

# SO<sub>2</sub> Processing During PASE

"Science is always wrong. It never solves a problem without creating ten more."

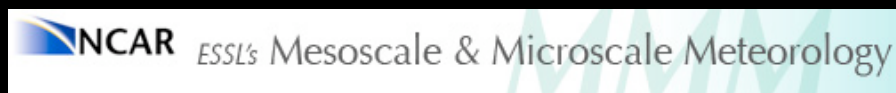
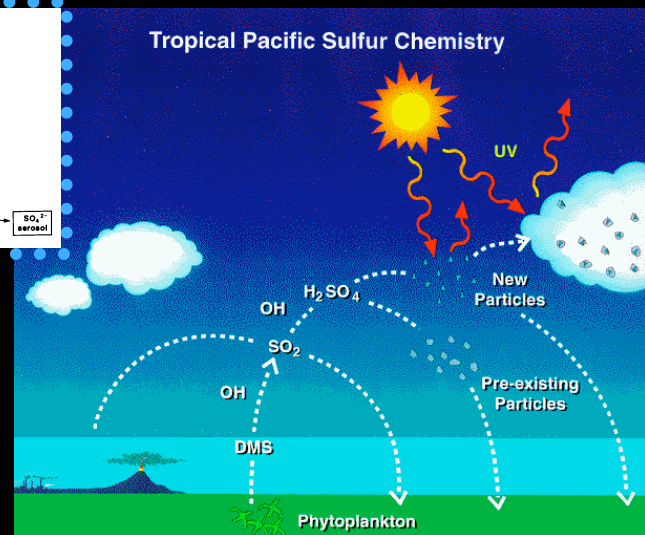
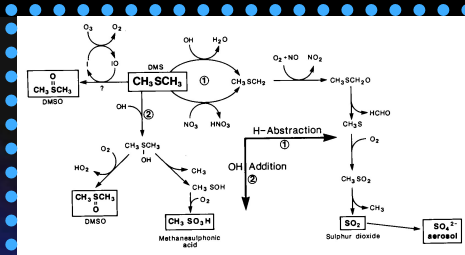
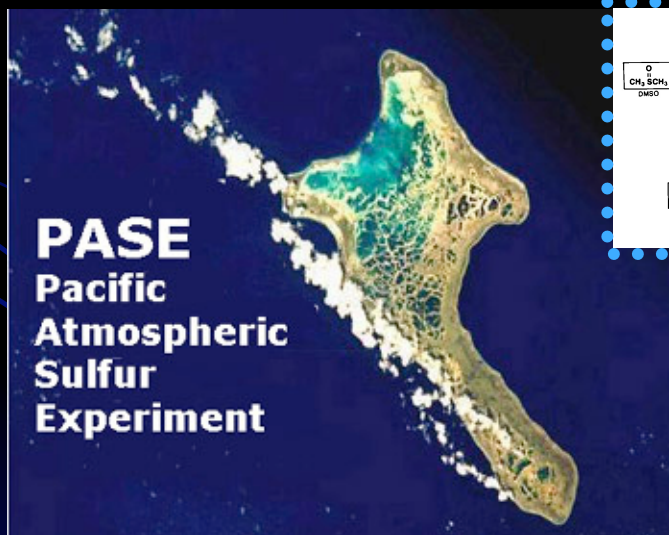
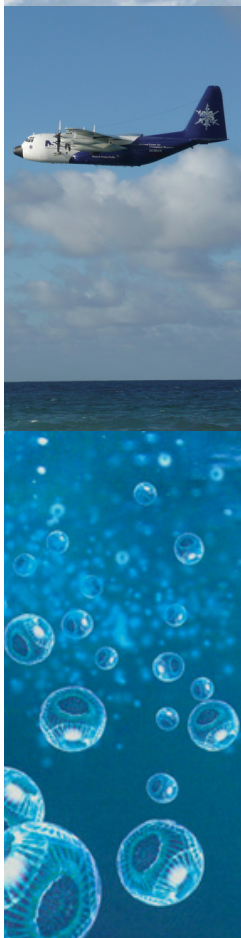
— George Bernard Shaw

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**Table 1. Global Budgets of Main Sulfur Species (TgS/yr) and Their Lifetimes (d)**

	<i>Liu et al., 2007</i>	Range of Other Models <sup>a</sup>		Estimated	Approximate	avg. of last
		Min	Max	Std. Deviation <sup>b</sup>	Diversity <sup>c</sup>	two columns
<b>DMS</b>						
Sources	26.1	10.7	27.6	4.3	11.7	8.0
Oxidation	25.9	10.7	27.6	4.3	11.7	8.0
Global Lifetime	2.4	0.5	3	0.6	---	
<b>SO<sub>2</sub></b>						
Sources	99.4	83.2	126	10.9	9.9	10.4
Anthropogenic	69	55	92	9.4	5.5	7.5
Ship	---	2.5	4.7	0.6	0.3	0.4
Volcanic	4.8	3.4	14	2.7	0.4	1.5
DMS Oxidation	25.6	10	25.7	4.0	11.5	7.8
Sinks	99.1	91.7	126	8.8	---	
Dry Deposition	27.6	14	56	10.7	8.8	9.8
Wet Deposition	4.9	0	20	5.1	1.6	3.3
Oxidation (-->SO <sub>4</sub> )	65.9	---	---	---	---	
Global Lifetime	2.4	0.6	2.6	0.5		
<b>NSS-Sulfate</b>						
Sources	65.8	---	---	---	---	
Direct Emissions <sup>d</sup>	---	1.1	4.6	0.9	1.8	1.3
Heterogeneous Ox.	49.3	24.5	57.8	8.5	10.8	9.7
Homogeneous Ox.	16.6	6.1	16.8	2.7	5.8	4.3
Sinks	65.7	---	---	---		
Dry Deposition	3.6	2.5	18	4.0	2.0	3.0
Wet Deposition	62.1	24	74	12.8	13.7	13.2
Global Lifetime	4.3	2.6	6.8	1.1	0.8	0.9

<sup>a</sup>Models included: *Langner & Rodhe [1991], Pham et al. [1995], Chin et al. [1996], Feichter et al. [1996], Graf et al. [1997], Lohmann and Feichter [1997], Chuang et al. [1997], Roelofs et al. [1998], Koch et al. [1999], Lohmann et al. [1999], Haywood & Boucher [2000], Rasch et al. [2000], Chin et al. [2000], Chuang et al. [2002], Rotstajn and Lohmann [2002], Easter et al. [2004], Koch et al. [2006], and Kloster et al. [2006].*

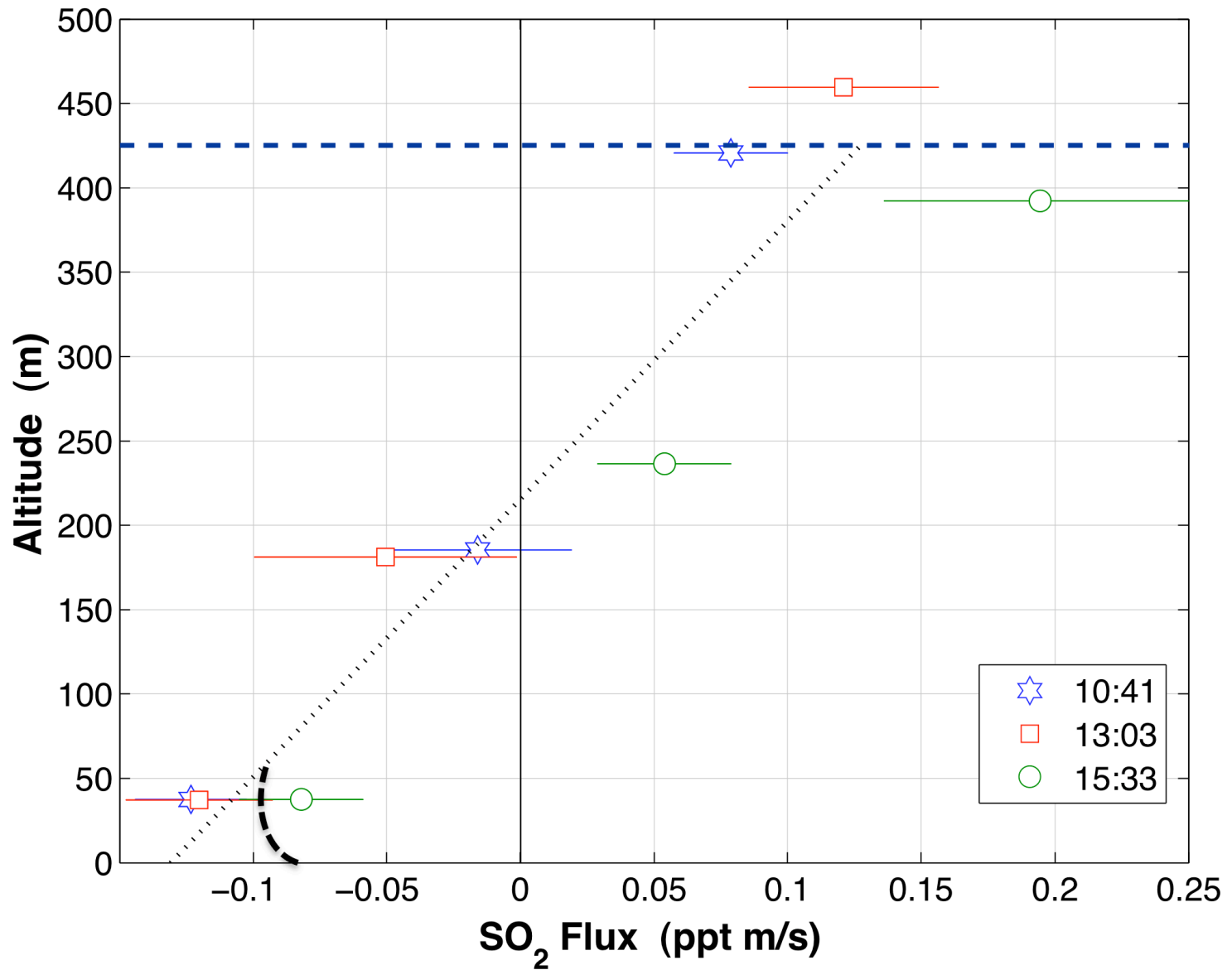
Anthropogenic Sources: fuel bound S, ore smelting, & biomass burning ~70 Tg S/yr

Largest uncertainties: SO<sub>4</sub> wet dep SO<sub>2</sub> dry dep SO<sub>2</sub> sources

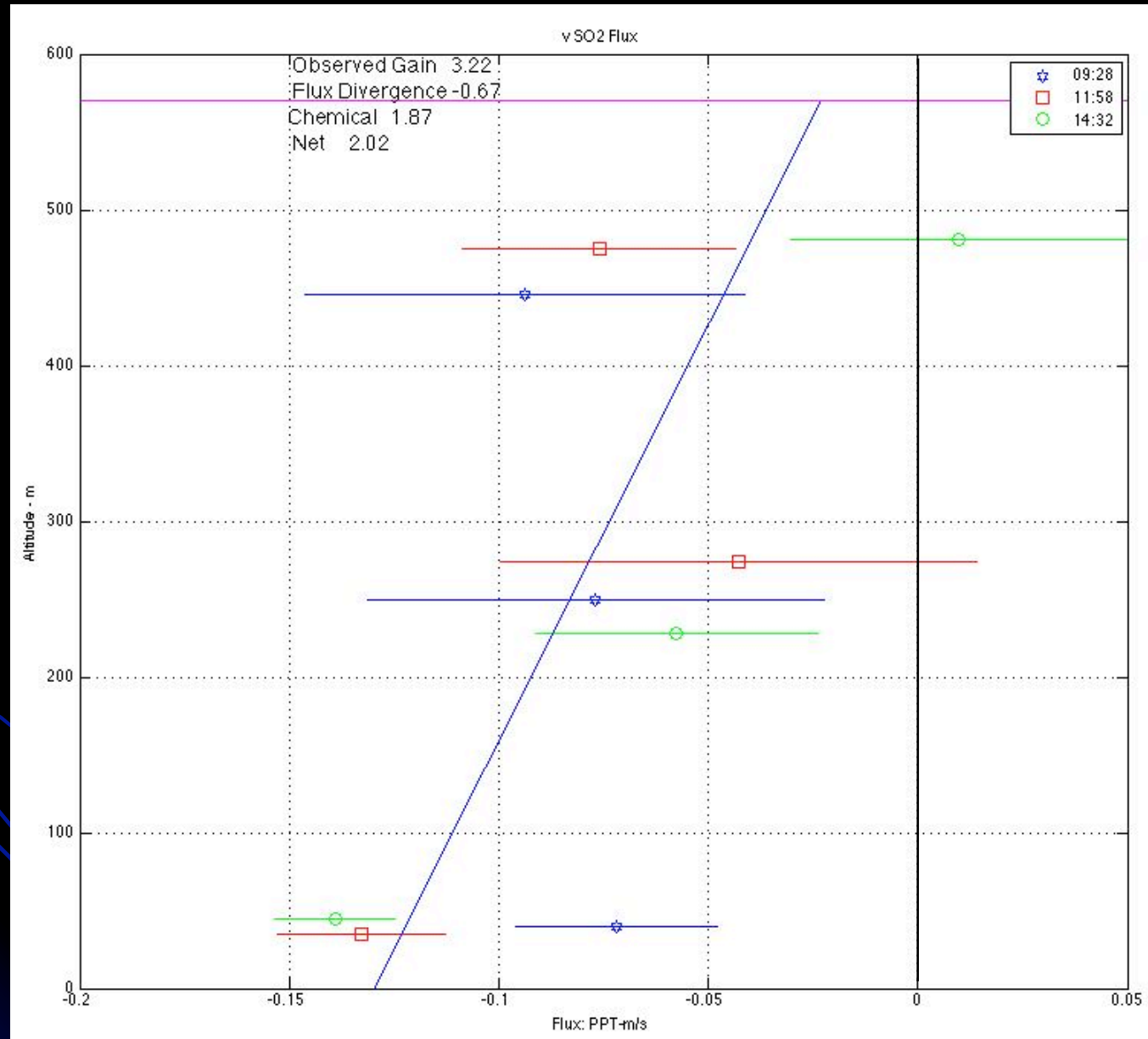
Oceanic DMS emissions: large (~25%) and very difficult to predict.

PASE can address: SO<sub>2</sub> dep & Hetero. Ox.

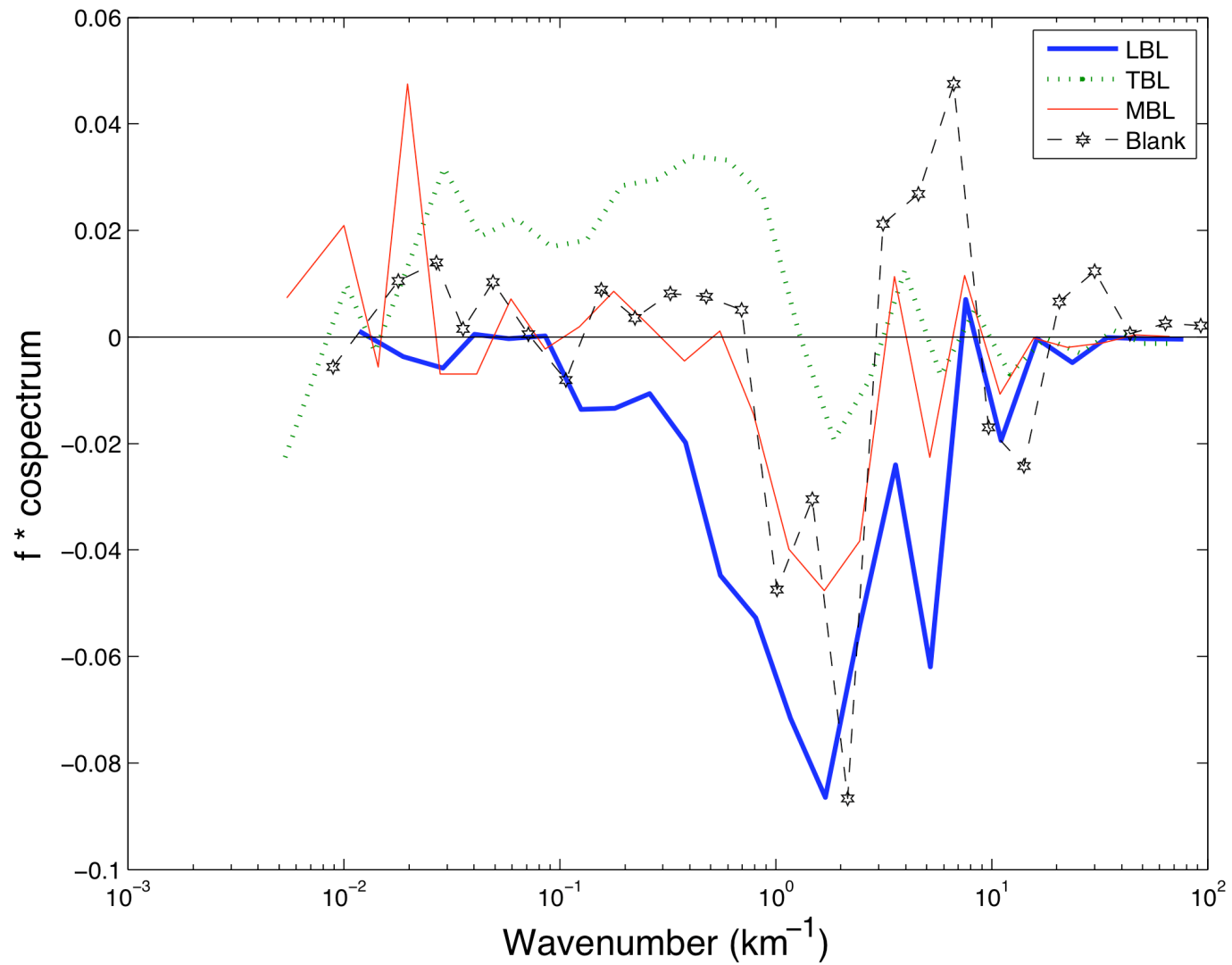
# RF02 SO<sub>2</sub> Flux Profiles



# Flux Profile RF08: BuL source

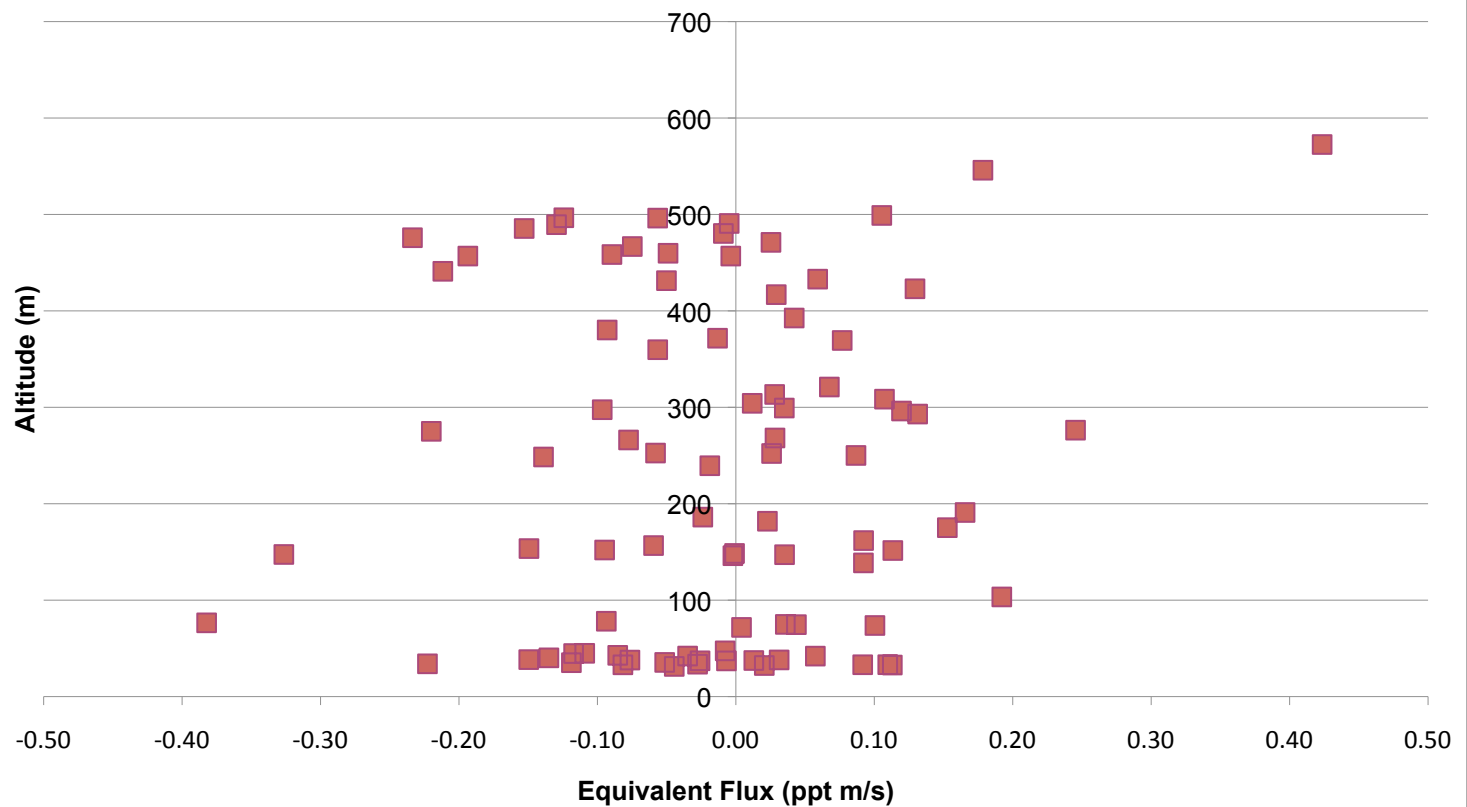


# SO<sub>2</sub> signal & blank, w Cospectra



# Statistics of Background 'flux'

Mean:  $-0.015 \pm 0.123$  ppt m/s



# SO<sub>2</sub> Budgets & Dry Deposition

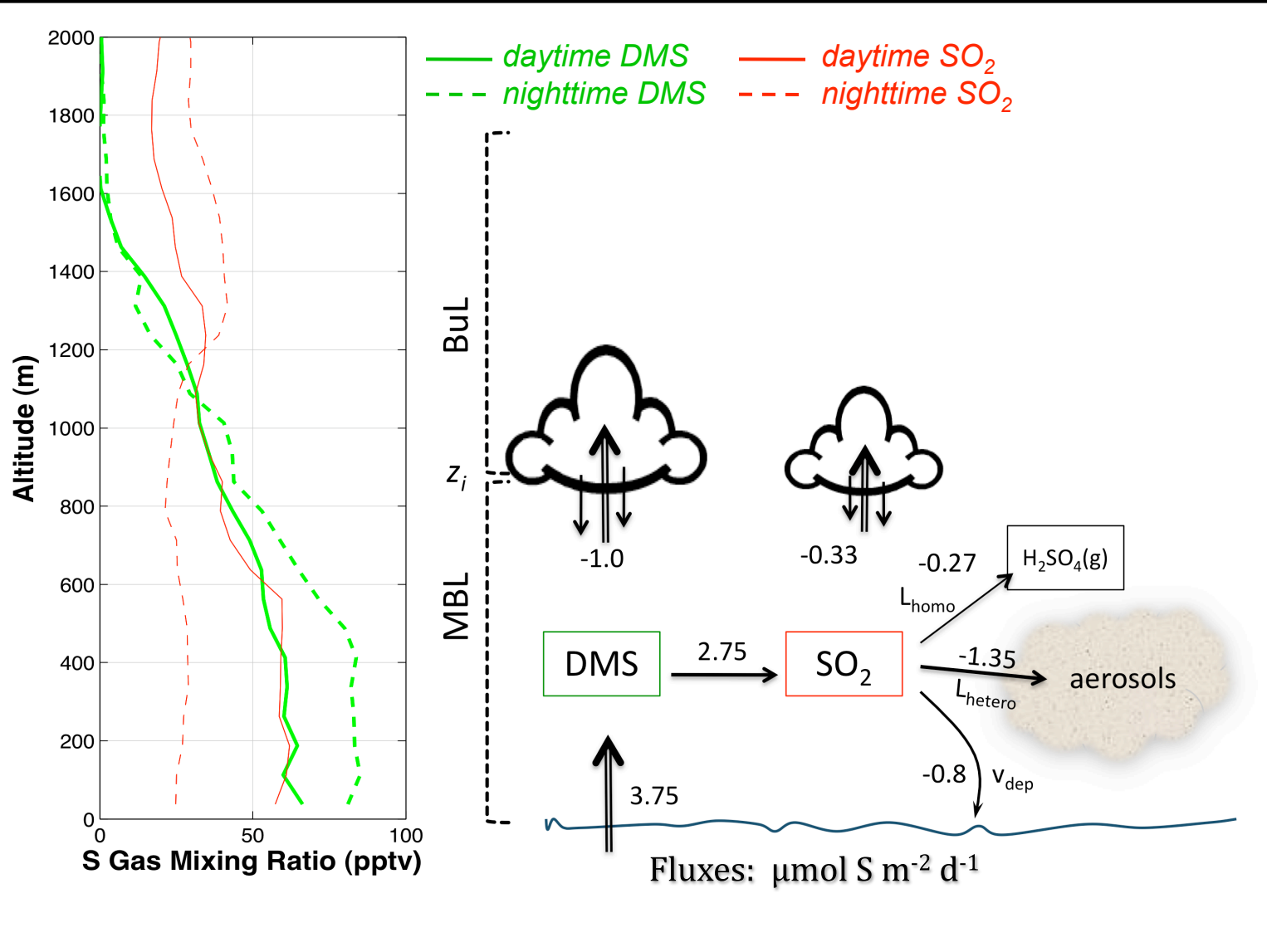
$$\frac{\partial[SO_2]}{\partial t} = -\frac{\partial}{\partial z} \langle w'[SO_2]' \rangle - \vec{U} \cdot \nabla[SO_2] + P_{DMS} - L_{chem}$$

Flight (RF)	SO <sub>2</sub> Trend (ppt/hr)	Flux Div. (ppt/hr)	Advection (ppt/hr)	DMS decay (ppt/hr)	Net Loss (ppt/hr)	Aerosol Loss (h <sup>-1</sup> )	V <sub>dep</sub> (cm/s)	[SO <sub>2</sub> ] (ppt)	Net Error (ppt/hr)	Surface Area (μm <sup>2</sup> /cm <sup>3</sup> )	Aerosol pH
1	0.00	-2.19	0.11	N/A	N/A		0.21	63.8	0.15		
2	0.76	-3.14	0.51	7.36	-3.97	-0.058	0.37	54.9	0.20	94.6	7.0
3	4.50	-1.81	-0.45	7.43	-0.67	0.002	0.31	55.2	0.25	98.1	N/A
5	4.96	-3.14	-0.38	11.23	-2.75	-0.033	0.43	57.4	0.48	90.8	6.9
6	-4.35	-2.10	-0.50	0.00	-1.75	-0.065	0.53	26.9		115.7	6.9
7	1.20	-3.23	-1.77	10.47	-4.27	-0.046	0.56	69.8	1.06	90.1	6.9
8	2.77	-0.46	1.81	5.88	-4.47	-0.095	0.33	40.7	0.21	58.8	7.2
11	2.60	-3.57	3.36	6.78	-3.96	-0.052	0.53	59.0	0.34	48.1	7.1
12	3.41	-0.63	0.73	7.39	-4.08	-0.078	0.35	43.9	0.23	87.3	7.0
13	-1.08	-0.59	0.63	0.00	-1.12	-0.040	0.29	28.1		121.0	6.8
14	5.85	-2.21	-2.06	11.29	-1.17	-0.002	0.46	68.1	0.33	91.9	6.4
<b>Day Avg</b>	<b>3.3</b>	<b>-2.3</b>	<b>0.2</b>	<b>8.5</b>	<b>-3.2</b>	<b>-0.045</b>	<b>0.42</b>	<b>56.1</b>	<b>÷ 0.39 = 6 d</b>	<b>82</b>	<b>6.9</b>
Std Dev (day)	1.8	1.2	1.8	2.2	1.5	0.034	0.09	10.2		18	0.3
Night Avg	-2.7	-1.3	0.1	0.0	-1.4	-0.052	0.41	27.5		118	6.9
<b>Diel Avg</b>	<b>0.3</b>	<b>-1.8</b>	<b>0.1</b>	<b>4.2</b>	<b>-2.3</b>	<b>-0.049</b>	<b>0.41</b>	<b>41.8</b>		<b>100</b>	<b>6.9</b>

$$P_{DMS} = \gamma \cdot k_{OH+DMS} [OH][DMS] \Rightarrow \gamma \geq 0.95$$

$$L_{chem} = \left\{ k_{OH+SO_2} [OH] + \left( \frac{1}{k_{mt}} + \frac{1}{k_{O_3} + k_{H_2O_2}} \right)^{-1} \right\} [SO_2]$$

# Diel Pattern of S Processing





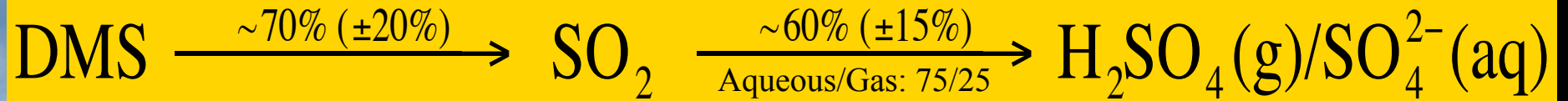
# Previous Estimates of $L_{\text{hetero}}$

**Table 4. Estimates of S(IV) Oxidation in Marine Boundary Layer**

Sulfur Uptake/Oxidation ( $\text{h}^{-1}$ )	Notes	Source
0.12 - 0.20	pH = 7	<i>Sievering et al. [1991]</i>
0.16	uptake limit	<i>Sievering et al. [1992]</i>
0.08 - 0.68	model	<i>Suhre et al. [1995]</i>
0.05 - 0.35	model (10 m)	<i>Mari et al. [1999]</i>
0.04	model	<i>Chen et al. [2000]</i>
0.06	NSS-SO <sub>4</sub> budget	<i>Huebert et al. [1996]</i>
0.02	SO <sub>2</sub> budget	<i>Bandy et al. [1996]</i>
0.02 - 0.04	NSS-SO <sub>4</sub> budget	<i>DeBruyn et al. [1998]</i>
0.015	SSA only	<i>Luria &amp; Sievering [1991]</i>
0.03	unconstrained	<i>Mihalopoulos et al. [2007]</i>
0.11	Approx. Average	<i>All Above</i>

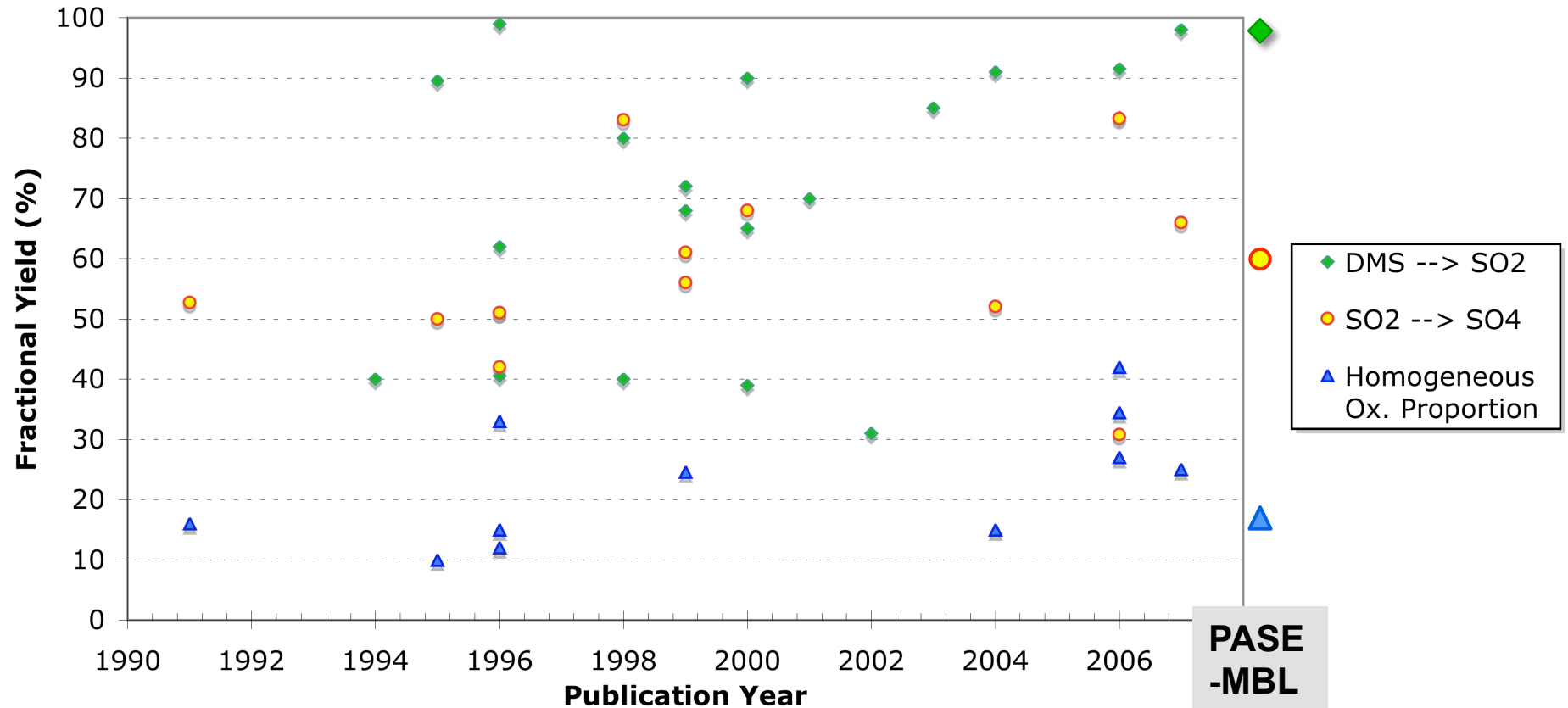
PASE Results:  $\sim 0$  to  $0.095 \text{ h}^{-1}$ , Average:  **$0.045 \text{ h}^{-1}$**

# Comparing to Consensus

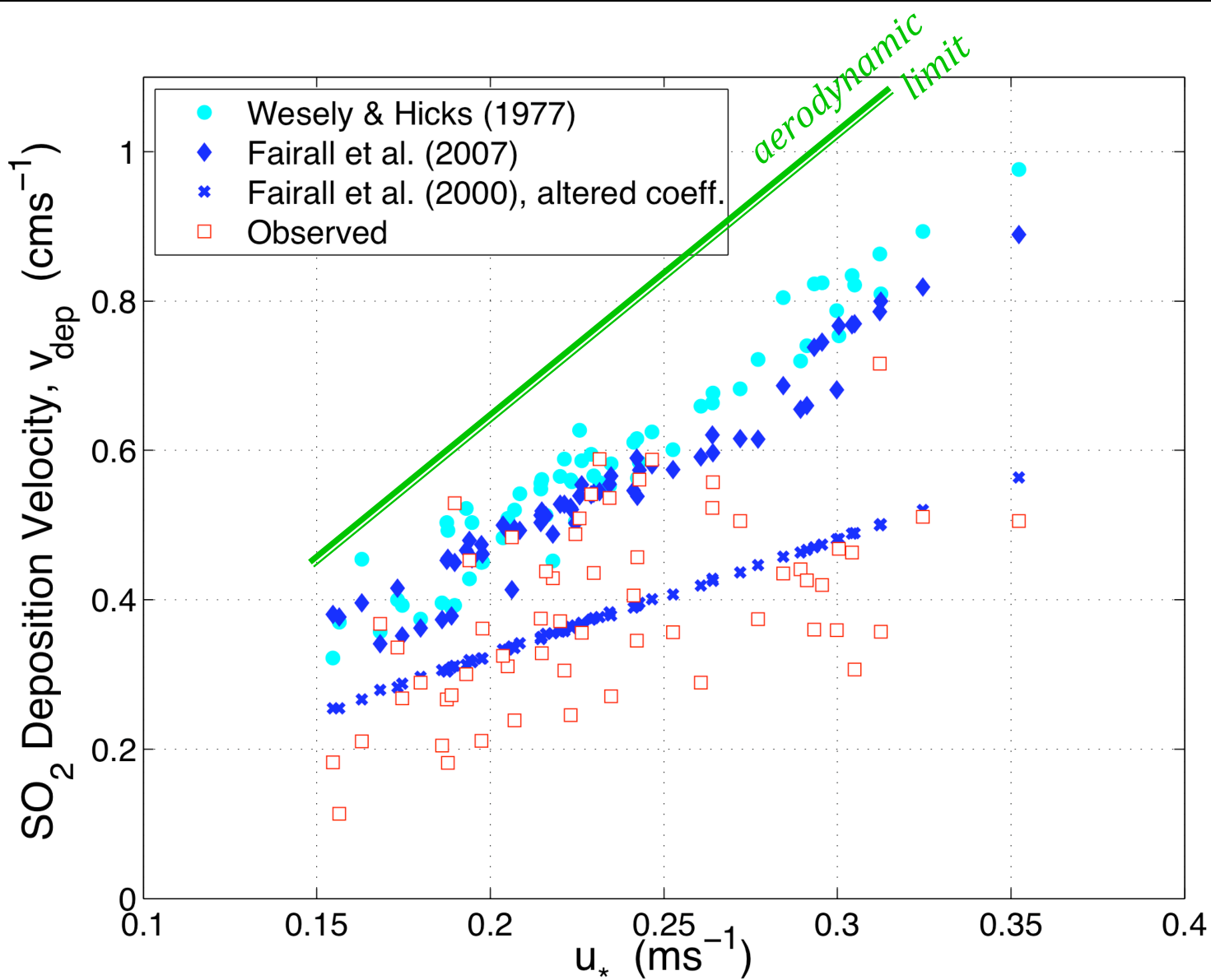


Number of refs:	19	13	11	
AVERAGES:	70.3	57.4	23.1	76.9
Standard Deviation:	22.4	14.9	10.4	10.4

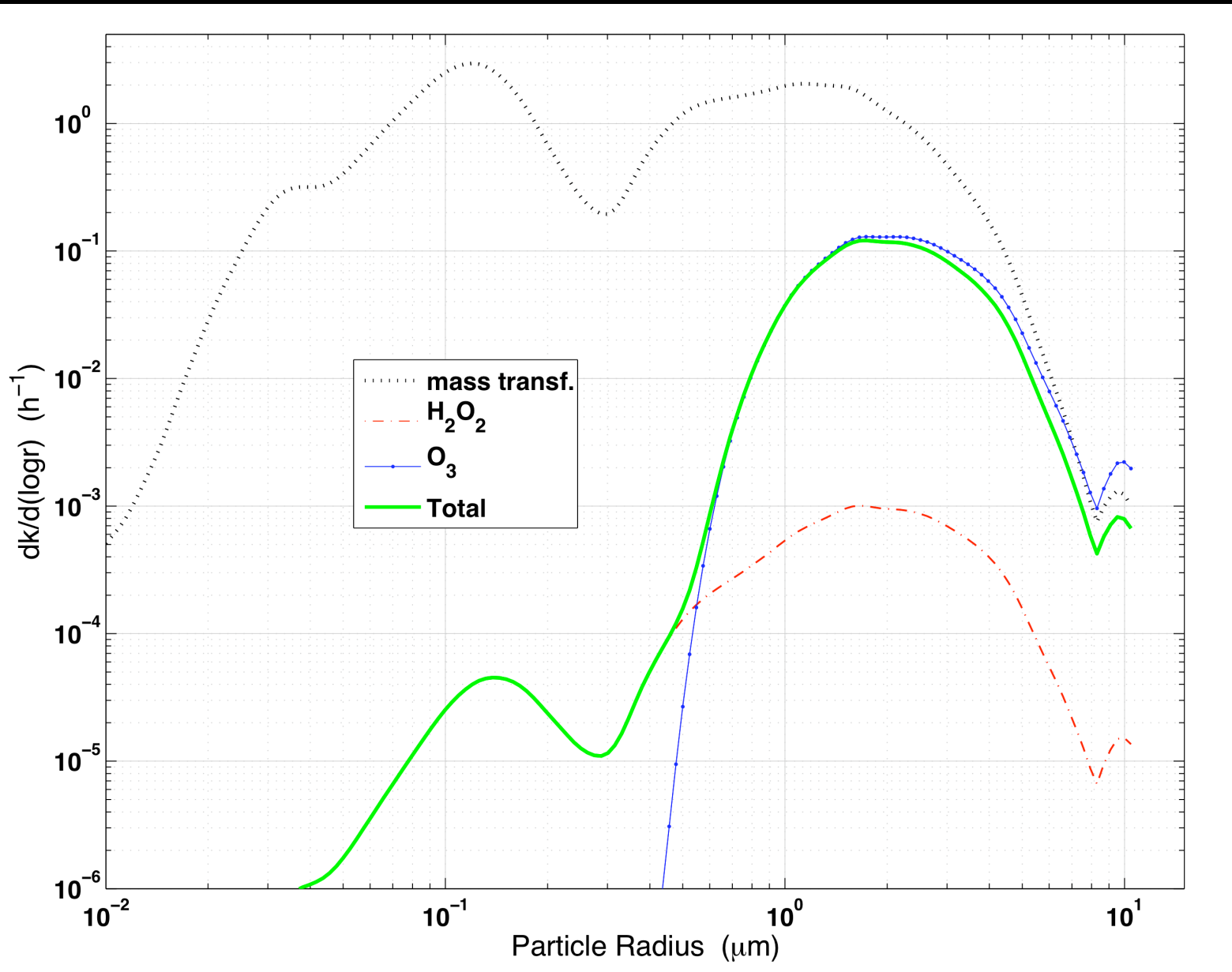
## Branching Ratios in Atmospheric Sulfur Oxidation



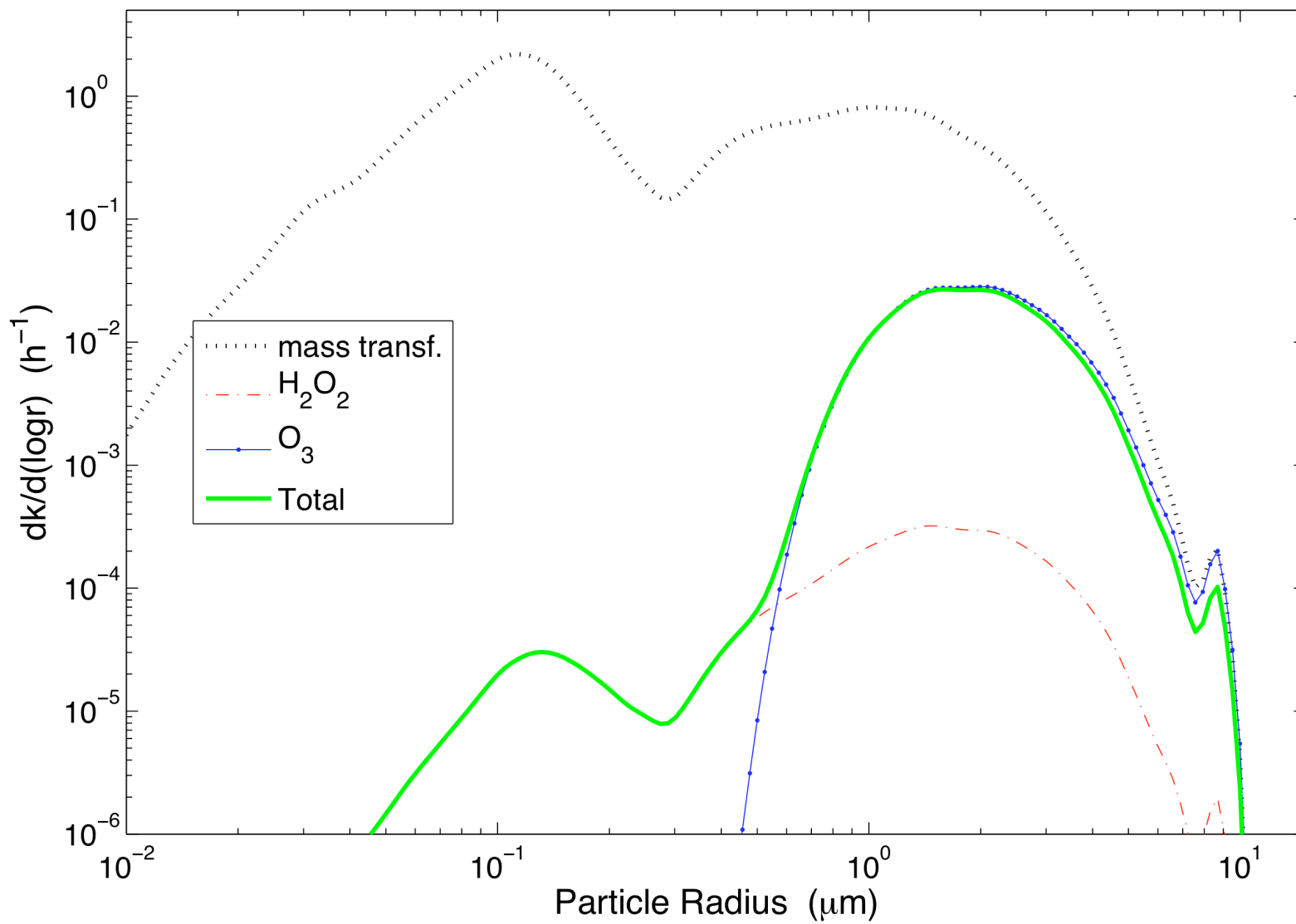
# Dry Deposition vs. Friction Velocity



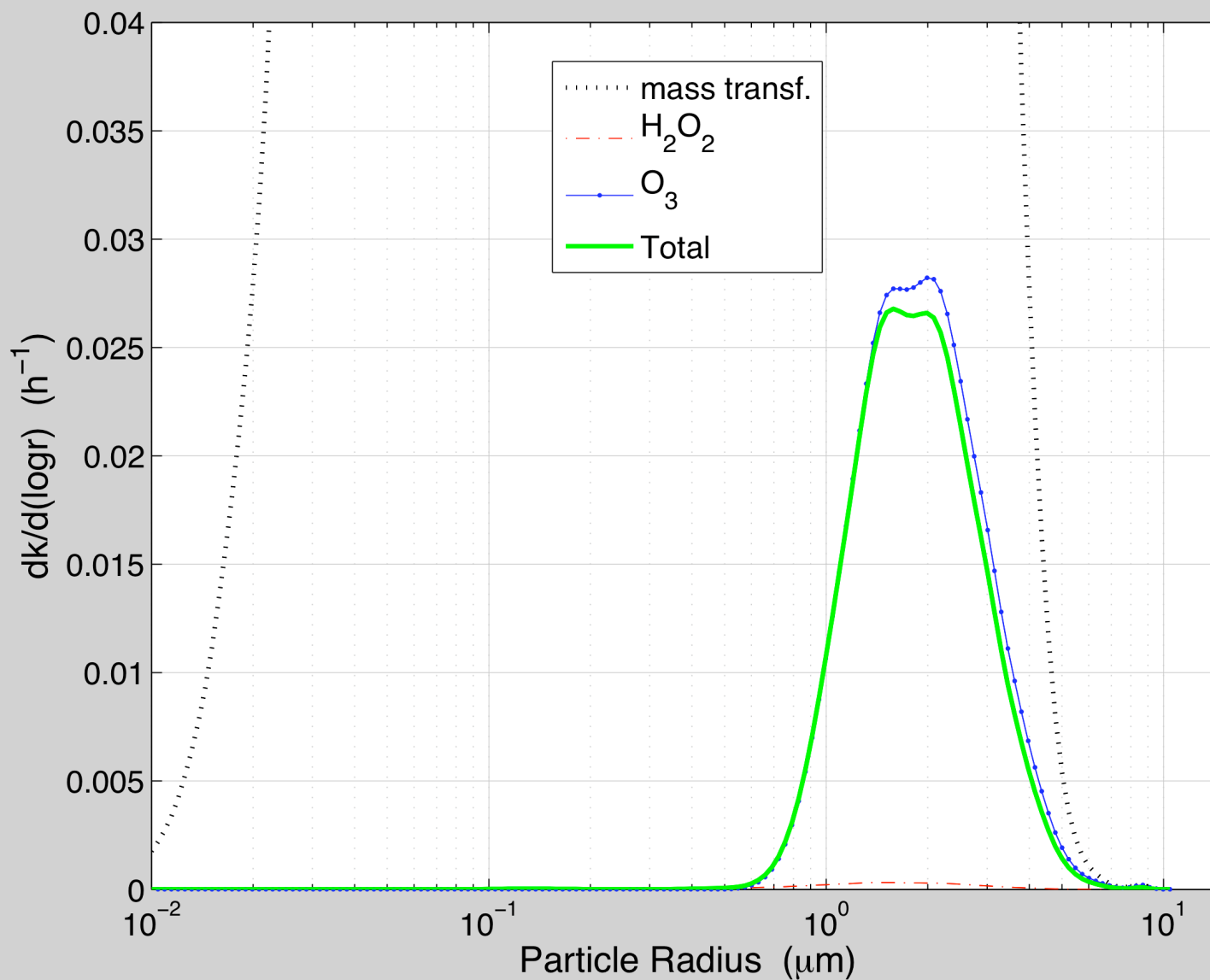
# Heterogeneous Processing: RF13



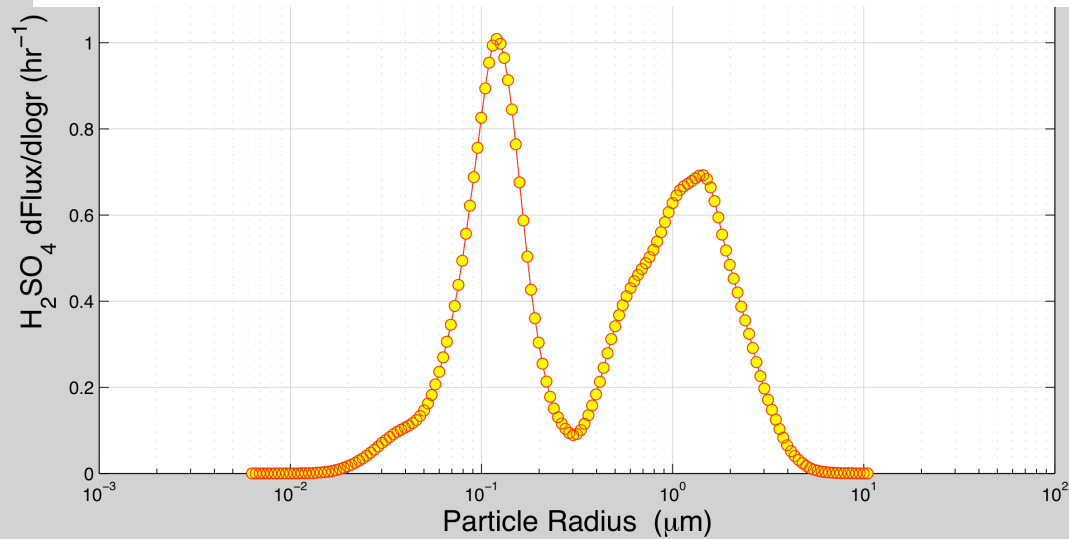
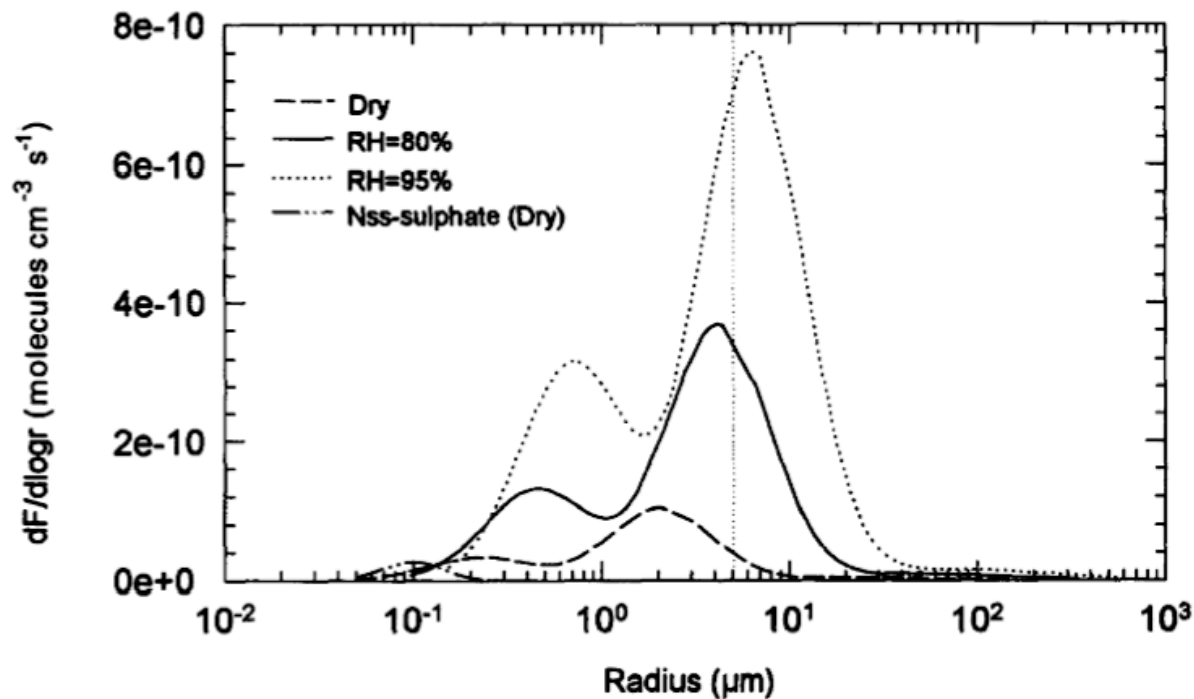
# Heterogeneous Processing: RF08



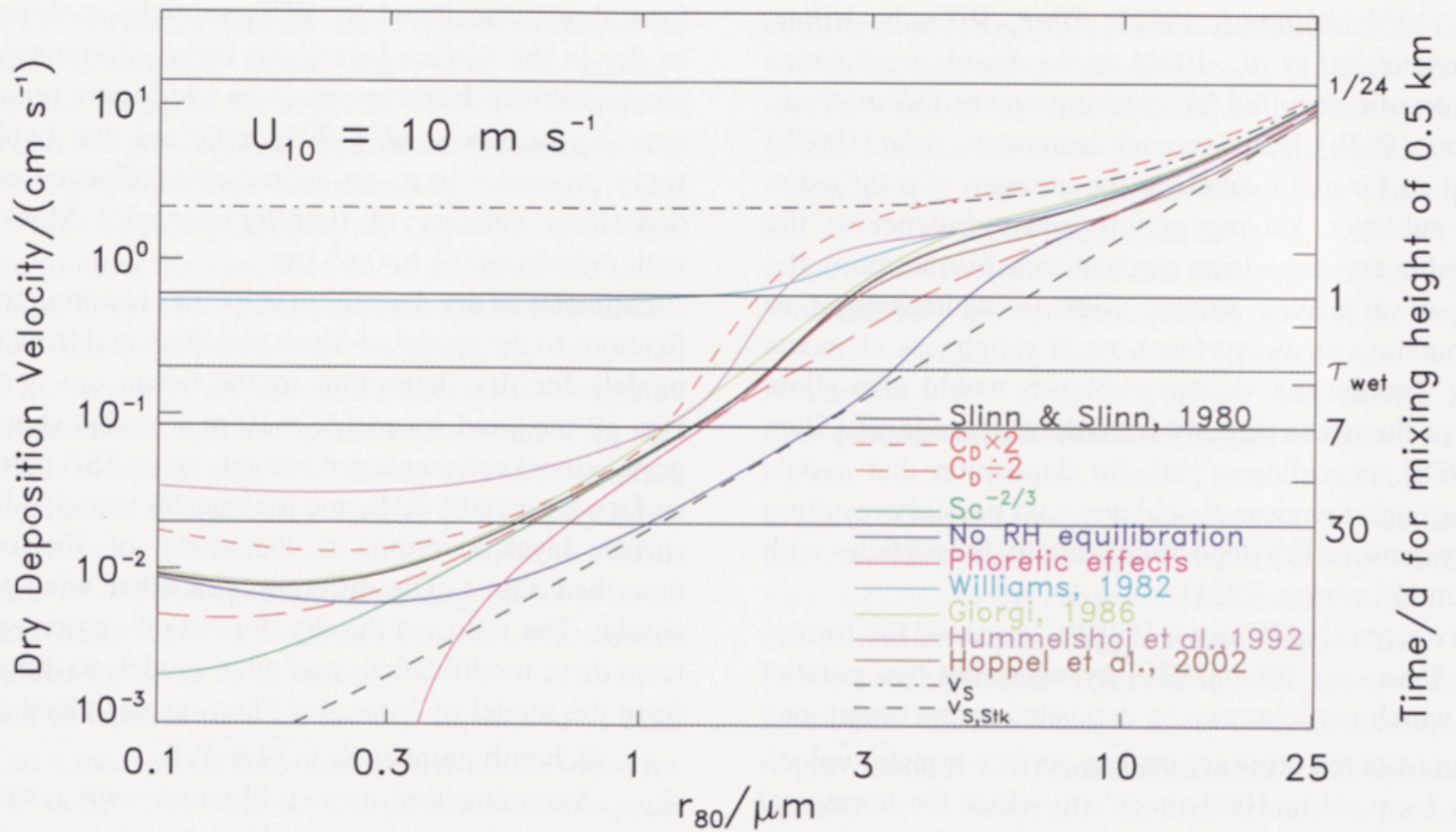
# Bulk of S(IV) oxidation on coarse mode



# H<sub>2</sub>SO<sub>4</sub> Estimated Uptake: O'Dowd & RF13



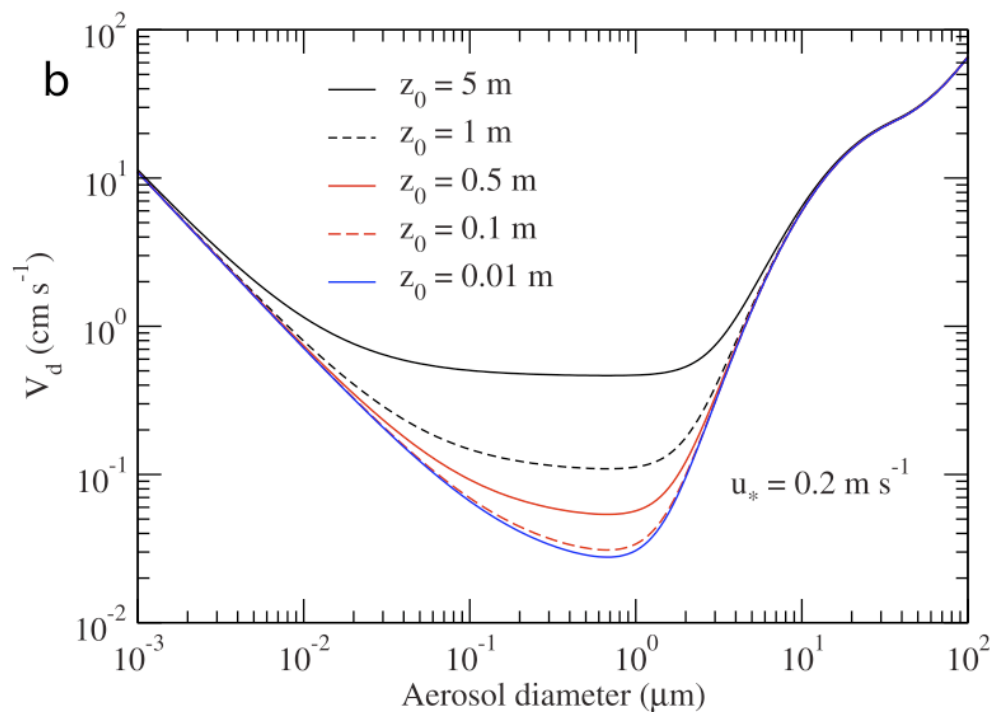
# Aerosol Dry Deposition



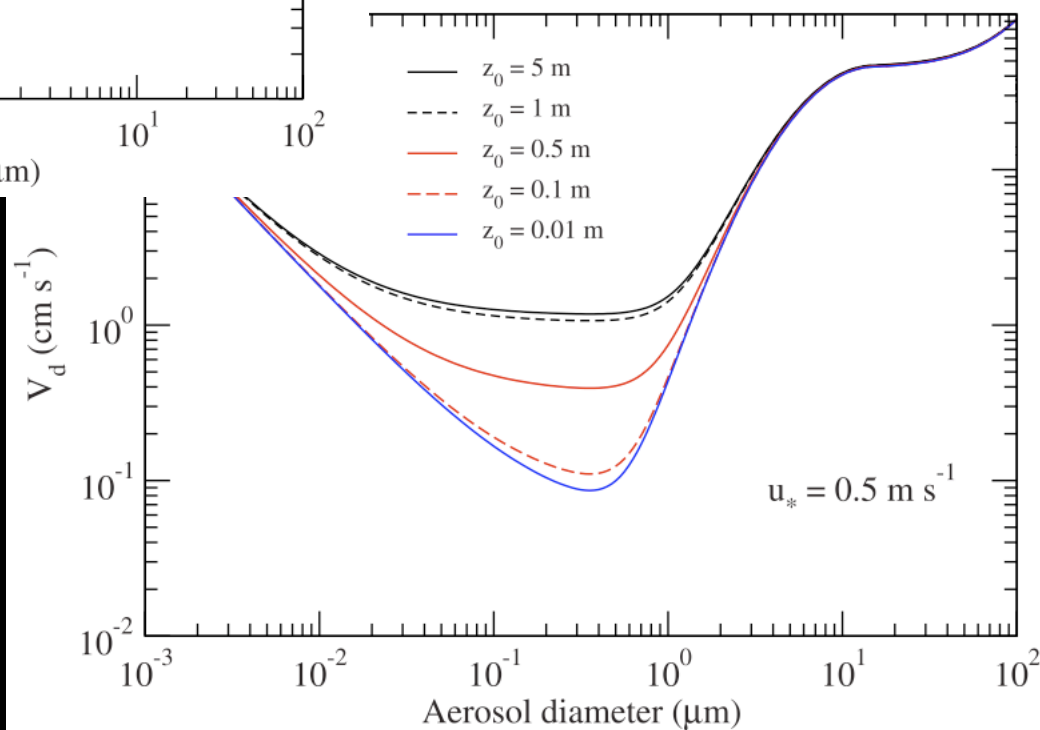
[Lewis & Schwartz, 2004]



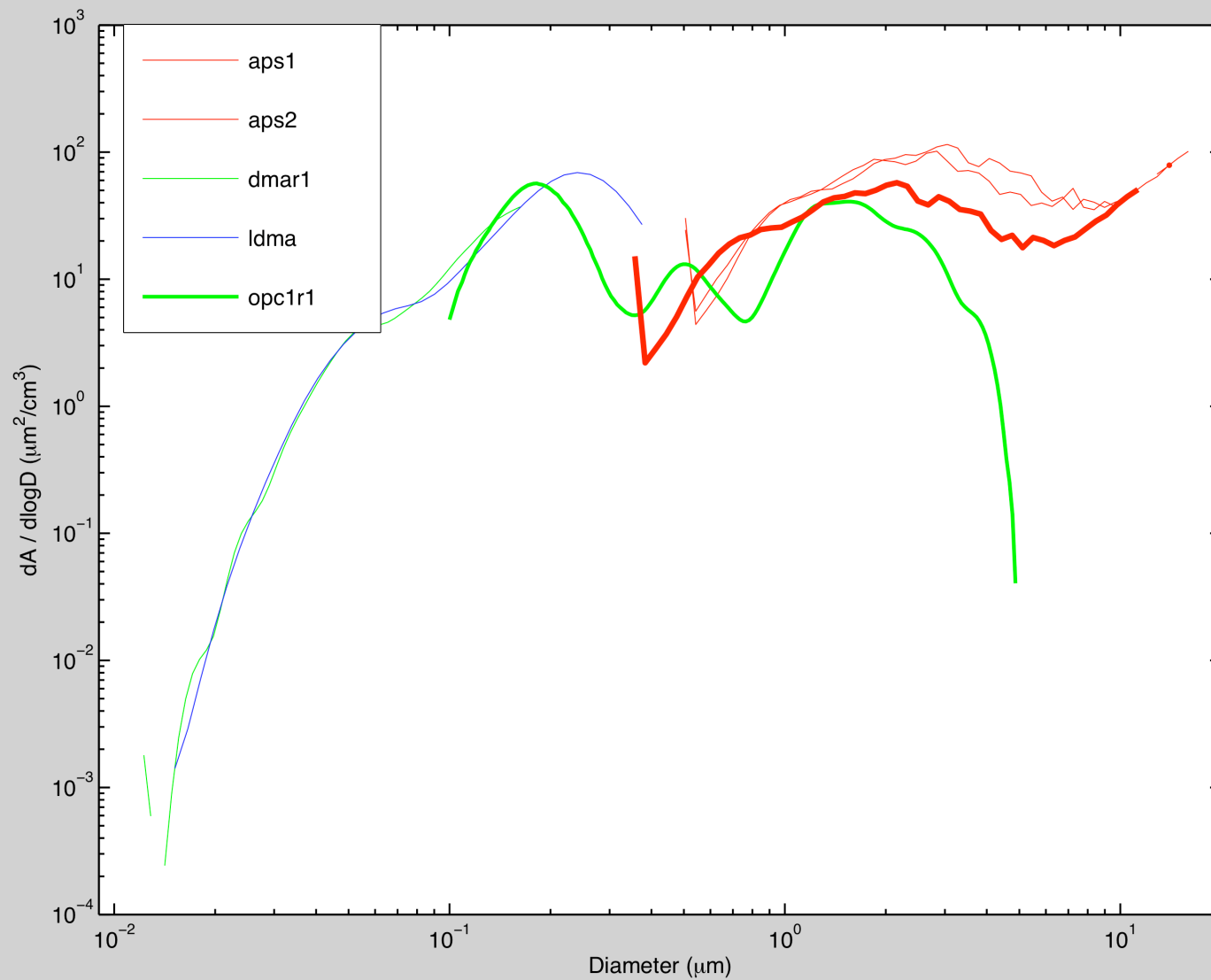
# Recent turbophoresis parameterization



[Feng, 2008]

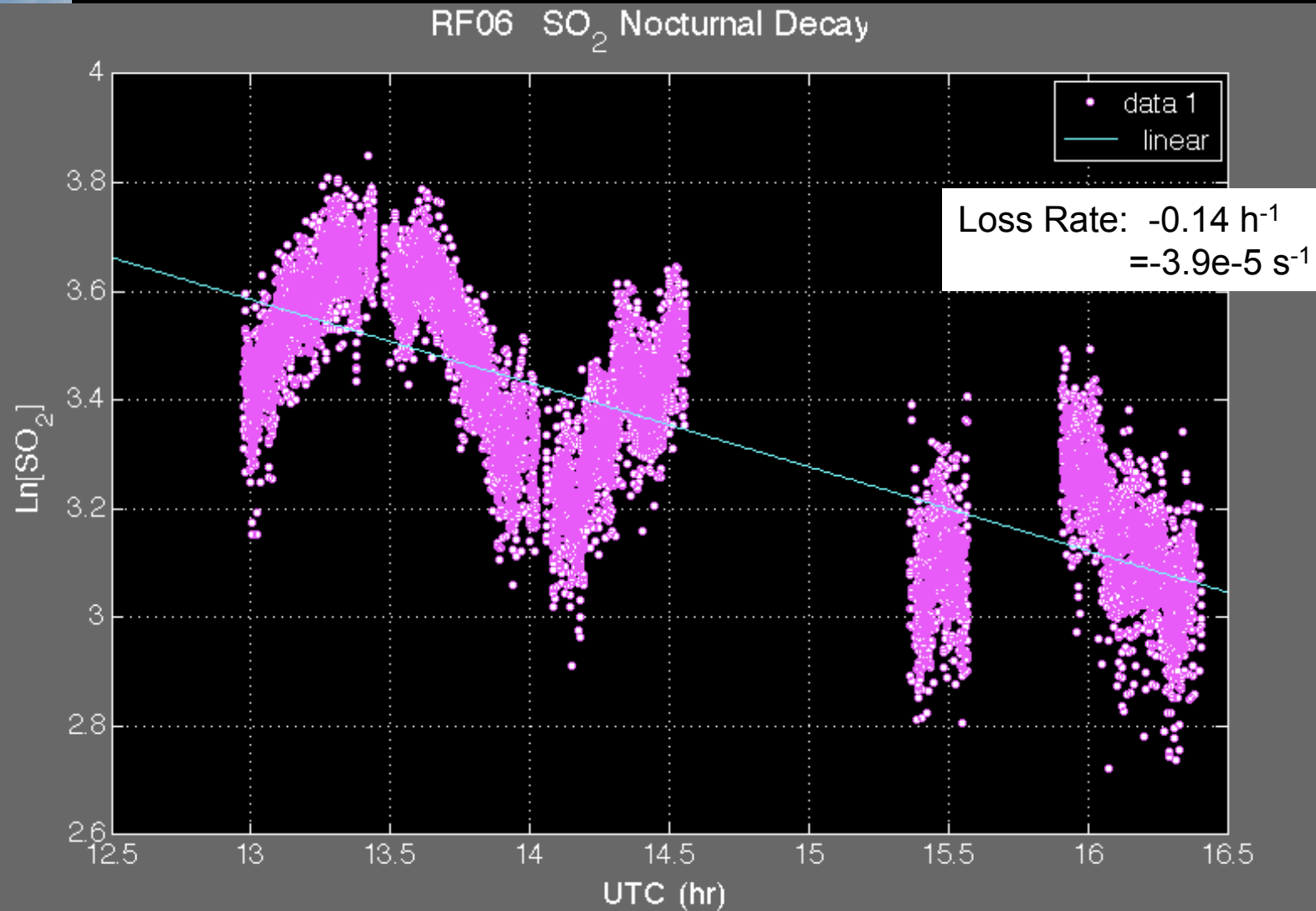


# Larger aerosols?

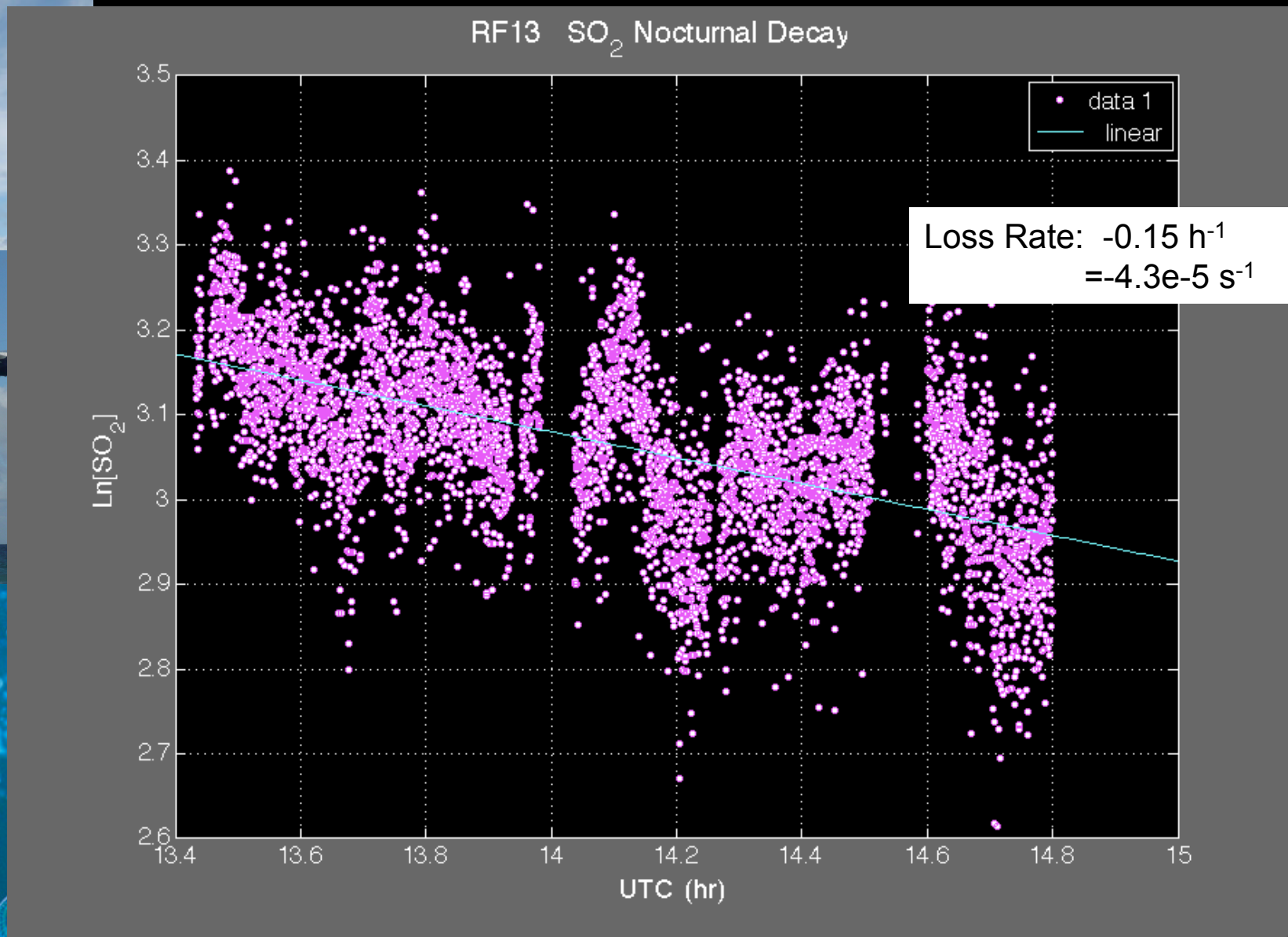


unnamed.fig, browseall.m, Yohei, 2008-02-27

# Inferring Heterogeneous Uptake



# Heterogeneous Loss on RF13



# Huebert's Complaint

GEOPHYSICAL RESEARCH LETTERS, VOL. 23, NO. 7, PAGES 737-740, APRIL 1, 1996

## Production and loss of methanesulfonate and non-sea salt sulfate in the equatorial Pacific marine boundary layer

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**Abstract.** We measured the concentrations of aerosol methanesulfonate (MSA) and non-sea salt sulfate (NSS) in the remote Pacific marine boundary layer (MBL) at Christmas Island (157° W, 2° N) in July and August of 1994. The project-average MSA displayed a distinct diurnal variation, decreasing to 8.6 ppt at sunrise and increasing to 12.1 ppt by sunset. The average NSS diurnal variation ranged from 196 ppt at sunrise to 235 ppt at sunset. Large-particle dry deposition may account for 10-20% of the observed nighttime decrease, with entrainment of cleaner free tropospheric air responsible for the rest. The entrainment velocity inferred from the nighttime decrease averaged  $0.5 \pm 0.2$  cm/s. A simple model suggests that NSS and MSA were produced at rates of about 74 and 6 ppt per day, respectively. Between 30 and 40 % of the daily dimethylsulfide (DMS) flux forms NSS and 3 % forms MSA.

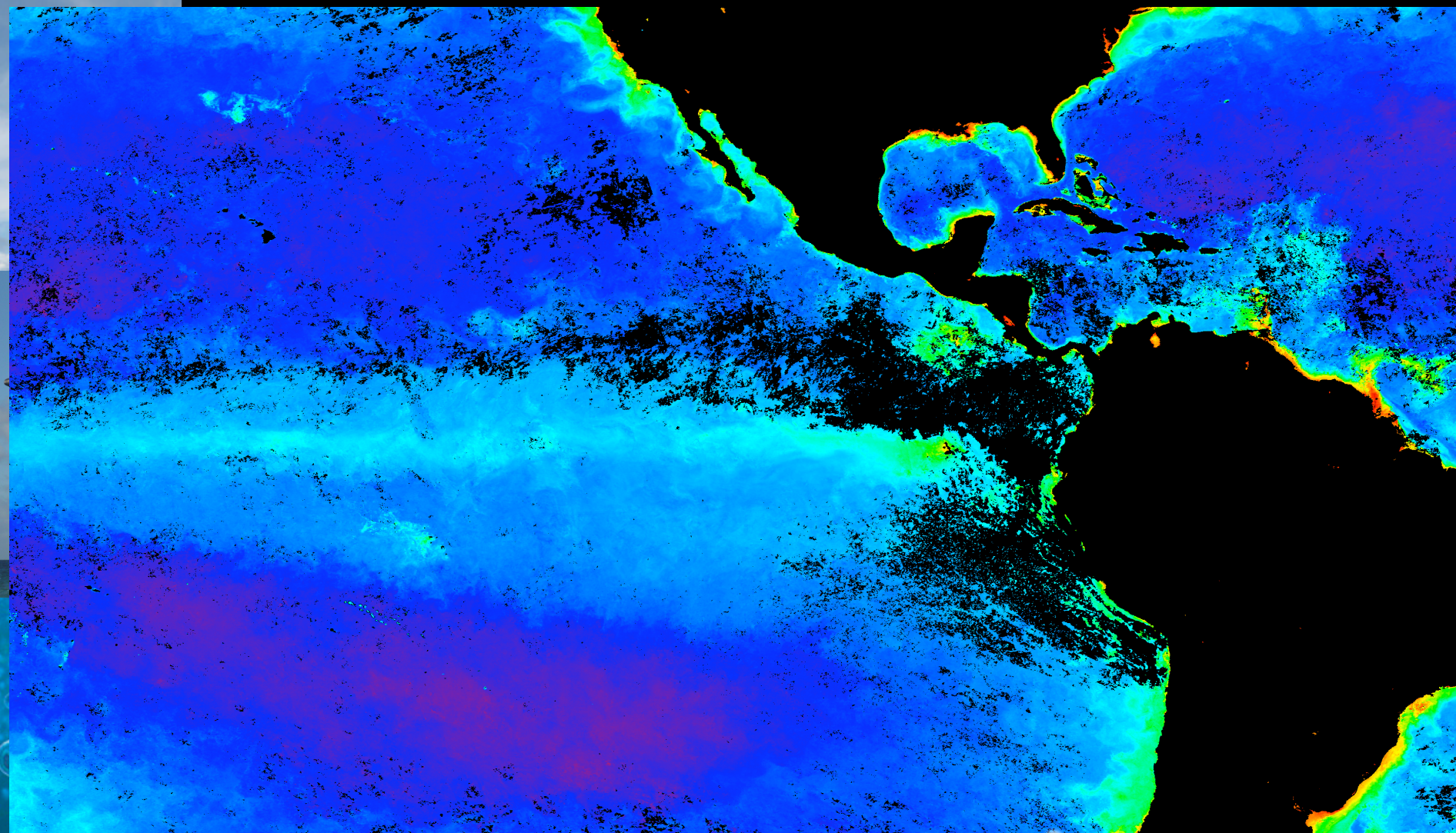
(157° W, 2° N), Republic of Kiribati. For most of the experiment our aerosol samplers were positioned atop a 30 m aluminum walkup tower situated about 10 m from the windward Pacific shoreline. We measured bulk concentrations of sulfate, MSA, sodium, and ammonium using open-faced 47 mm Teflon filters (Gelman Zefluor, 1  $\mu$ m pore size). Samples were extracted and analyzed by ion chromatography [Huebert *et al.*, 1995]. The chemical size distribution of aerosols was measured with a pair of MOUDI impactors [Marple *et al.*, 1991]. To assess the depth of the MBL and its state of mixing, we launched between 4 and 8 balloons per day from an NCAR CLASS sounding system.

### Results and Discussion

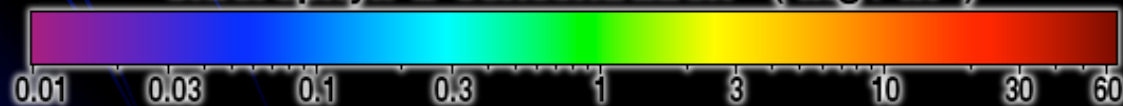
Our discussion will focus on two intensive sampling

PASE Results: ~42% DMS Flux  $\rightarrow$  NSS  
~30 ppt of NSS in 12 h day

# Primary Productivity During PASE (Aug)



Chlorophyll *a* concentration ( $\text{mg} / \text{m}^3$ )



# Tropical Precipitation by Satellite

