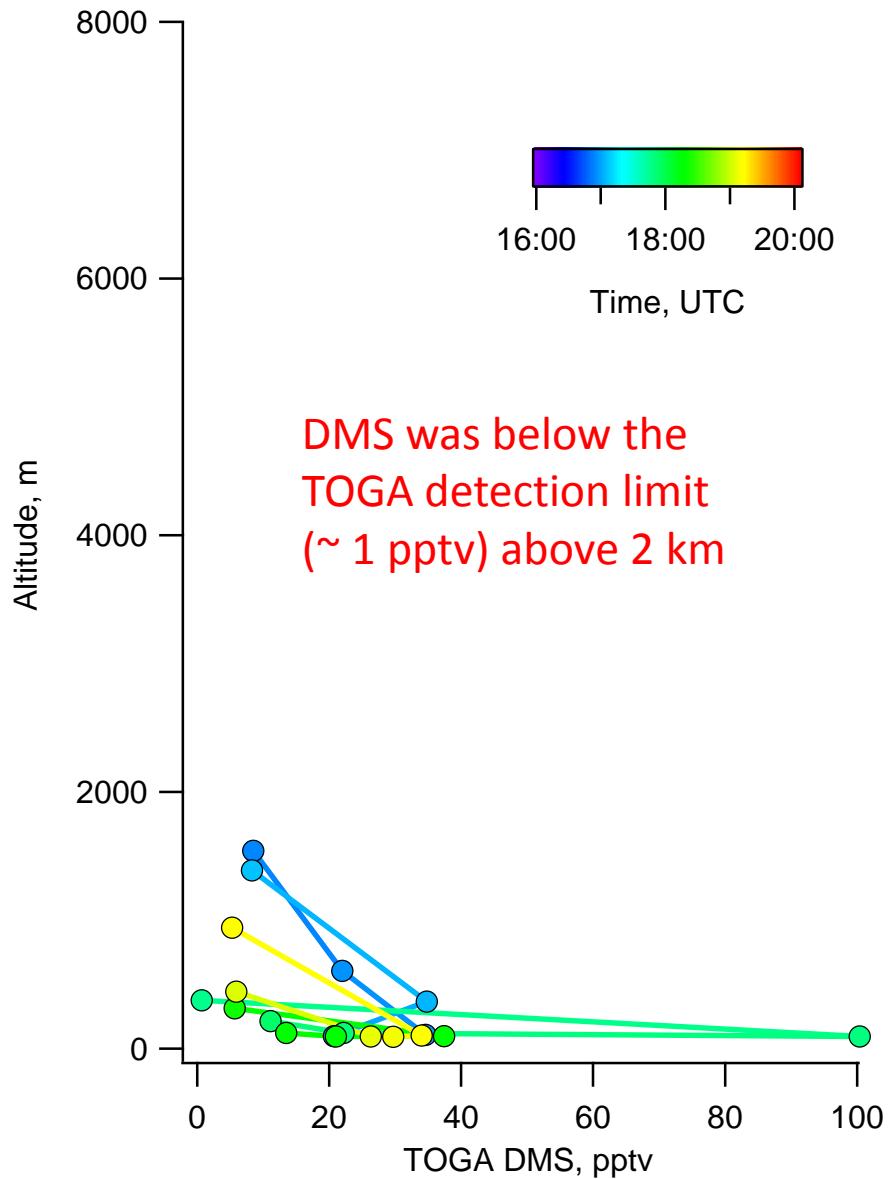
# RF01: 8 profiles off the west coast of Peru

CAM-chem organohalogens and dibromomethane in good agreement with measurements.



# RF01: 8 profiles off the west coast of Peru

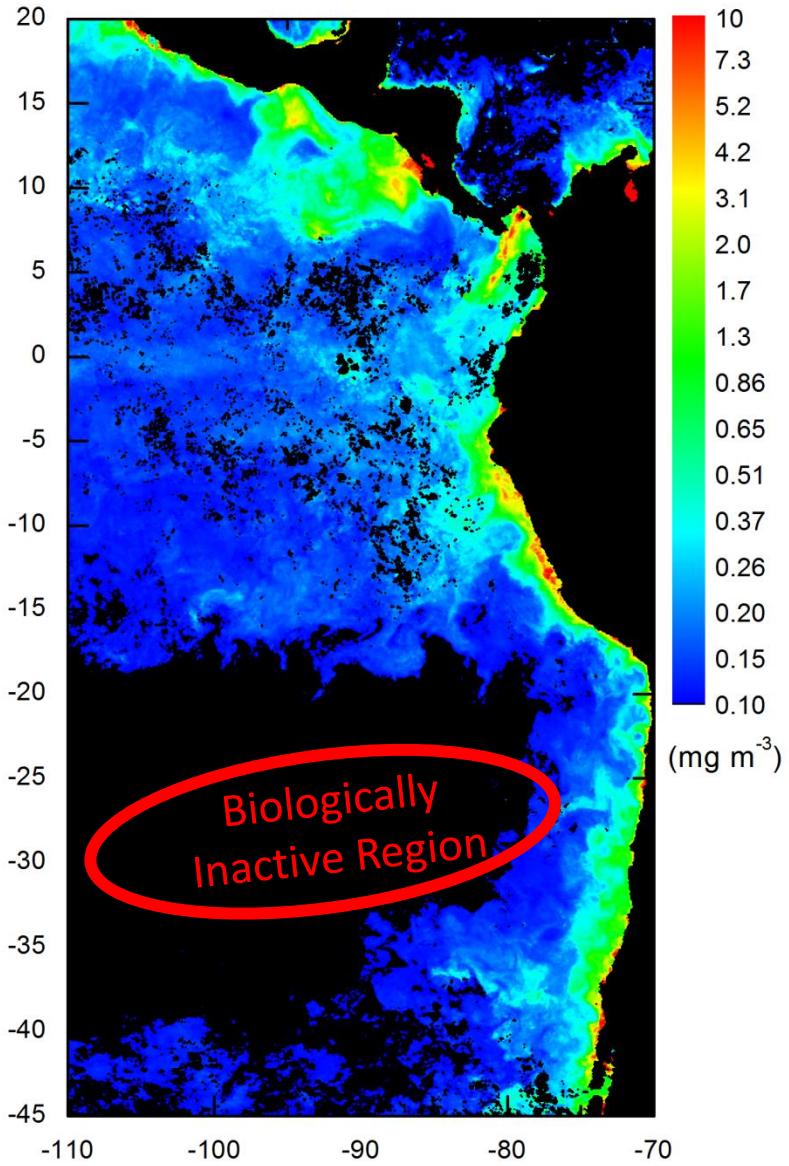
CAM-chem organohalogens and DMS in good agreement with obs.



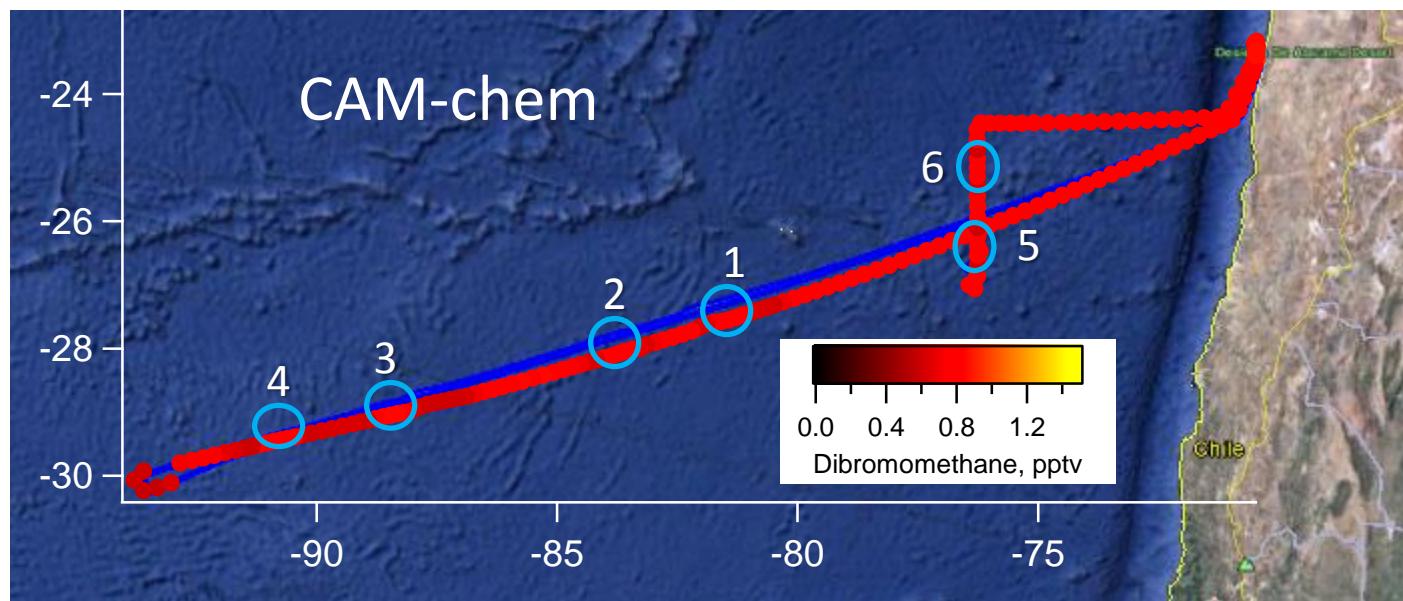
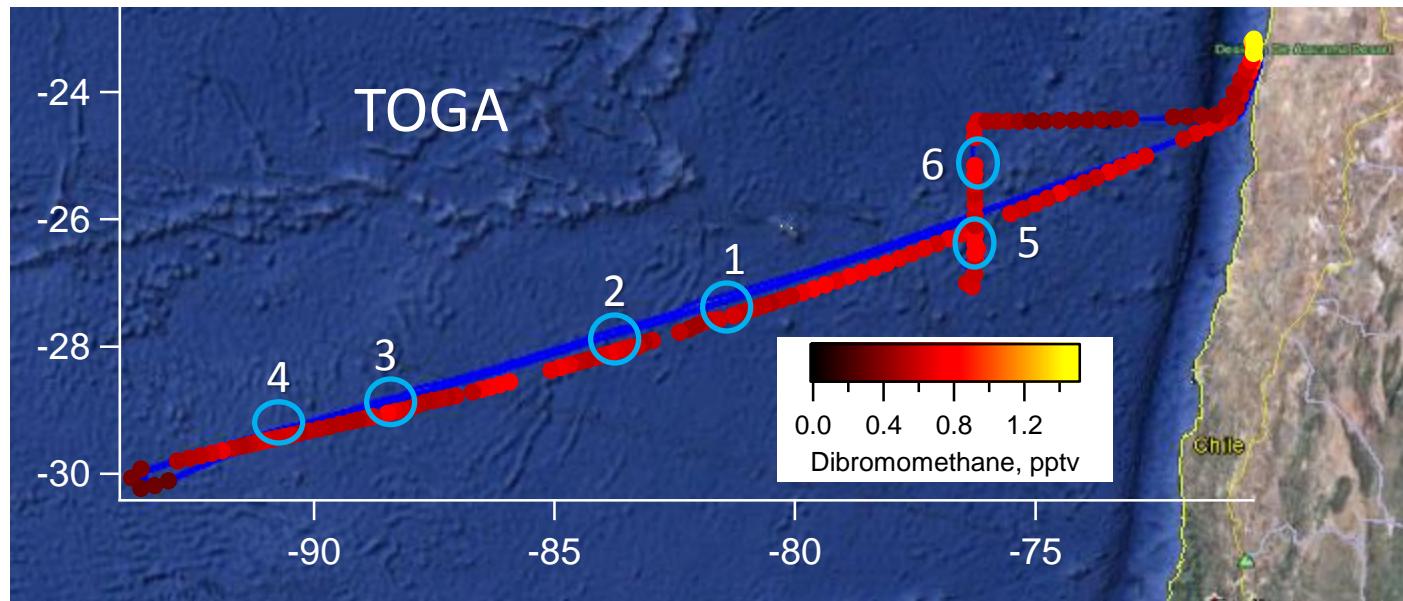
# RF05 – oligotrophic ocean



MODIS Chlorophyll-a

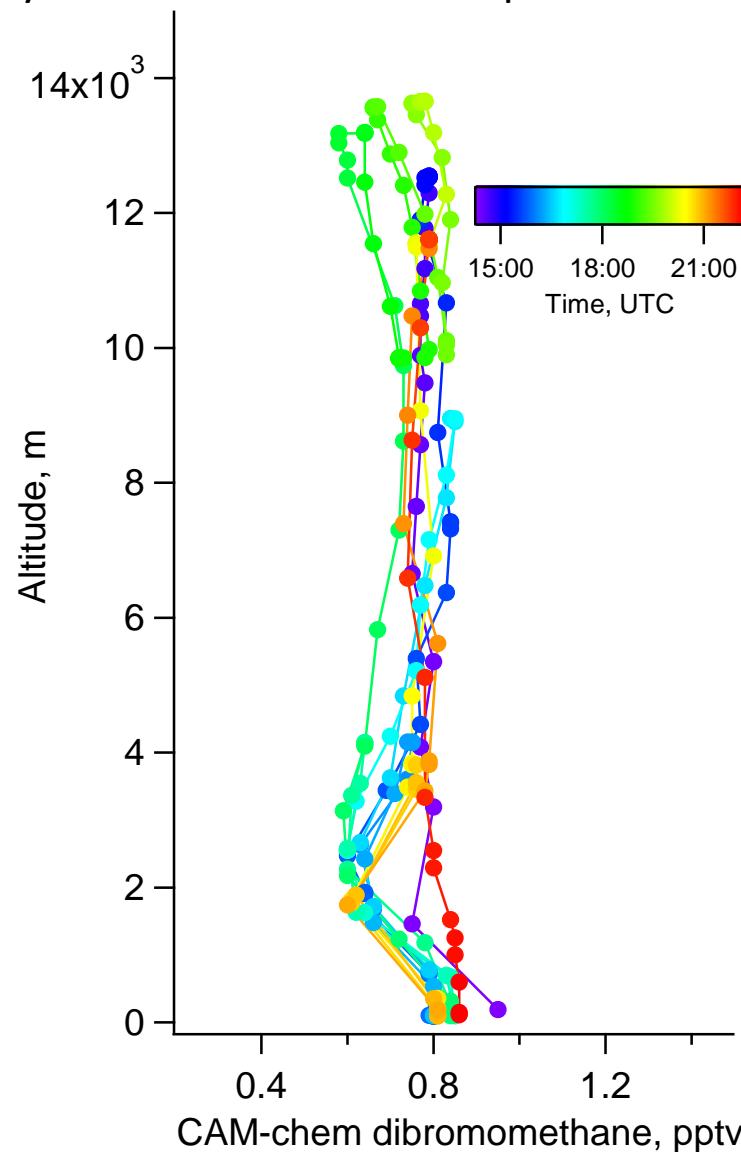
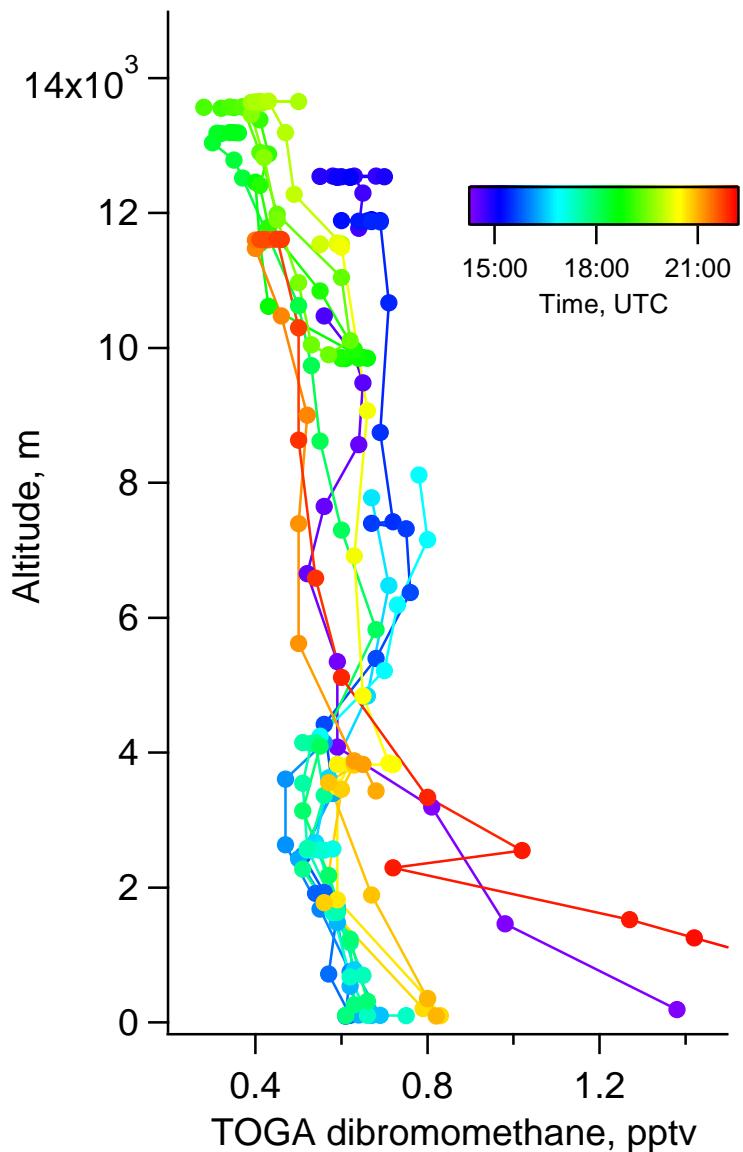


# RF05 – oligotrophic ocean



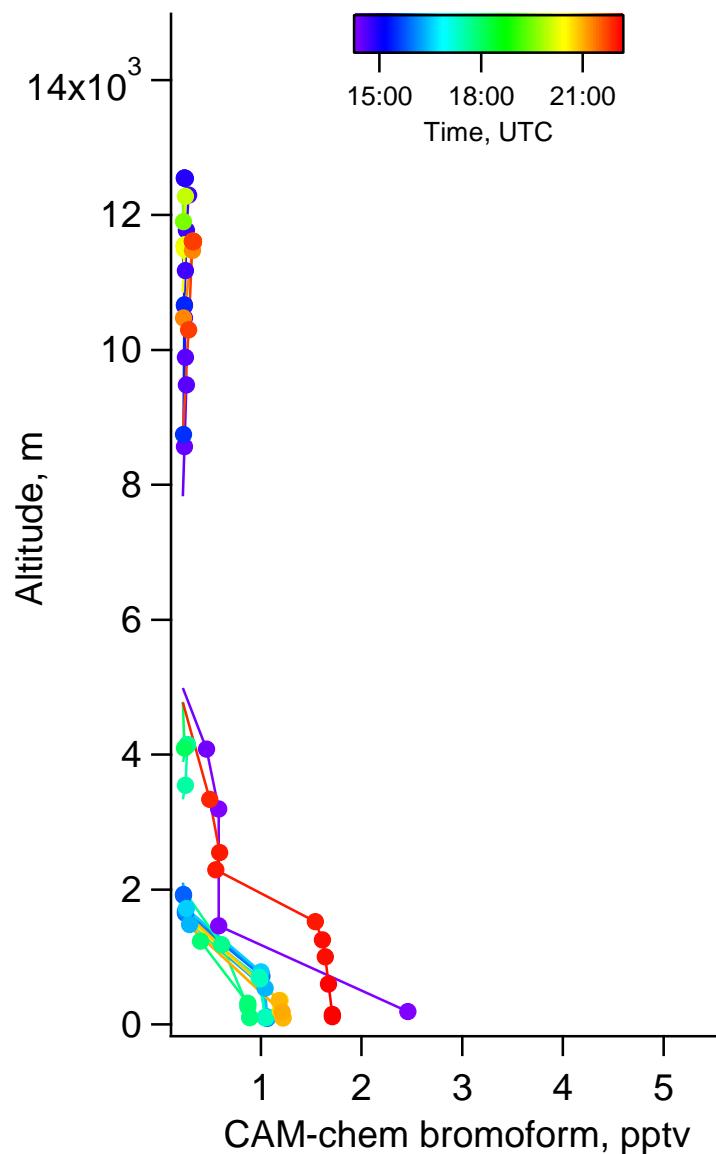
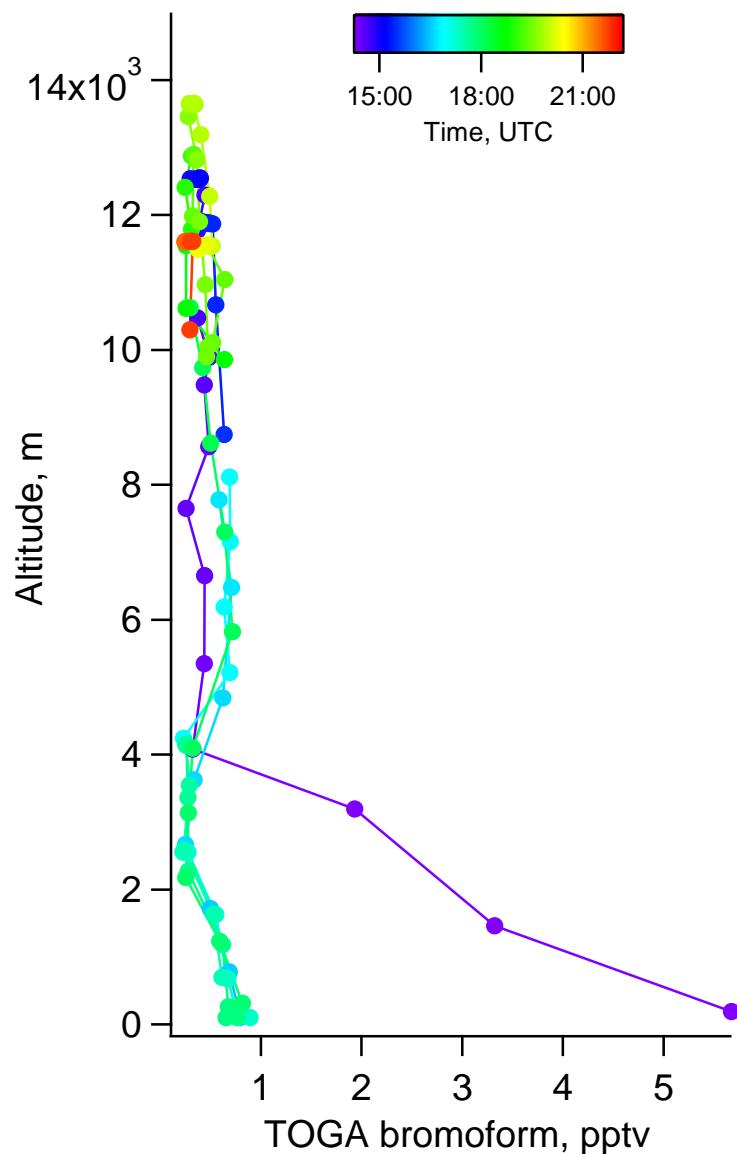
## RF05 – oligotrophic ocean

Model and obs. Are very similar, with the exception of take-off and landing.  
Dibromo in MBL runs 1 – 4 were slightly lower than CAM-chem predicts.



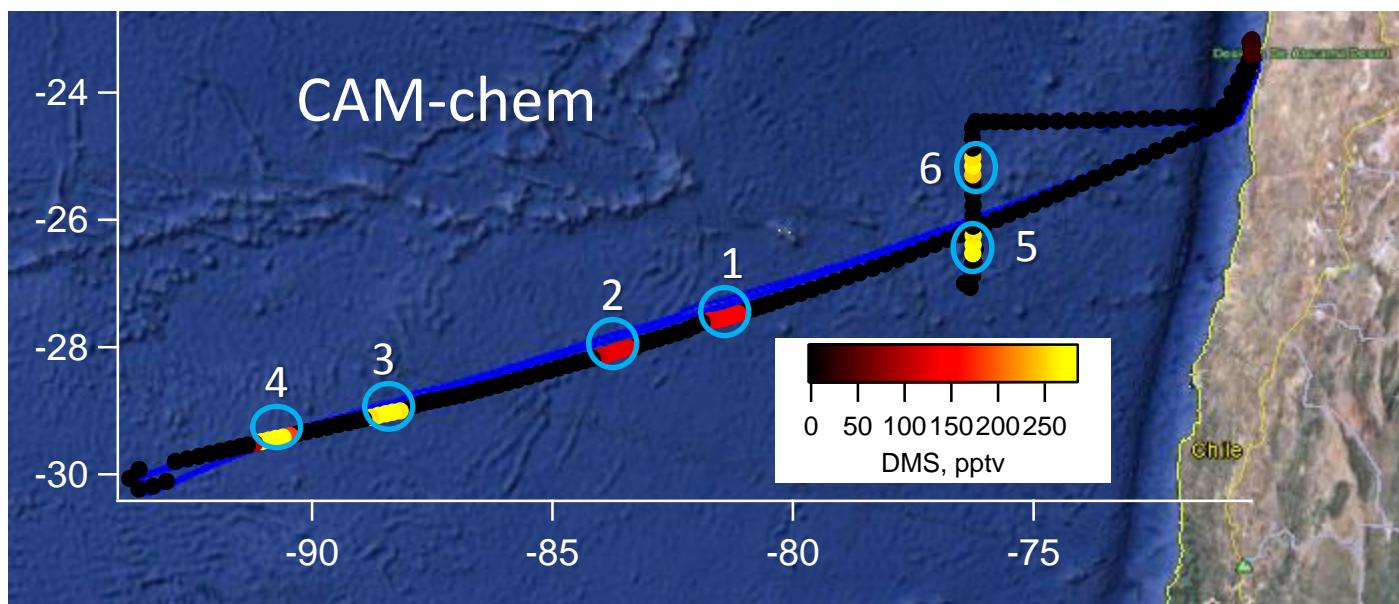
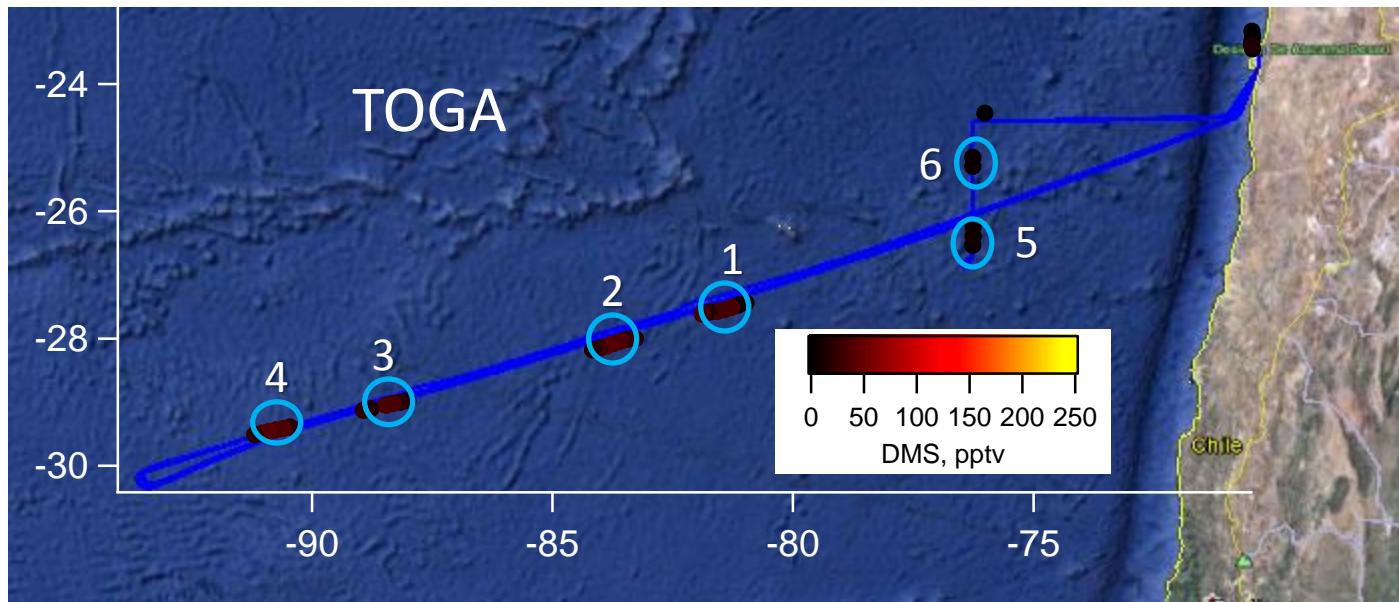
## RF05 – oligotrophic ocean

Model and obs. very similar, and relatively low ( $\sim 1$  ppt) in the oligotrophic ocean MBL.



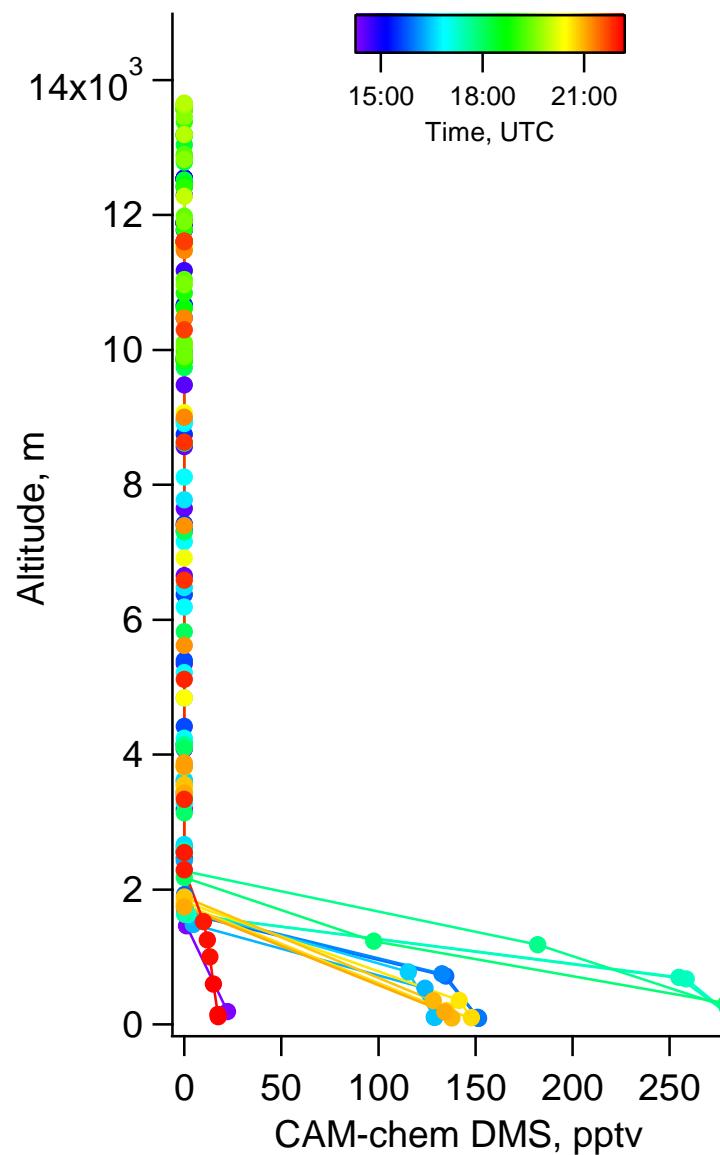
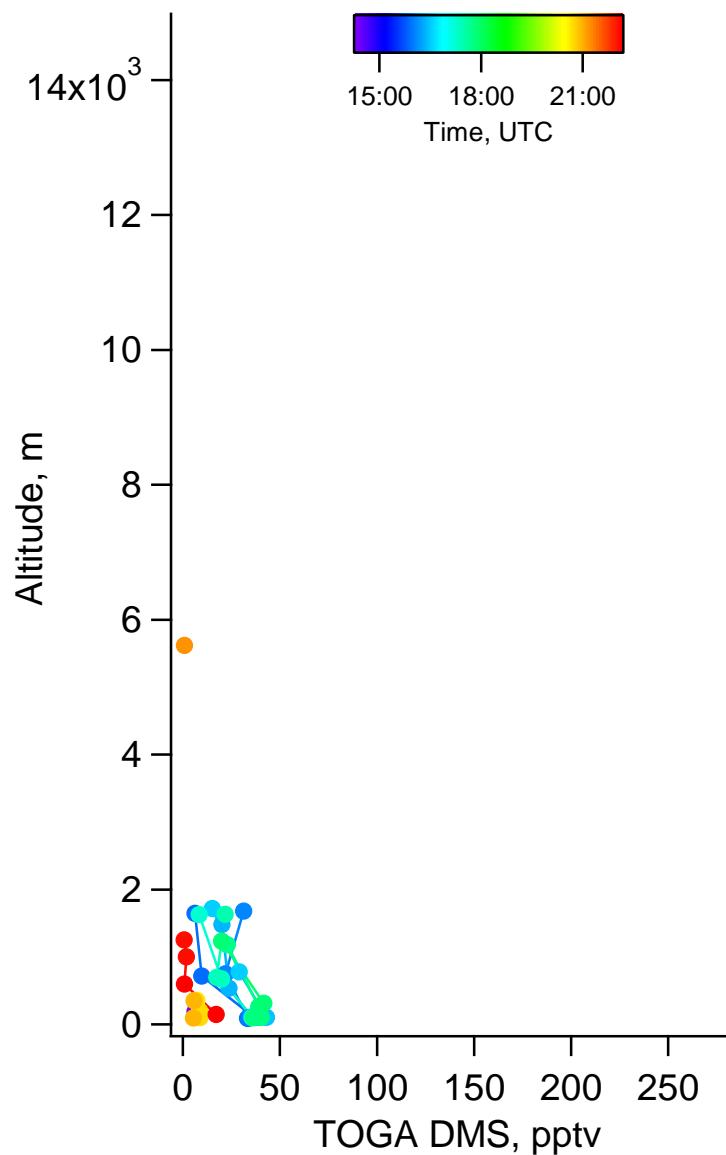
# RF05 – oligotrophic ocean

TOGA DMS  
Is significantly  
Lower than  
CAM-chem  
DMS in all  
MBL runs.

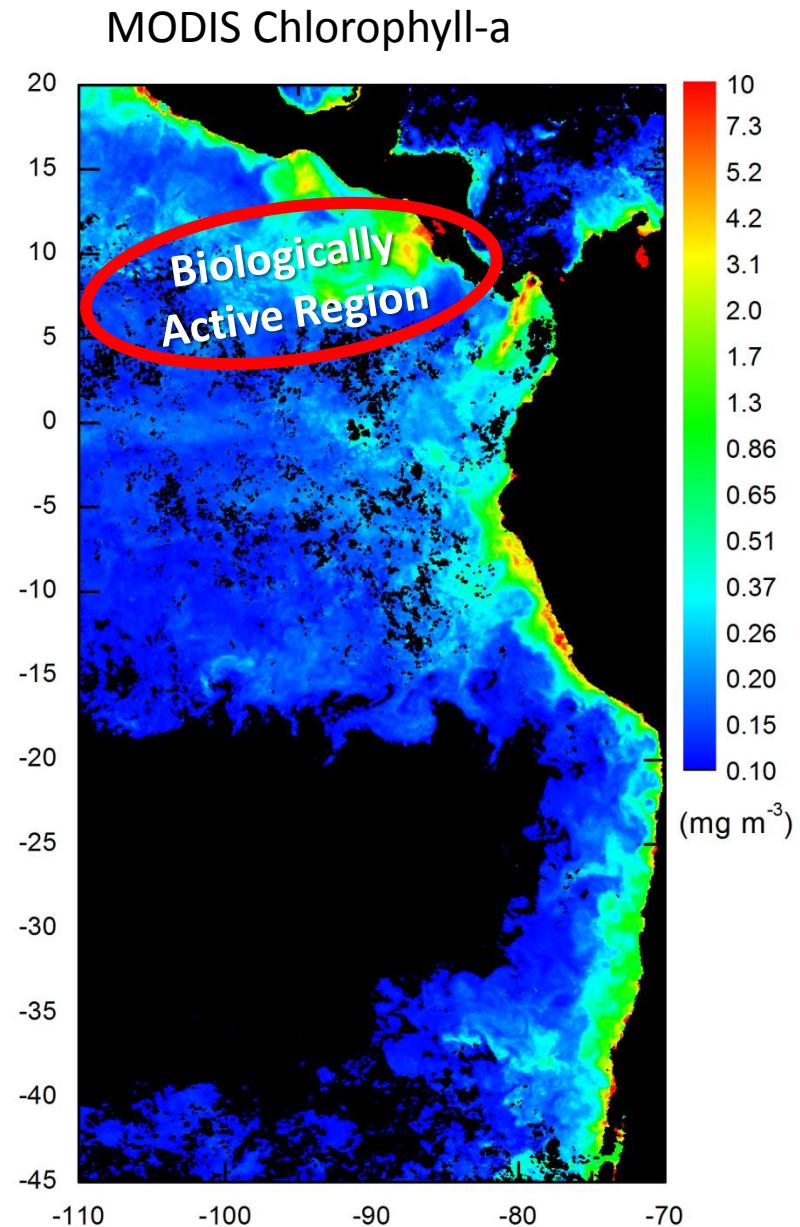


# RF05 – oligotrophic ocean

Model DMS >> TOGA DMS. Largest disagreement at the furthest two MBL runs.

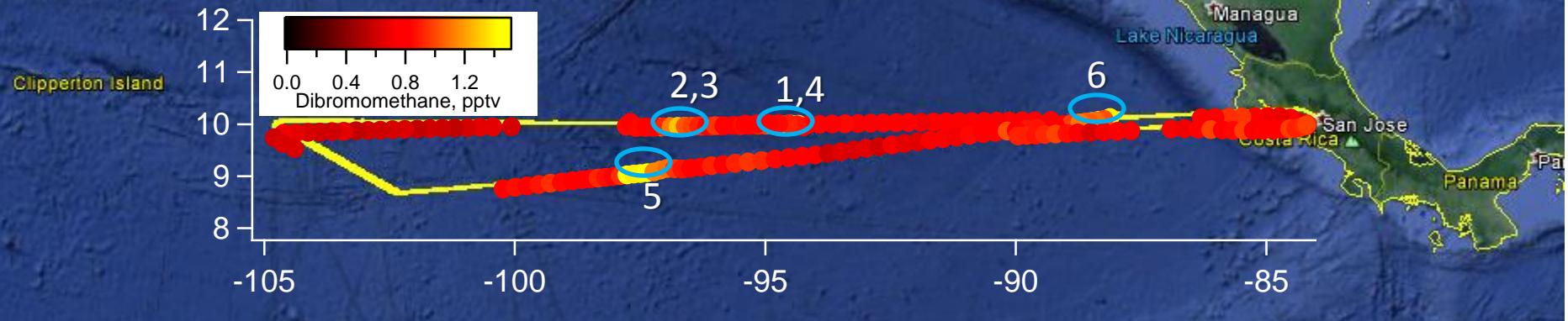


# RF12 – biologically active region/convection MBL

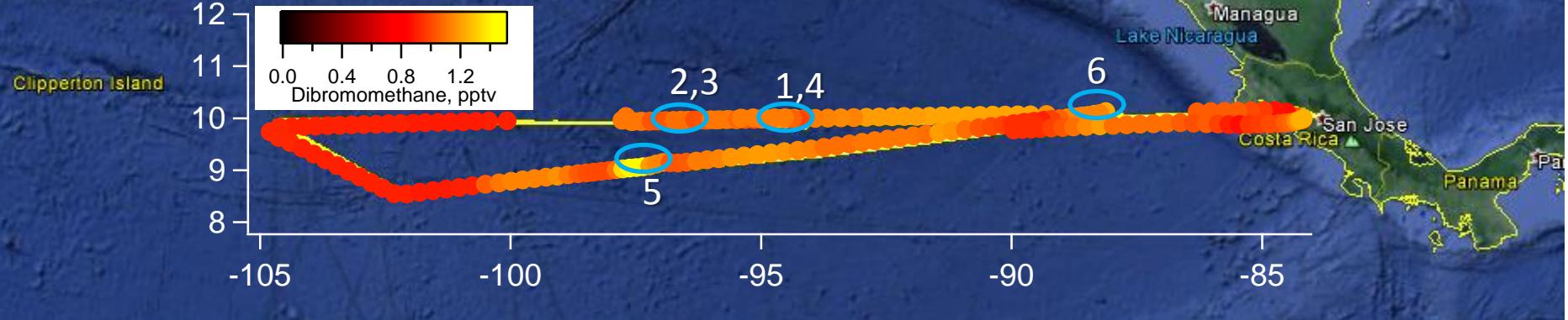


## RF12 – biologically active region

TOGA

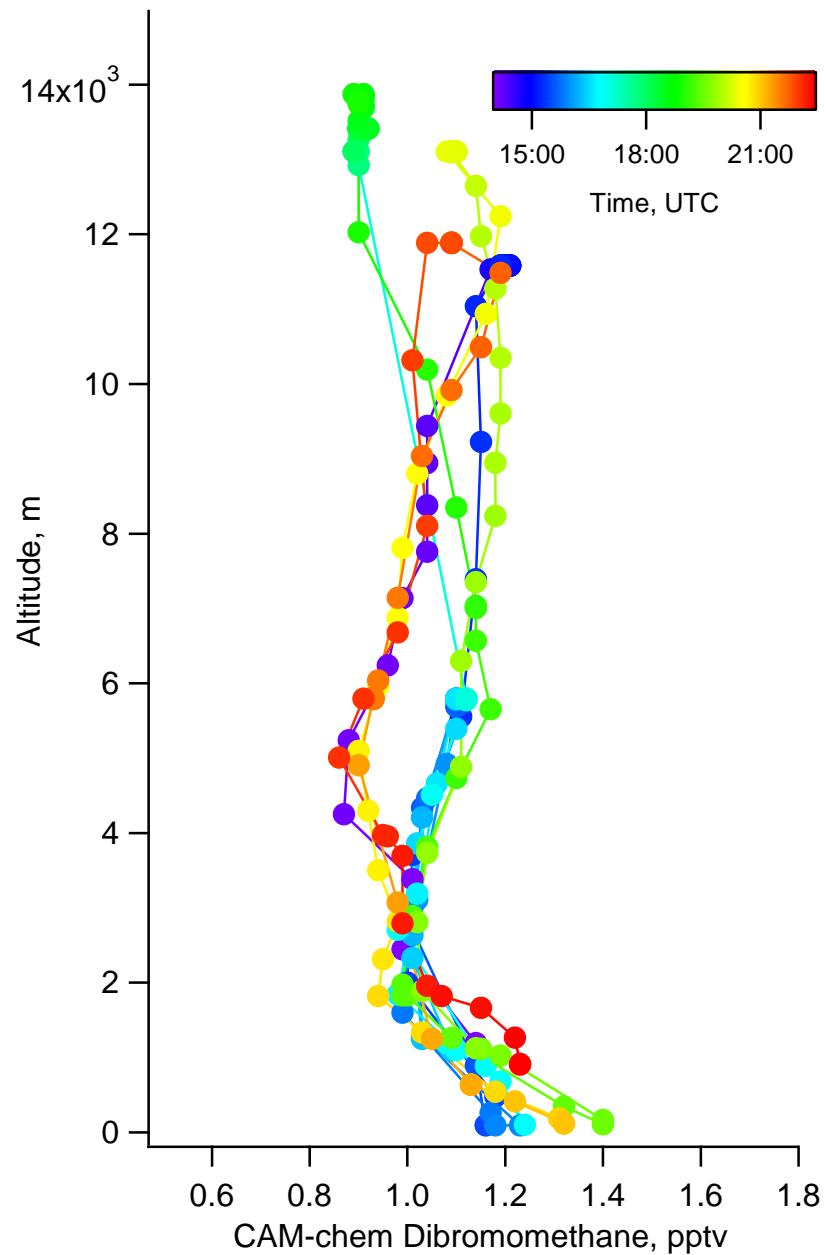
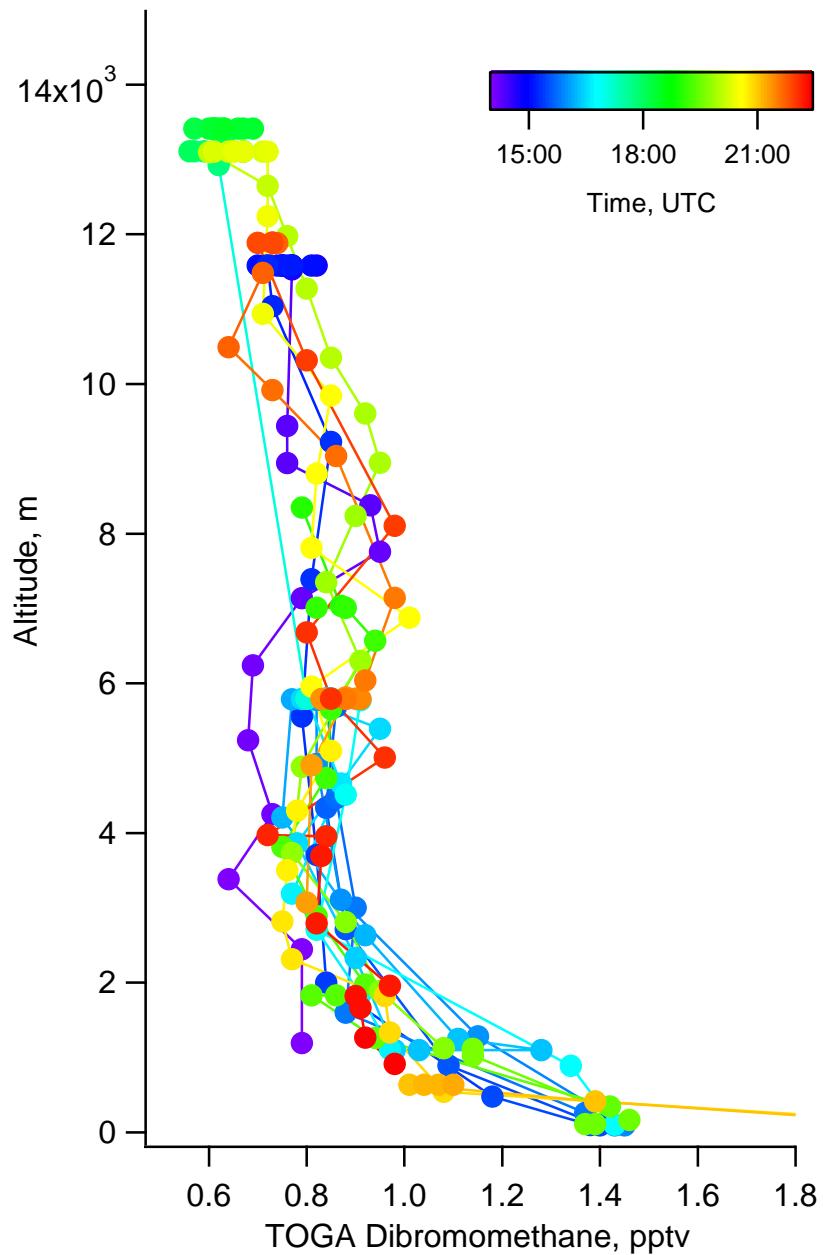


CAM-chem



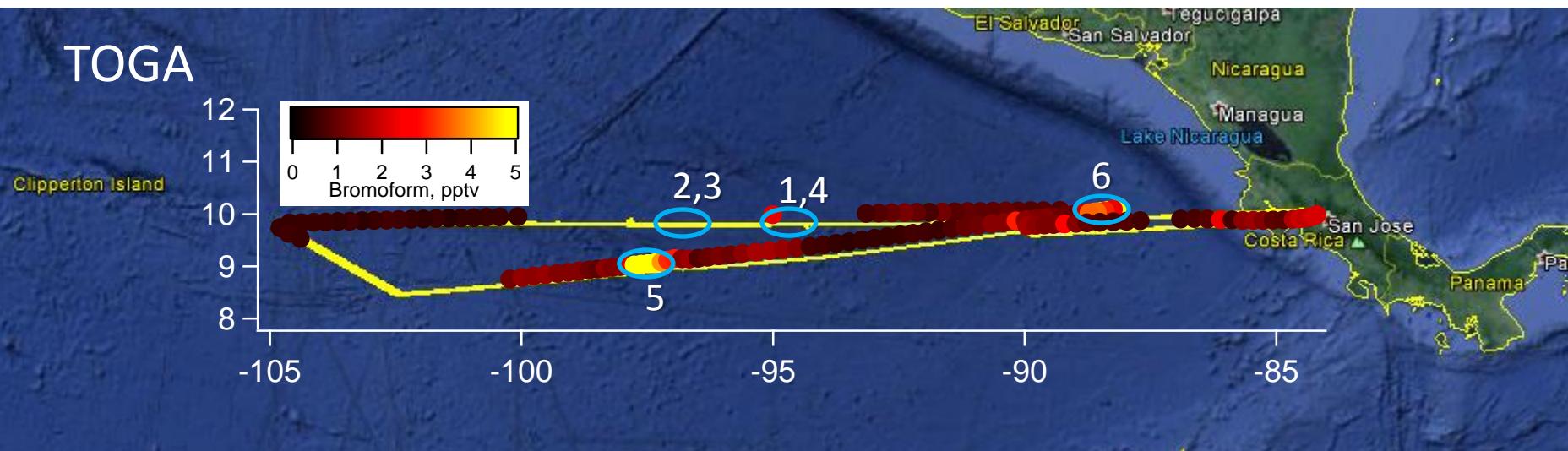
# RF12 – biologically active region

Observations are a little higher than the model at the surface, and lower aloft.

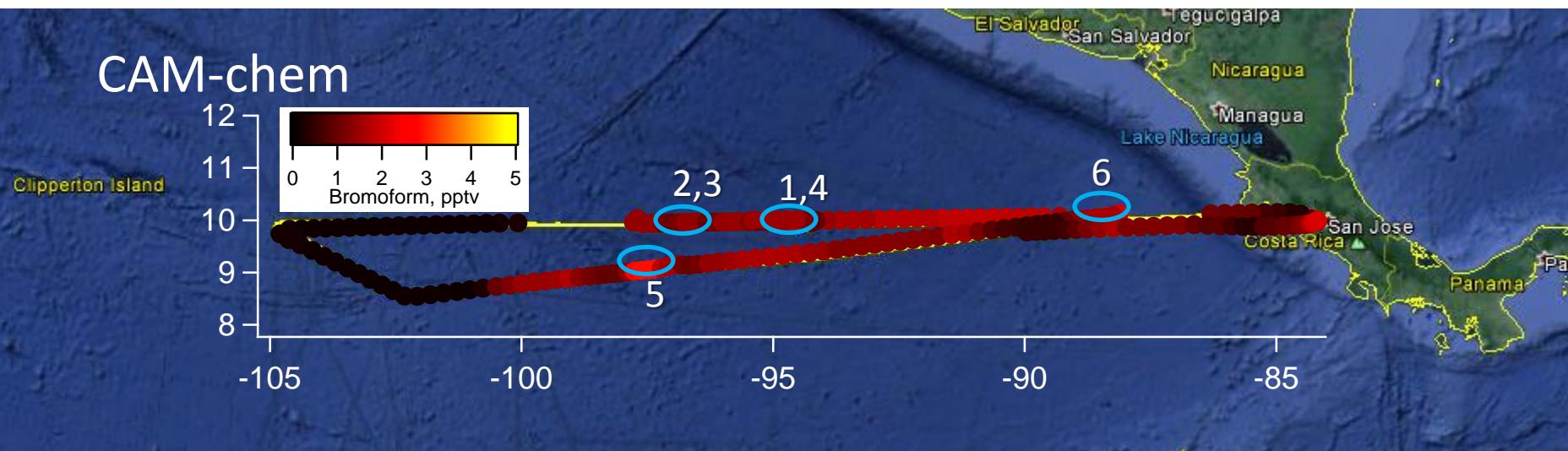


## RF12 – biologically active region

TOGA

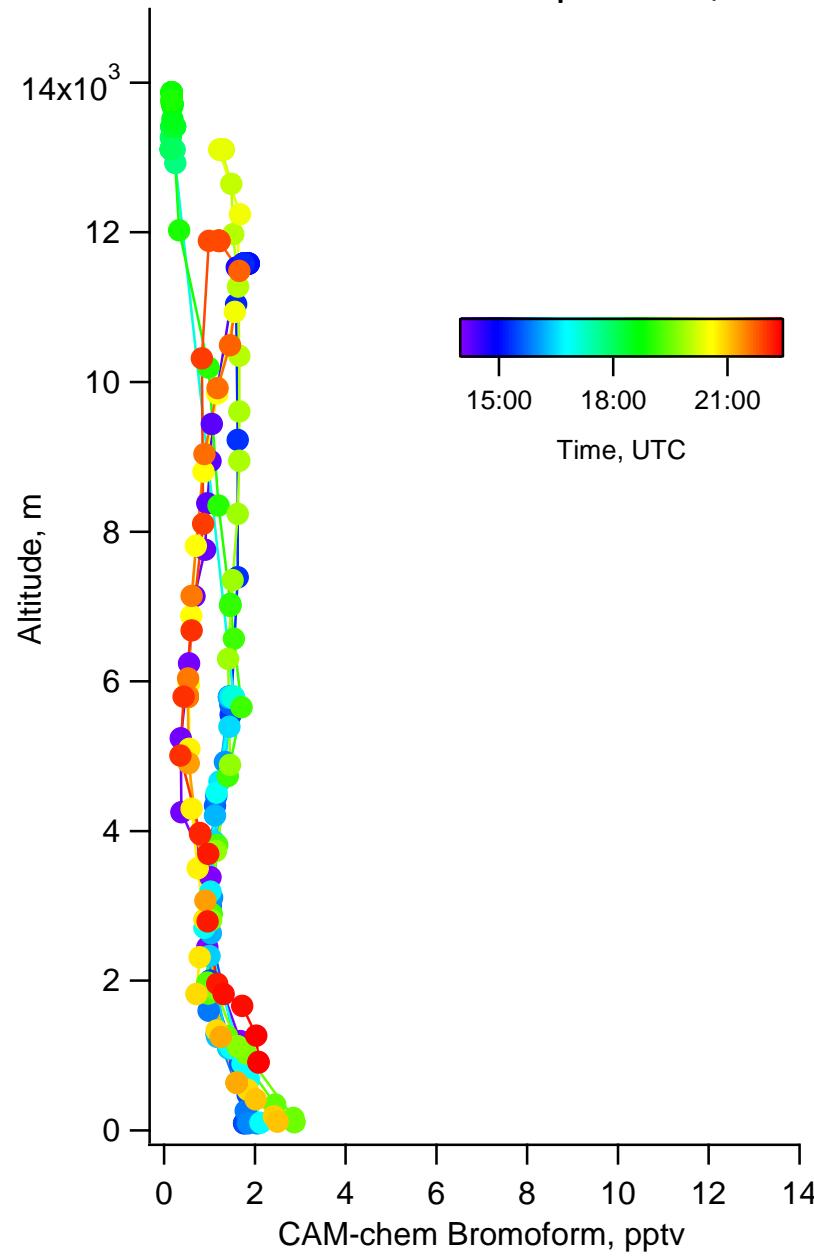
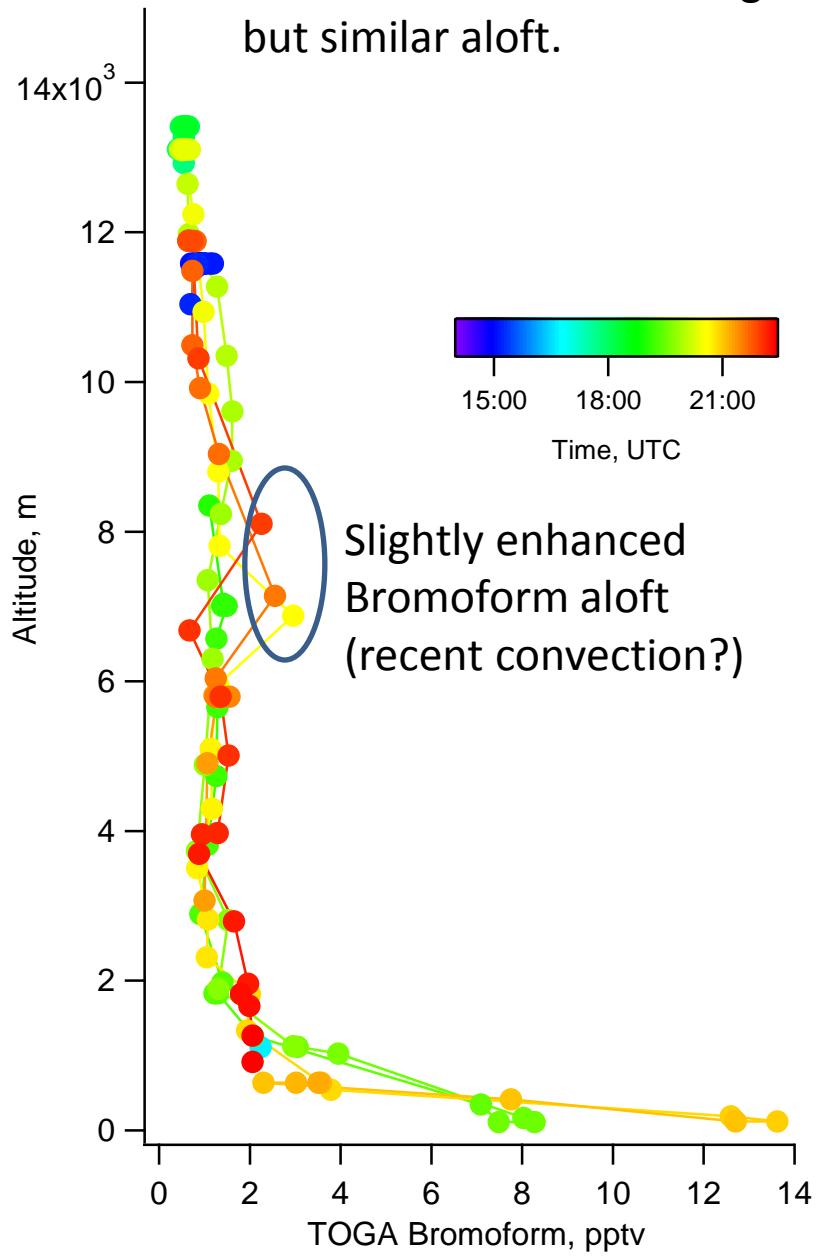


CAM-chem

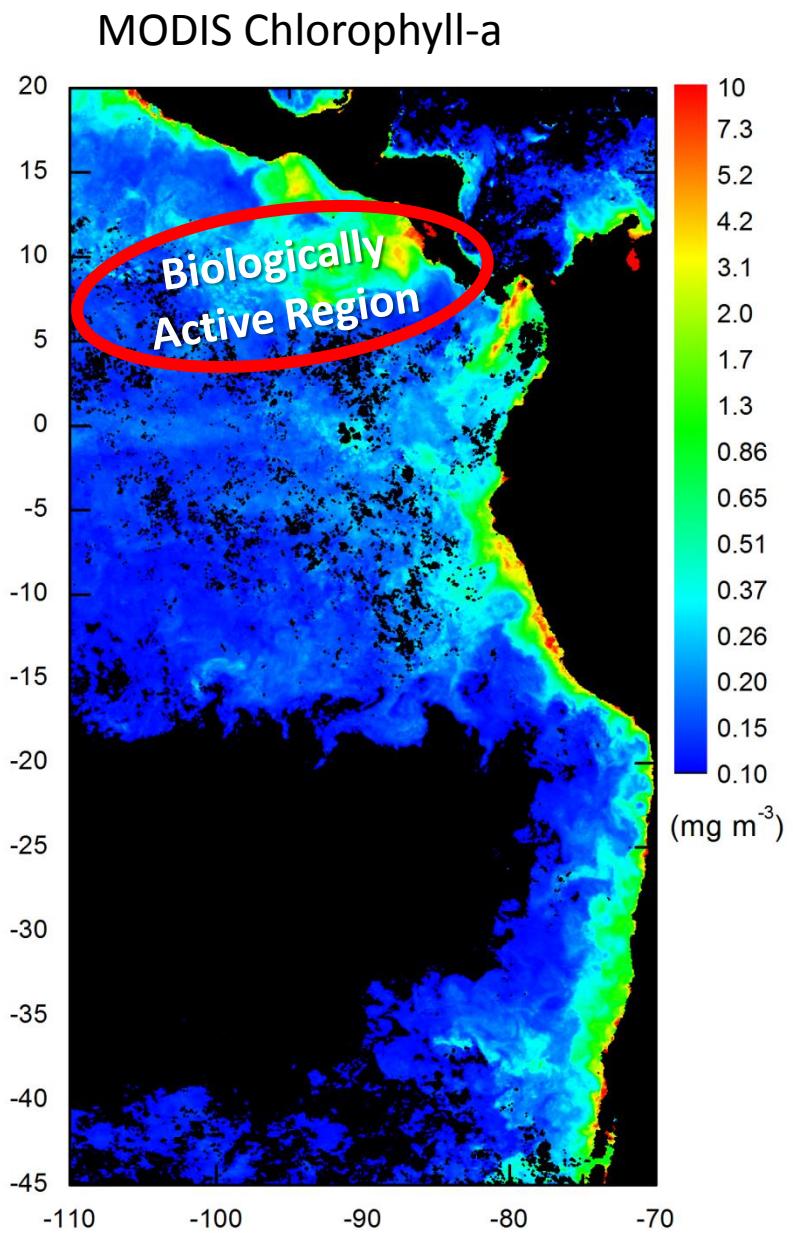


## RF12 – biologically active region

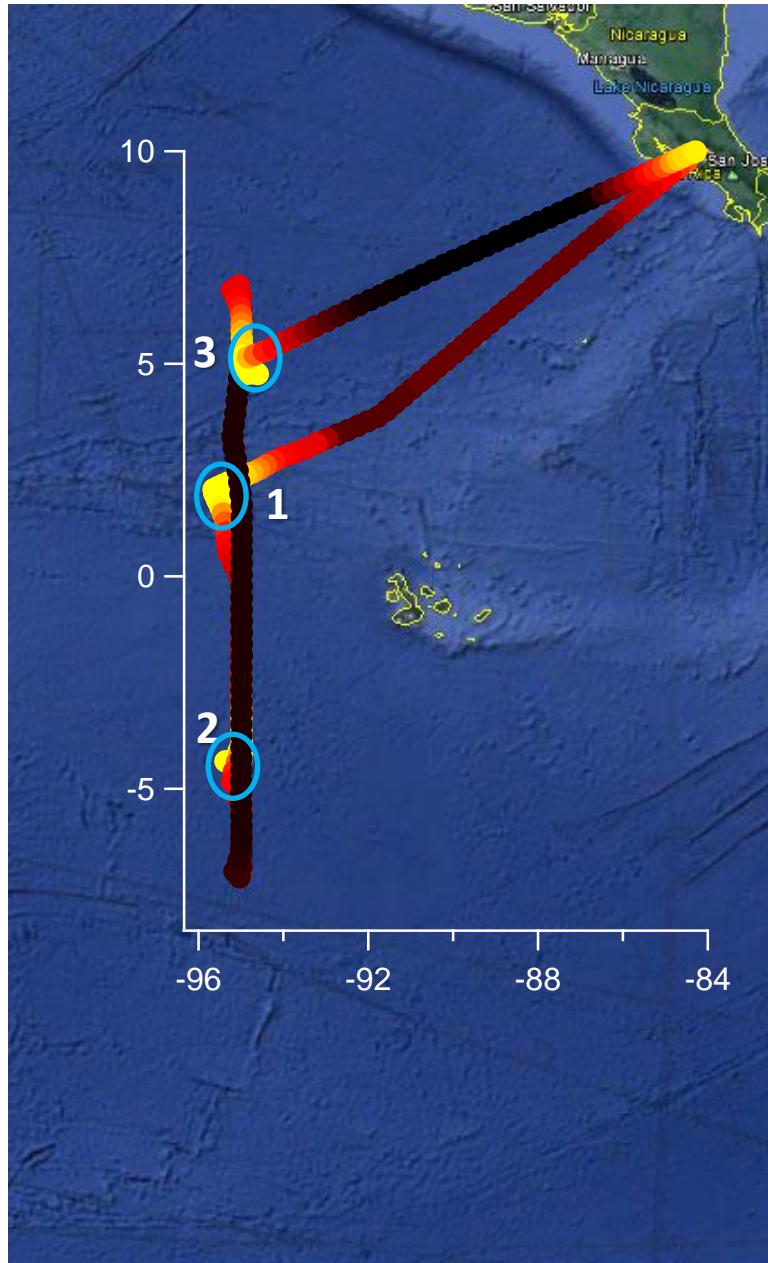
Observations are much higher at the surface than the model predicts, but similar aloft.



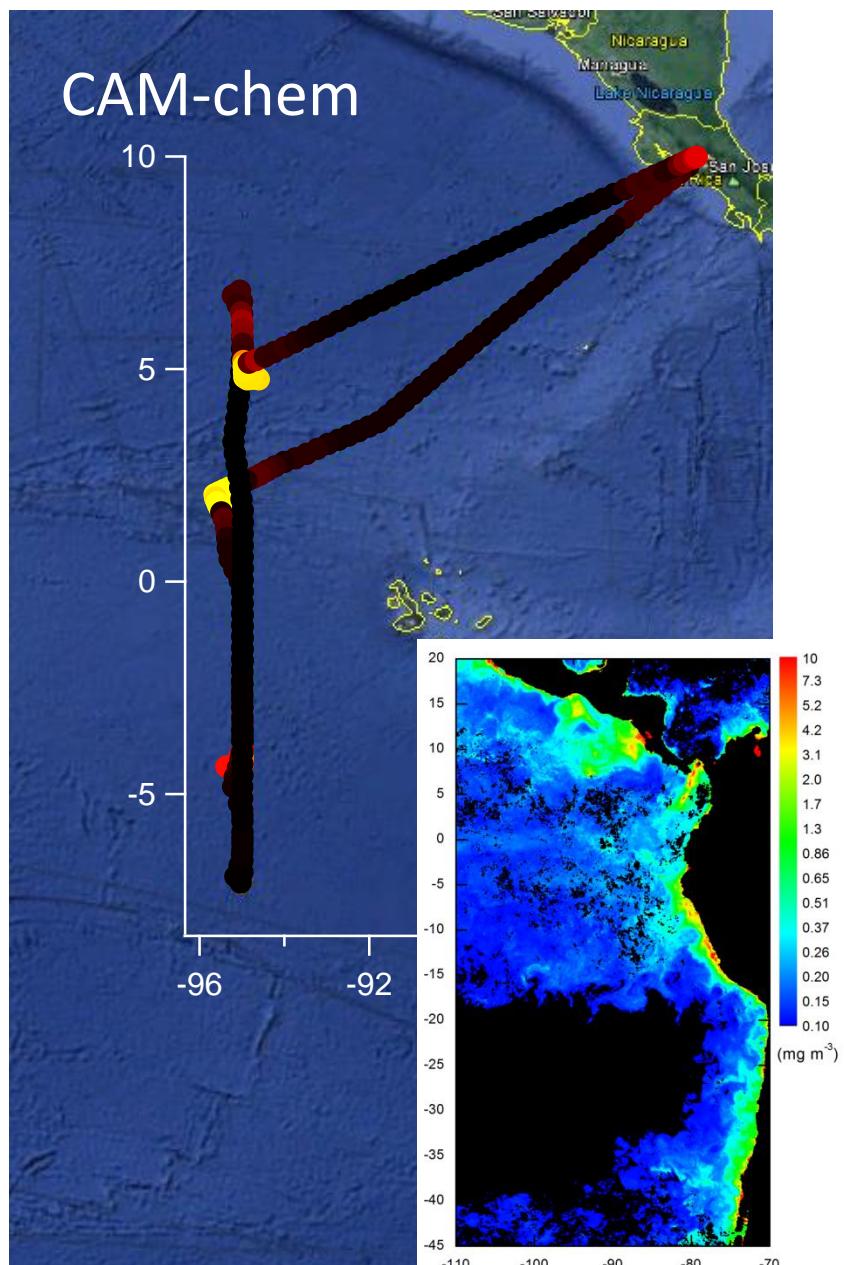
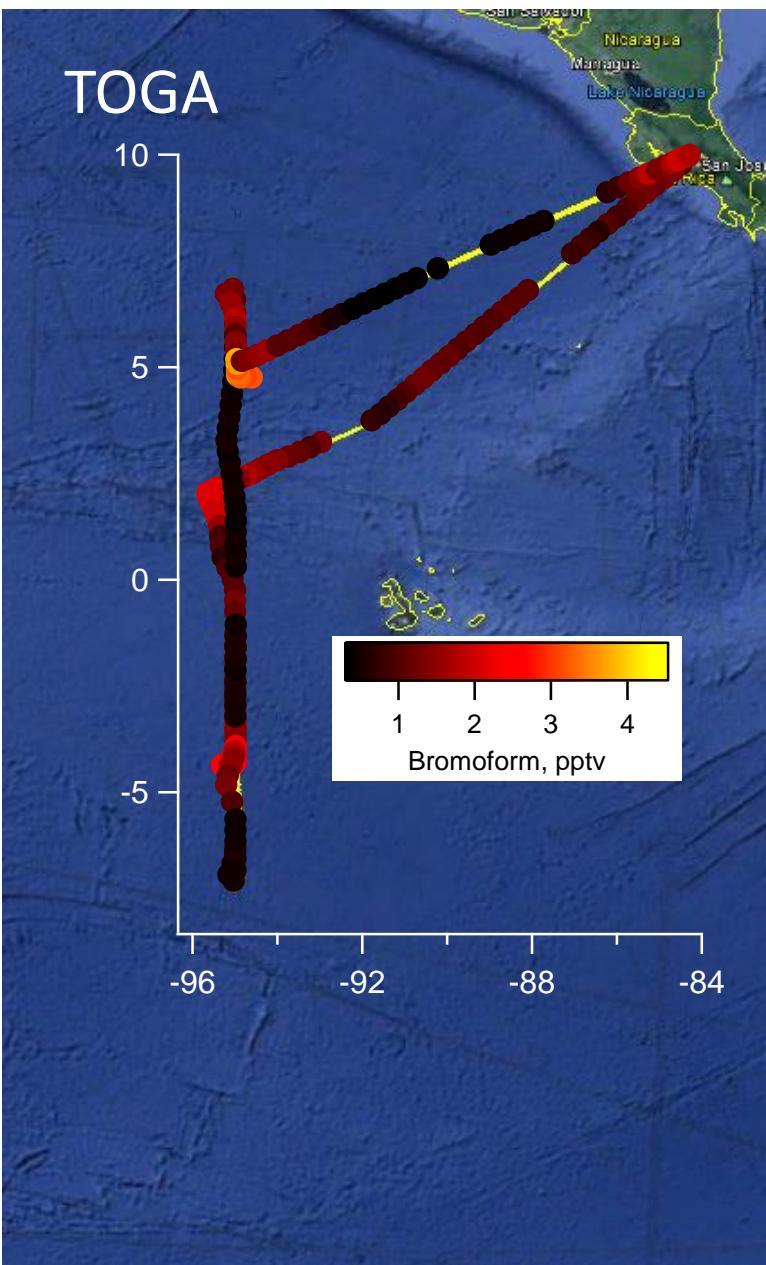
# RF15 – biologically active region



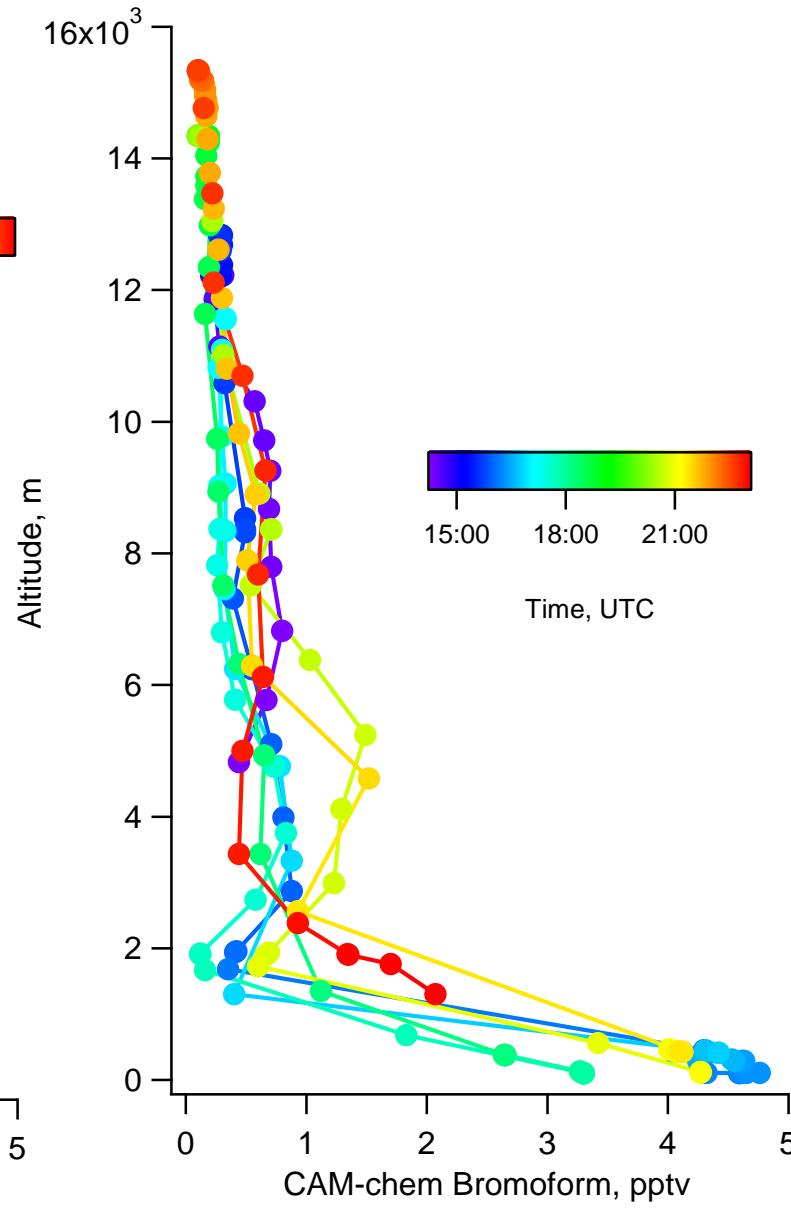
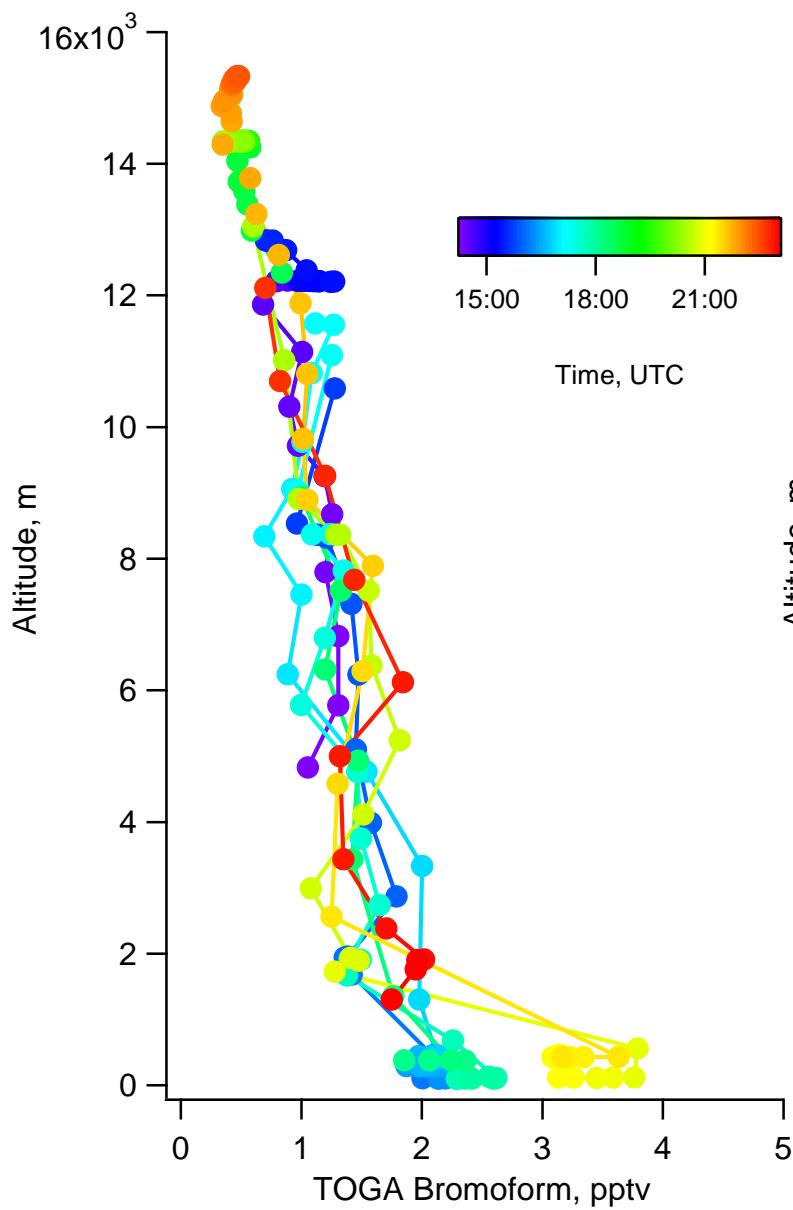
## RF15 – biologically active region



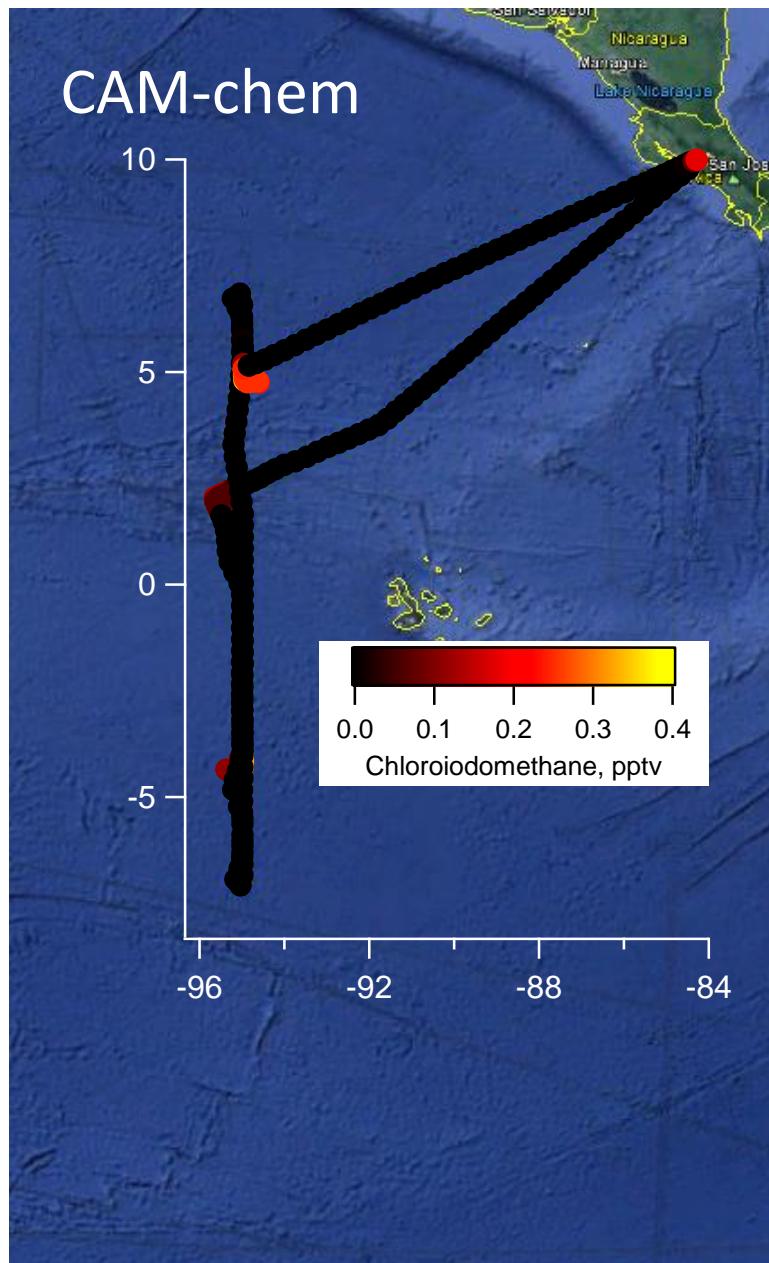
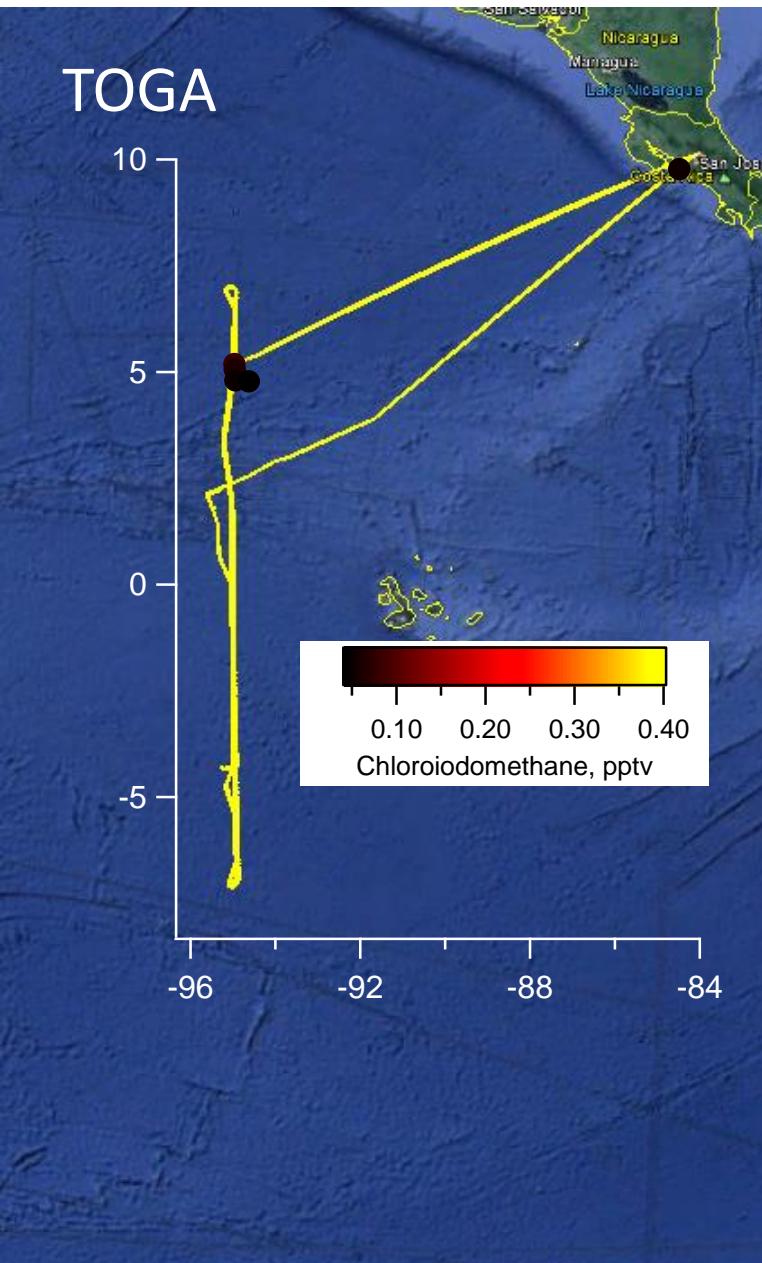
## RF15 – biologically active region



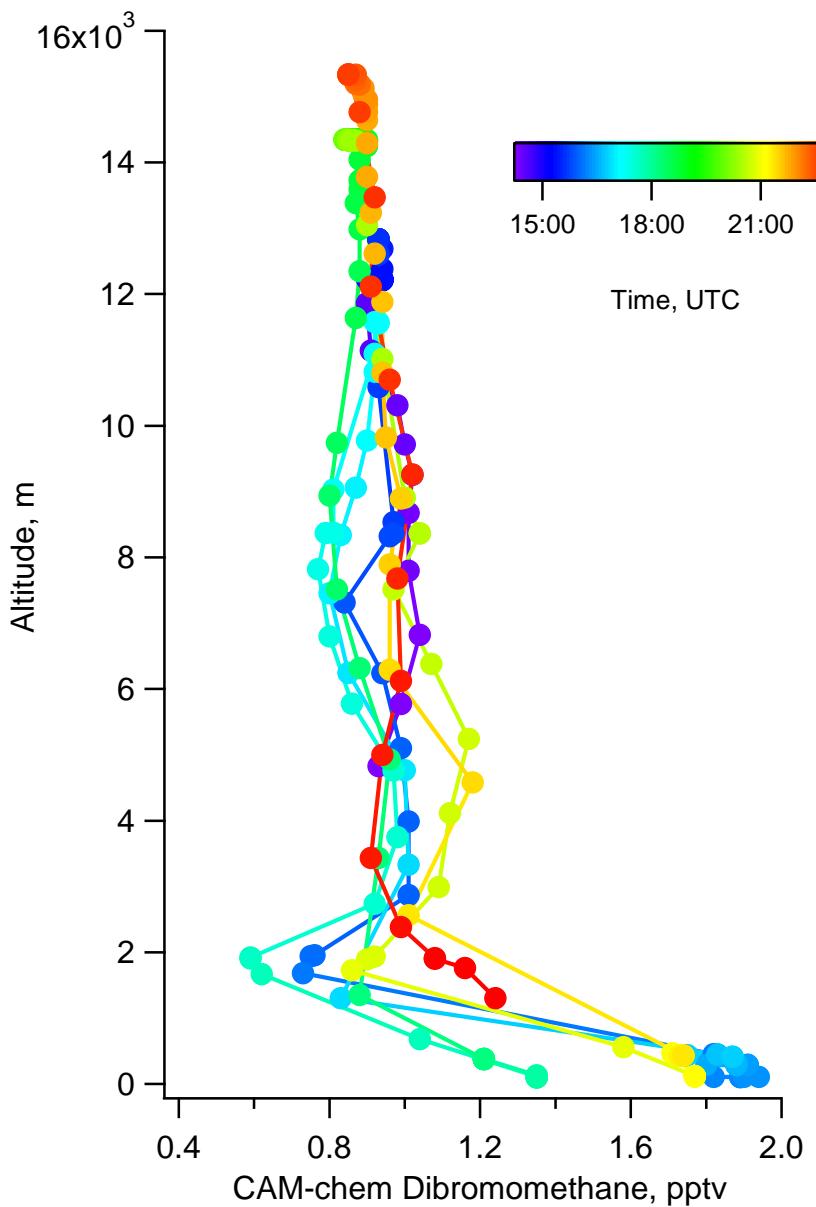
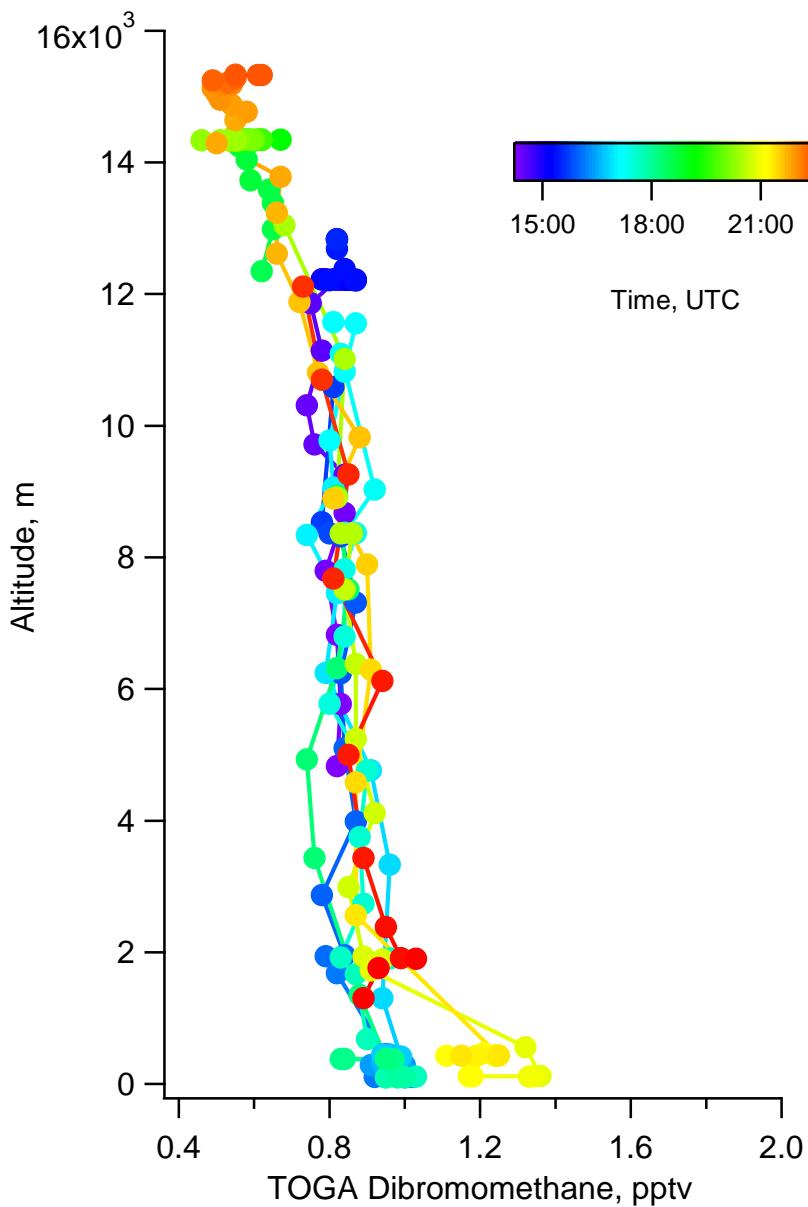
## RF15 – biologically active region



## RF15 – biologically active region



## RF15 – biologically active region



# Anthropogenic/Biomass Burning/Ocean Tracers

n-butane

benzene

toluene

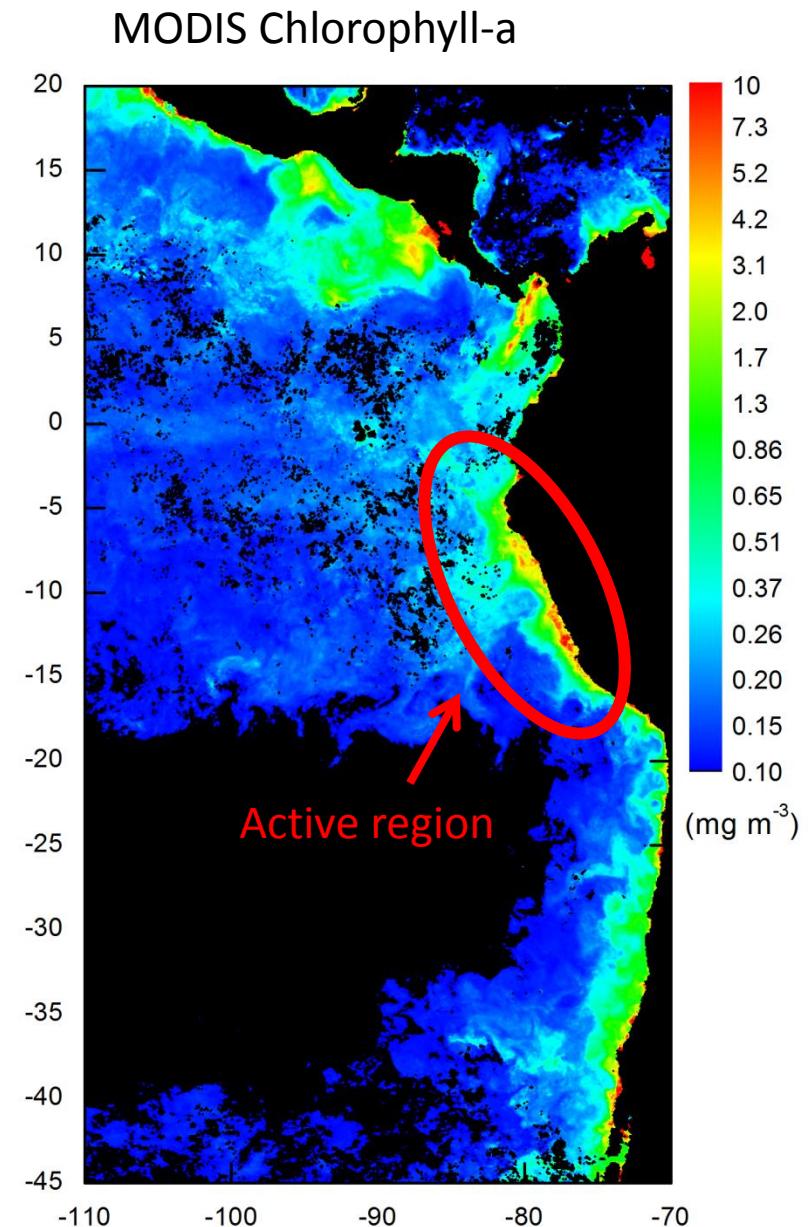
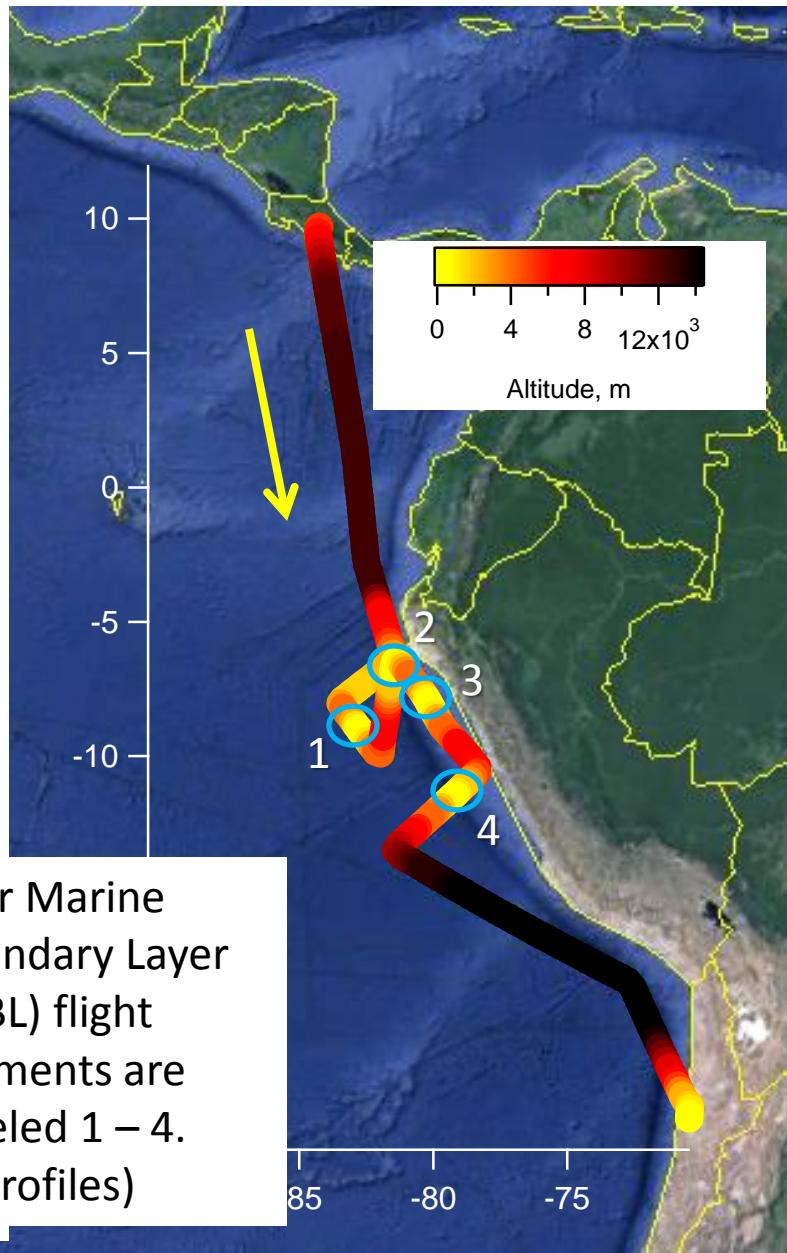
acetylene (CAM-chem)

acetone\*, acetonitrile

Formaldehyde\*, Acetaldehyde\*

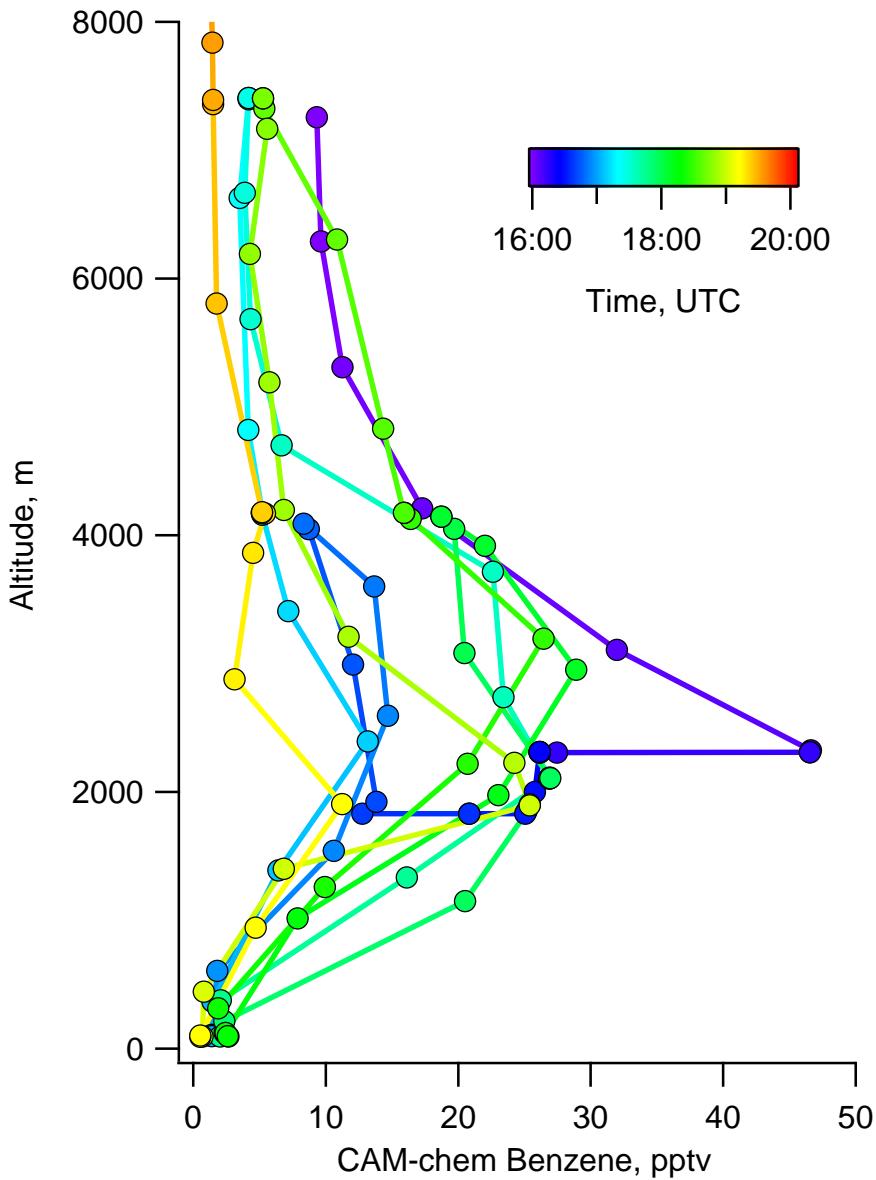
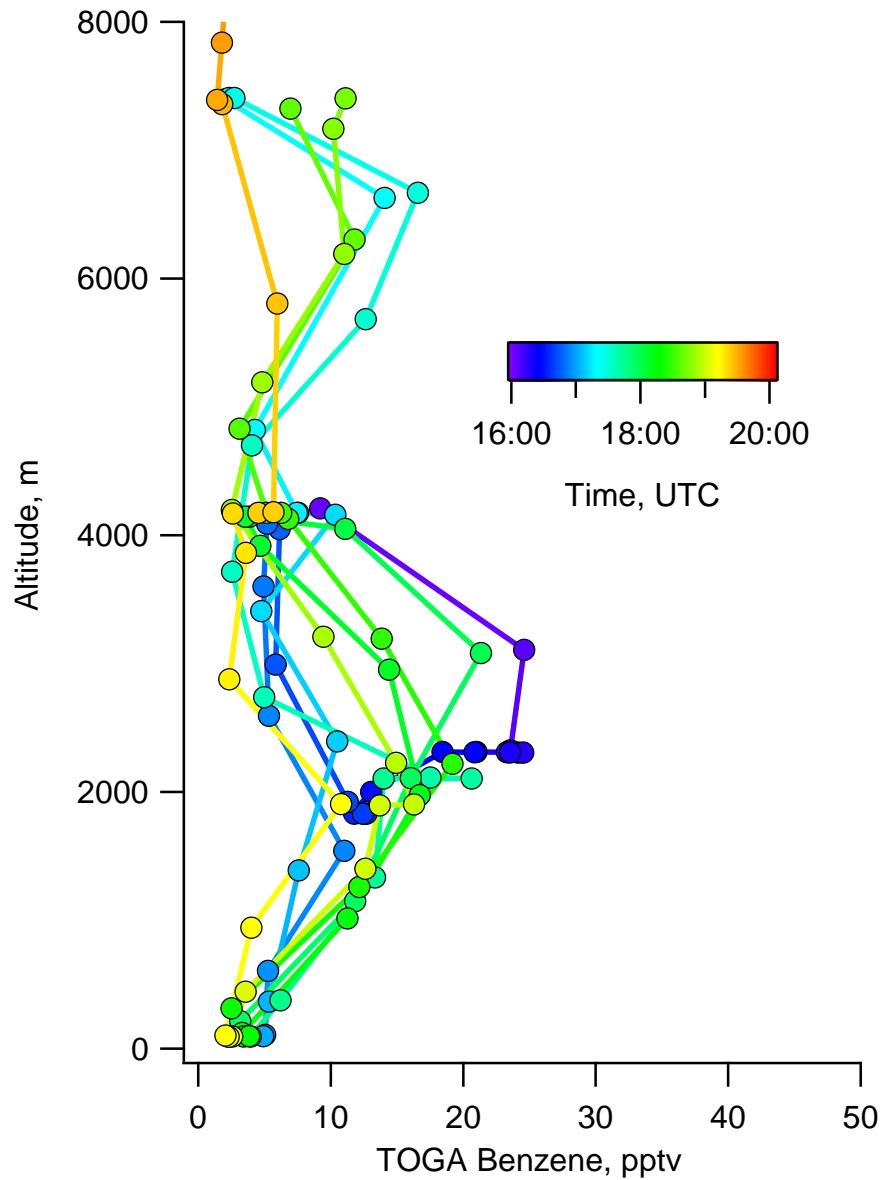
\*also have other natural sources

# RF01 – coastal emissions, continental outflow



# RF01: 8 profiles off the west coast of Peru

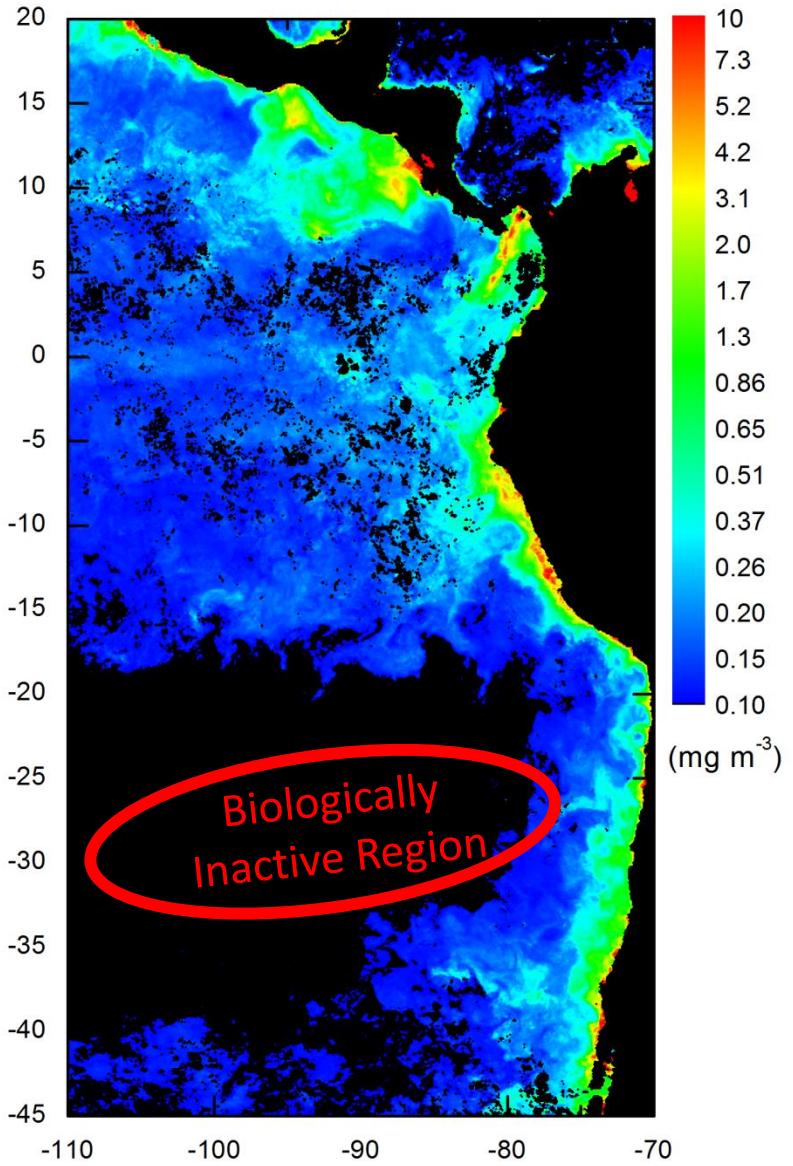
CAM-chem captured the pollution layer aloft at  $\sim 2\text{-}3\text{ km}$



# RF05 – oligotrophic ocean

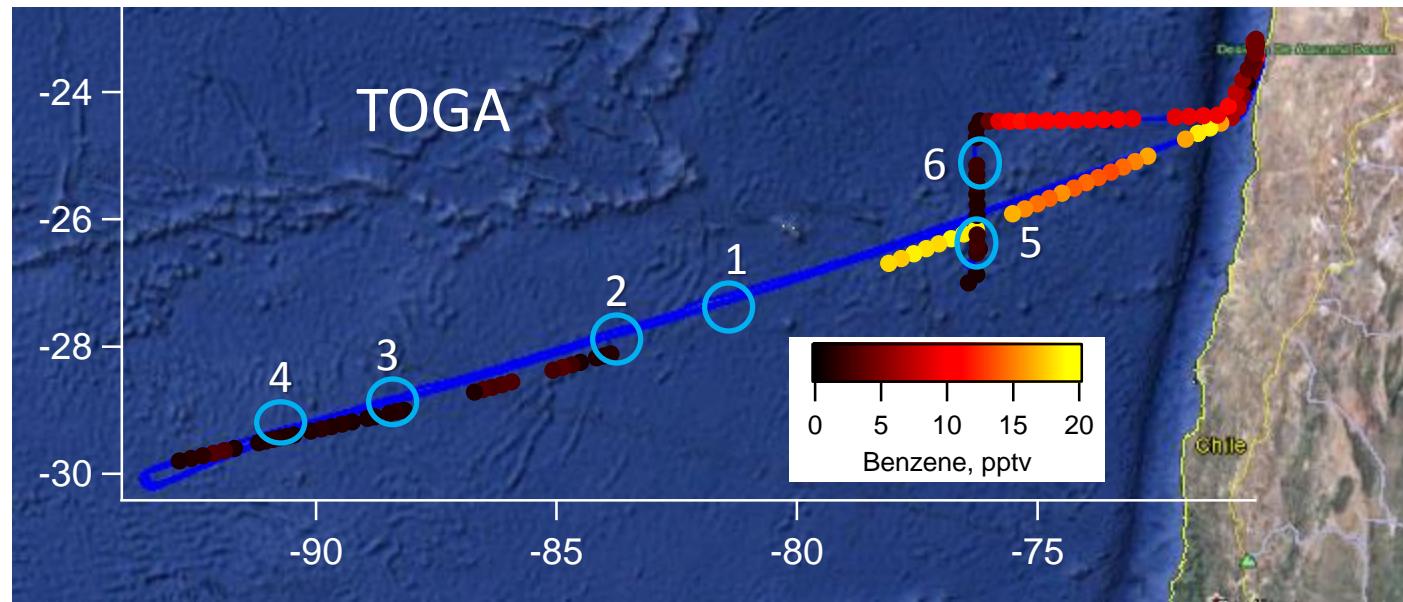


MODIS Chlorophyll-a

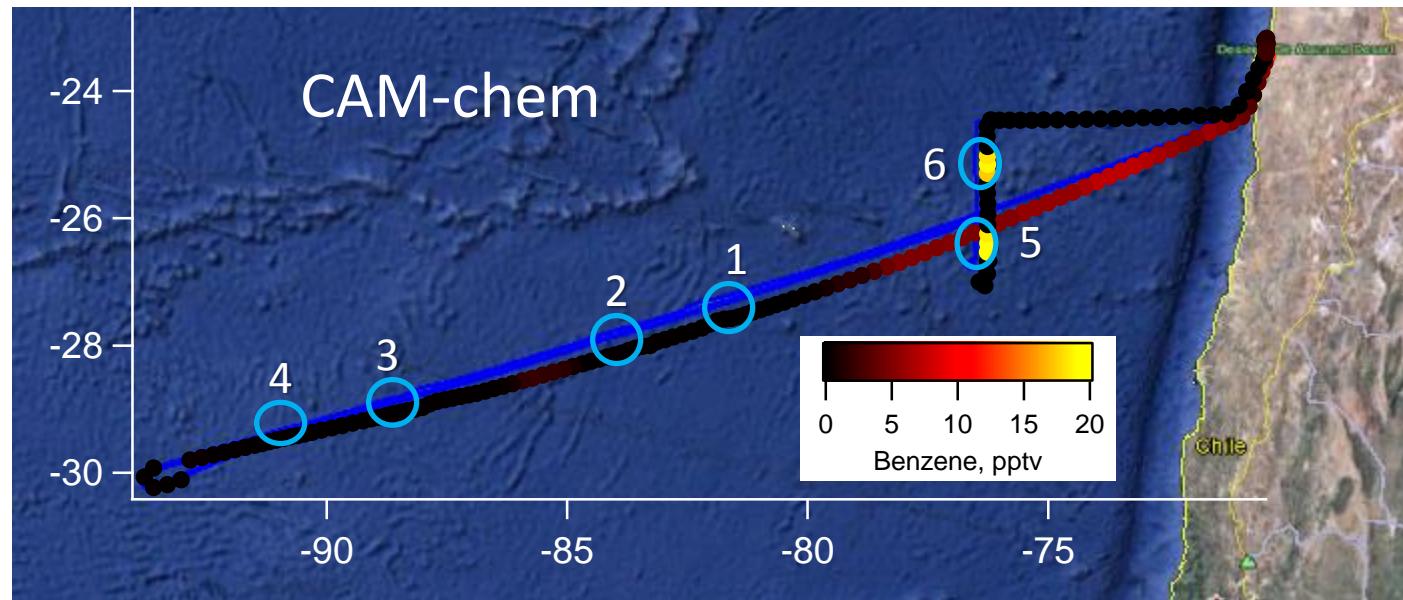


# RF05 – oligotrophic ocean

TOGA benzene

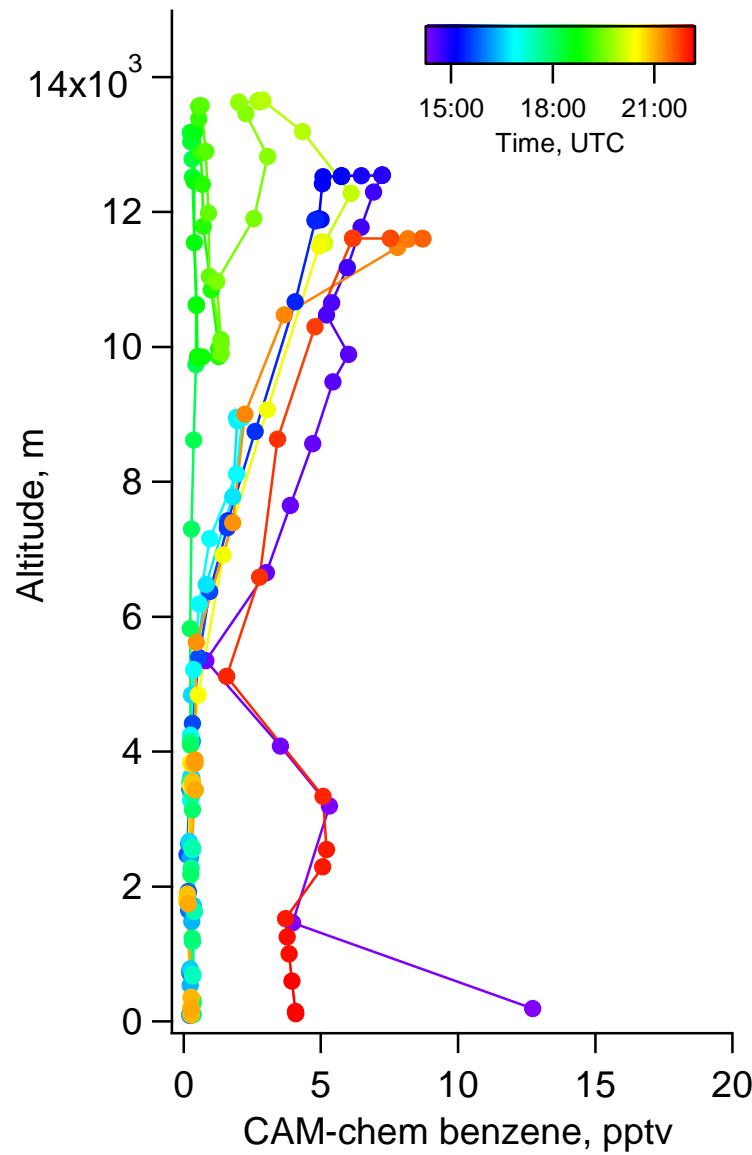
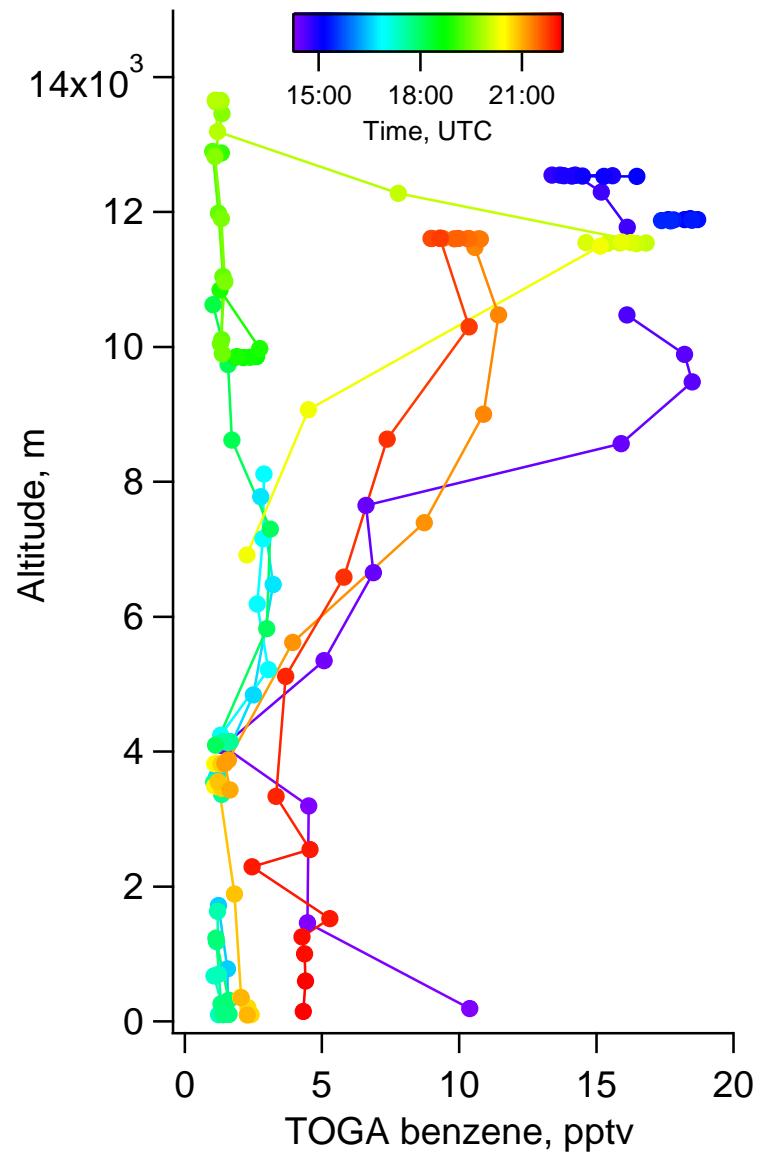


CAM-chem

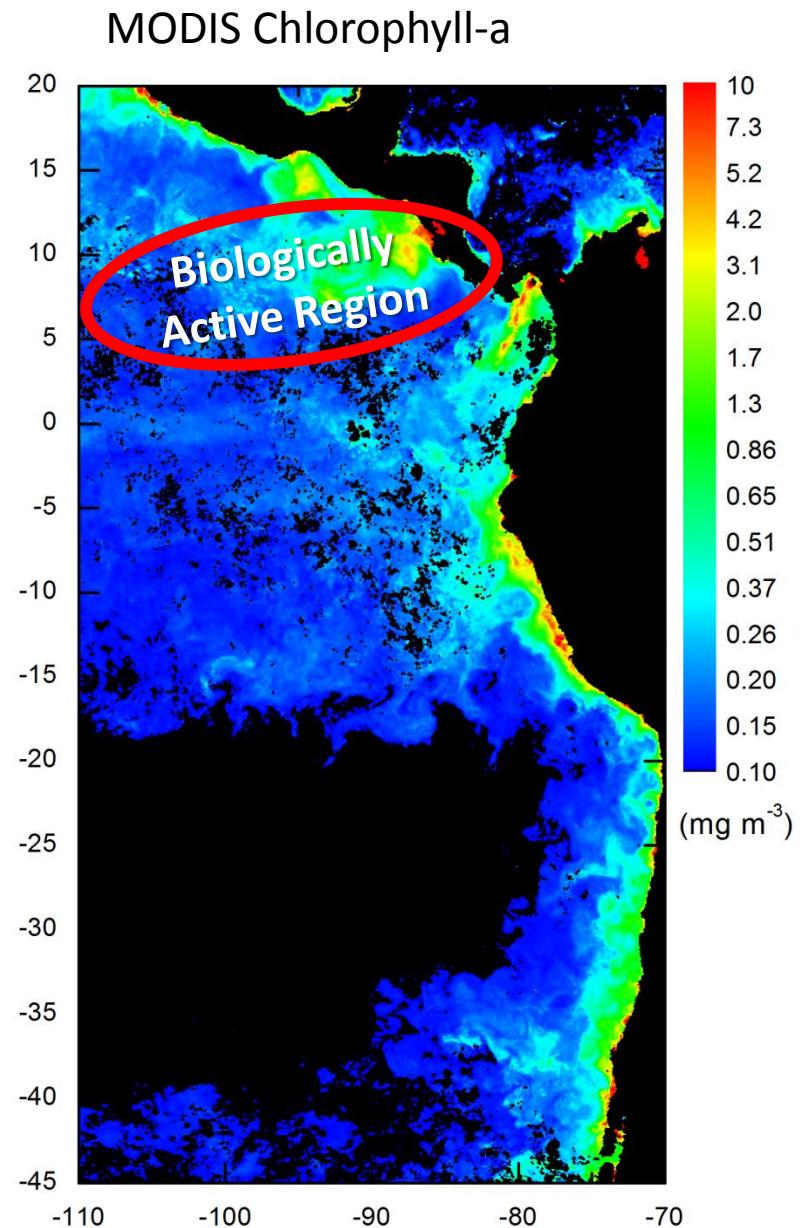


# RF05 – oligotrophic ocean

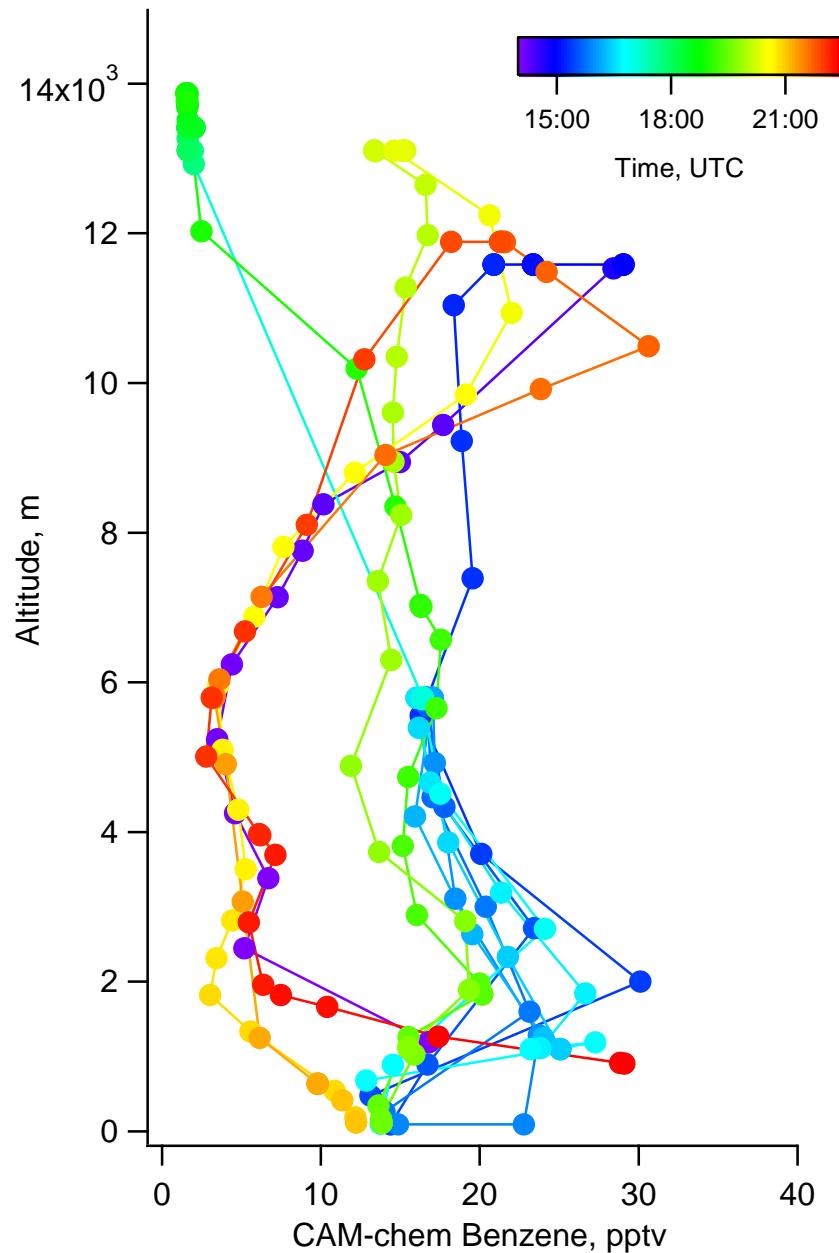
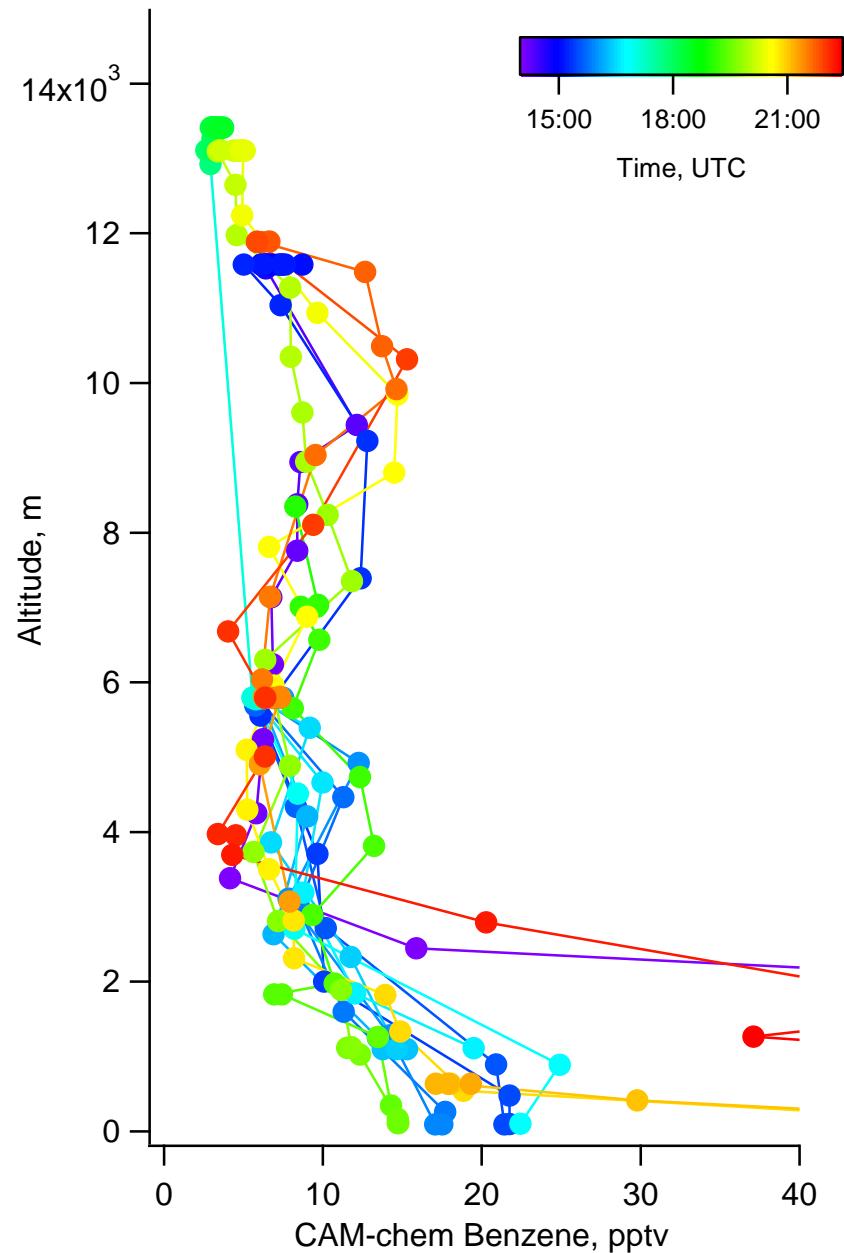
benzene



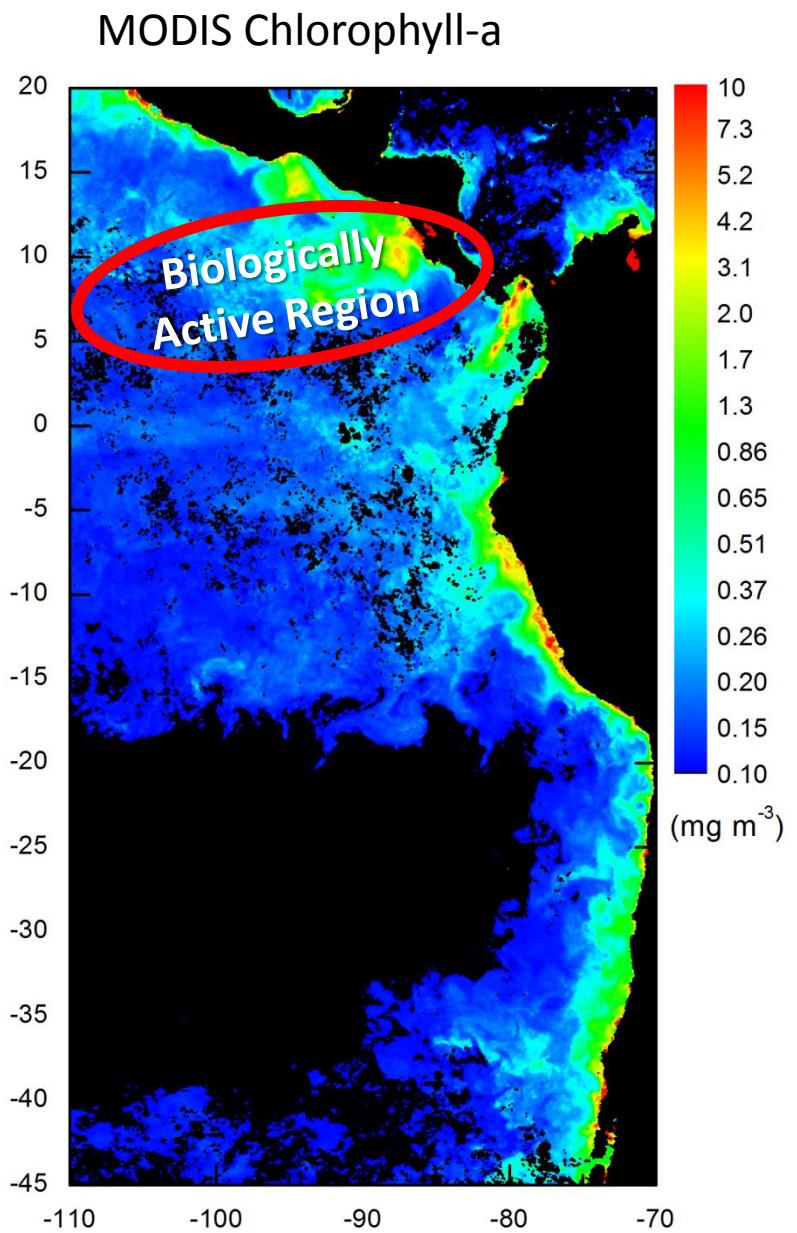
# RF12 – biologically active region/convection MBL



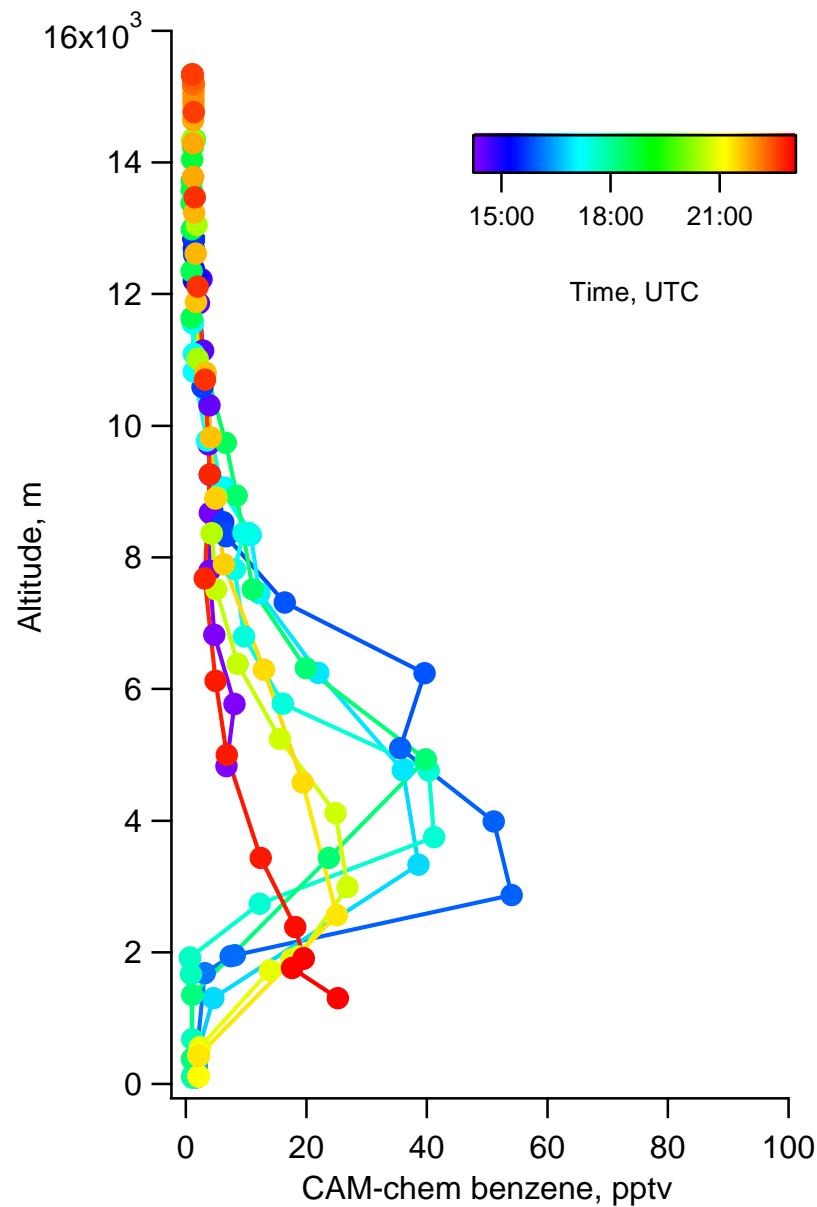
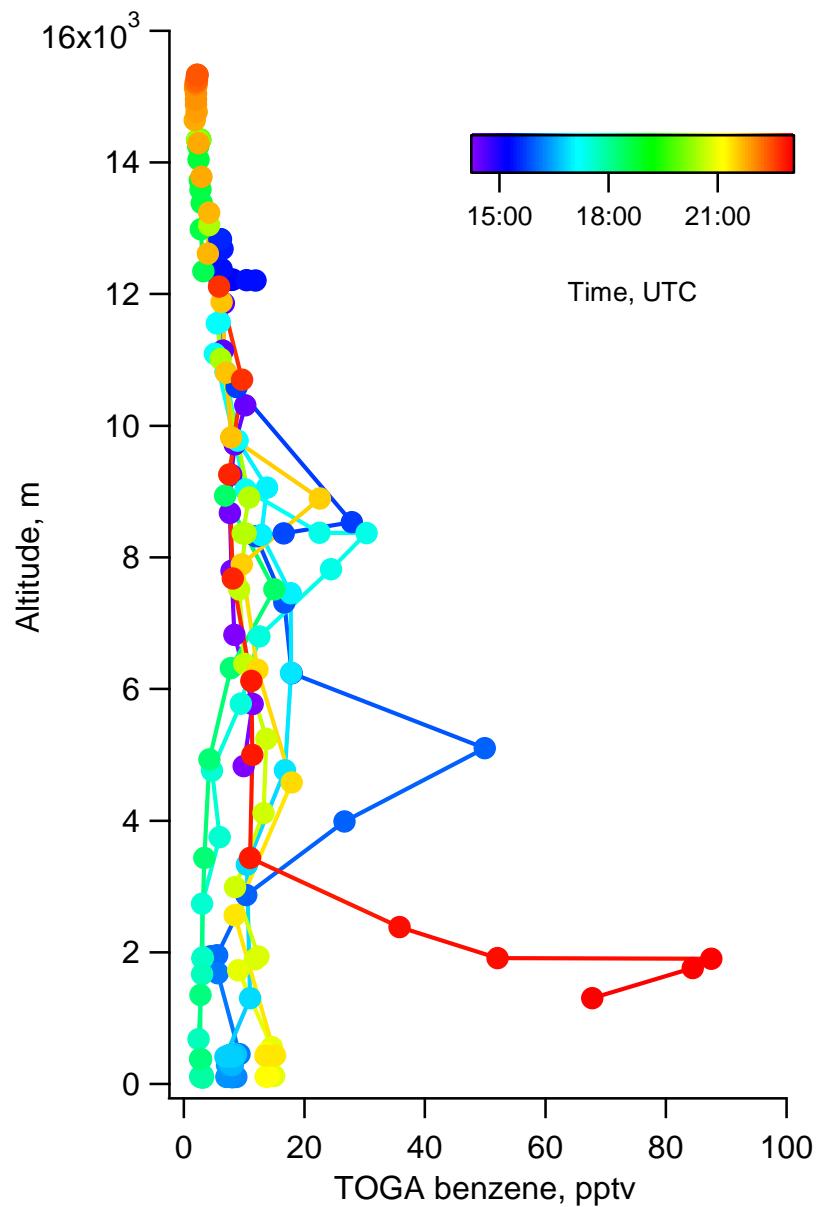
# RF12 – biologically active region



# RF15 – biologically active region



## RF15 – biologically active region



# Oxygenated VOCs

acetone ( $\text{CH}_3\text{COCH}_3$ )

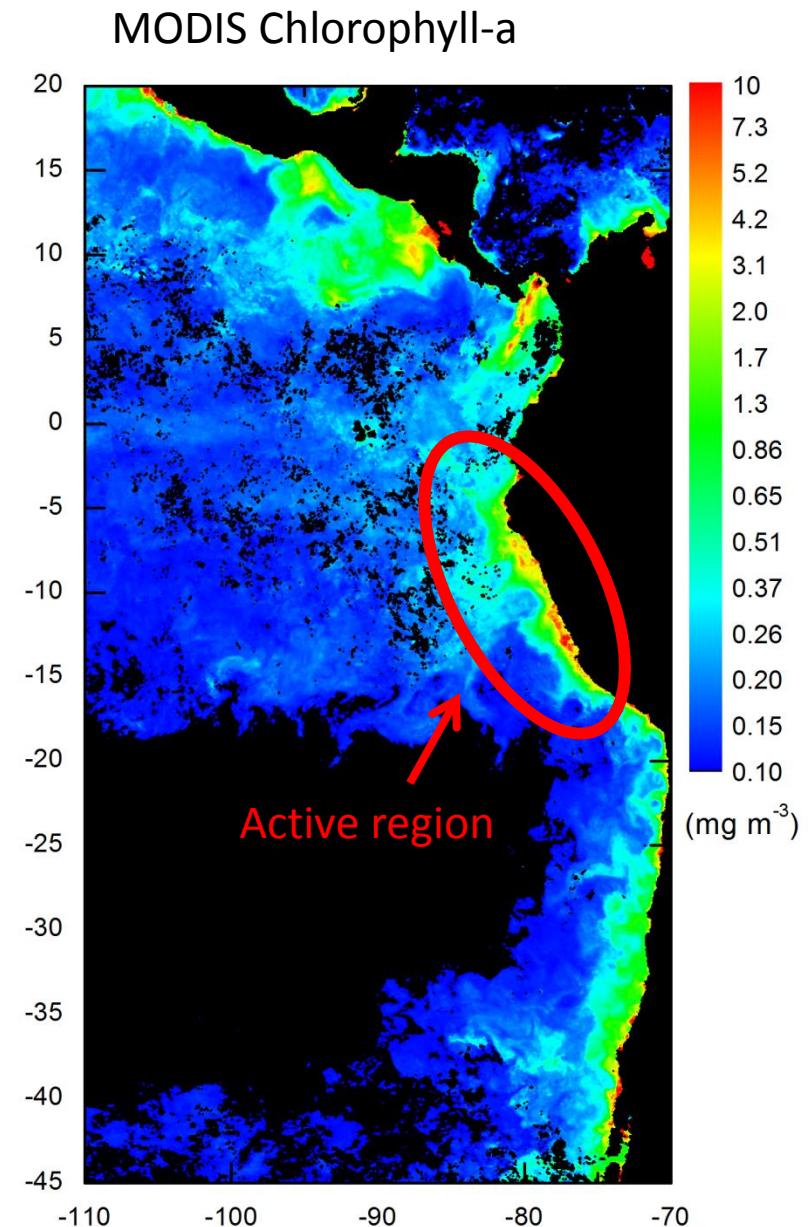
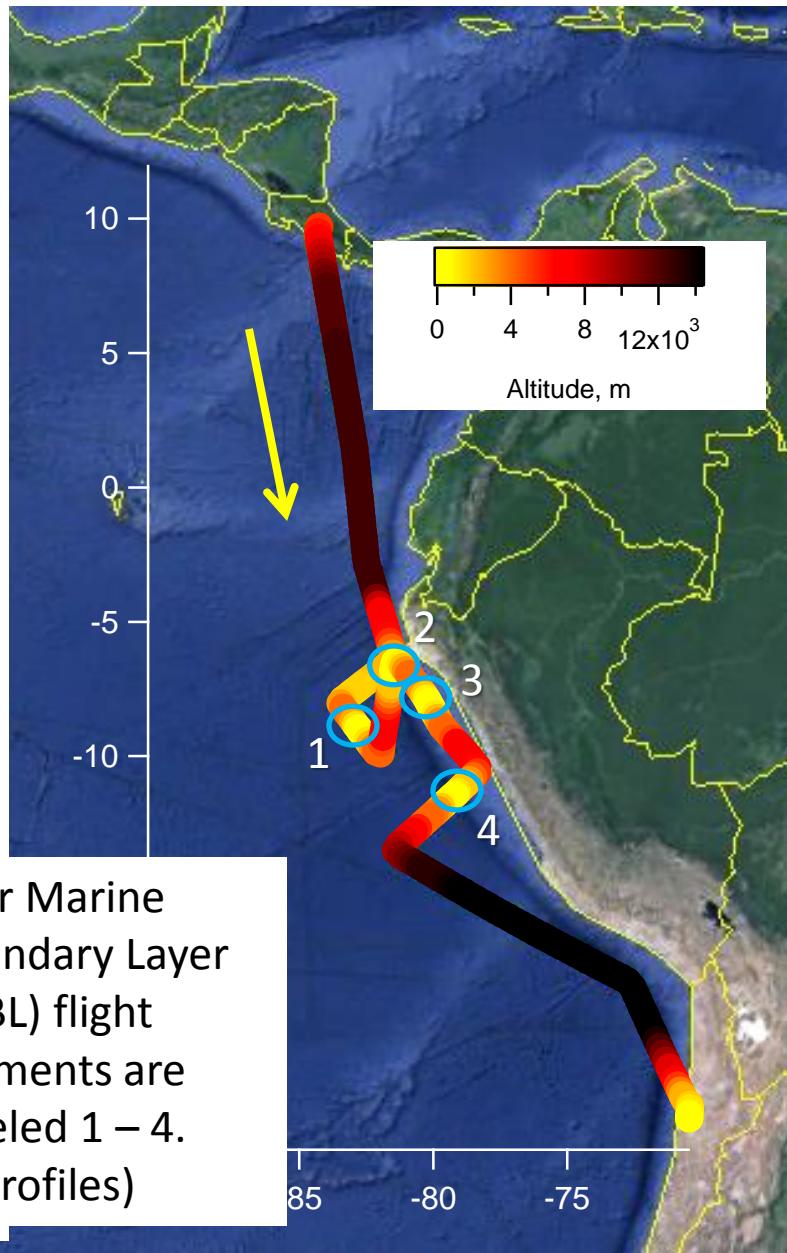
formaldehyde ( $\text{HCHO}$ )

acetaldehyde ( $\text{CH}_3\text{CHO}$ )

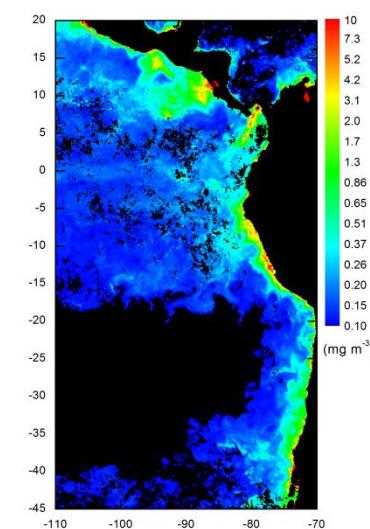
methanol ( $\text{CH}_3\text{OH}$ )

Butanone/Methyl Ethyl Ketone (MEK)

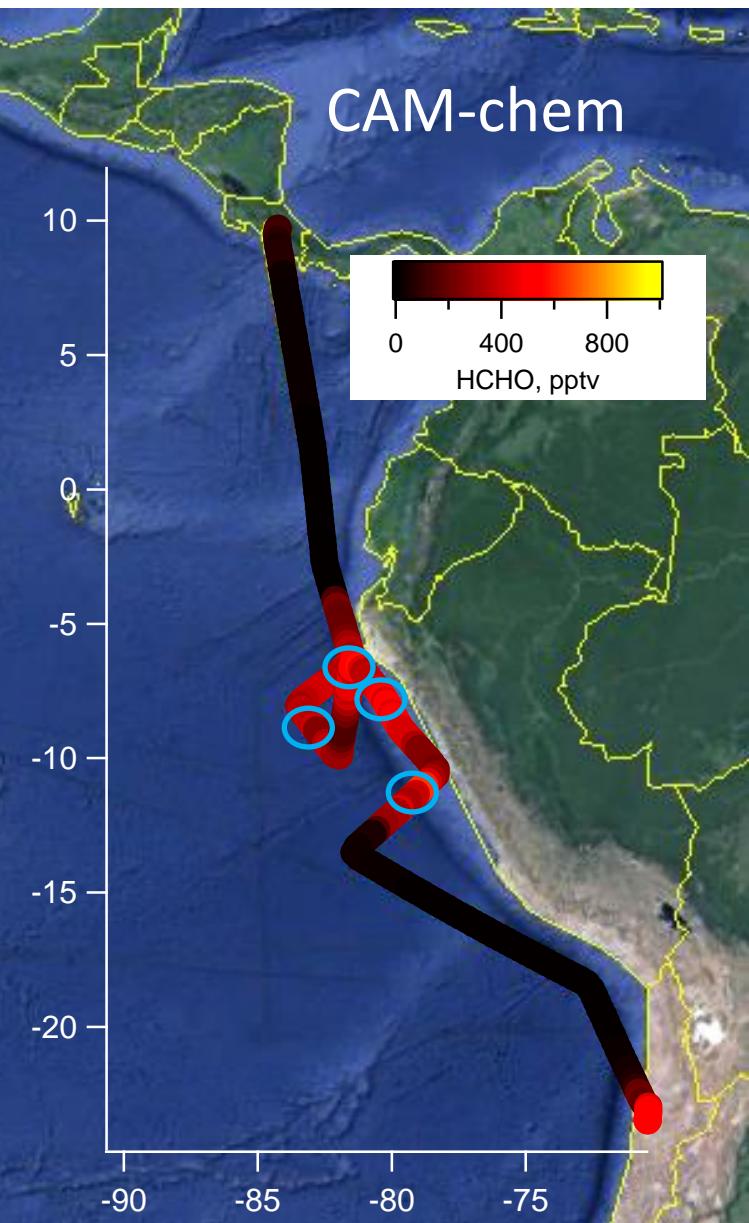
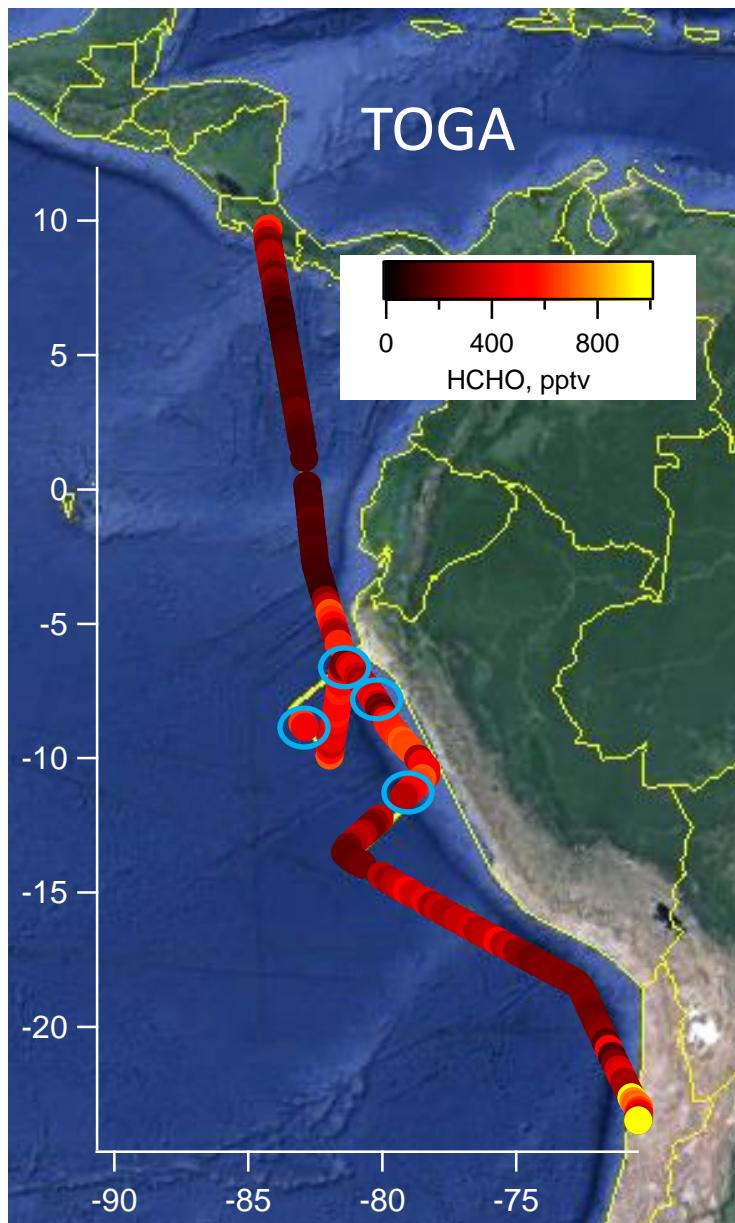
# RF01 – coastal emissions, continental outflow



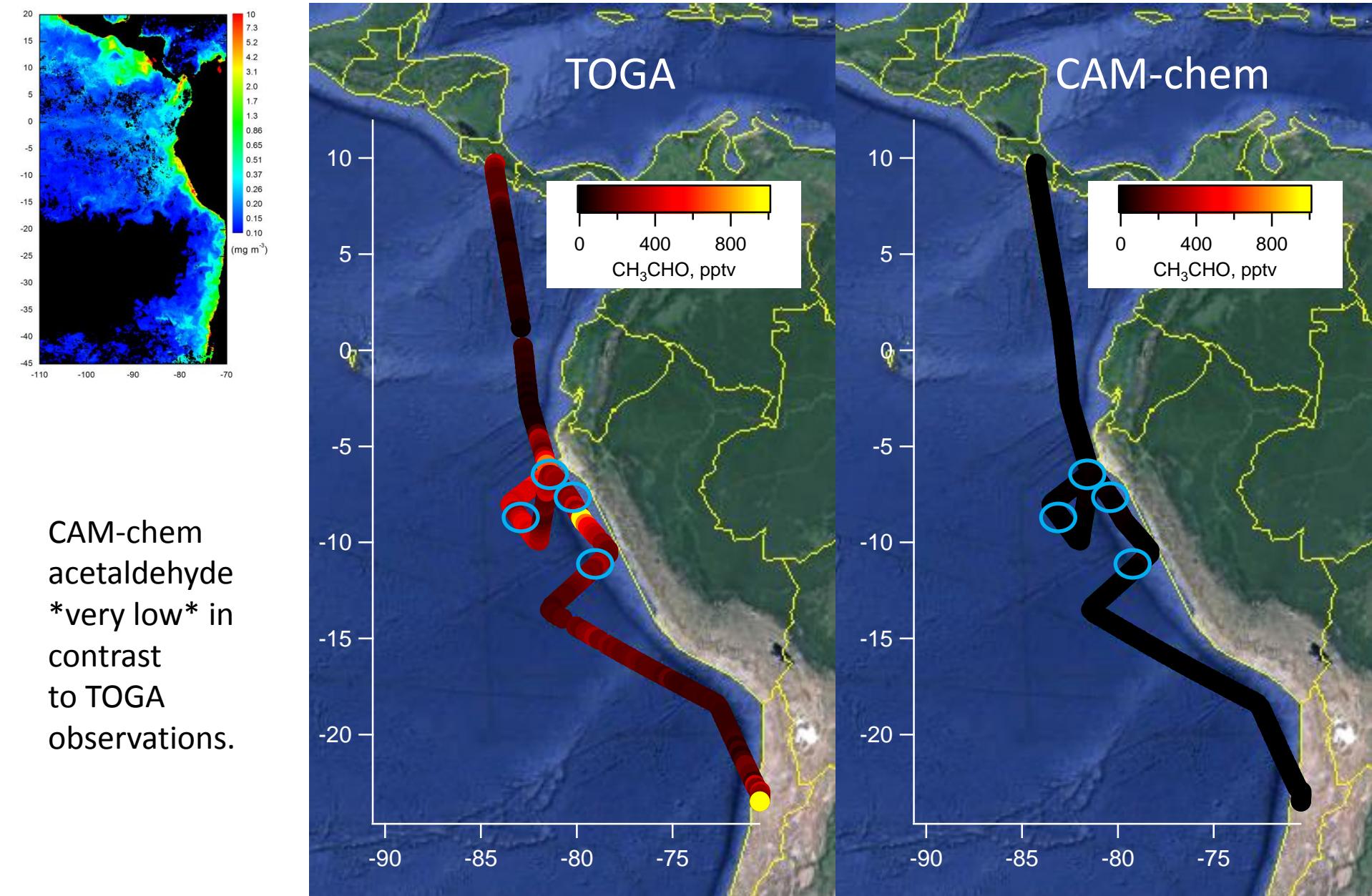
# From RF01



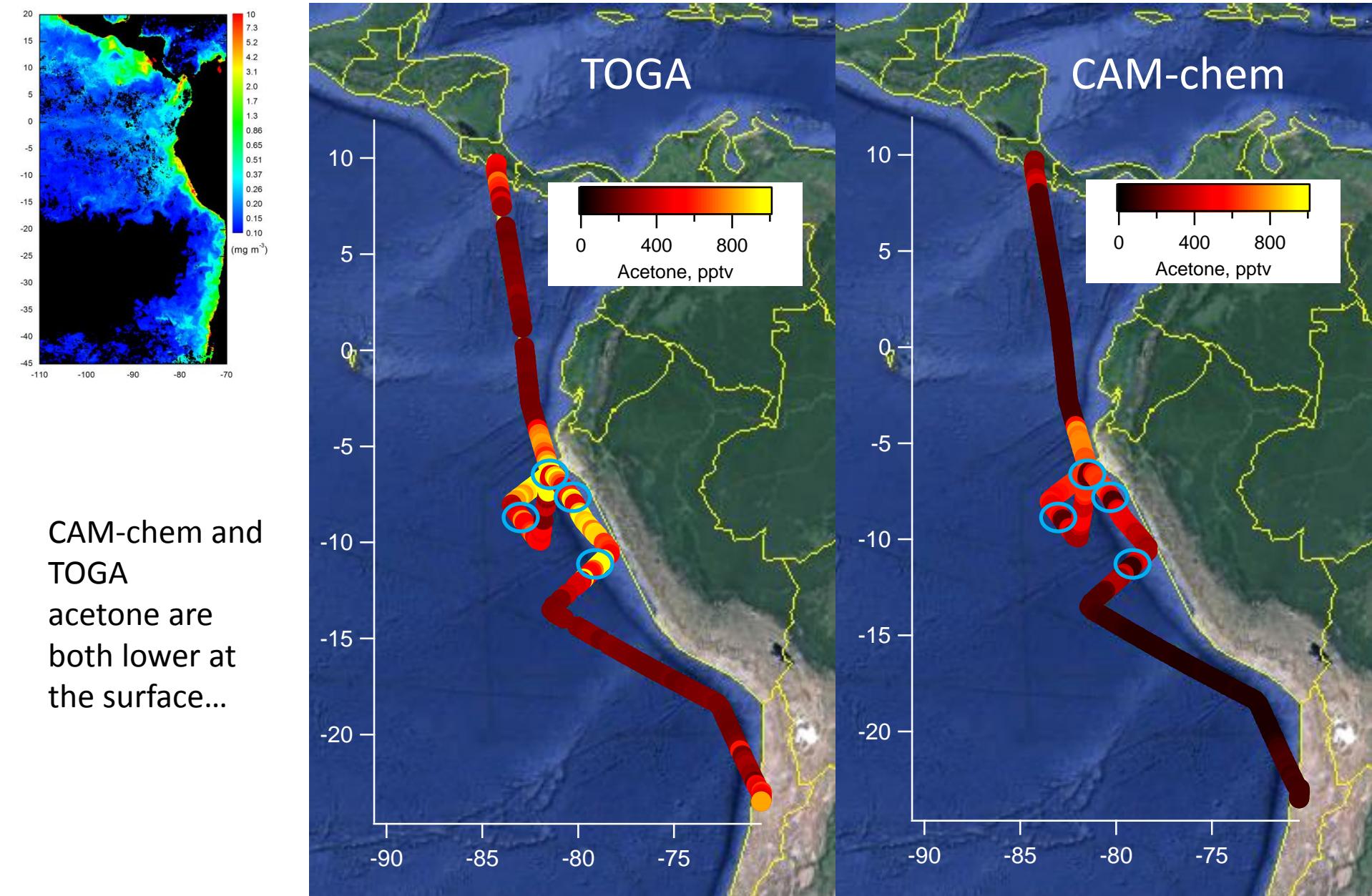
CAM-chem  
HCHO appears to  
capture  
The pollution off  
the coast, but is  
lower than TOGA  
in high altitude  
flight segments.



# From RF01

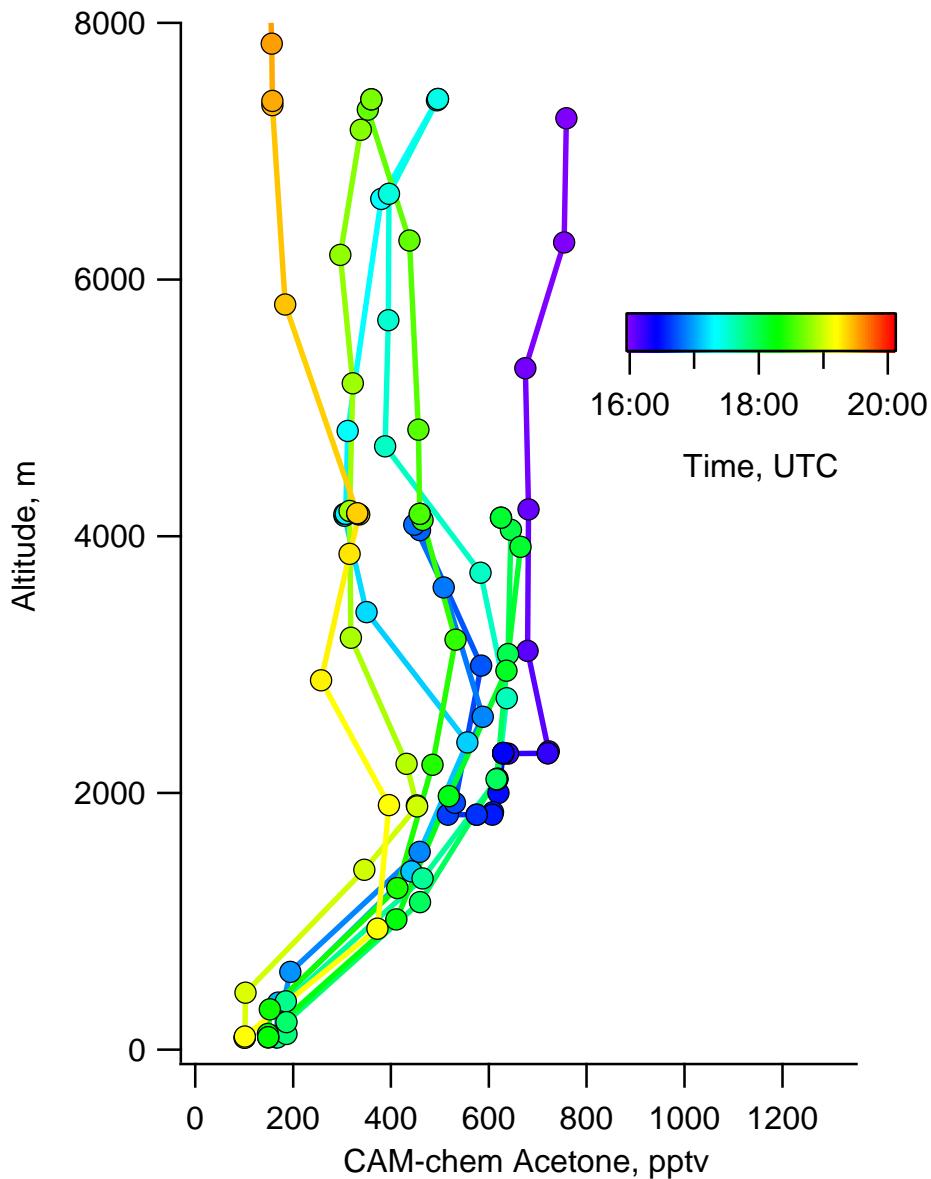
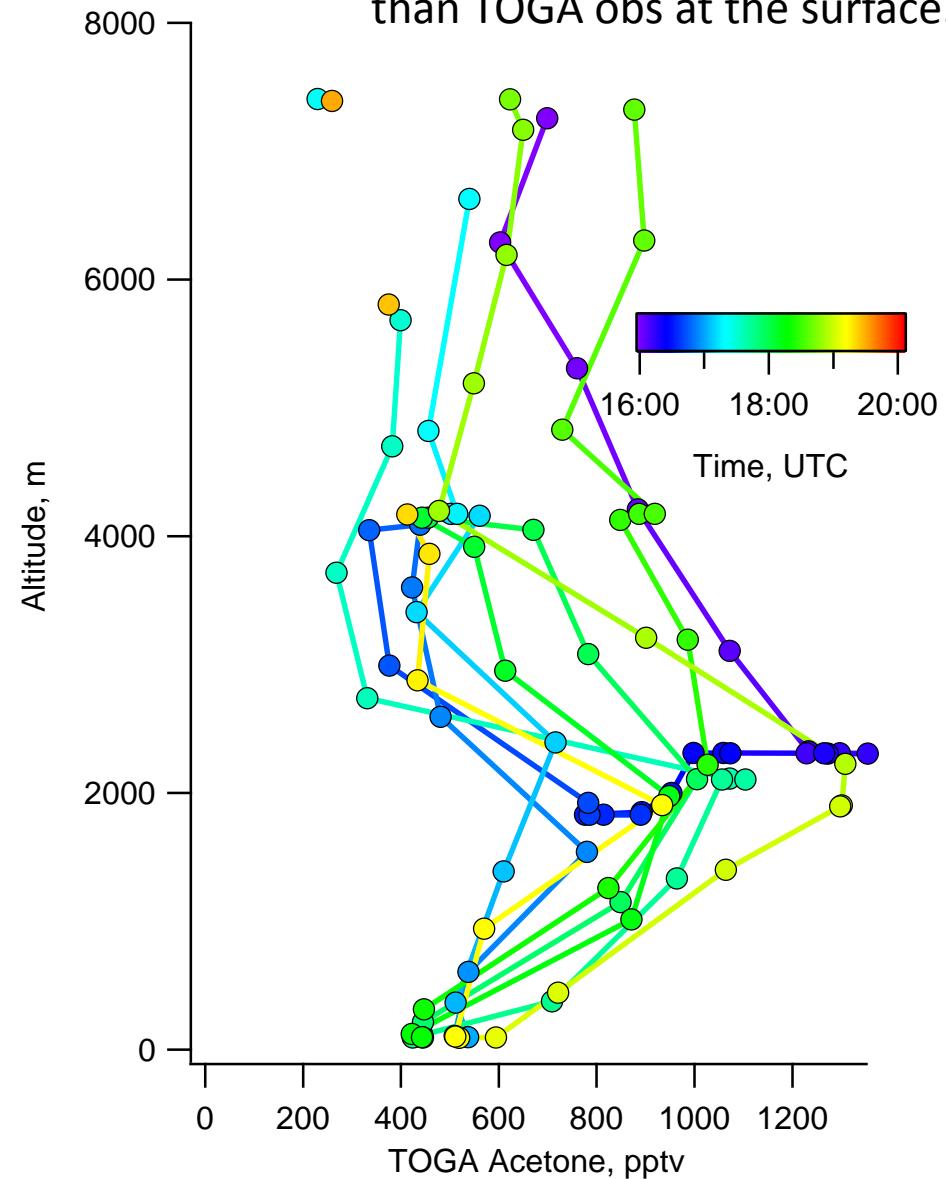


# From RF01



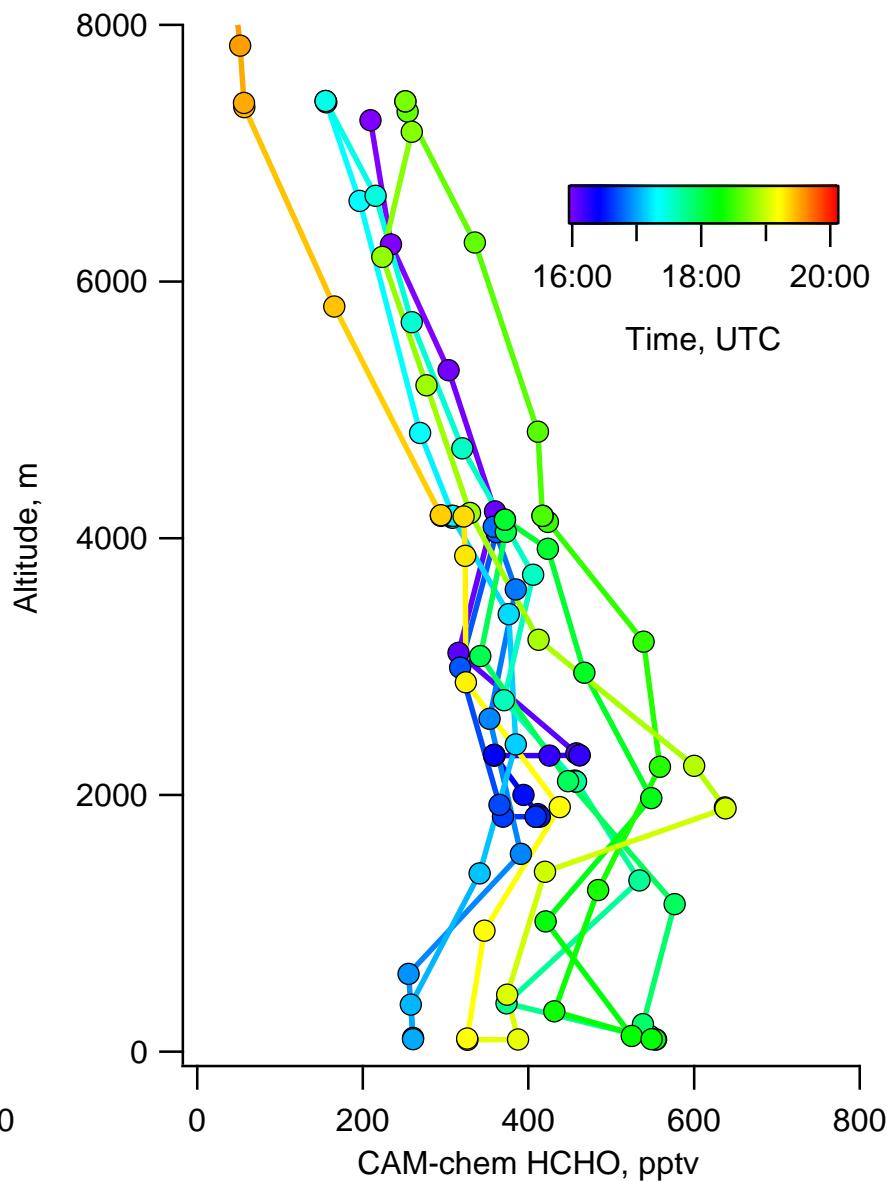
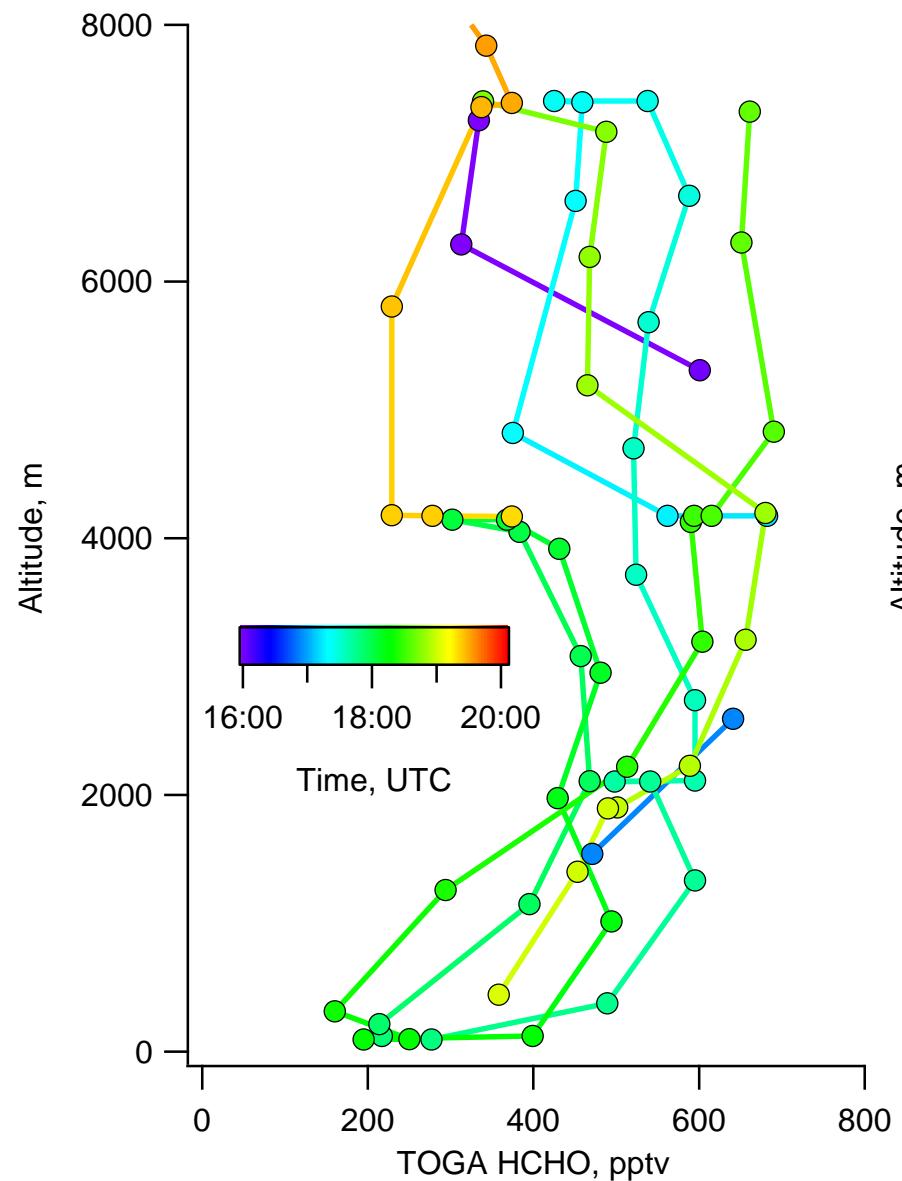
# RF01: 8 profiles off the west coast of Peru

CAM-chem captured the pollution layer aloft at  $\sim$  2-3 km, but is lower than TOGA obs at the surface.



# RF01: 8 profiles off the west coast of Peru

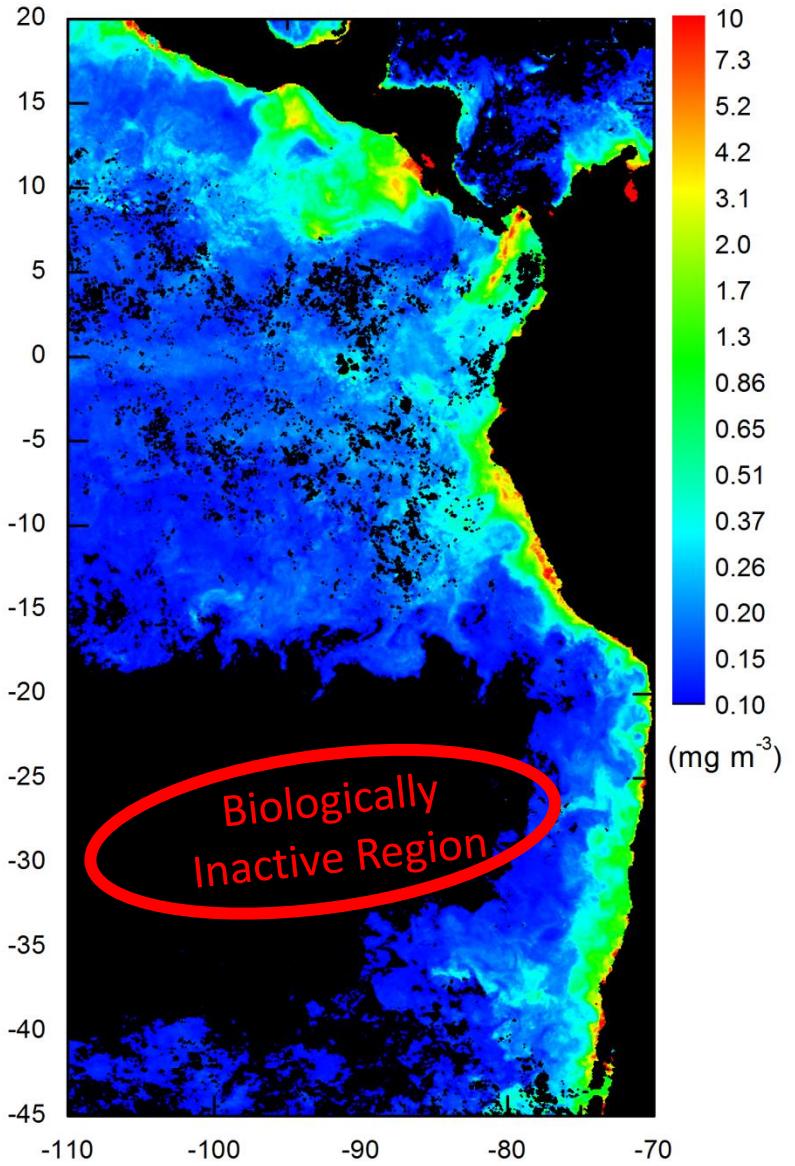
CAM-chem HCHO agrees nominally with TOGA.



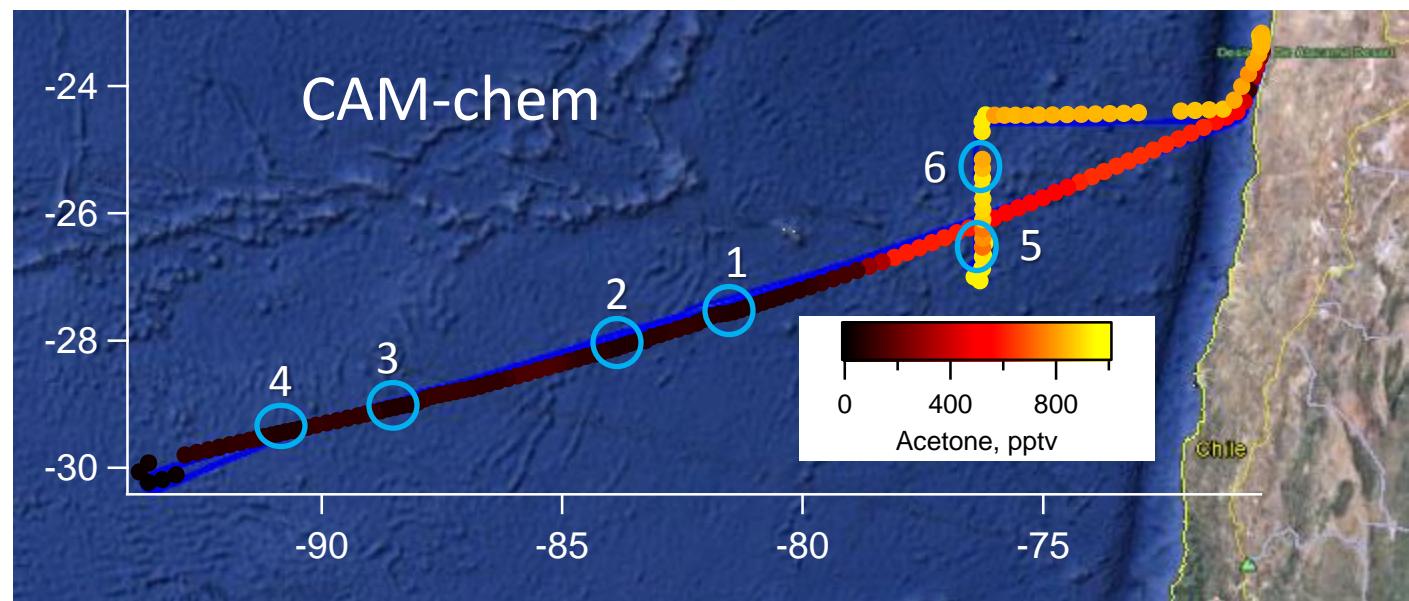
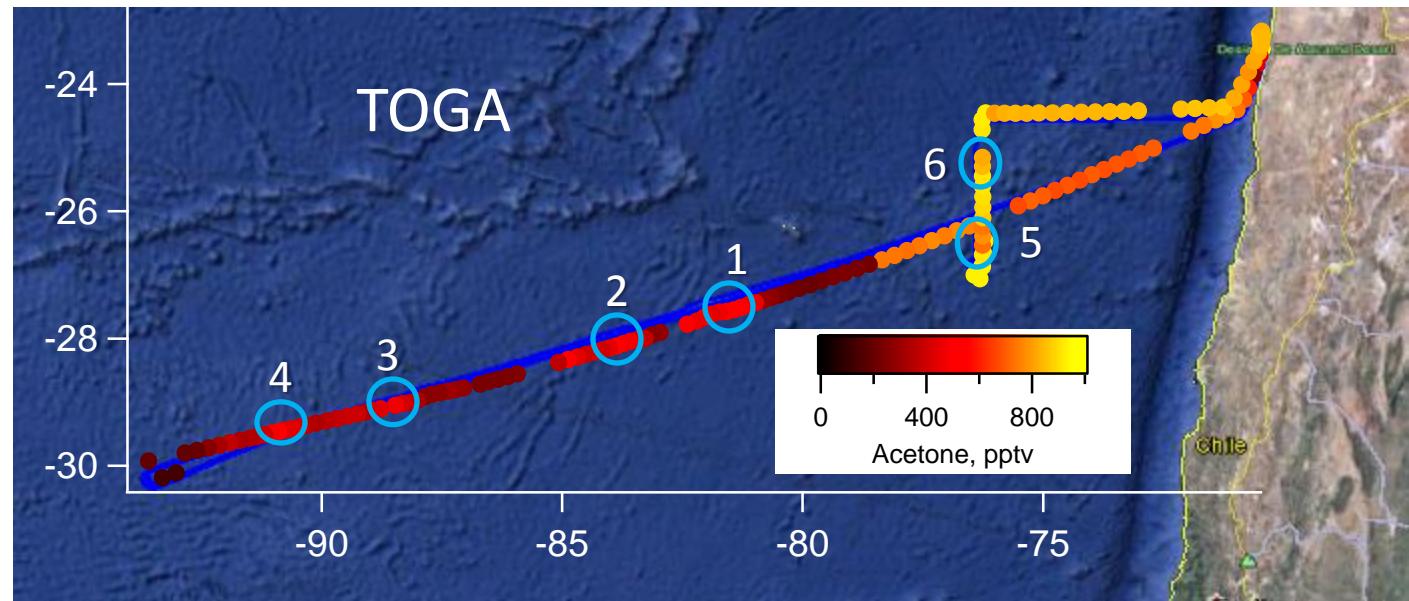
# RF05 – oligotrophic ocean



MODIS Chlorophyll-a

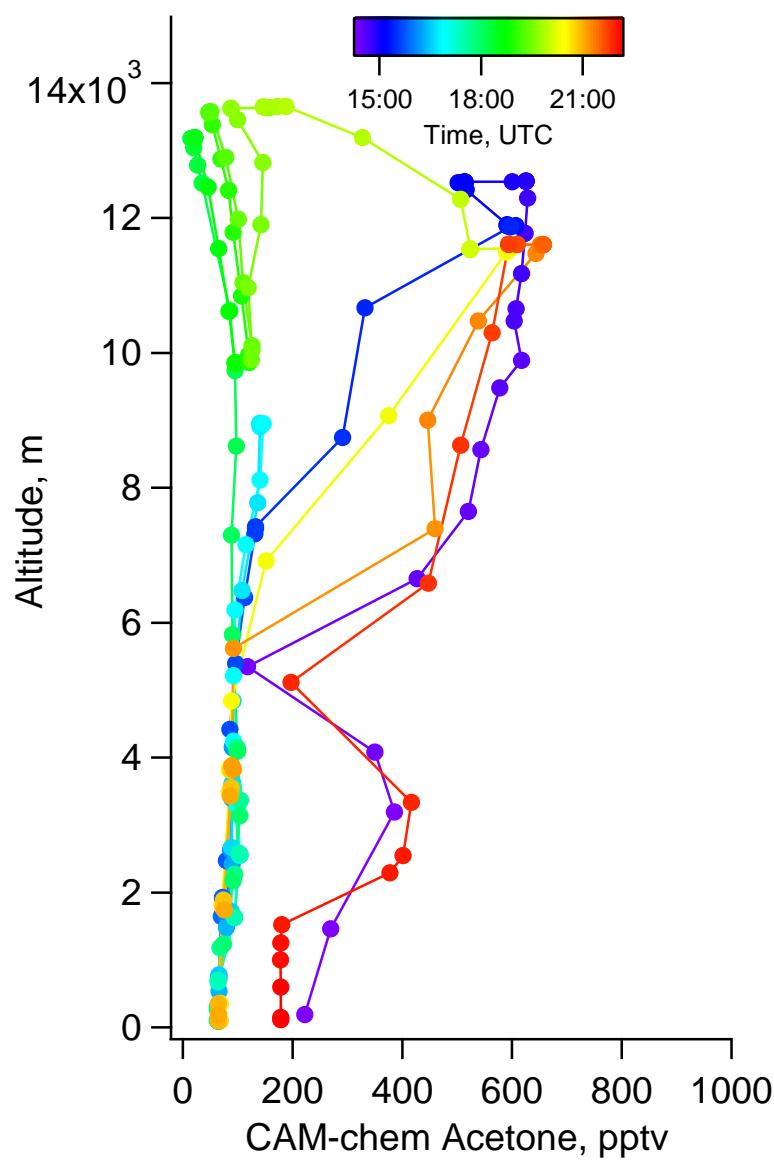
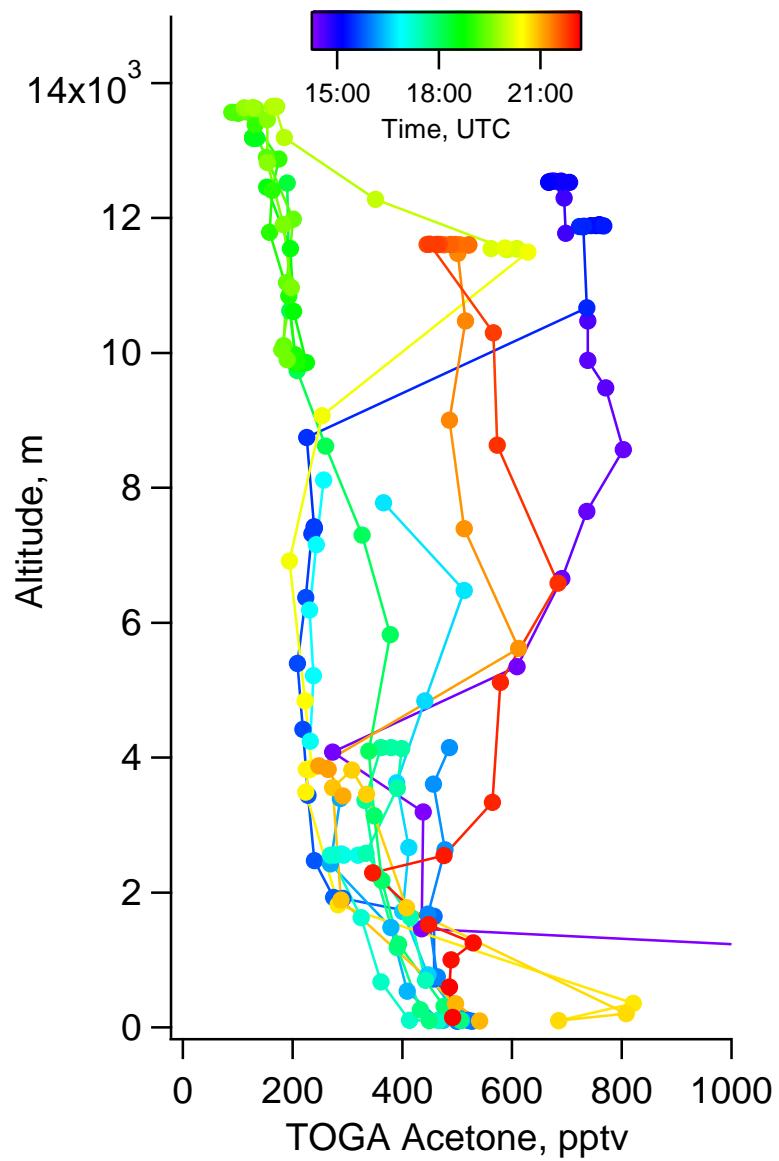


# RF05 – oligotrophic ocean

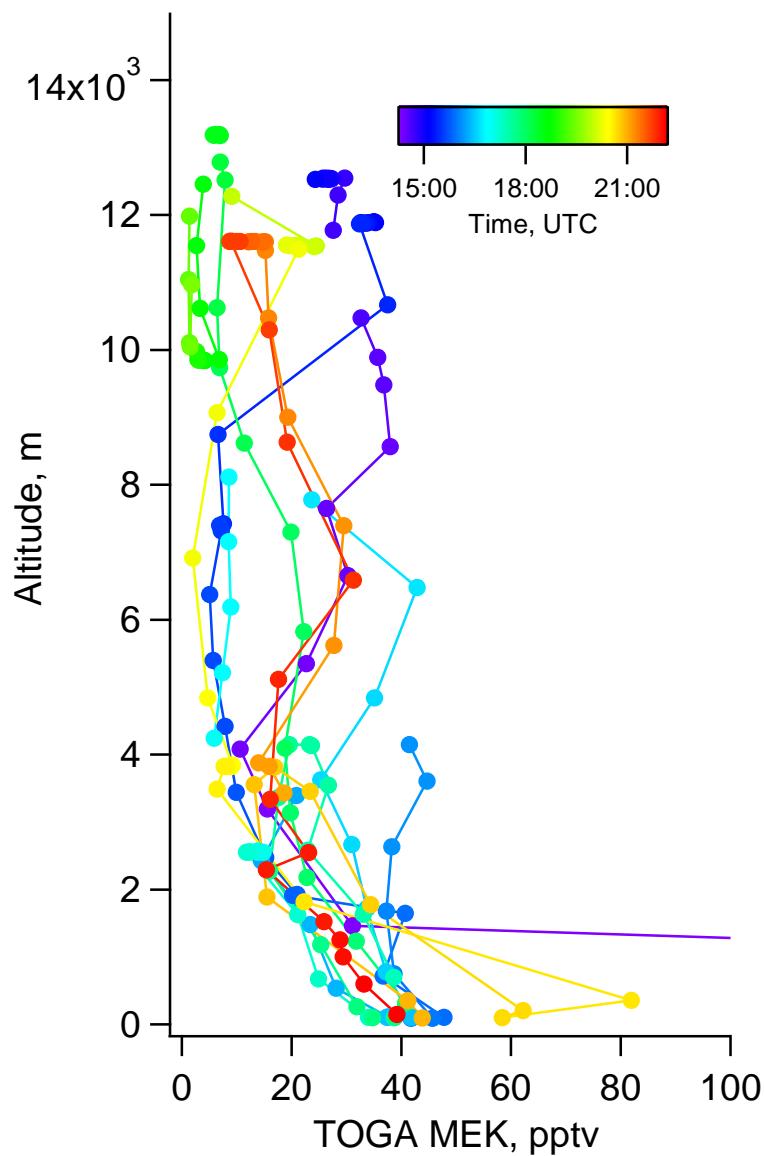


# RF05 – oligotrophic ocean

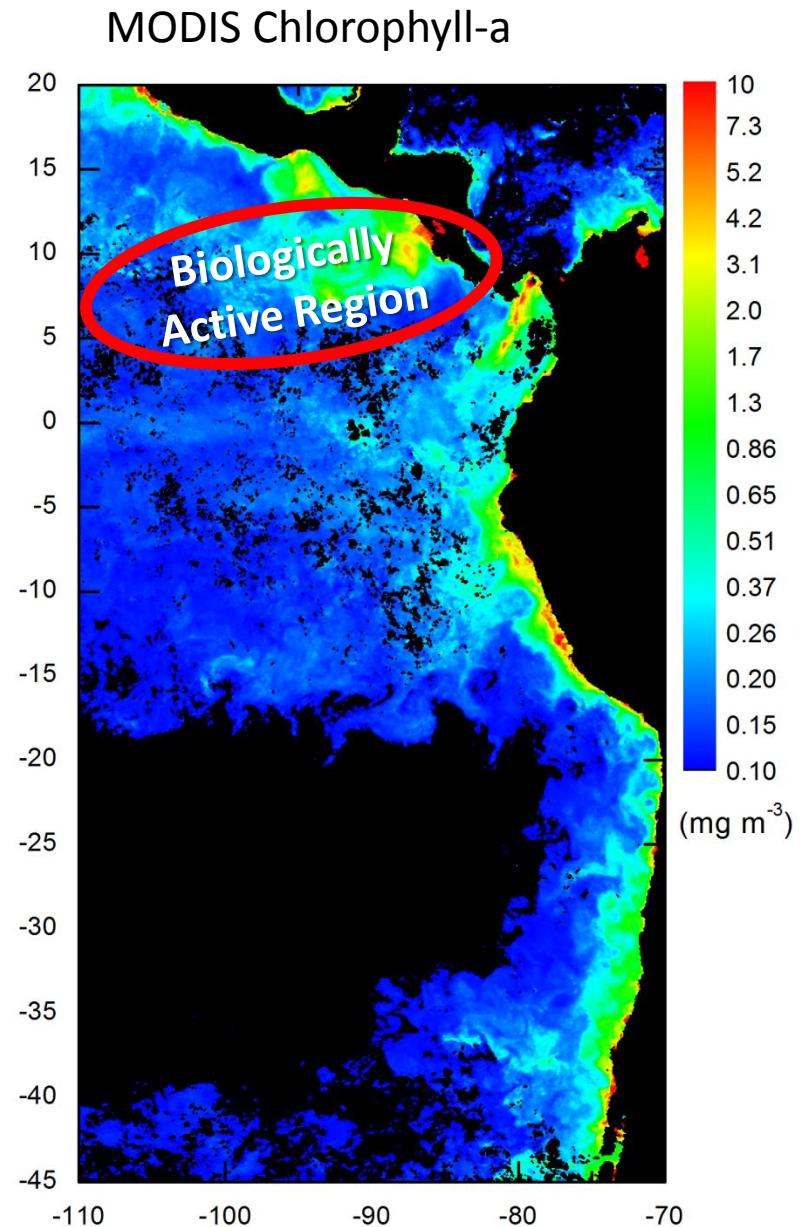
CAM-chem acetone << TOGA obs. near the oligotrophic surface and the transition layer.



# RF05 – oligotrophic ocean

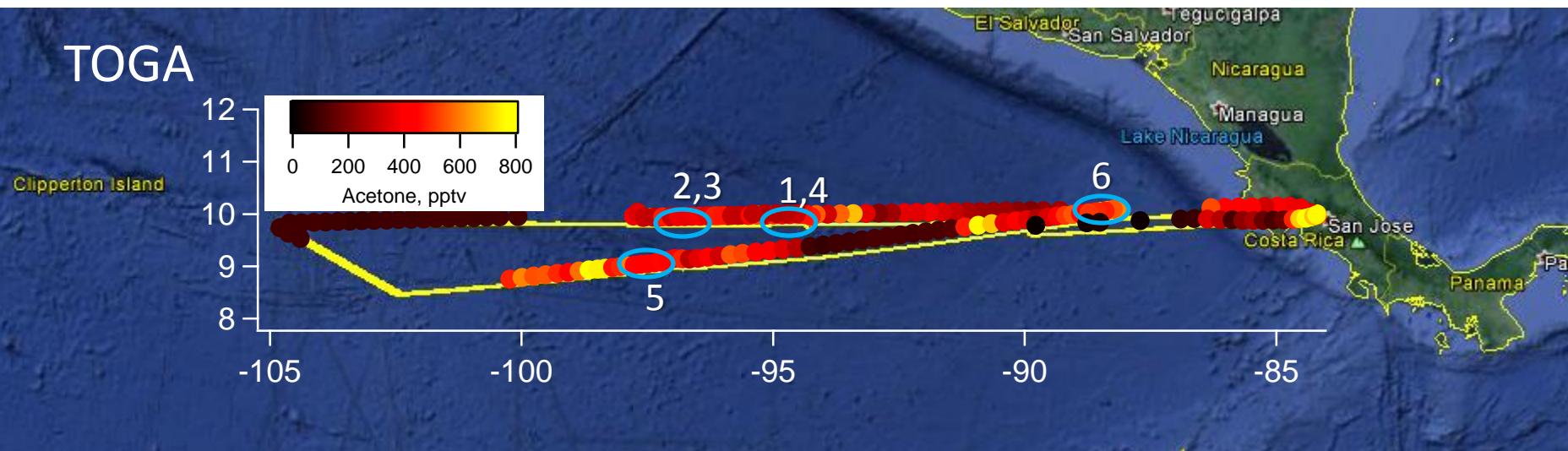


# RF12 – biologically active region/convection MBL

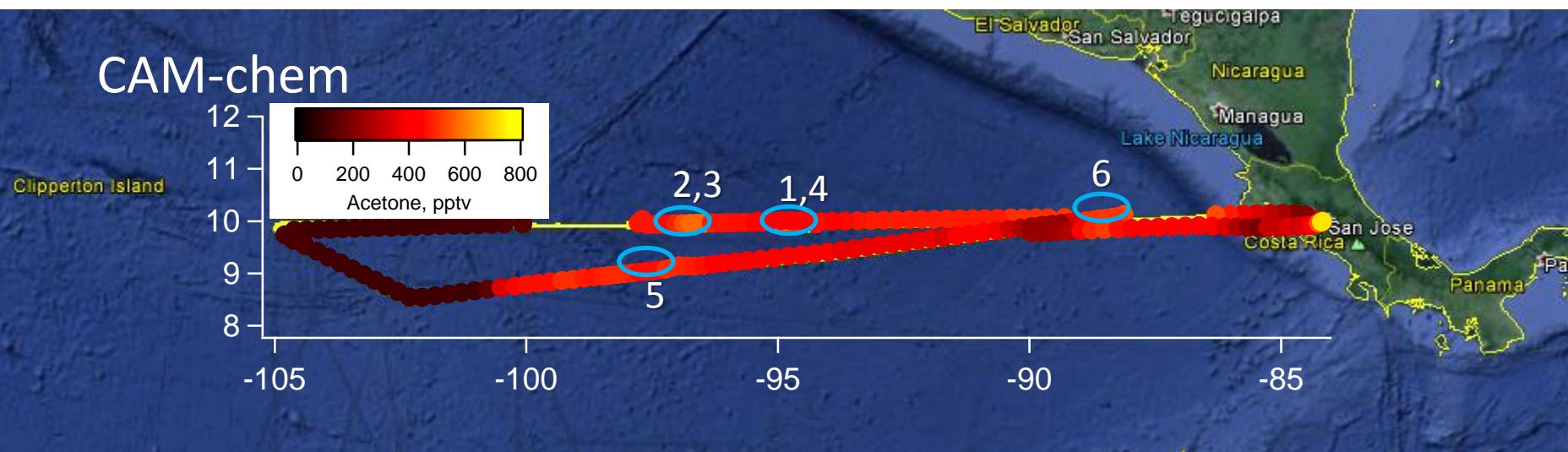


## RF12 – biologically active region

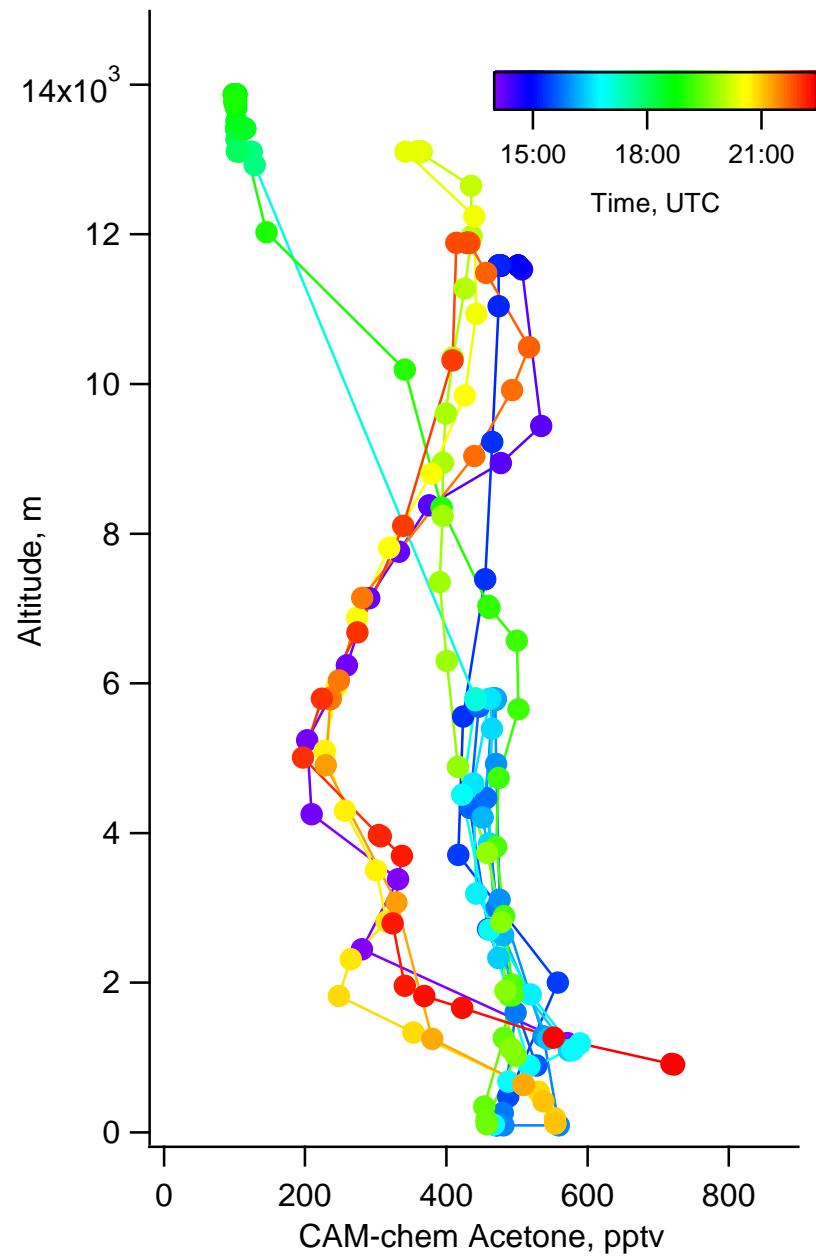
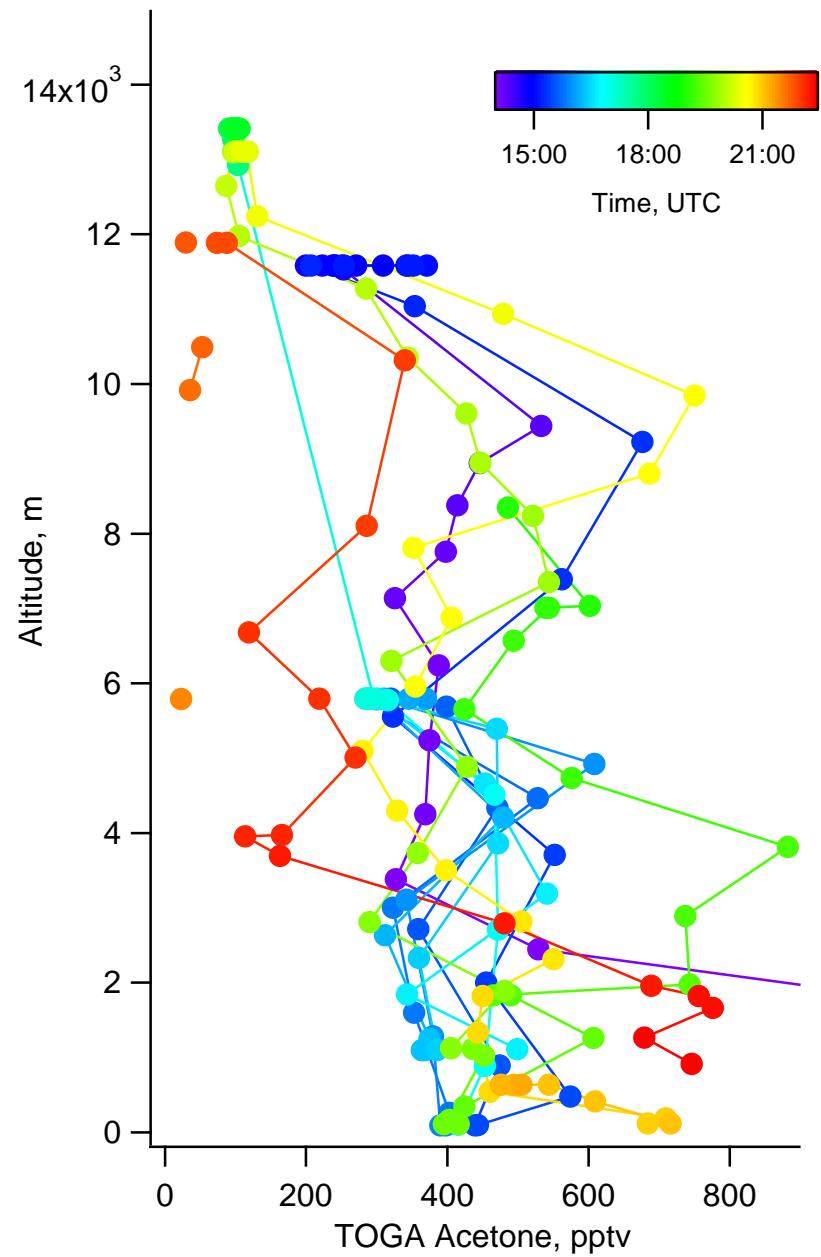
TOGA



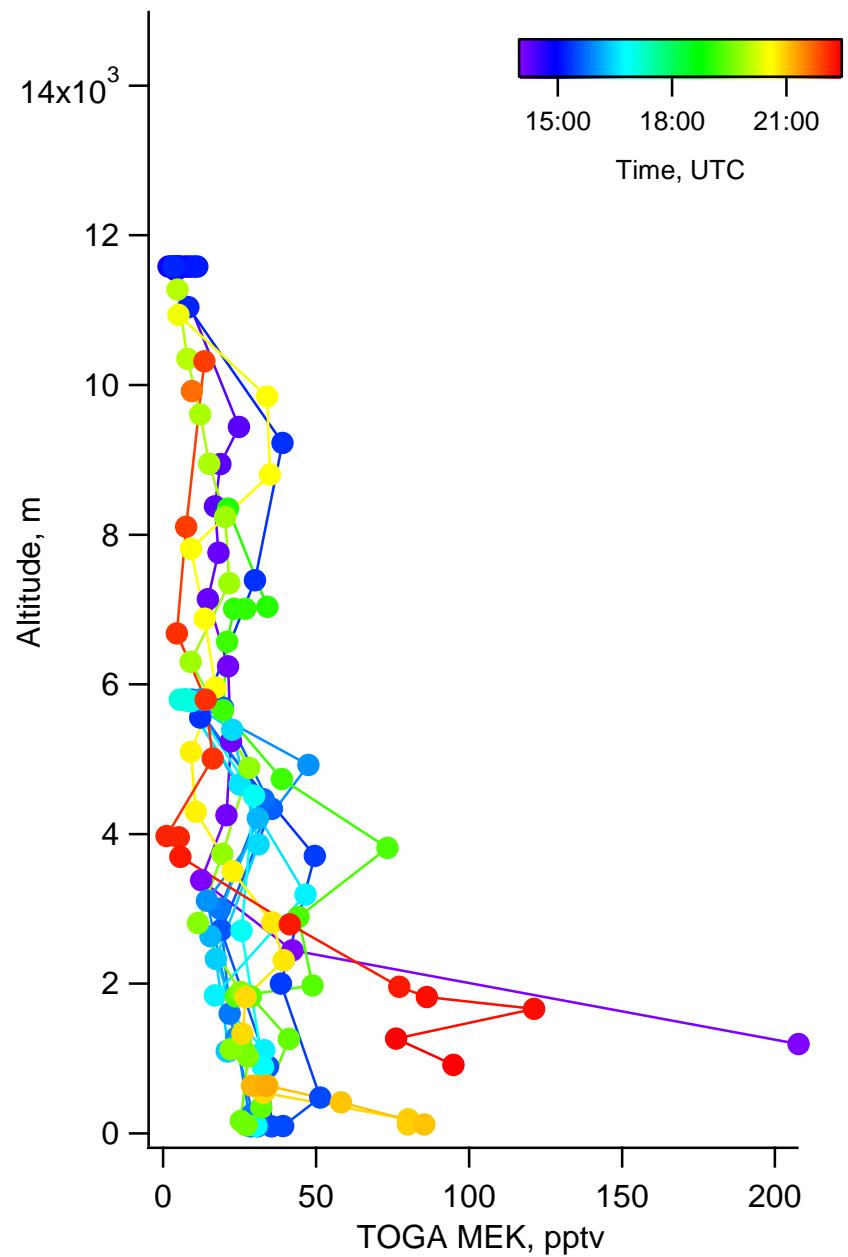
CAM-chem



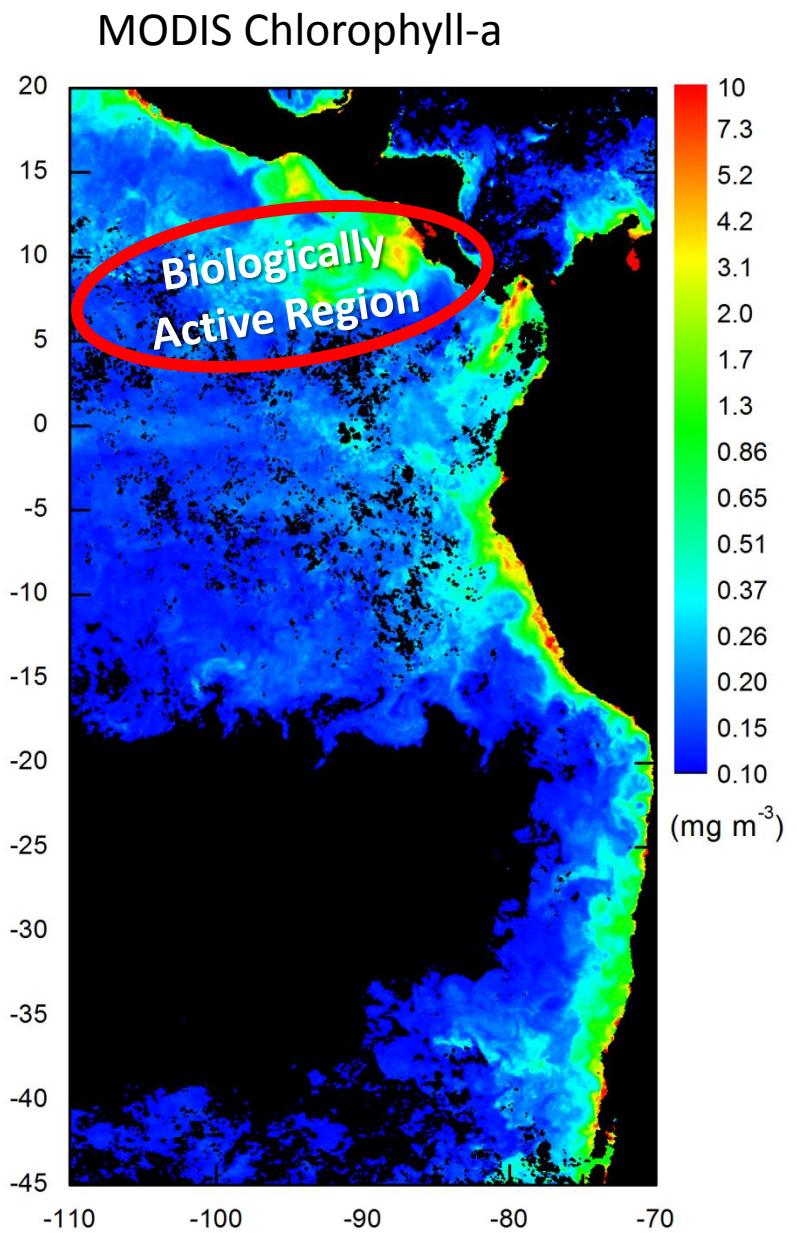
## RF12 – biologically active region



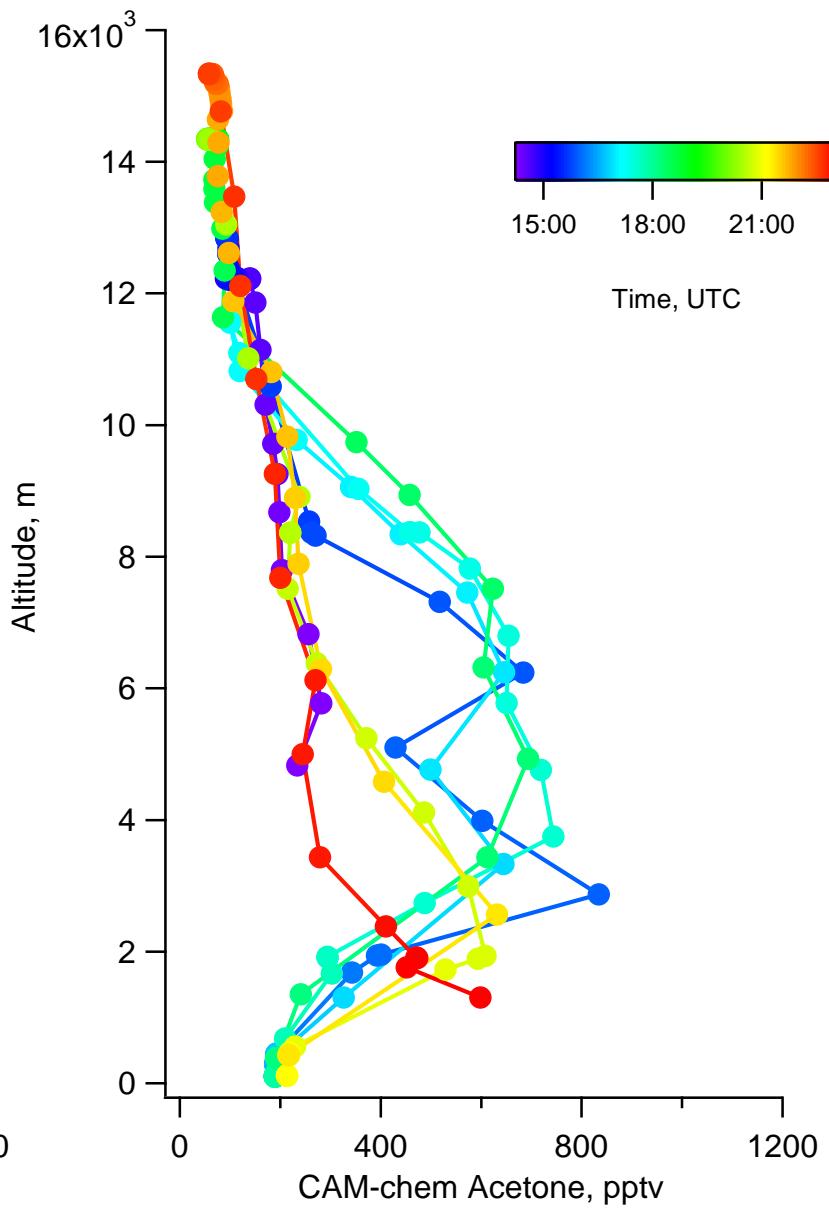
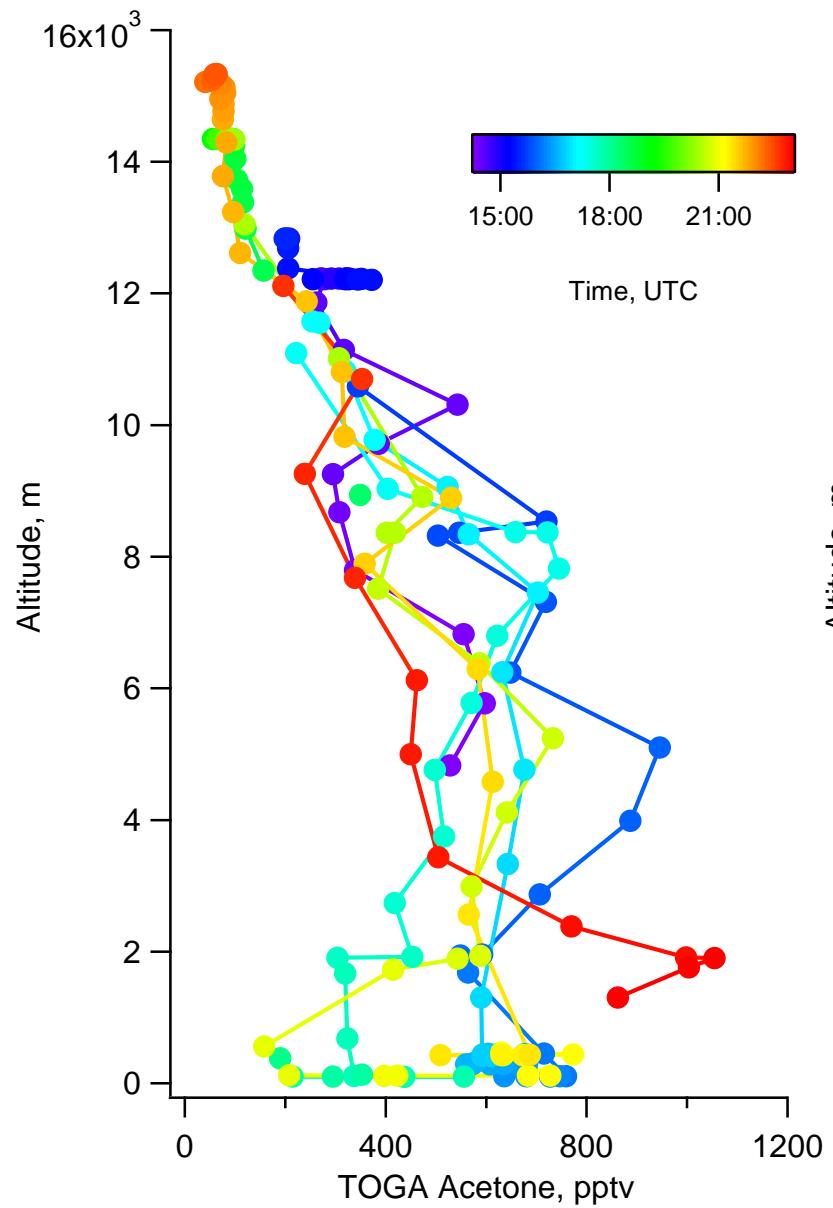
# RF12 – biologically active region



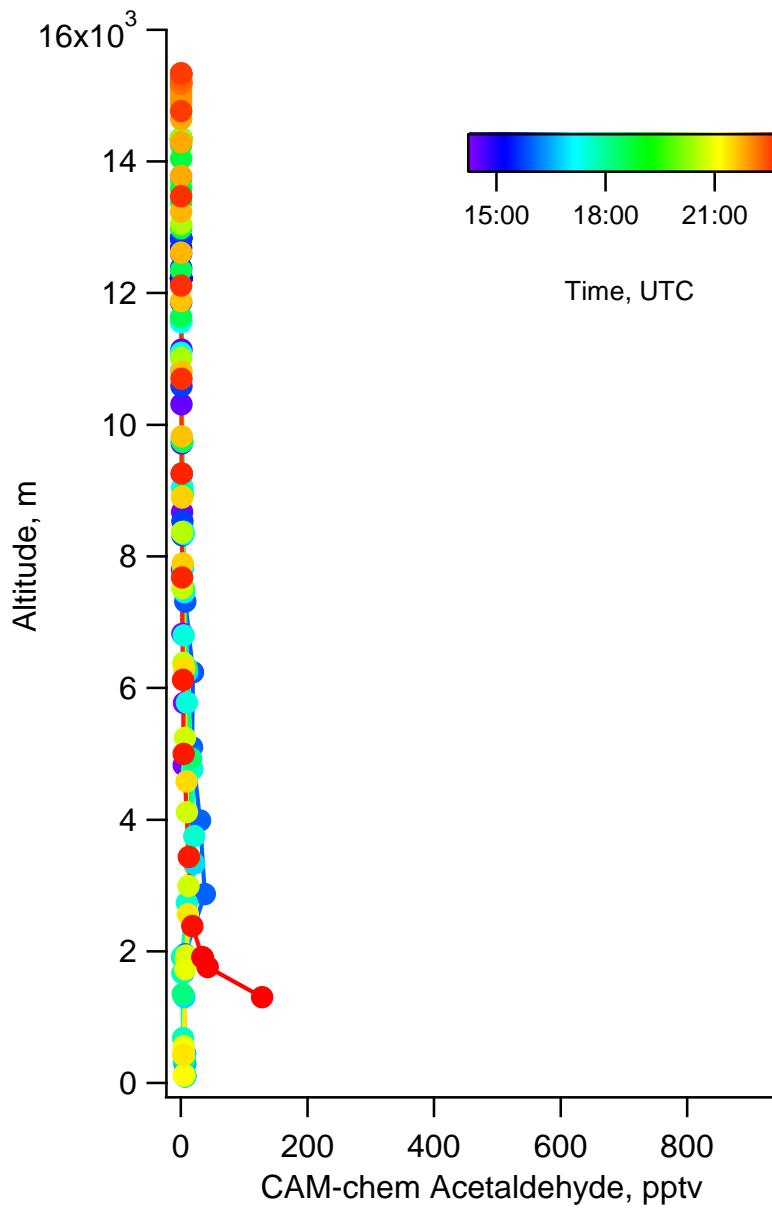
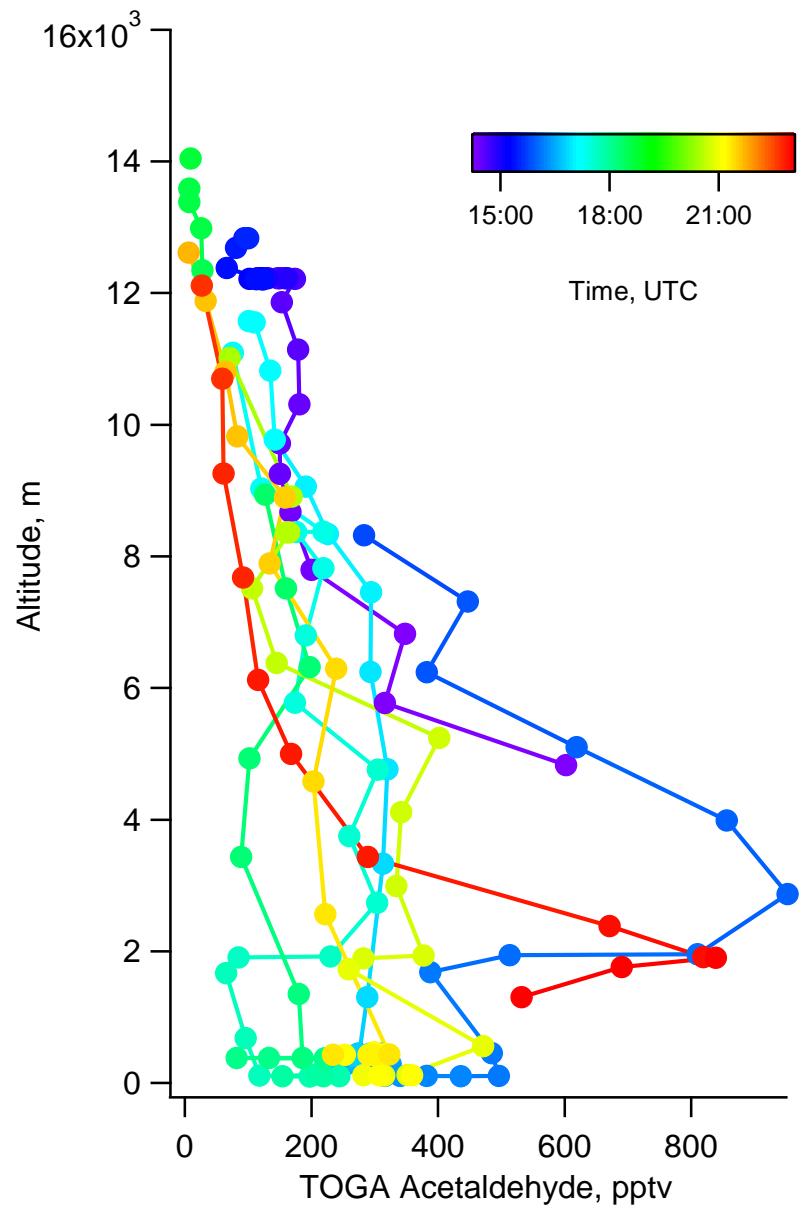
# RF15 – biologically active region



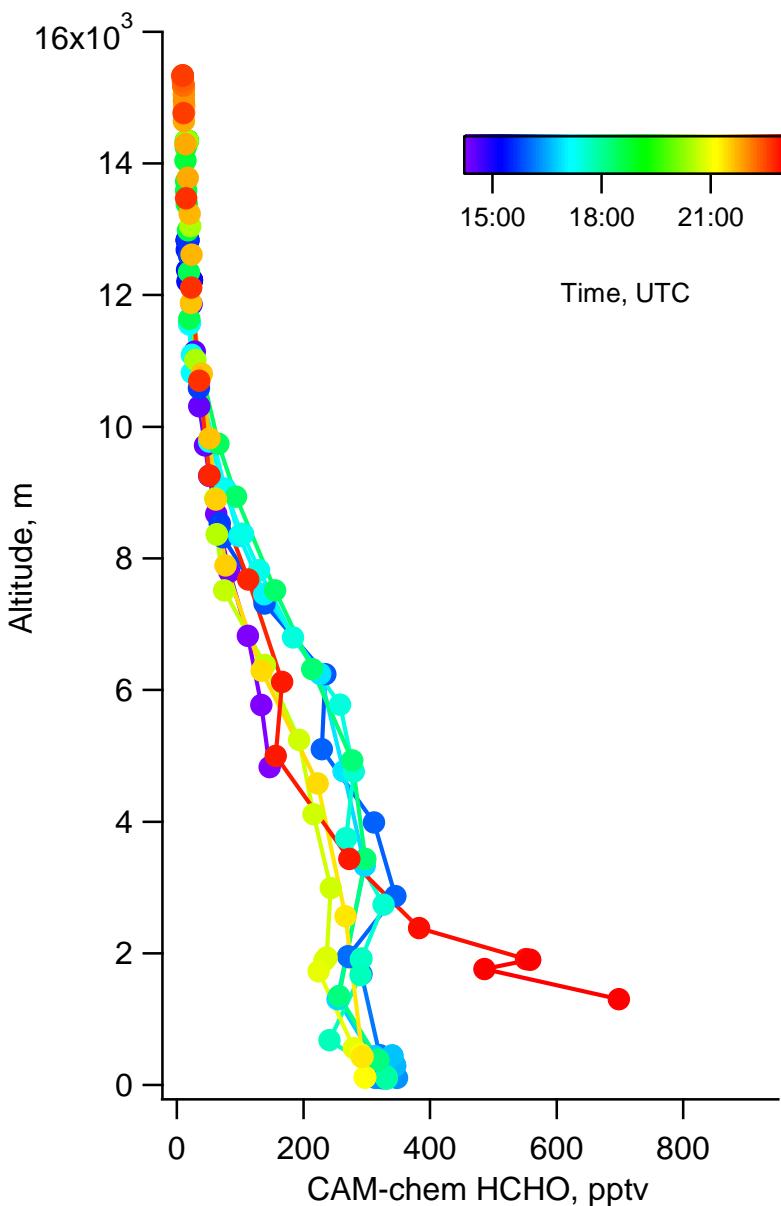
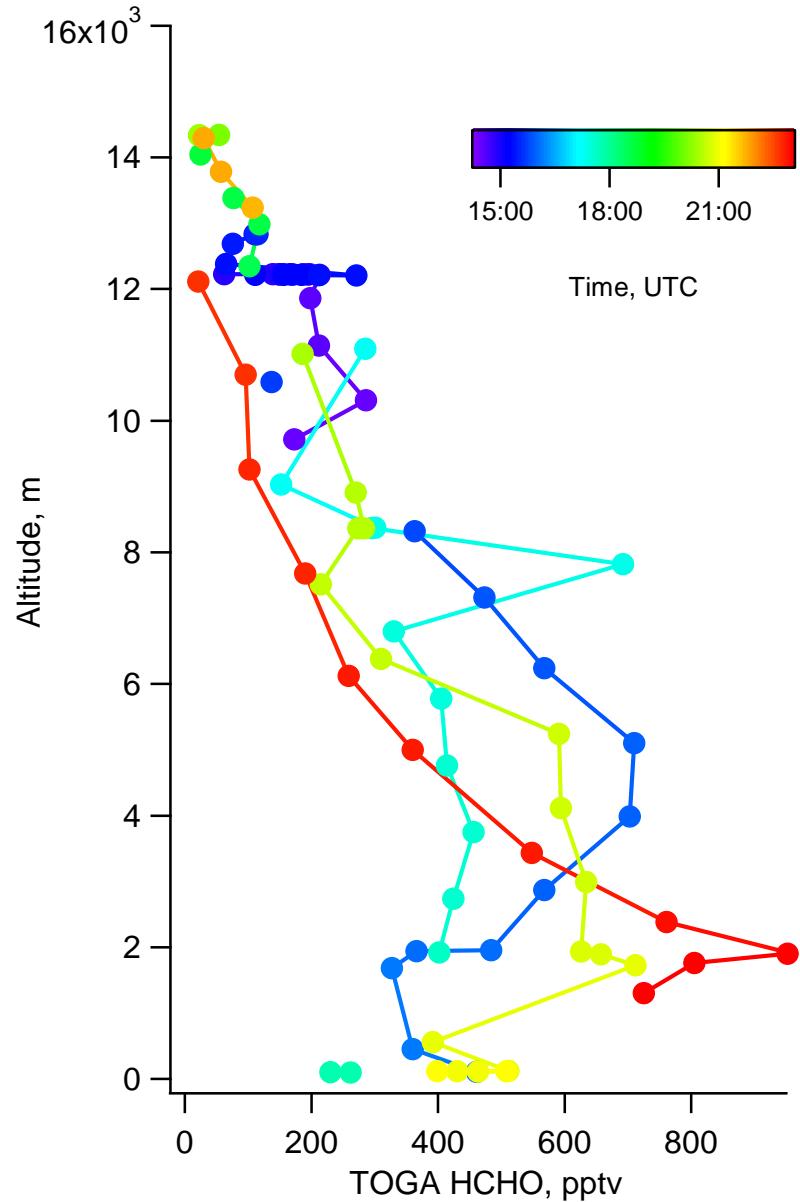
## RF15 – biologically active region



## RF15 – biologically active region



## RF15 – biologically active region



# Next Steps

- TOGA/CAM-chem comparison – remainder of flights
- Assess organohalogen and DMS emission inventory
- Investigate OVOC production/loss over ocean
- Convection of species emitted from the ocean
- TORERO relevant papers:
  - OVOCs in the remote troposphere
  - TOGA paper
  - Comparison of observed organohalogens and recent CAM-chem model results