### PASE December 2009

#### Ship & Aircraft Plumes

#### Flights RF10 & 1st Stack RF09

**Peroxide Budgets** 

# I. Ship & Aircraft Plumes

- Unique opportunity ... plume aerosol photochemistry ... pristine bkg air.
- High  $NO_x$  ... active halogens ... DMS oxidation impact.
- Dawn to mid-morning transition.
- Transport and dilution.

# Ship Plumes Roadmap

- State possible cases from 4 flights.
- Describe what we are using as plume criteria.
- Show VOS shiptracks and HYSPLIT<sup>1</sup> back trajectories.
- Plume dilution estimates.
- Photochemical expectation.
- Chemical interpretation.
- Outstanding issues in the chemistry & transport.

<sup>1</sup>The authors gratefully acknowledge the NOAA Air Resources Laboratory (ARL) for the provision of the HYSPLIT transport and dispersion model and/or READY website (http://www.arl.noaa.gov/ready.html) used in this publication.

# I. Plume Case Identification

Two part process to identify cases Type I - Observed chemistry and aerosol spikes; Is there a VOS<sup>1</sup> ship consistent with trajectories?

Type II - From VOS ship tracks & trajectory estimates; Are there missed chemical and aerosols plumes?

While it is relatively "easy" to see plumes in the chemistry, did we actually note/observe all we should have seen based upon VOSs.

<sup>1</sup>VOS - volunteer observing ships

### **Noted Plume Cases**

#### Type I

- RF13 multiple encounters after 15:55 UTC Highly likely, 3-day transport,
- RF04 ~20.53 UTC

#### RF07 - ex. ~23.95 & ~24.21 UTC

Plausible from CN,  $SO_2$  and  $O_3$ , but difficult match to VOS

#### <u>Type II</u>

#### RF01 - ~20.71, ~21.02, ~21.46, ~22.99 UTC

Probable from ship track, 1.5-day transport, mixed in with aircraft events.

Possible, 2-day transport but aloft, convective flight.

### Plume Phenomenon example: RF13 ~19:57-20:04, ~20:09



### Plume Phenomenon - 2



### Plume Phenomenon - 3



 $SO_2 \times 10^1$  DMS x  $10^1$  H<sub>2</sub>O<sub>2</sub> CH<sub>3</sub>OOH

### **RF13 Plume Locations**



### **RF13 Plume Trajectories & Tracks**



#### Is 3 days okay based upon plume dilution?

#### Dilution estimate Method A:

- 1. Simple power law relationship in Z and Y.
- 2. Coefficients from multiple ship track expts.
- 3. Added limit on Z;  $Z_{max} = 500$  m.
- 4. Plume cross section area: A(t) = Z(t)Y(t).
- 5. from von Glassow et al. [2003].

#### Dilution estimate Method B:

- 1. LES study using FIRE and BOMEX data.
- Dilution rate scaled to convective turnover time t\*; t\* = Zi/w\*; w\*=f(integrated bouyancy flux).
- 3.  $D(t) = a(t^*/t)^b = dln(A_{plume})/dt$ .
- 4. from Chosson et al. [2008].

### Ship Plume Dilution $\Delta$ Estimates



### RF13 Plume Maximum "Deltas"

Time	Alt.	03	S02	DMS	CN	HO	H2SO4	CH3OOH	RO2
15.99	300	-1.00	15	na	500	0.7	0.0	na	
17.02	148		150	-40	350	1.5	3.0	na	-5.0
17.22	499	-0.75	115	-20	250	1.0	3.0	-75	-8.0
18.72	313	1.00	115	-25	225	2.6	15.0	-120	-10.0
19.43	37	2.00	115	-25	225	5.0	17.0	-100	-12.0
20.04	380	2.00	75	-15	160	5.0	10.0	-180	-9.0
20.15	380	0.25	10	0	40	0.5	2.0	-50	-6.0
possible									
17.62	499	-0.25	0	0	25	0.0	0.0	-50	-1.5

# Plume Event Chemistry Summary

- Increased SO<sub>2</sub>
- Increased CN; fine = cold = hot; non-volatile
- Increased O<sub>3</sub> often but not always
- Increased HO (day)
- Increased H<sub>2</sub>SO<sub>4</sub> (day)
- Decreased RO<sub>2</sub> and interpolated orgO<sub>2</sub> (day)
- Decreased CH<sub>3</sub>OOH (day)
- Decreased DMS often but not always
- $HO_2$  and  $H_2O_2$  relatively unaffected
- MSA relatively unaffected

Plume Photochemical Expectation<sup>1</sup>

Ship emits NOx, SO<sub>2</sub>, CO, VOCs, CH<sub>2</sub>O, and "soot" and sulfate fine particles

Three stages of chemistry based on NOx

<u>Daytime</u>

stage one - early plume dispersion with NO<sub>x</sub> >1000 ppb and NO/NO<sub>2</sub>/O<sub>3</sub> photostationary state applies, O<sub>3</sub> depletion.

- stage two intermediate plume dispersion with  $1 < NO_x < 1000$  ppb, O<sub>3</sub> recovery, enhanced HO and HNO<sub>3</sub> and H<sub>2</sub>SO<sub>4</sub> production, and HO<sub>2</sub> and RO<sub>2</sub> are small, and hydroperoxide production is absent.
- stage three long-range plume dispersion with  $NO_x < 1$  ppb, HO is maximized, HO<sub>2</sub> and RO<sub>2</sub> radicals return, VOC oxidation becomes important, net O<sub>3</sub> production and possible hydroperoxide formation.

<sup>1</sup>Chen et al. [2005]; von Glasow et al. [2003]; Song et al. [2003a&b]; Karamchandani and Seigneur [1999]; Karamchandani et al. [1998; 2000];

### Plume Photochemical Expectation cont.

#### Nighttime

- stage one early plume dispersion with NO<sub>x</sub> >1000 ppb and O3 titration as NO converted to NO<sub>2</sub>,
- stage two intermediate plume dispersion with 1 < NO<sub>x</sub> < 1000 ppb, NO, NO<sub>2</sub>, NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> chemistries are effective, and
- stage three long-range plume dispersion with NO<sub>x</sub> < 1 ppb, NO<sub>2</sub> dominates NO<sub>x</sub>, some NO<sub>3</sub> and negligible N<sub>2</sub>O<sub>5</sub>.

# **Power Plant and Ship Plume**

- Expected photochemistry of ship plumes as they dilute is consistent with our observed changes in SO<sub>2</sub>, HO, H<sub>2</sub>SO<sub>4</sub> and O<sub>3</sub>. CN?
- In stage 2 and 3, dilution with background air should set peroxide levels to background values, ∆ROOH=0.
- HO<sub>2</sub> and RO<sub>2</sub> should both decrease which in turn leads to decreased peroxide production.

Karamchandani et al. [1998, 1999, 2000], Karamchandani and Seigneur, [1999], von Glasow et al. [2003], Chen et al. [2005], and Song et al. [2003a&b]

### Unresolved "Howevers":

#### Q1: How is CH<sub>3</sub>OOH reduced in ship plume impacted ambient air relative to H<sub>2</sub>O<sub>2</sub>, which does not appear to change?

Plume chemistry is enriched in  $NO_x$ , and depleted in  $HO_2$  and  $CH_3OO$  and  $RO_2$  in general. From the standpoint of hydroperoxide production there should be little production of either  $H_2O_2$  or  $CH_3OOH$  within a plume. Hence, the  $H_2O_2$  or  $CH_3OOH$  concentrations should be that of the diluent air.

Q2: How does HO<sub>2</sub> remain constant while RO<sub>2</sub> is reduced as expected in plume air?

### Do Loss Rxns in Plume Ans. Q1?

HO or other speculative reactant, e.g. CI from NO<sub>3</sub>/N<sub>2</sub>O<sub>5</sub>/HNO<sub>3</sub>

 $CH_{3}OOH + HO \longrightarrow H_{2}O + CH_{3}OO$   $CH_{3}OOH + HO \xrightarrow{O_{2}} H_{2}O + CH_{2}O + HO_{2}$   $CH_{3}OOH + Cl \longrightarrow HCl + CH_{3}OO$   $CH_{3}OOH + Cl \xrightarrow{O_{2}} HCl + CH_{2}O + HO_{2}$   $H_{2}O_{2} + HO \longrightarrow H_{2}O + HO_{2}$   $H_{2}O_{2} + Cl \longrightarrow HCl + HO_{2}$ 

	НО	C1					
$H_2O_2$	$1.7 \times 10^{-12}$	$4.1 \times 10^{-13}$ (4x slower than HO)					
CH <sub>3</sub> OOH	$7.4(5.5) \times 10^{-12}$	$5.7(5.9) \times 10^{-11}$ (7x faster)					

Table: *ks* for ROOH reactions with HO and Cl, cm<sup>3</sup> molec<sup>-1</sup> s<sup>-1</sup>

### Simple Loss Example

 $\begin{array}{ll} [H_2O_2]_0 = 1000 & k_1 = 1.7 \times 10^{-12} \\ [CH_3OOH]_0 = 1000 & k_2 = 7.4 \times 10^{-12} \\ [HO]_p = 2 \times [HO]_{bkg} \sim 10^7 \end{array}$ 

### $[CH_3OOH]_t:[H_2O_2]_t = exp(-\{k_2 - k_1\} [HO]t)$

From HO loss alone, it would take ~1 hr after mixing to deplete  $CH_3OOH$  relative to  $H_2O_2$  by 100 to 200 ppt.

# Does $CH_3OO$ cycling to $HO_2$ by $NO_x$ in Plume Ans. Q2?

 $\begin{array}{c} CH_{3}OO + HO_{2} \longrightarrow CH_{3}OOH + O_{2} \\ CH_{3}OO + NO \longrightarrow CH_{3}ONO_{2} \\ CH_{3}OO + NO \xrightarrow{O_{2}} \rightarrow CH_{2}O + NO_{2} + HO_{2} \\ CH_{3}OO + NO_{2} \longrightarrow CH_{3}OONO_{2} \\ CH_{3}OO + NO_{3} \longrightarrow CH_{2}O + HO_{2} + NO_{2} \\ \hline CH_{3}OO + RO_{2} \longrightarrow CH_{2}O + RO + O_{2} \\ \hline CH_{3}OO + RO_{2} \longrightarrow CH_{2}O + RO + O_{2} \\ \hline CH_{3}OO + RO_{2} \longrightarrow CH_{2}O + ROH + O_{2} \\ \hline CH_{3}OO + RO_{2} \longrightarrow CH_{3}OH + RO + O_{2} \end{array}$ 

net:  $CH_3OO + NO_x \rightarrow CH_2O + RNO_x + HO_2$ and with

 $CH_2O + hv \rightarrow CO + HO_2 + H_2$ 

# Could Halogen Chemistry Ans Q1 and Q2 As Well?

- Plumes are active high NO<sub>x</sub> environments with HNO<sub>3</sub> NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub>.
- RF13 plumes cross dawn to mid-morning sun.
- Q3: Is there evidence of halogen and NOx chemistries in CH<sub>3</sub>OOH, DMS or other species?

### Plume "Punch" List

- Aerosol evolution piece of story.
- Directed plume photochemical simulations.
- Evidence for a halogen story.

# II. Recover Flights RF10 & RF09

- Regain Fluxes for 2 of 13 flights.
- Spikes in some RAF data channels.
- Detect and eliminate ramps and jumps.
  - High pass filter.
  - Threshold defined for each variable.
- Strategy for data fill-in going forward.
- Ex. Spectra & Flux results using a simple fill strategy.

### ATX Spike Example: index~436150 (25 hz)



### WIC "Spike Train" Example: index ~ 424250 (25Hz)





# Simple Fill Strategy

- Applies to removed Spikes & Plumes:
  RF01, RF02, RF09, RF10, RF12, & RF13
- Procedure
- 1) 75% data coverage within 200 s segment.
- 2) Determine perturbations; demean and detrend segment.
- 3) Linear interpolation of perturbation across gap (no discernable difference with zero perturbation fill).

# Vertical Cospectra & Flux Comparisons

- By flight, by time, by altitude
- By Variable
  - Temperature, potential temperature (T,P)
  - Water, virtual potential temperature (T,P,q)
  - Speed (momentum:  $[U^2+V^2]^{1/2}$ )
  - Chemical: DMS,  $SO_2$ , and  $O_3$
- kCo(w'S') & ∑Co(w'S')

### Virtual Potential Temperature Recalculated using MRLA with fill procedure





![](_page_30_Figure_0.jpeg)

![](_page_31_Figure_0.jpeg)

![](_page_31_Figure_1.jpeg)

RF09 is on the positive edge of the range defined by other flights.

RF10 is on the negative edge of the range defined by other flights.

# Going Forward on Fill Strategies

- $\checkmark \cdot 1$ ) zero fill
- ✓ 2) linearly interpolated values
- Singular spectrum analysis (Schoellhamer, 2001; Kondrashov and Ghil, 2006)
- 4) Lomb-Scargle periodogram (Hocke and Kämpfer, 2009)
- □ 5) ARMA automated method (Broersen, 2006)

### **Backup Slides**

# **RF01 Plume Study**

![](_page_34_Figure_1.jpeg)

### **RF04 Plume Study**

![](_page_35_Figure_1.jpeg)

### **RF07 Plume Study**

![](_page_36_Figure_1.jpeg)

### **Plume Halogen Radical Activation**

e.g., Finlayson-Pitts [2003]; Aldener et al. [2006]:

$$\begin{aligned} ClONO_{2(g)} + NaCl_{(aq)} &\longrightarrow Cl_{2(g)} + NaNO_{3(aq)} \\ NO_{3(g)} + NaCl_{(aq)} &\longrightarrow Cl_{(g)} + NaNO_{3(aq)} \\ N_2O_5 + NaCl_{(aq)} &\longrightarrow ClNO_{2(g)} + NaNO_{3(aq)} \end{aligned}$$

e.g., Finlayson-Pitts [2003]; Finlay and Saltzman [2008]:  $O_3 + Br_{(aerosol)}^{-} \xrightarrow{bulk, hetero}$   $BrONO_2 + Br_{(aerosol)}^{-} \xrightarrow{}$  $BrNO_2 + Br_{(aerosol)}^{-} \xrightarrow{}$ 

e.g., Nissenson et al. [2008]:

$$HO_{(g)} + Cl^{-}_{(aq,sfc)} \longrightarrow 0.5Cl_{2(g)} + HO^{-}_{(aq)}$$

# w, Speed Cospectrum

![](_page_38_Figure_1.jpeg)

![](_page_39_Figure_0.jpeg)

![](_page_39_Picture_1.jpeg)

![](_page_39_Picture_2.jpeg)

RF09 is central within the range defined by other flights.

RF10 is central within the range defined by other flights.

### w Ozone Cospectrum

![](_page_40_Figure_1.jpeg)

![](_page_41_Figure_0.jpeg)

### Ozone

RF09 is within the range defined by other flights.

RF10 does not have fast ozone data available.

### w,H<sub>2</sub>O Cospectrum

![](_page_42_Figure_1.jpeg)

![](_page_43_Figure_0.jpeg)

![](_page_43_Figure_1.jpeg)

RF09 is central within the range defined by other flights.

RF10 is at the negative edge of the range defined by the other flights.

![](_page_44_Figure_0.jpeg)

![](_page_44_Figure_1.jpeg)