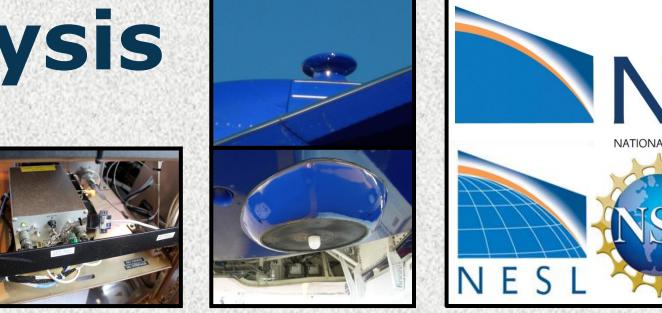


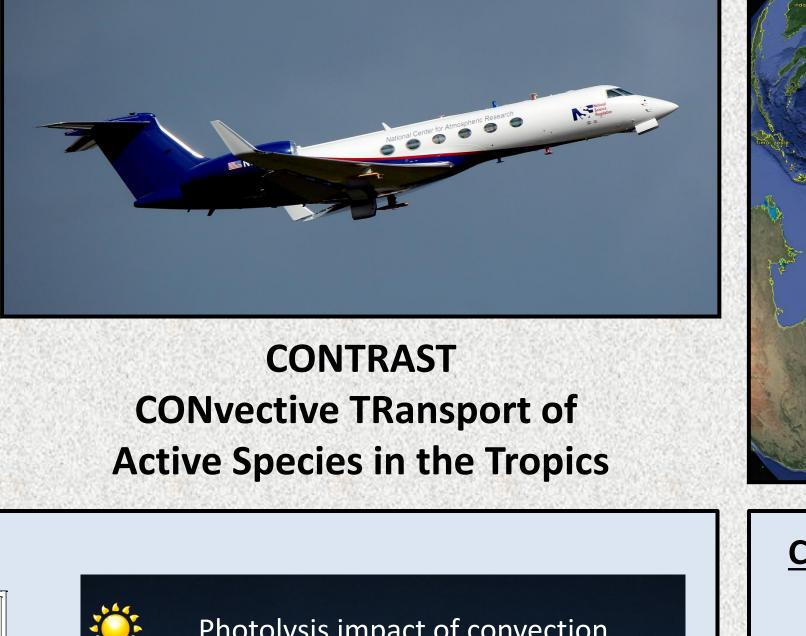
# **Cloud and aerosol influence on photolysis** frequencies during CONTRAST

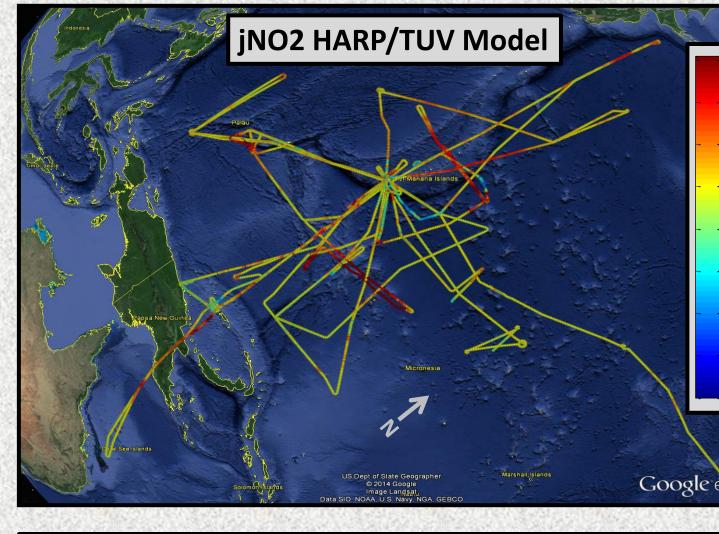
Samuel R. Hall, Kirk Ullmann



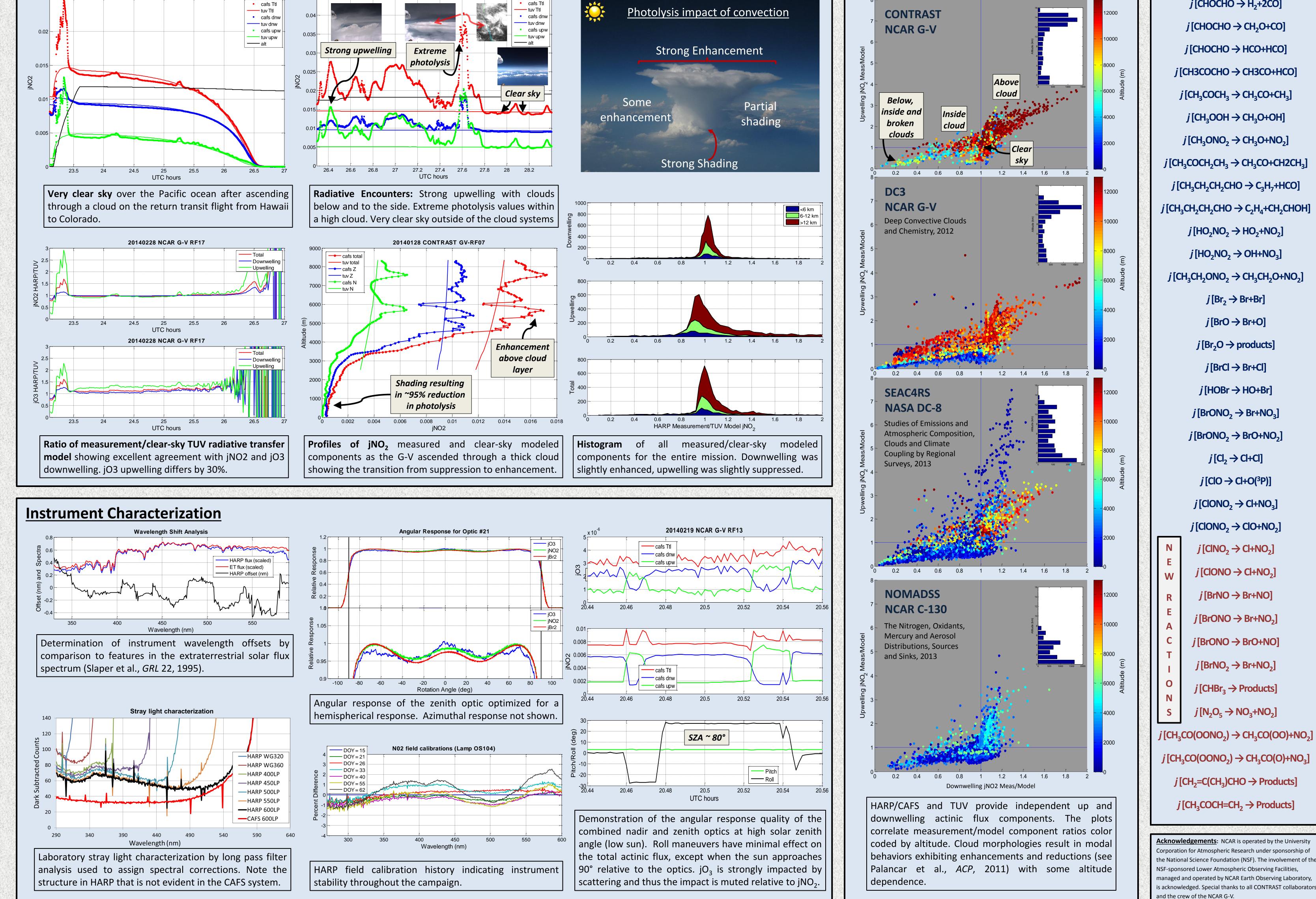
### Summary

Clouds and aerosols influence the lifetime of photolytically active chemical species in the atmosphere through scattering and absorption. Below and within optically thick convective clouds, shading reduces the photochemistry thus permitting vertical motion within the cloud to transport short-lived species from the boundary layer to the free troposphere and episodically to the lower stratosphere. Conversely, near highly scattering cloud tops, photochemistry is greatly accelerated. In such rapidly evolving environments, accurate photolysis frequencies are required to study the chemical processes and evolution of distributed species. During the CONTRAST experiment, photolysis frequencies were calculated from measurements of spectrally resolved actinic flux by the HIAPER Airborne Radiation Package (HARP) on the NCAR G-V aircraft. Statistical correlations of the data reveal modal behavior that could help assess cloud fields in global and regional chemistry models.



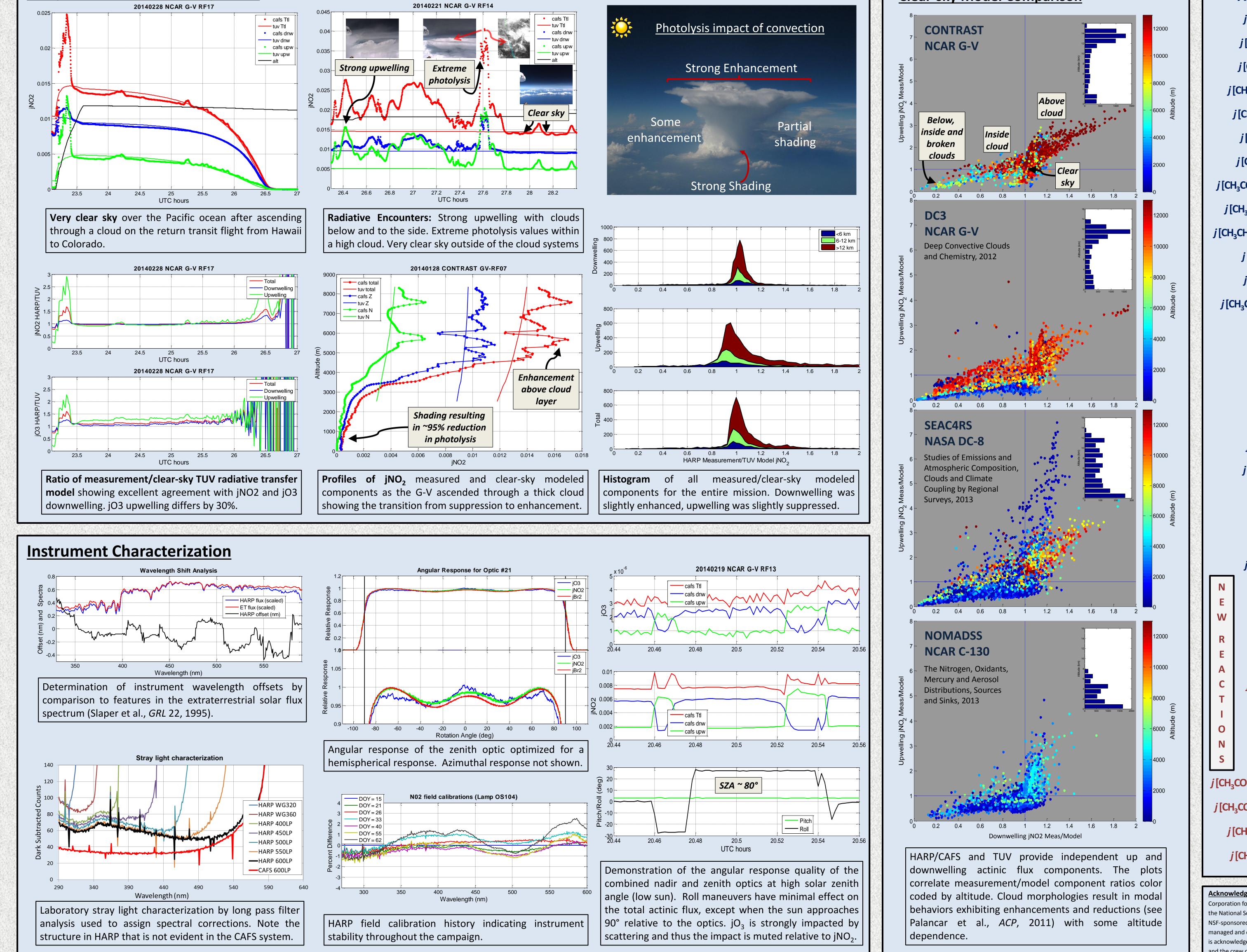


## **Clear-sky Model Comparison**



**Photolysis Frequencies**  $j[O_3 \rightarrow O_2 + O(^1D)]$  $j[NO_2 \rightarrow NO+O(^3P)]$  $j[H_2O_2 \rightarrow 2OH]$  $j[HNO_2 \rightarrow OH+NO]$  $j[HNO_3 \rightarrow OH+NO_2]$  $j[CH_2O \rightarrow H+HCO]$  $j[CH_2O \rightarrow H_2+CO]$  $j[CH_3CHO \rightarrow CH_3+HCO]$  $j[C_2H_5CHO \rightarrow C_2H_5+HCO]$ j [CHOCHO  $\rightarrow$  H<sub>2</sub>+2CO]

## **Comparisons to TUV clear sky model**





j [CHOCHO  $\rightarrow$  HCO+HCO] j [CH3COCHO  $\rightarrow$  CH3CO+HCO] j [CH<sub>3</sub>COCH<sub>3</sub>  $\rightarrow$  CH<sub>3</sub>CO+CH<sub>3</sub>] j [CH<sub>3</sub>OOH  $\rightarrow$  CH<sub>3</sub>O+OH] j [CH<sub>3</sub>ONO<sub>2</sub>  $\rightarrow$  CH<sub>3</sub>O+NO<sub>2</sub>] j[CH<sub>3</sub>COCH<sub>2</sub>CH<sub>3</sub>  $\rightarrow$  CH<sub>3</sub>CO+CH2CH<sub>3</sub>] j [CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CHO  $\rightarrow$  C<sub>3</sub>H<sub>7</sub>+HCO] j [CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CHO  $\rightarrow$  C<sub>2</sub>H<sub>4</sub>+CH<sub>2</sub>CHOH]  $j[HO_2NO_2 \rightarrow HO_2 + NO_2]$  $j[HO_2NO_2 \rightarrow OH+NO_3]$ j[CH<sub>3</sub>CH<sub>2</sub>ONO<sub>2</sub>  $\rightarrow$  CH<sub>3</sub>CH<sub>2</sub>O+NO<sub>2</sub>]  $j[Br_2 \rightarrow Br+Br]$ j [BrO  $\rightarrow$  Br+O]  $j[Br_2O \rightarrow \text{products}]$ j [BrCl  $\rightarrow$  Br+Cl]  $j[HOBr \rightarrow HO+Br]$  $j[BrONO_2 \rightarrow Br+NO_3]$  $j[BrONO_2 \rightarrow BrO+NO_2]$  $j[Cl_2 \rightarrow Cl+Cl]$ j [CIO  $\rightarrow$  CI+O(<sup>3</sup>P)] j [CIONO<sub>2</sub>  $\rightarrow$  CI+NO<sub>3</sub>]

 $j[CINO_2 \rightarrow CI+NO_2]$ 

j [BrNO  $\rightarrow$  Br+NO]