

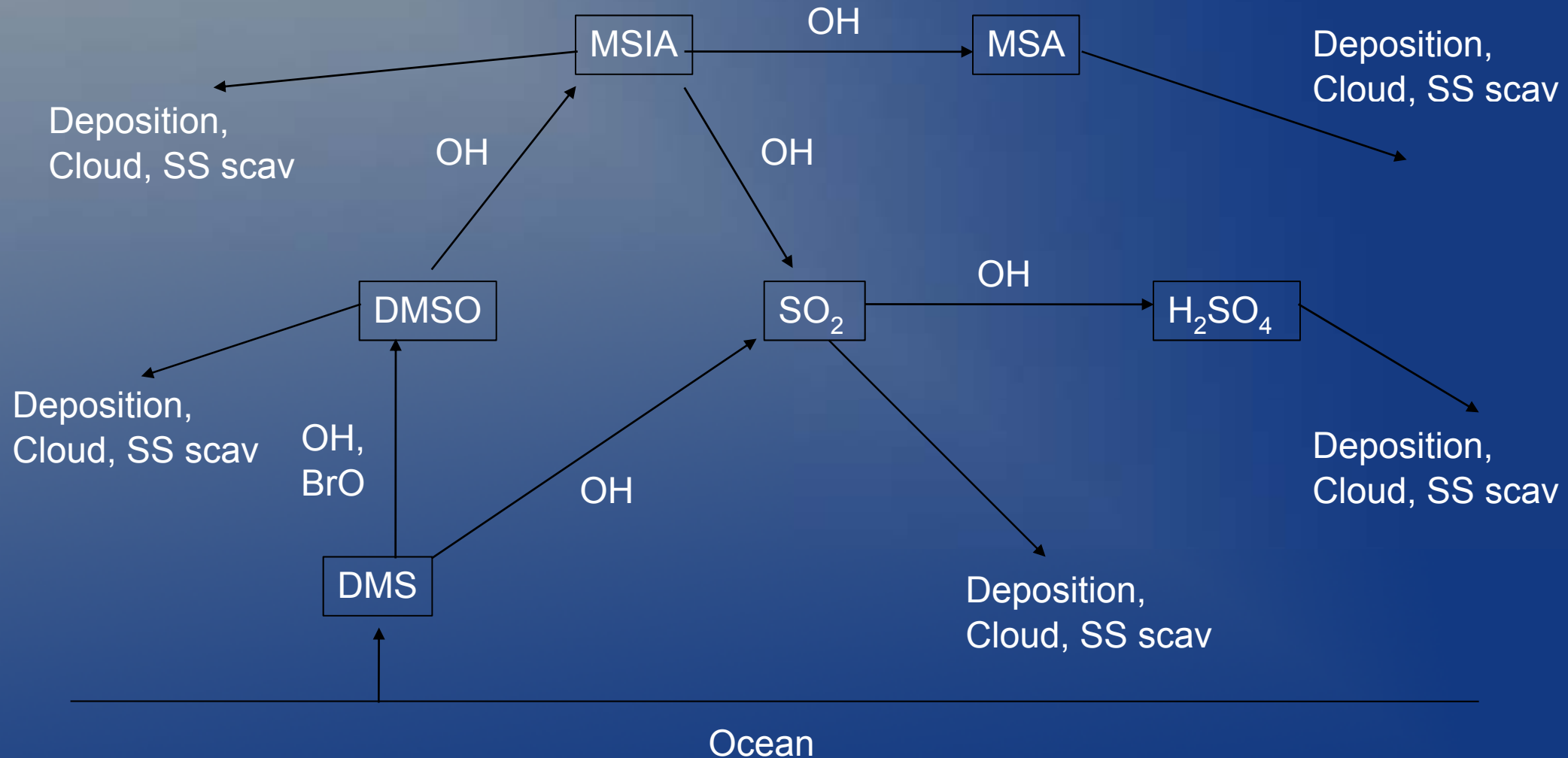
1-D Modeling of DMS and SO₂ Chemistry in the Equatorial Pacific

B. Gray, D. Gu, Y. Wang,
and the PASE Science Team

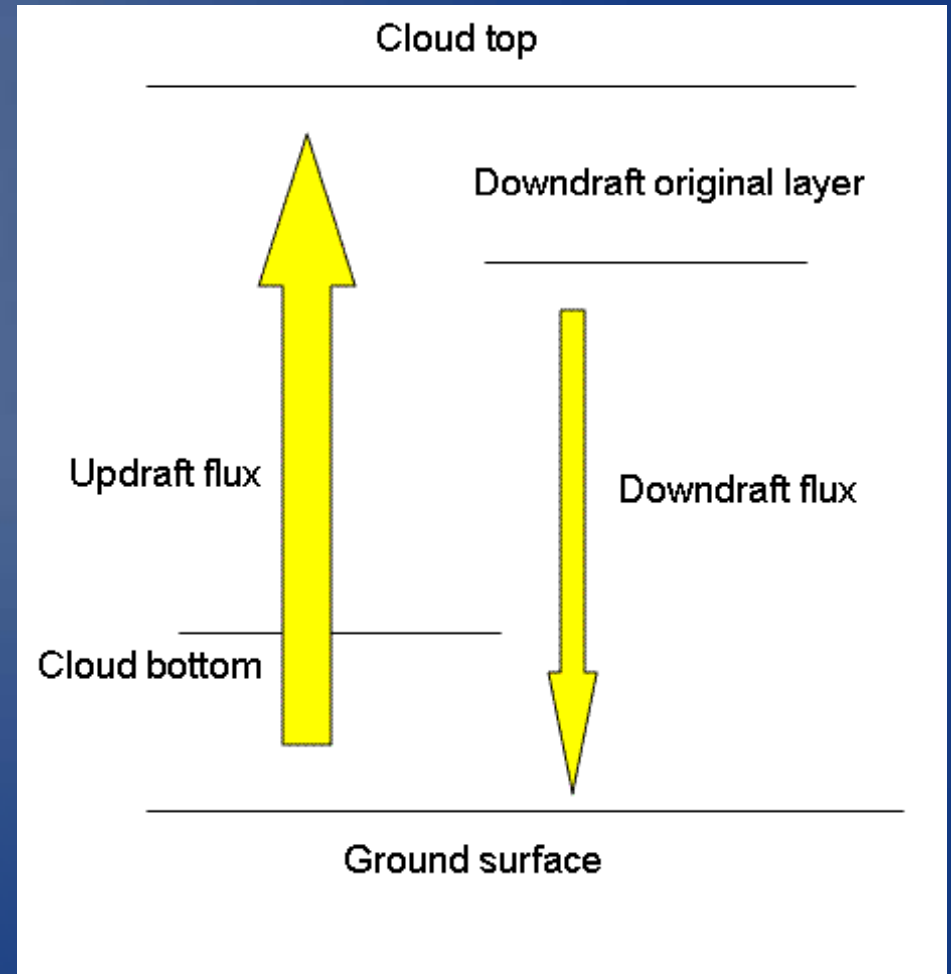
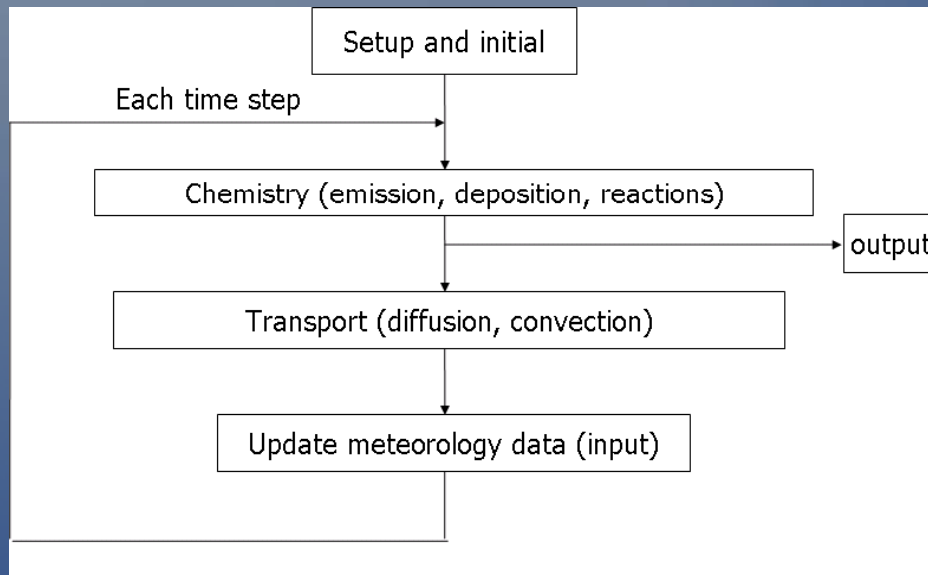
Research Focus

- Prior Studies → Mostly Box Models
 - No convection, diffusion
 - No separation of SO₂ loss to sea-salt scavenging, dry deposition
 - No cloud losses for DMS products
- This Study → 1-D Model
 - Regional chEmical trAnsport Model (REAM) plus WRF meteorological fields
 - PEM Tropics-A and -B data used to constrain diffusion
 - Cloud and sea-salt losses calculated

DMS Chemistry



Model Setup

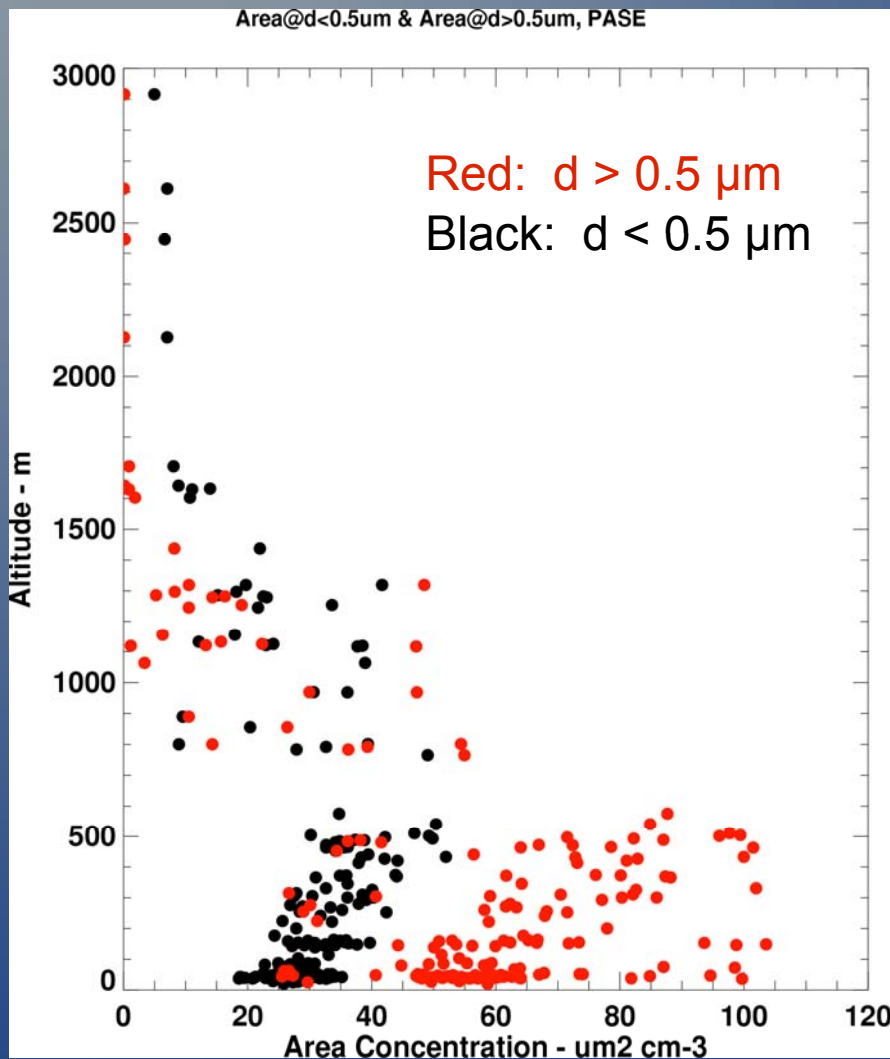


Model constrained with PASE observations of CO and O₃

Modeled Chem and Phys Processes

- Sulfur species: DMS, SO₂, DMSO, MSIA, H₂SO₄, MSA
- DMS concentration set in lowest layer with diurnally varying boundary condition
- DMS loss via OH, BrO
- Chemical prod/loss for other species
- Sinks for soluble species
 - Dry deposition (BL only)
 - Sea-salt scavenging (BL and BuL)
 - Cloud scavenging (BuL only)

Sea-Salt Concentration



- Plot of area concentration for large and small sizes
- Clearly shows three vertical regions
- Size distribution associated with each pair
- Size data used to calculate scav rate

SO₂ Scavenging via Sea Salt

- SO₂ taken up by sea salt, reacts with O₃ in aqueous phase
 - Reaction strongly pH dependent
 - Shuts down at pH ~ 5.5
- Ocean pH buffered by excess alkalinity
 - Sea-salt assumed to have ocean composition
 - pH unaffected by acid addition until excess alkalinity consumed; pH quickly declines after
 - Available alkalinity: 0.07 eq mol per kg
- One mole of SO₂ consumes 2 eq mol alkalinity
- SO₂ loss driven by sea-salt VOLUME flux

SO₂ Scavenging via Sea Salt

$$\frac{dF}{dR} = V_D \square \frac{dN}{dR}$$

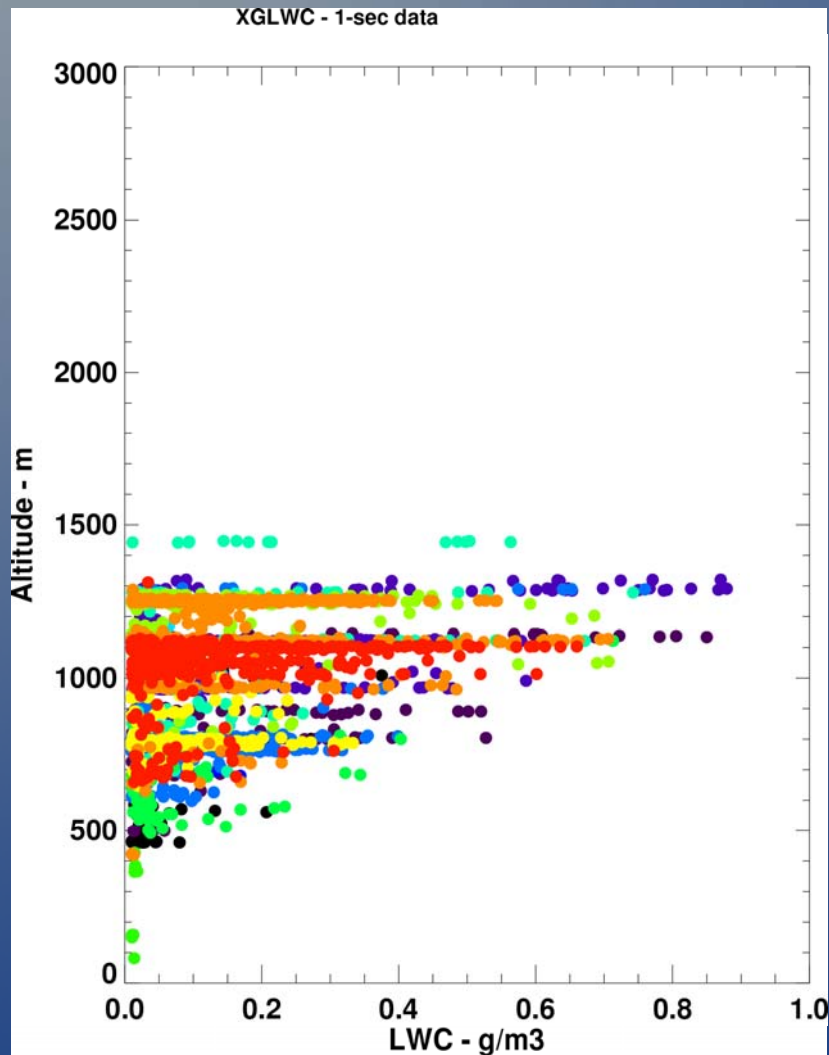
$$\frac{dF_V}{d\log D} = V \square V_D \square \frac{dN}{d\log D}$$

- Convert particle flux eqn to volume flux eqn
- Multiply volume flux by alkalinity concentration
- SO₂ scav rate is half of alkalinity flux
- BL: 10x10¹² molec cm⁻² day⁻¹
 - Global avg: 50 – 200x10¹²

SS Scav for Other Solubles

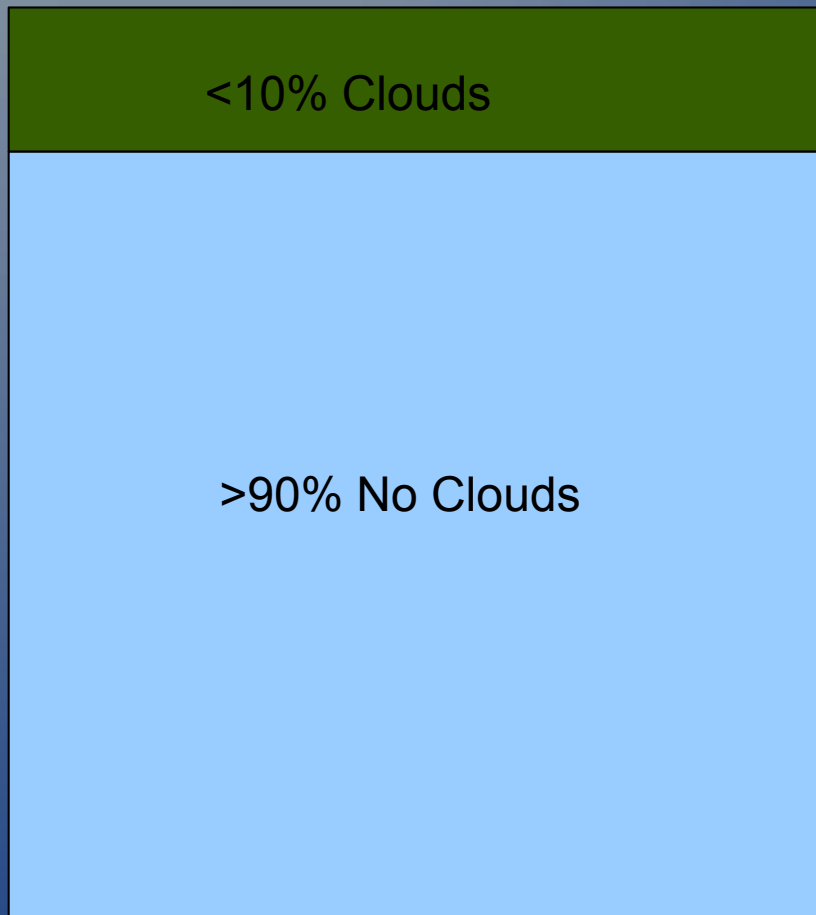
- Other solubles: DMSO, MSIA, H₂SO₄, MSA
- Assumption: Saturation not reached
- These limited by kinetics, not available alkalinity
 - Use Danneke formulation for flux rate to sea-salt surface
 - Loss rate depends on surface area concentration
- BL rates (s⁻¹): H₂SO₄ = 1.4x10⁻³, MSA = 3.8x10⁻⁴

Cloud Scavenging



- Assumption: Soluble species scavenged instantly upon contact with cloud
- From XGLWC sensor, $LWC > 0.01$ g/m³ for $< 10\%$ of BuL
- Cloud scav driven by BuL mixing time

Cloud Scavenging

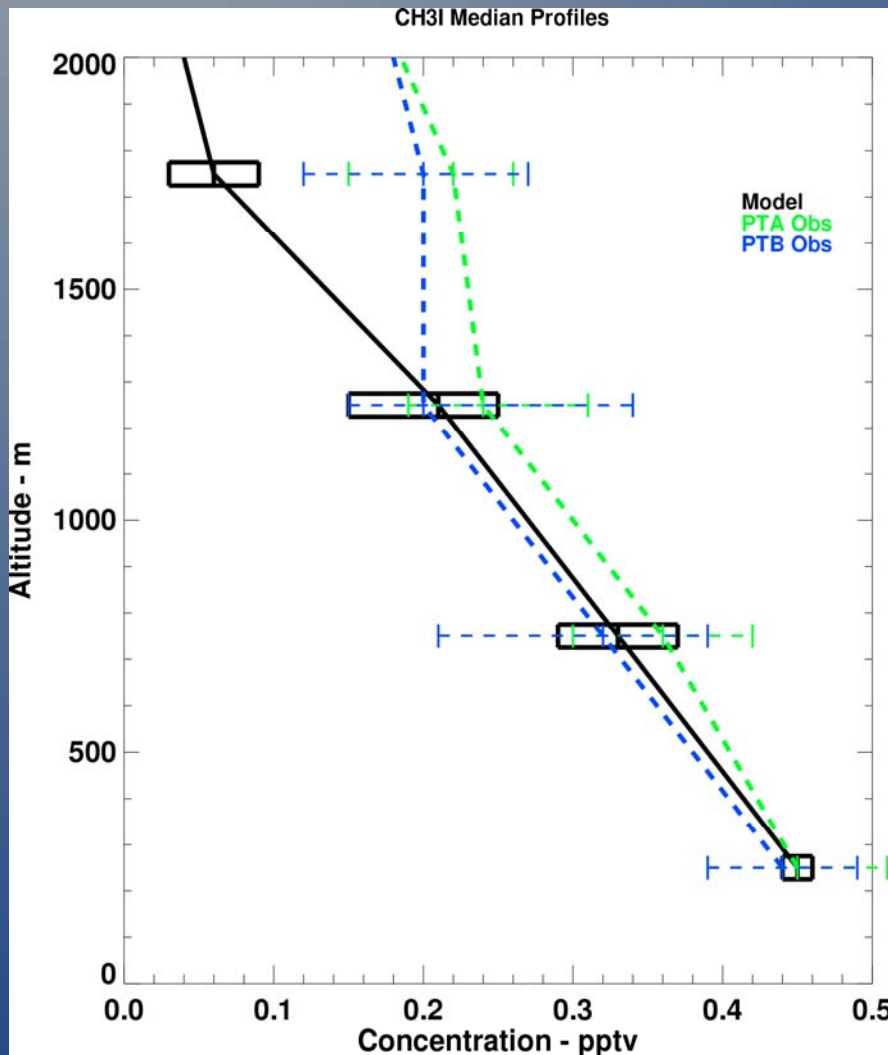


- Mixing timescale within a layer: $\tau = h^2/2K_z$
- If <10% clouds, h will not vary much, so scav rate \sim constant
- $\tau \sim 1.5$ days in BuL

PASE Data

- PASE data used for daytime comparison to model
- One model per each PASE flight used
 - Models constrained with PASE CO, O₃
 - BuL mixing constrained with PEM-TA, TB CH₃I
- Flights included: 2, 3, 5, 8, 9, 12
 - RF01 → No DMS
 - RF04 → Different region
 - RF06, RF13 → Night flights
 - RF07 → Too short

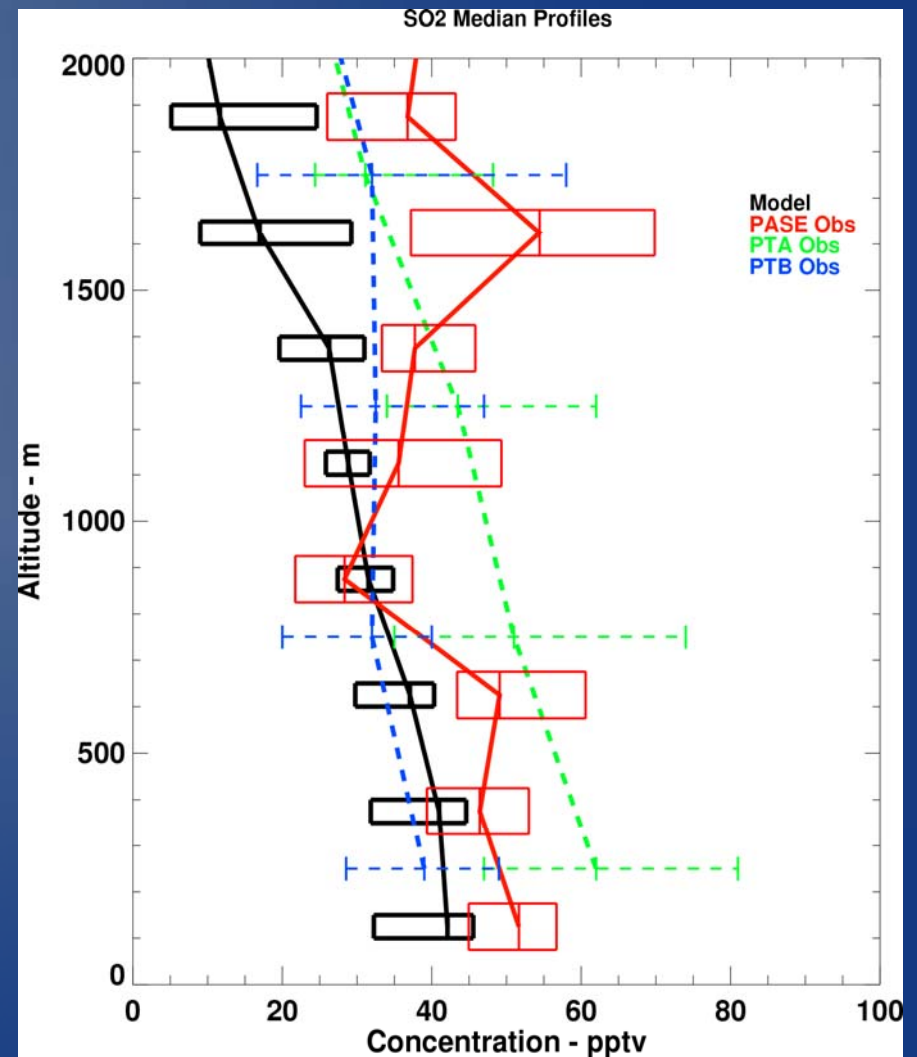
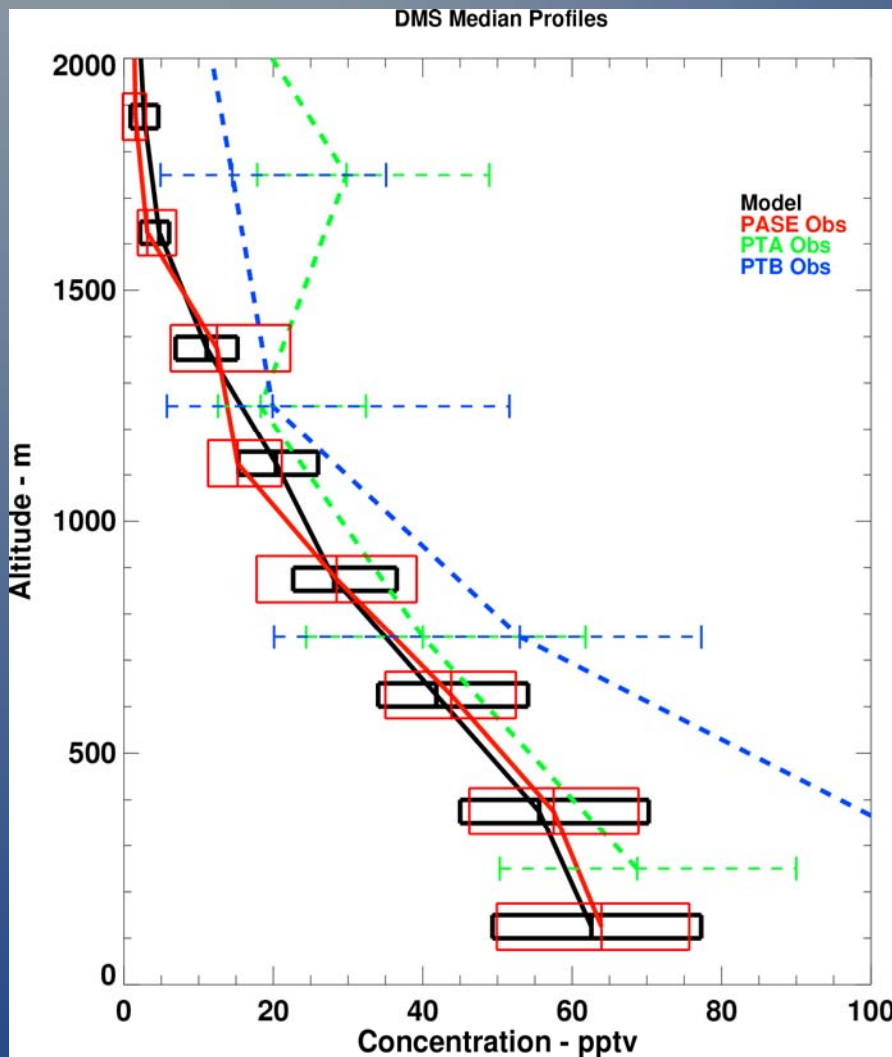
Diffusion, Convection



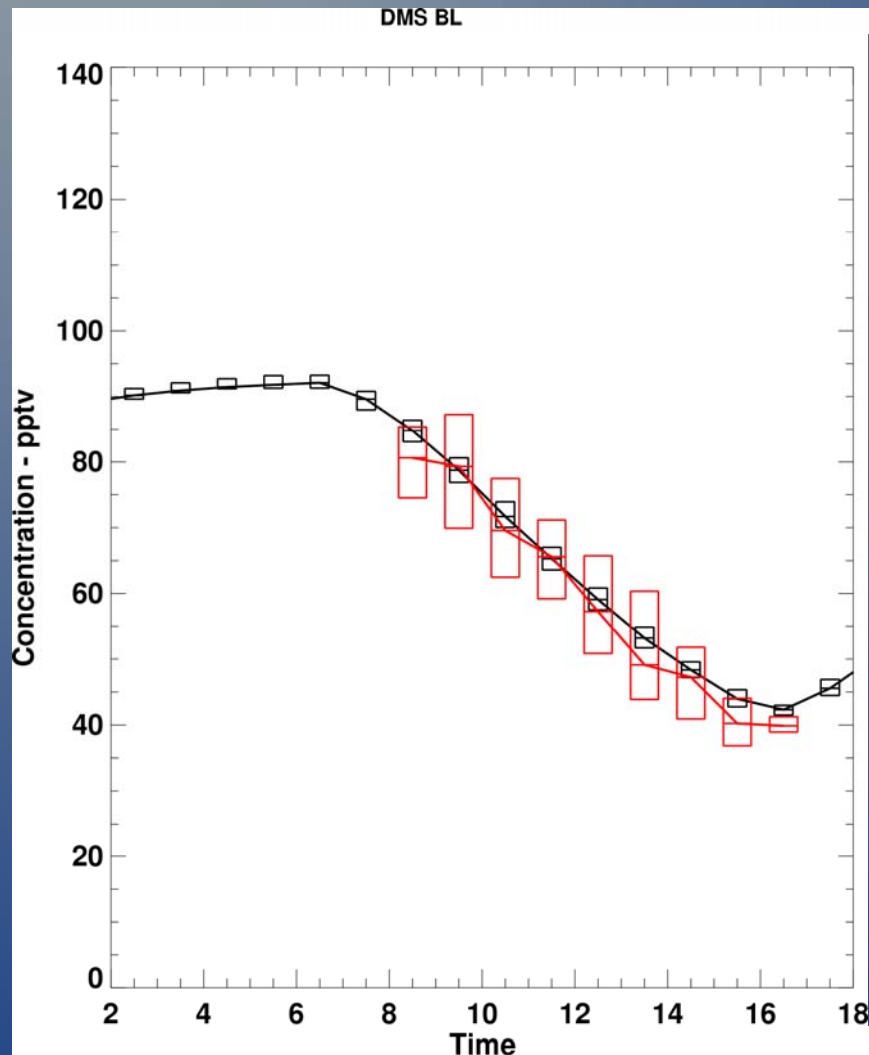
- BuL diffusion determined via CH_3I comparison to PEM Tropics-A, -B
 - BL diffusion determined by WRF
 - BuL $K_z = 3 \text{ m}^2/\text{sec}$
- Shallow convection calculated by WRF
 - Model appears to be low compared with

Results for DMS, SO₂

(using WRF-calculated Shallow Convection)

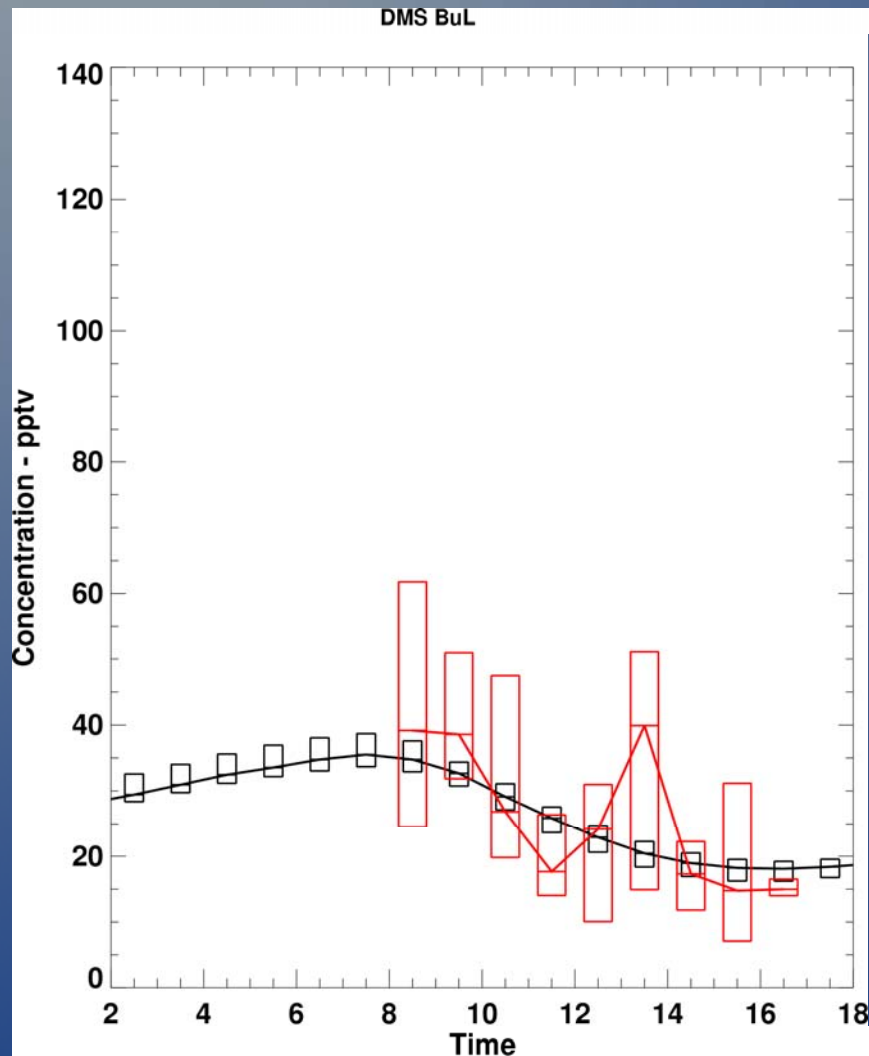


BL DMS



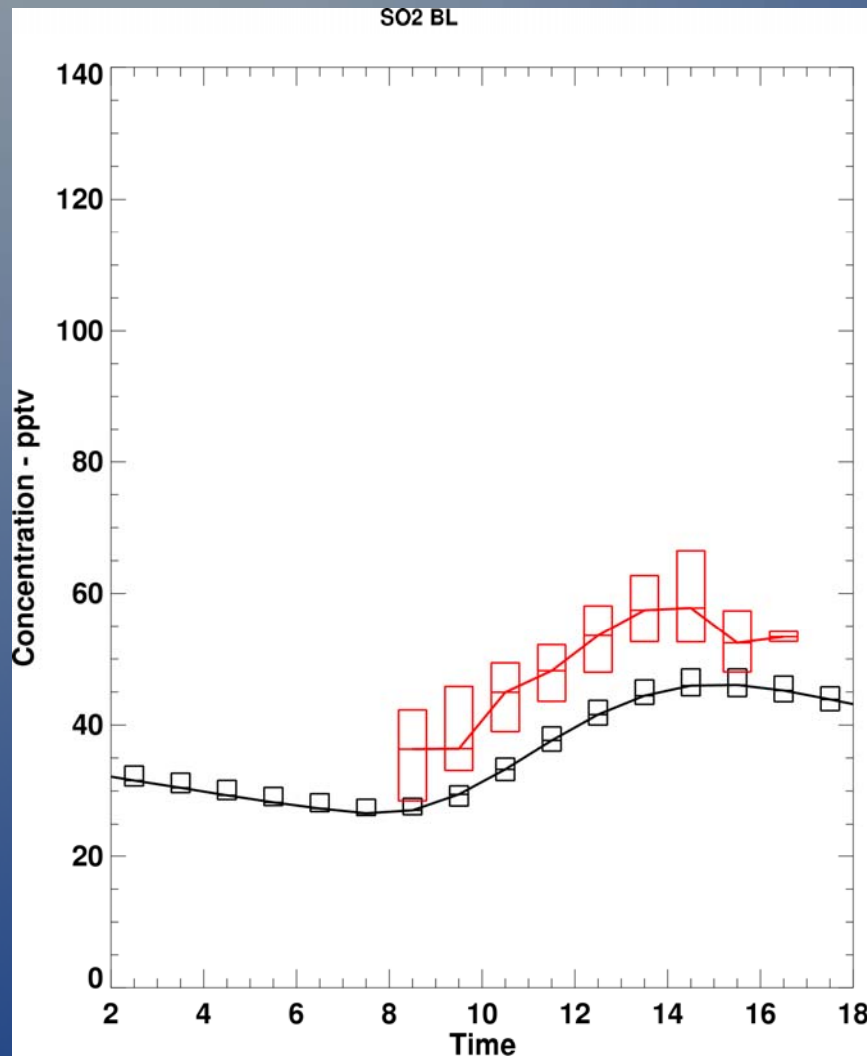
- Six model layers comprise BL
- Value in lowest layer via diurnally varying boundary condition

BuL DMS



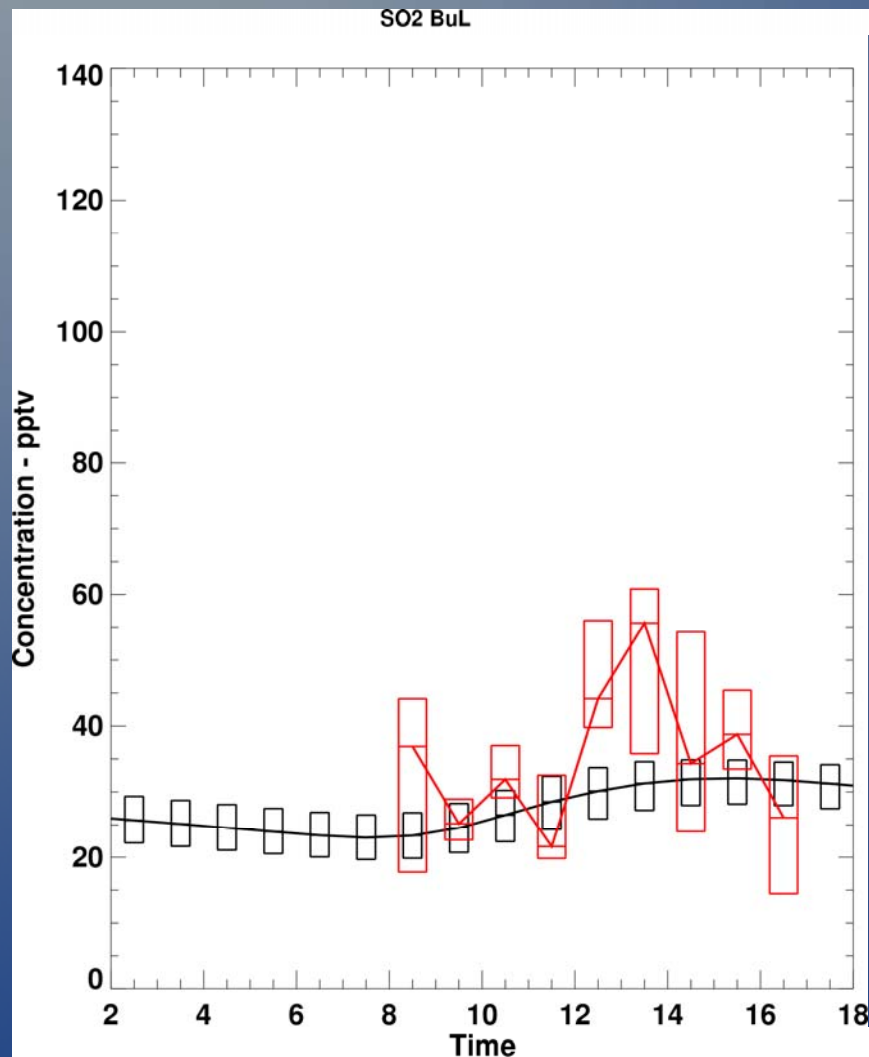
- Model simulates PASE BuL obs reasonably well
- Significant variability in BuL observations

BL SO₂



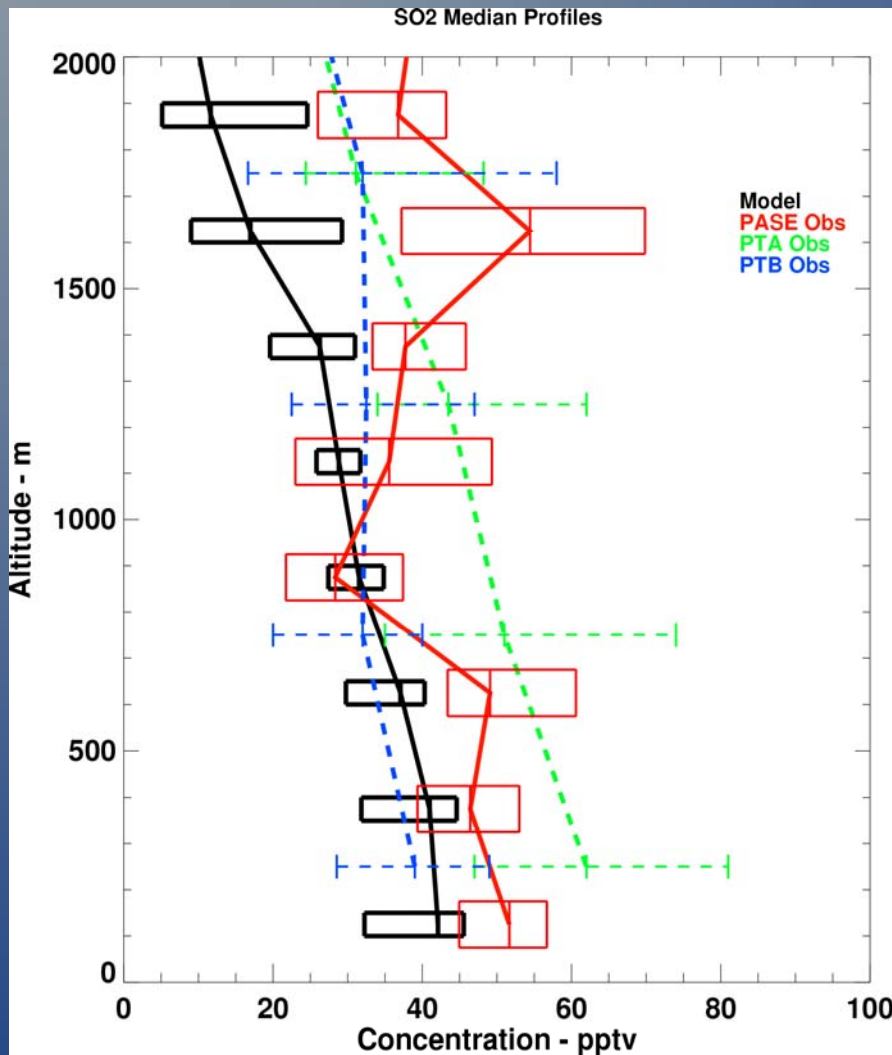
- Model underestimates SO₂ concentration
- Lifetimes:
 - SS Scav: 5 days
 - Chem: 7 days
 - Deposition if V_D is 0.1 cm/s: 6 days

BuL SO₂



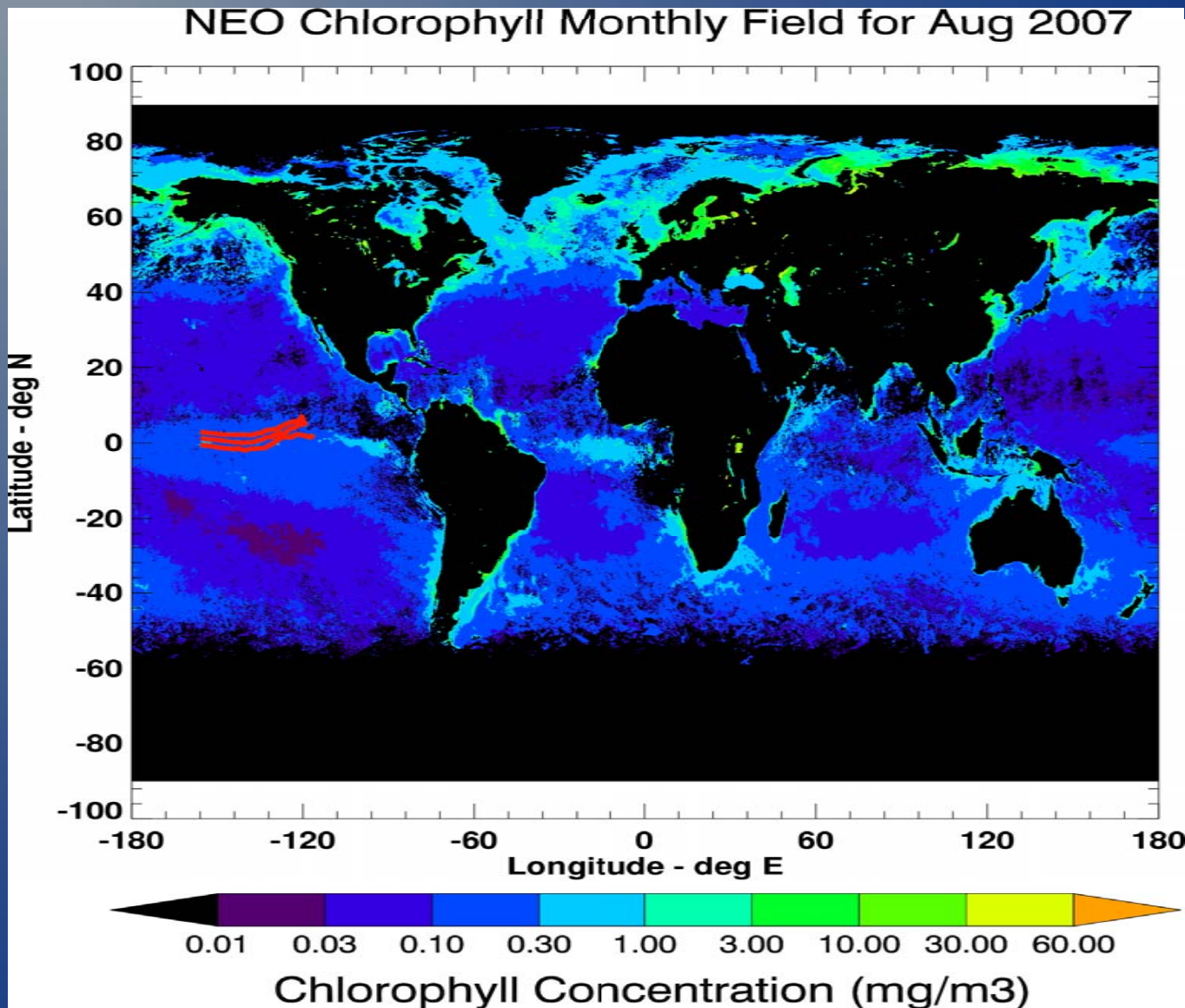
- Model slightly low vs observation
- Lifetimes:
 - SS Scav: 30 days
 - Chem: 7 days
 - Clouds: 1.5 days

Hypotheses for SO₂ in Lower FT

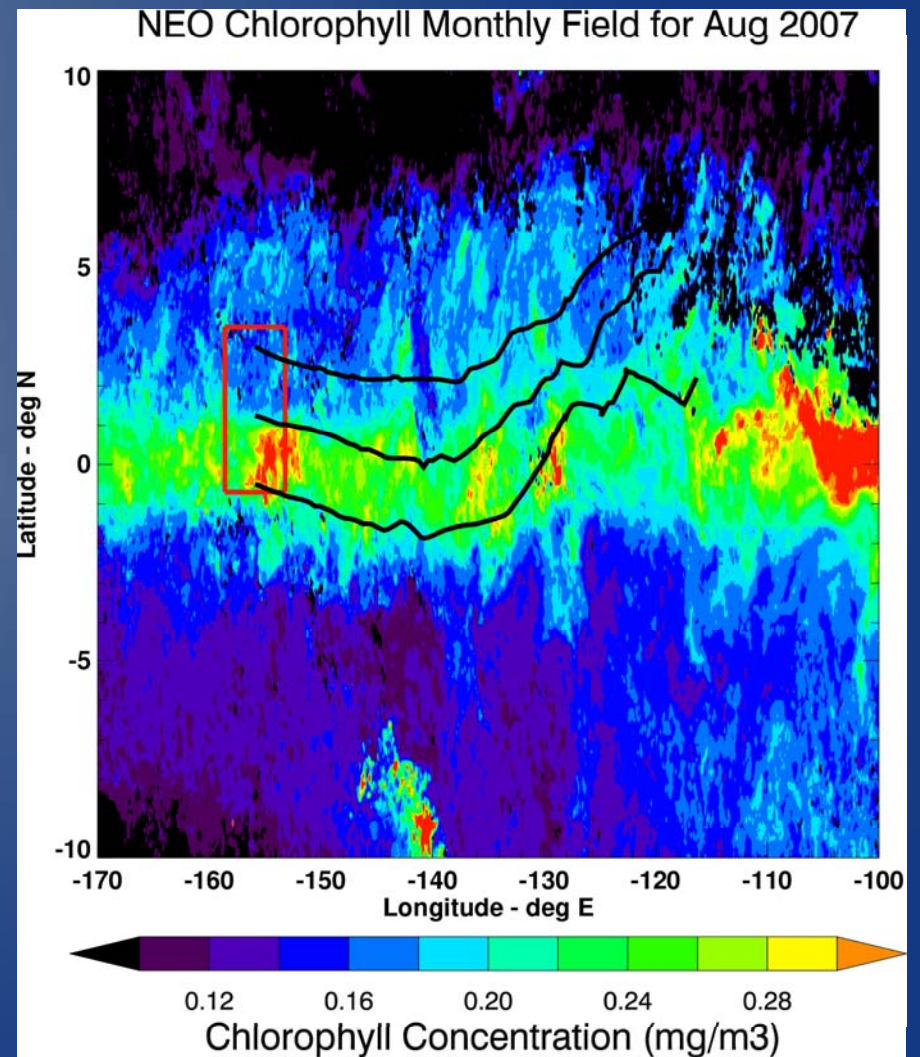
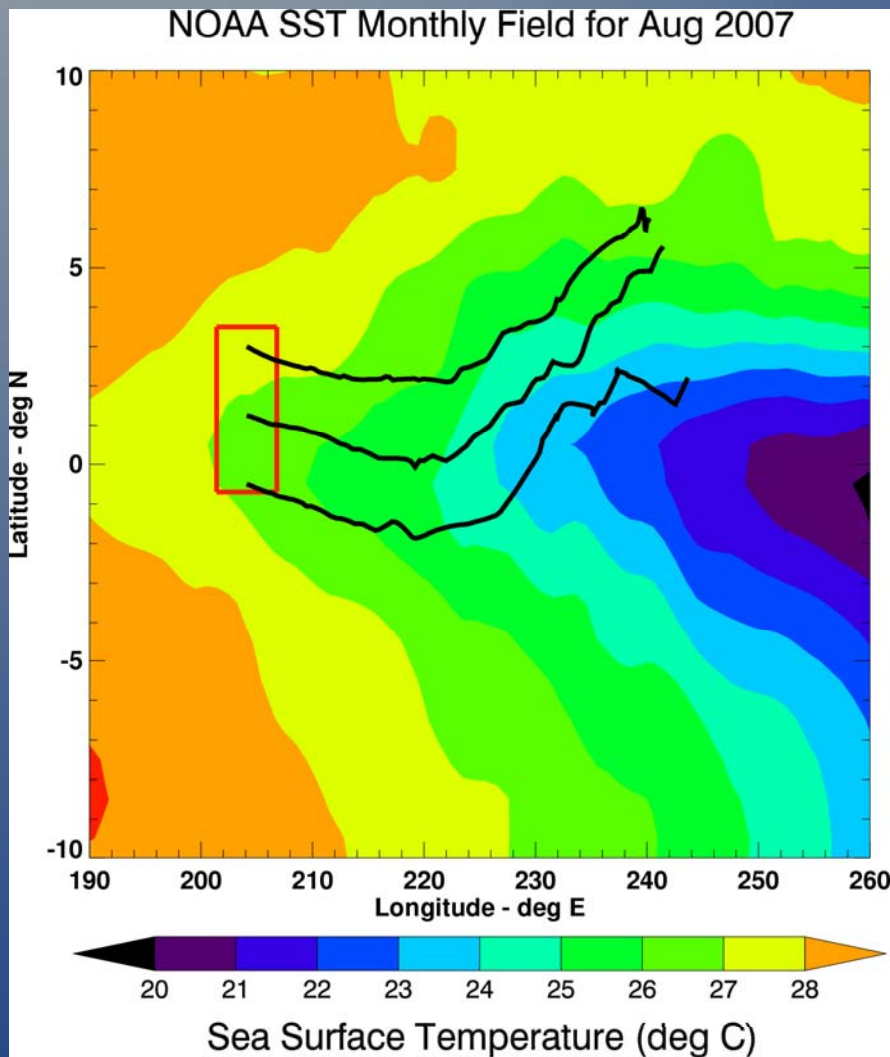


- Advection from an area with a higher DMS flux
- Convection much higher than predicted by WRF

Chlorophyll Map with 5-day HYSPLIT Backtrajectories



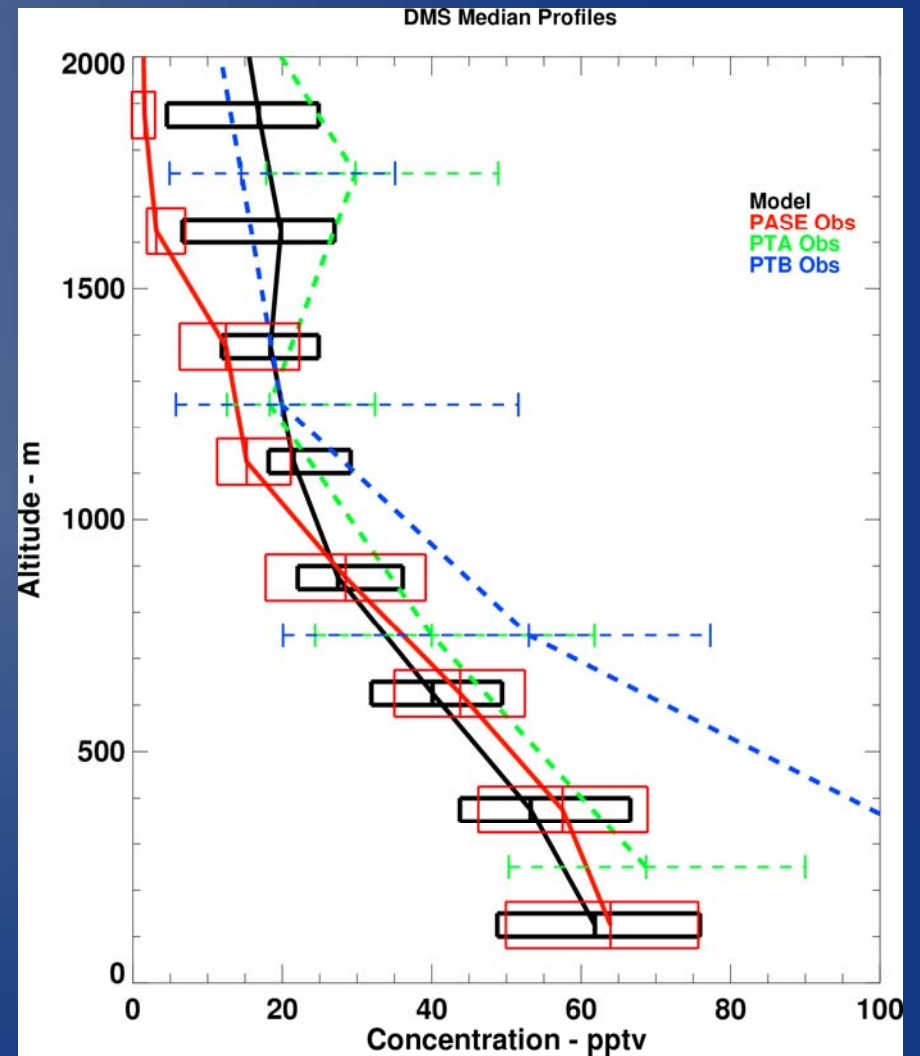
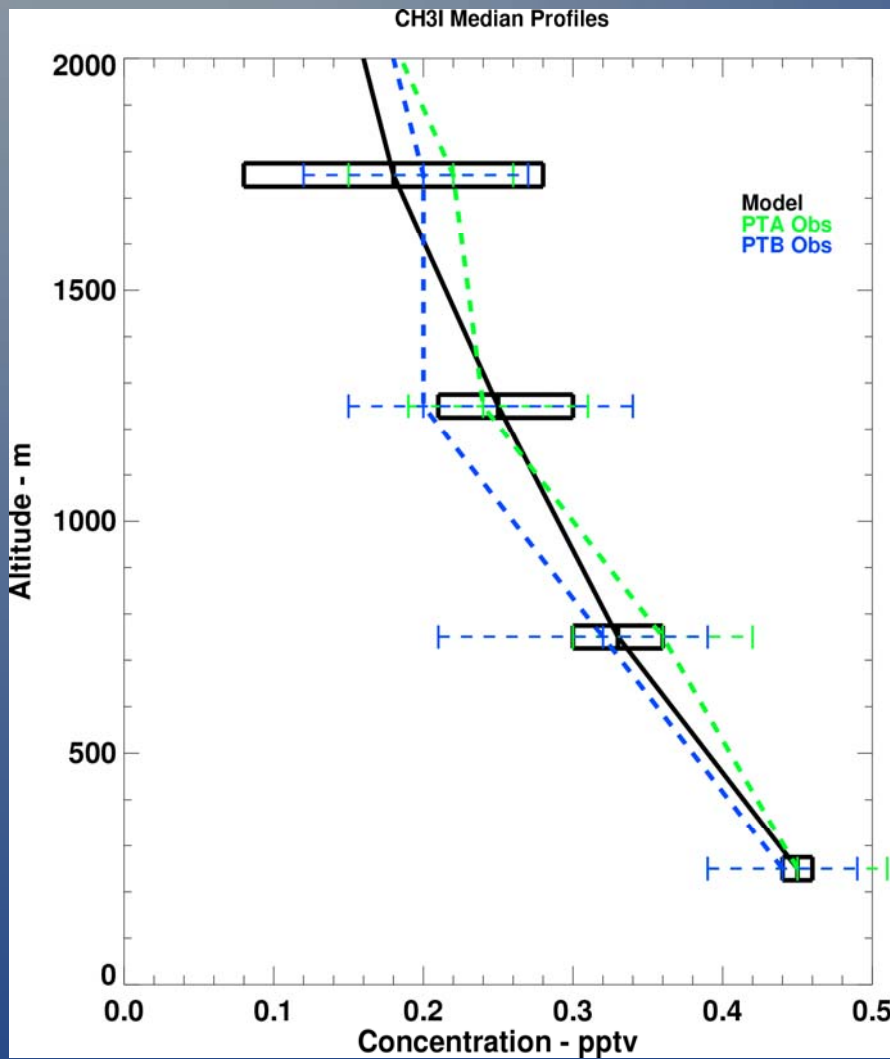
SST and Chlorophyll MAPS



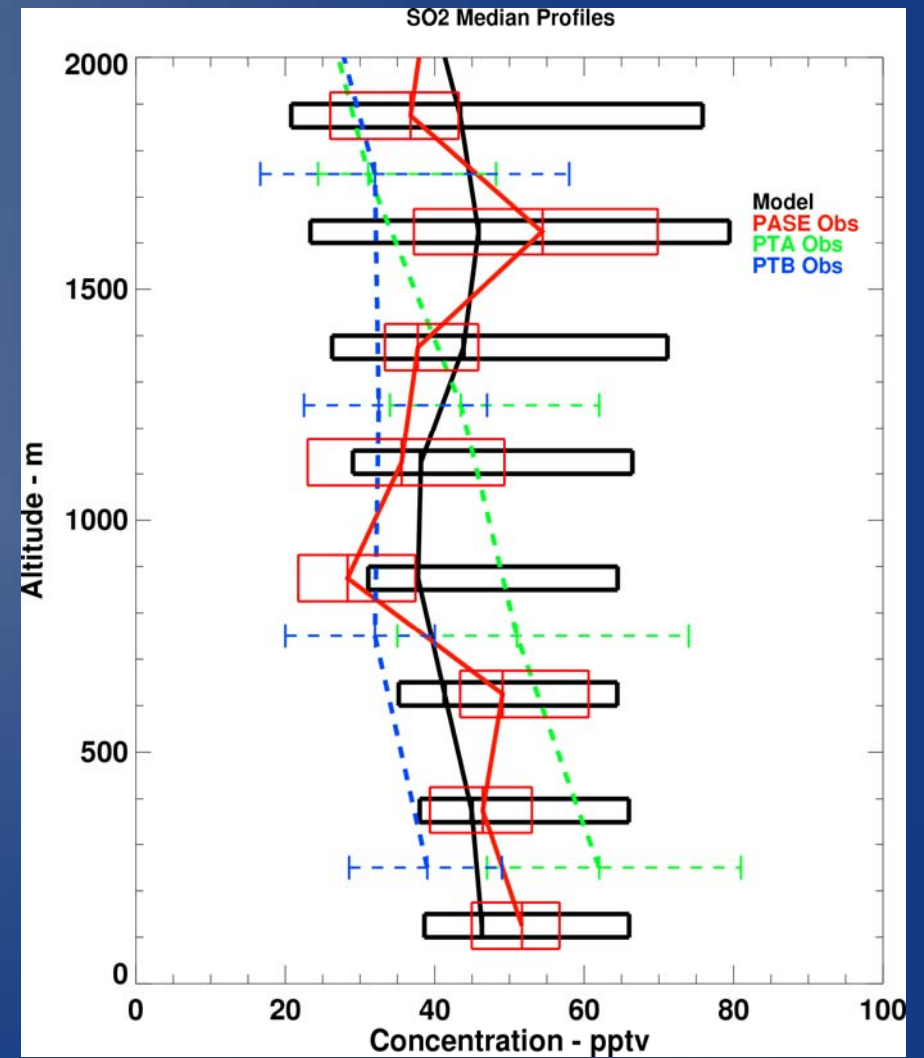
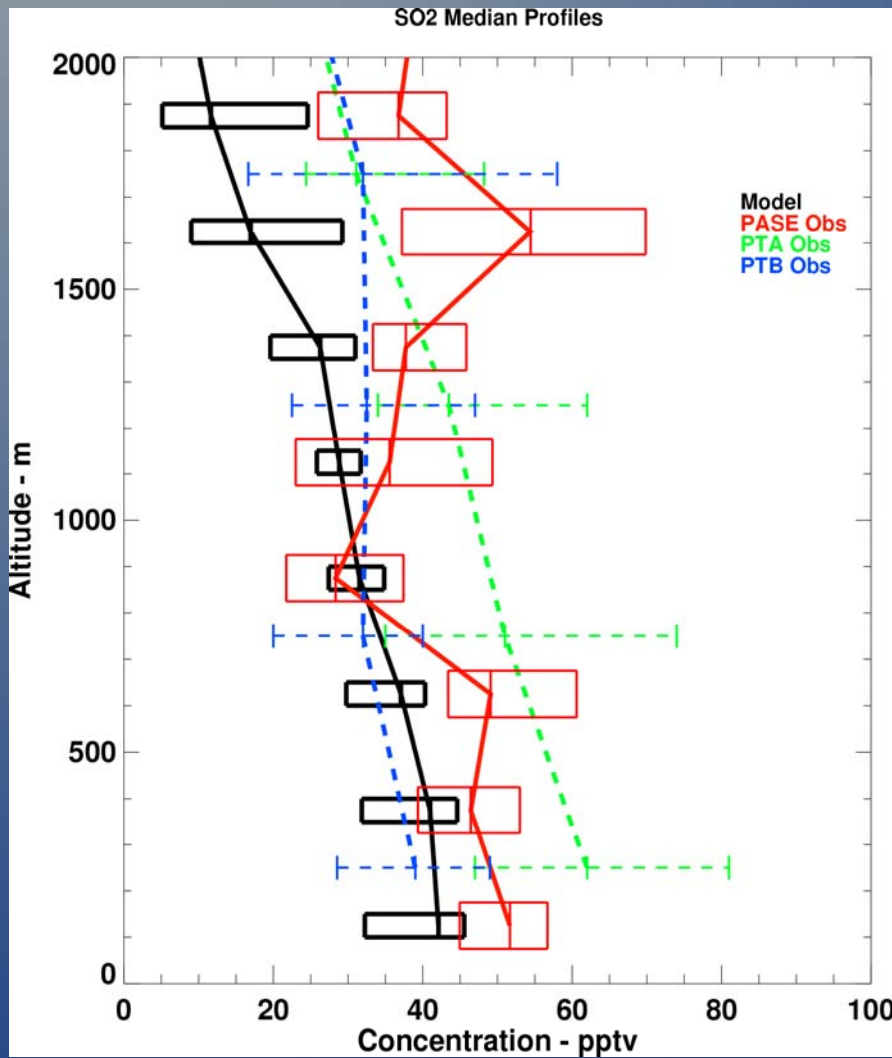
Summary Regarding Advection

- HYSPLIT indicates advection from cooler waters, which could indicate greater upwelling
- However, this area does not have higher chlorophyll concentration, so it *may* not have higher DMS productivity
- Conley et al. (2009): Advection vector for PASE generally perpendicular to DMS gradient
 - On average, advection not large DMS source during PASE

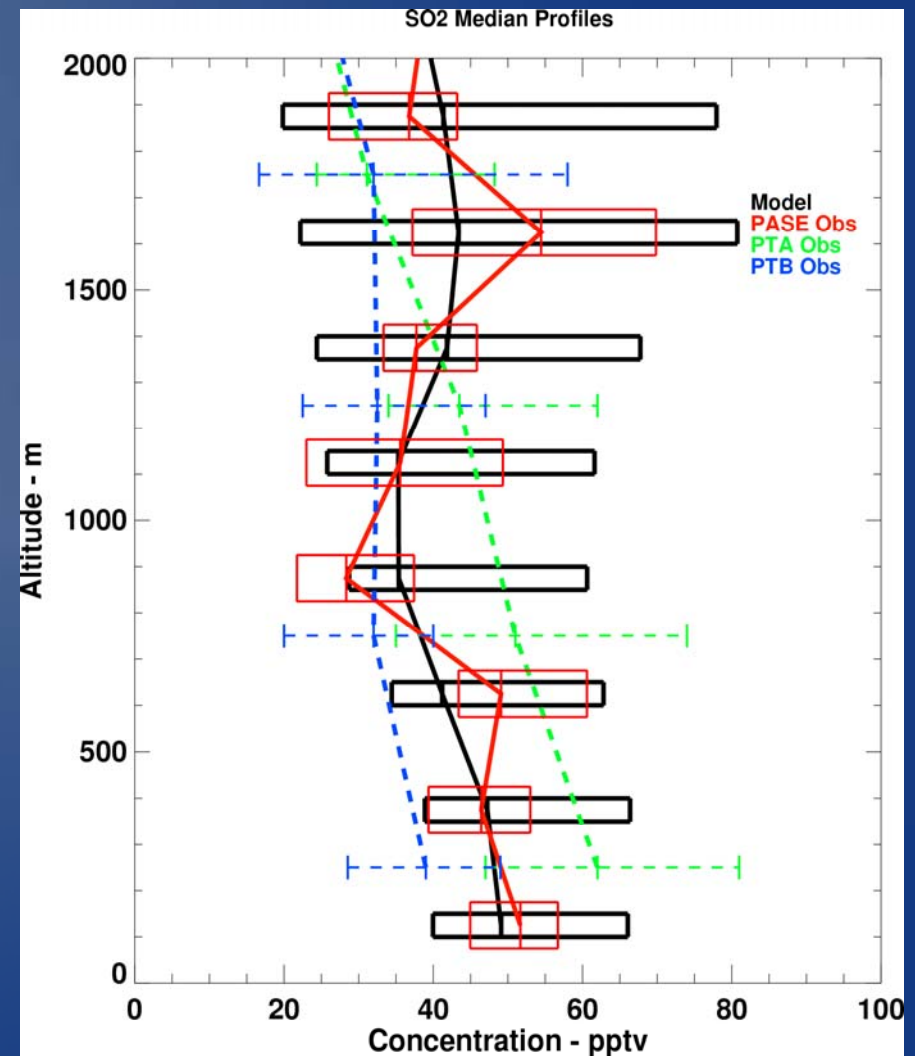
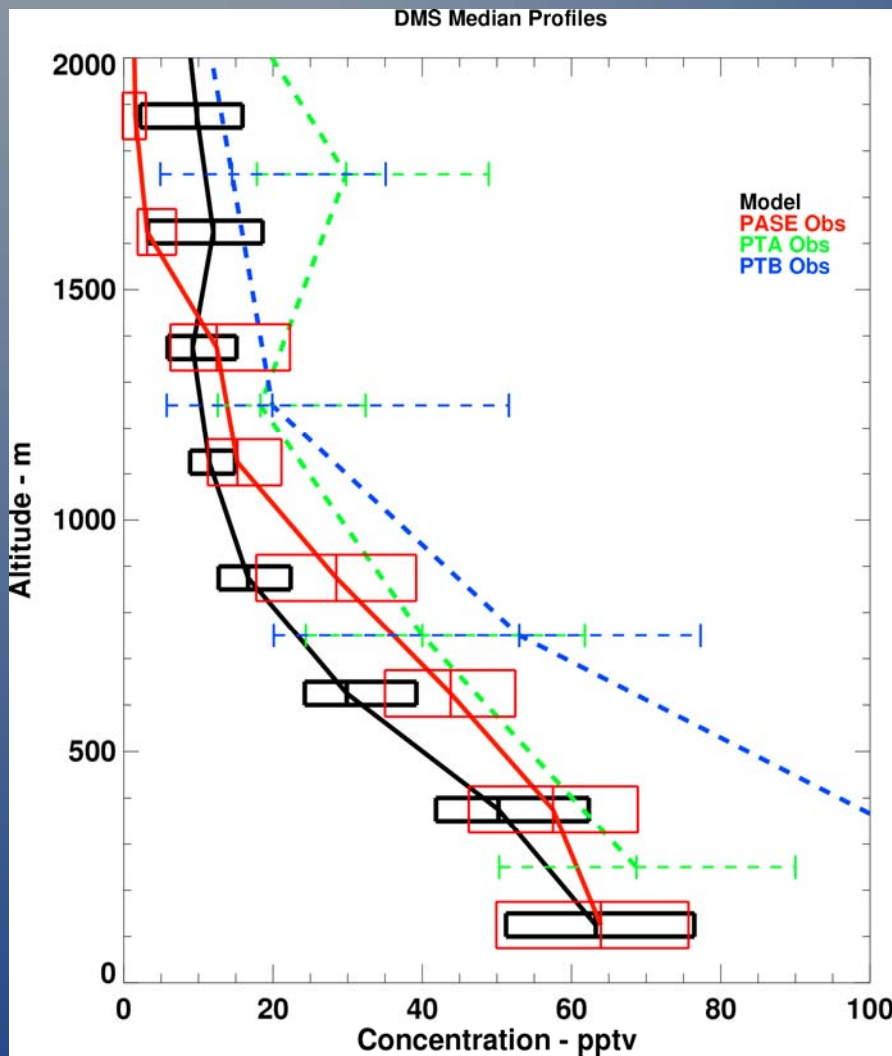
10X WRF Convection



SO₂: 1X vs 10X Conv



DMS, SO₂ 10X Conv and 2pptv BrO

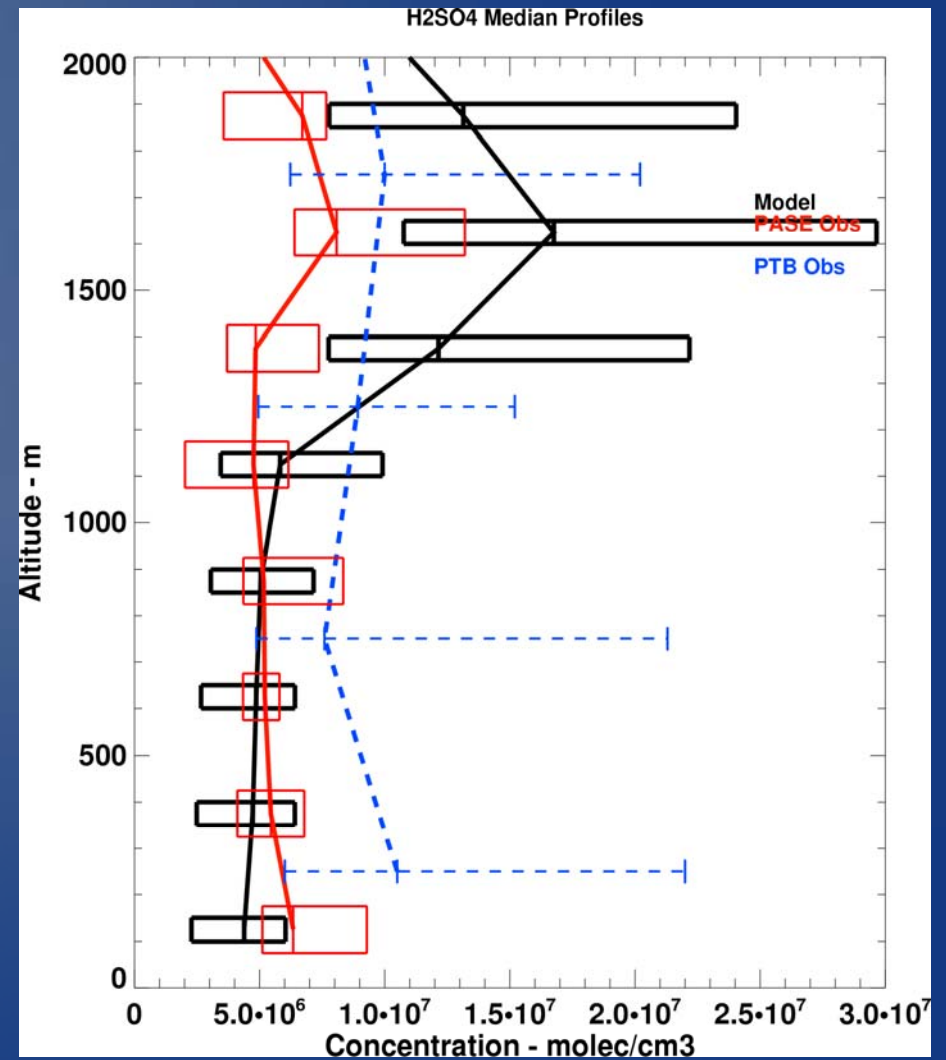
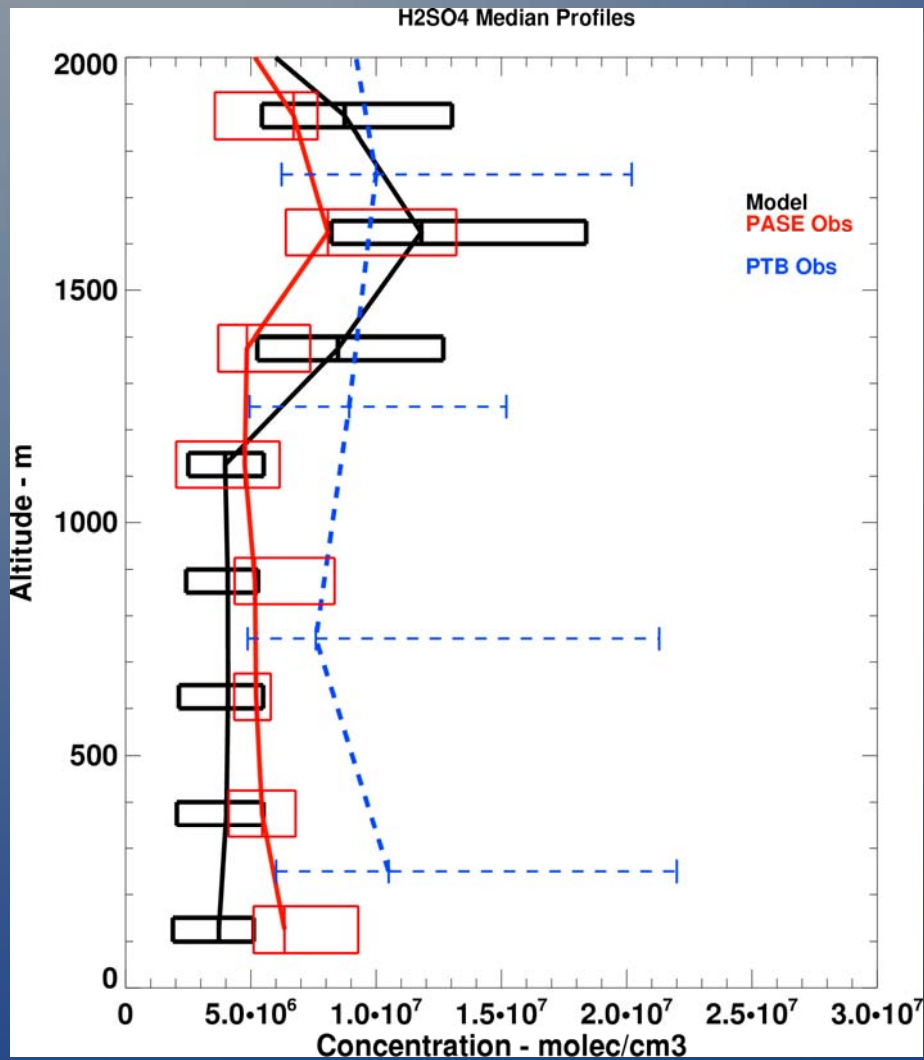


Summary

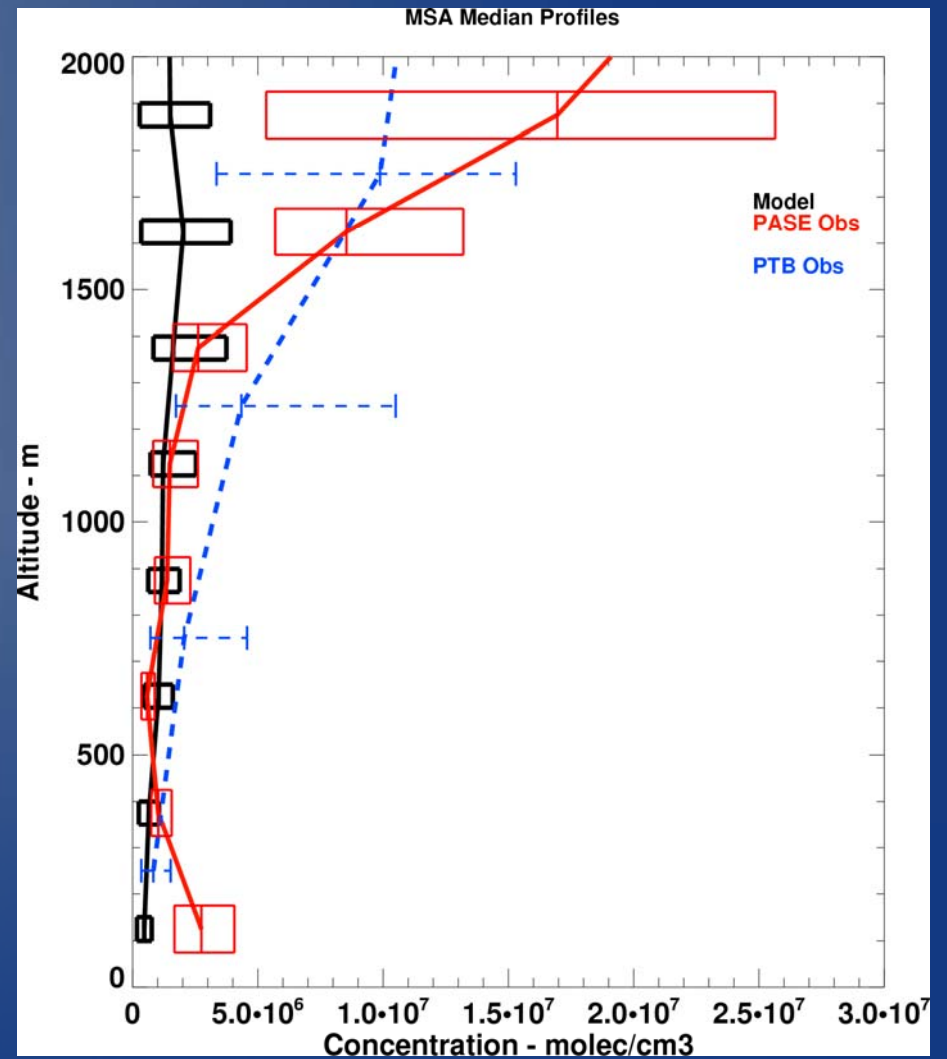
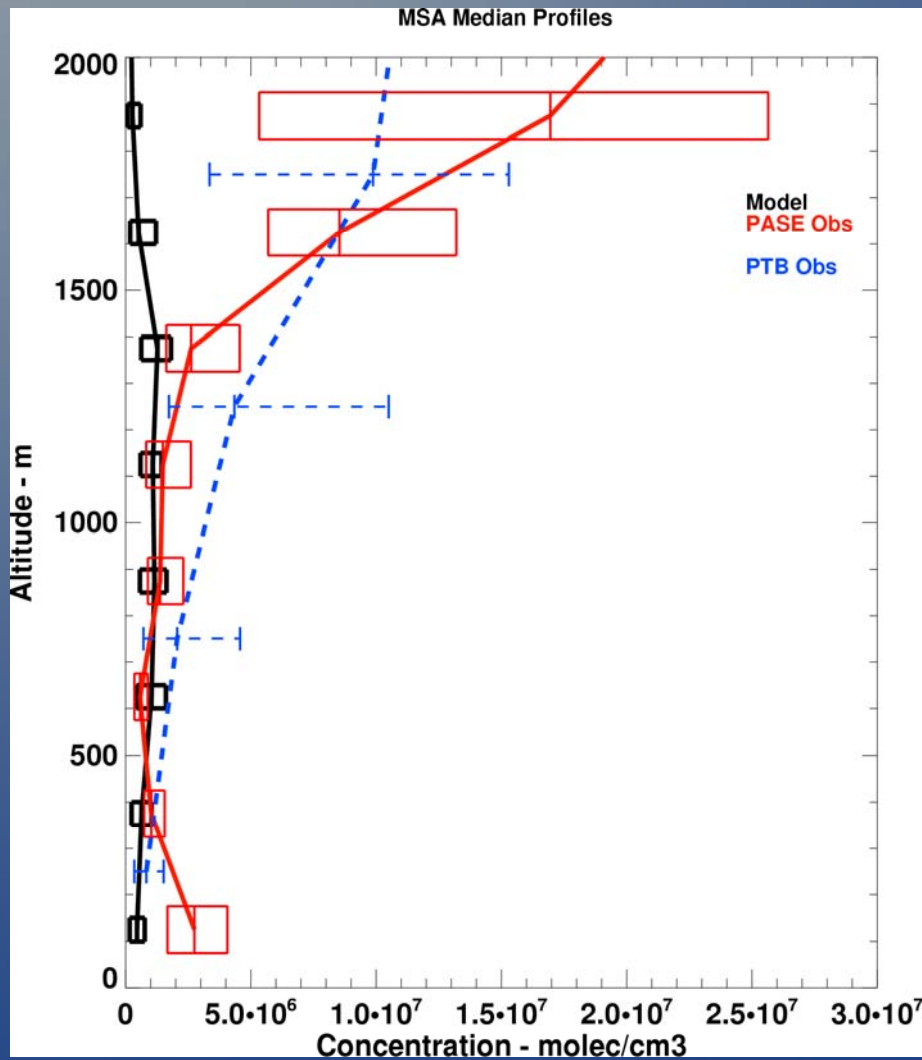
- Primary SO₂ loss mechanism in the CBL is cloud scavenging
 - tau_chemistry = 7 days
 - tau_deposition = 6 days (Vd = 0.1 cm/sec)
 - tau_ss in BL = 5 days, tau_ss in BuL = 30 days
 - tau_cloud = 1.5 days
- High SO₂ concentration in lower FT strongly suggests high shallow convection or advection from stronger DMS production region
 - High DMS in shallow conv offset by BrO?

Future Work

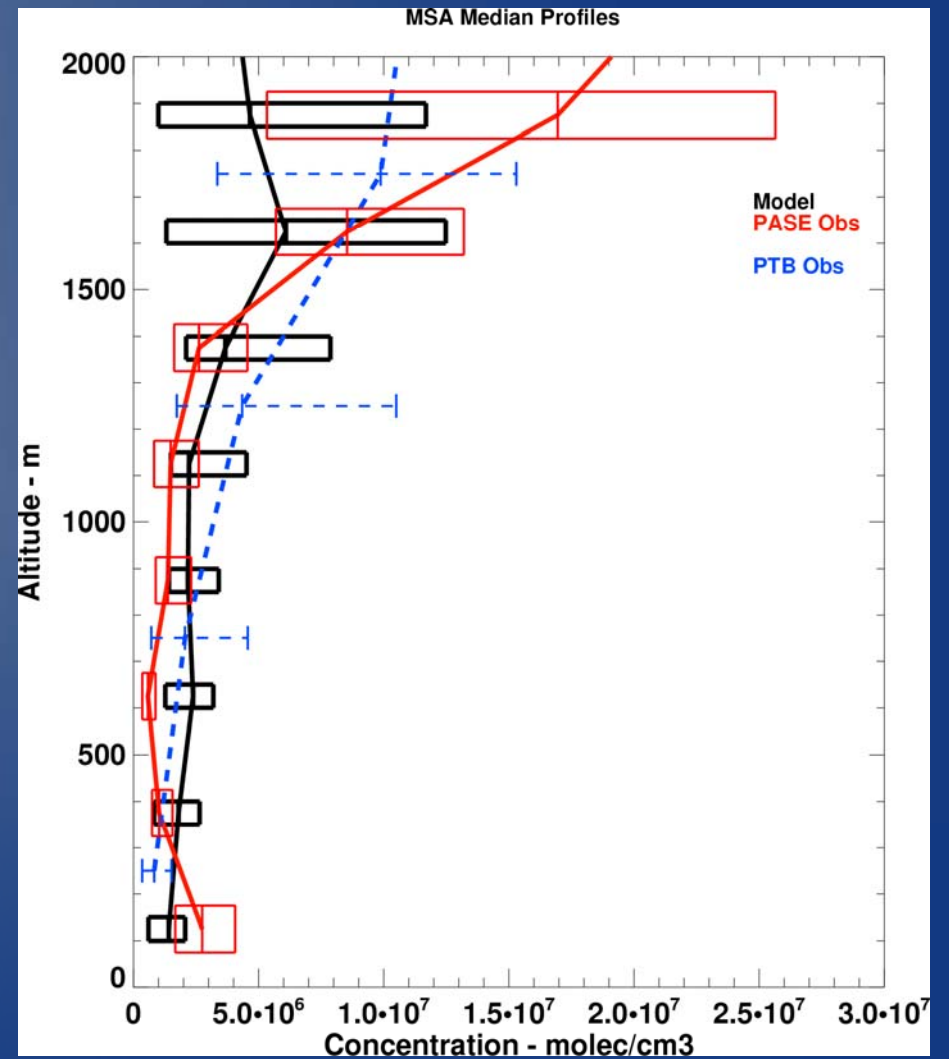
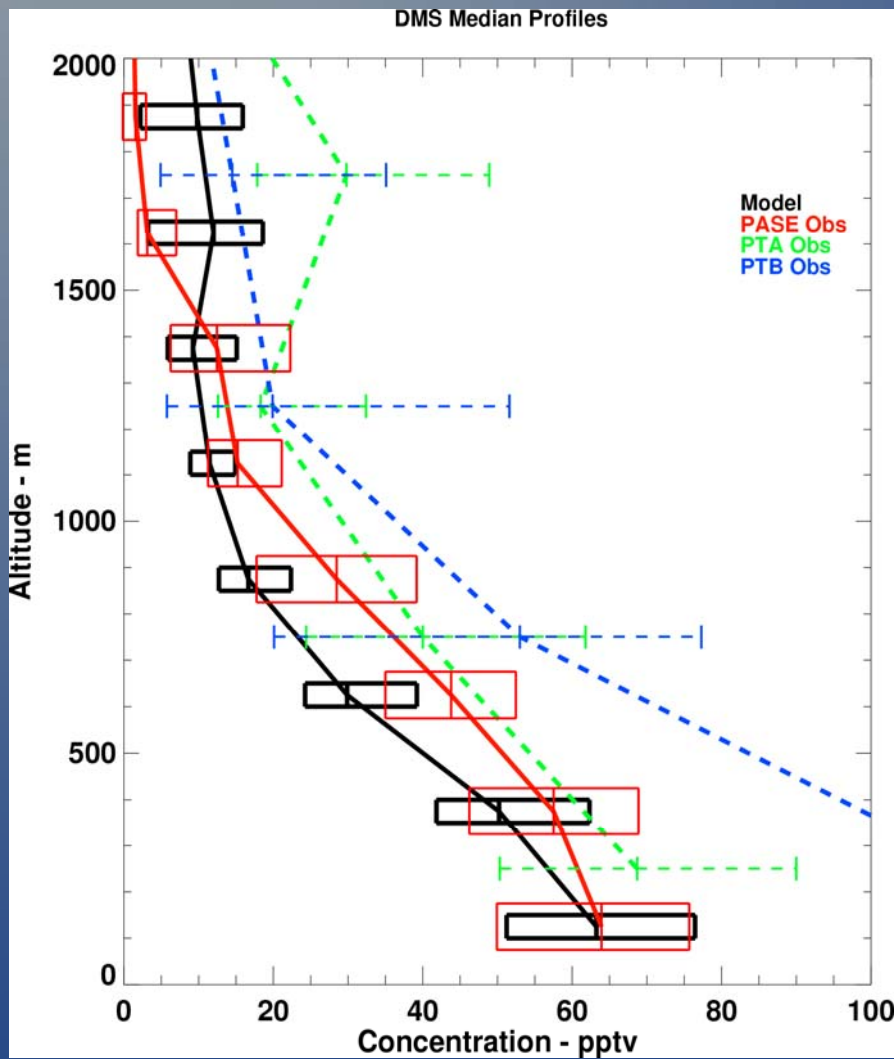
H₂SO₄: 1X vs 10X Conv



MSA: 1X vs 10X Conv



10X Conv with 2pptv BrO



Summary

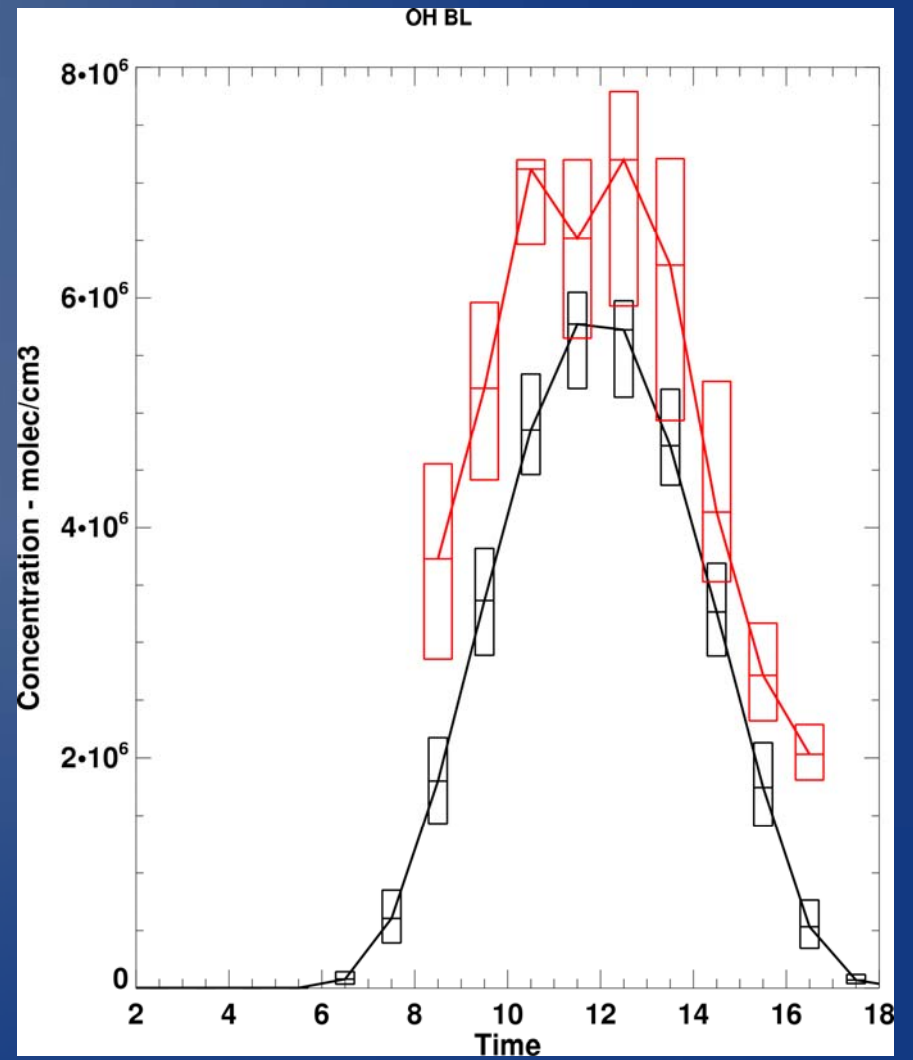
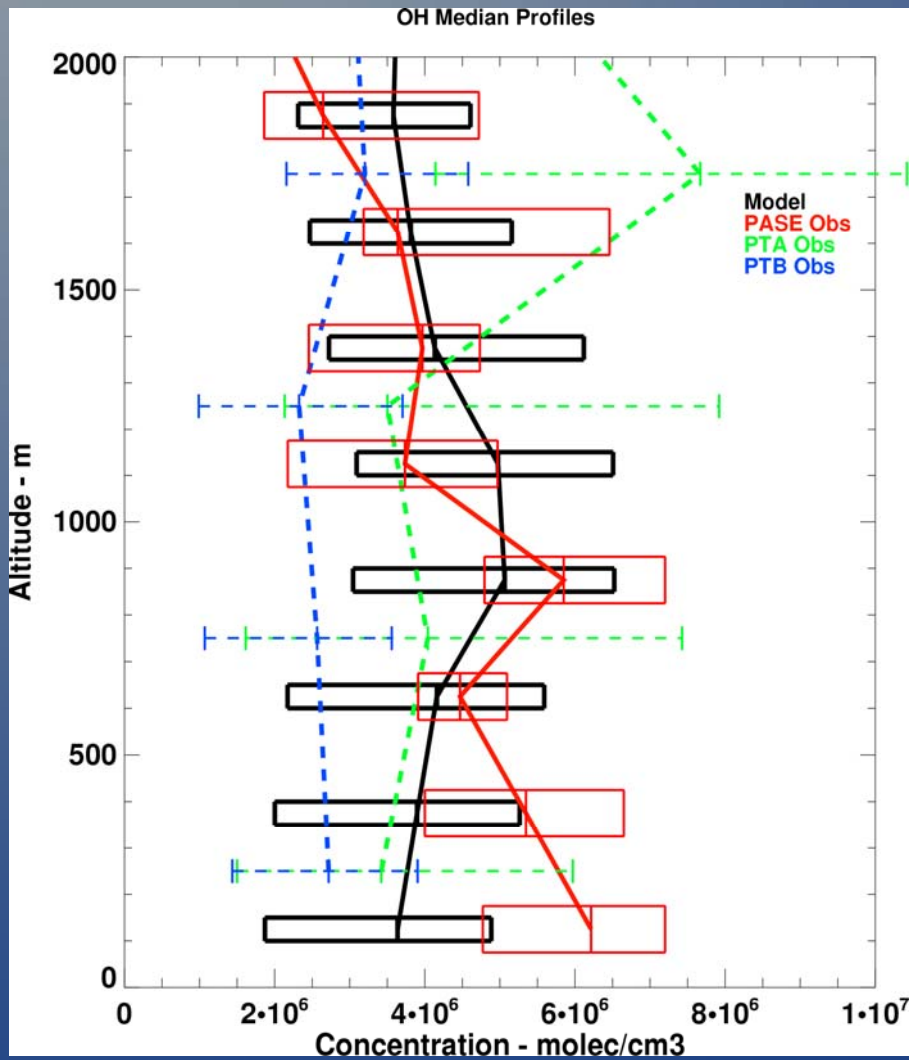
- Possible source for MSA near ocean surface
 - BrO
- Hypotheses for enhanced MSA in lower FT
 - Halogen chemistry
 - Transport from high-DMS region
 - pH dependence in aerosol sink
 - Sharp pH decrease with increasing altitude?
 - Need to model NH_3 and H_2SO_4 uptake

Supplementals

Summary (continued)

- Even with 10X convection, model MSA in lower FT much too small
- However, addition of 2 pptv BrO to all layers greatly improves DMS and MSA comparison with 10X convection; H₂SO₄ worse
- MSA concentration still too small with 10x conv and 2 pptv BrO
 - More halogen chemistry in lower FT?
 - Cloud source for MSA from upper BuL?

OH



SST

